# Design, Synthesis and SAR Evaluation of Novel Benzoxa-[2,1,3]-diazole Amino Acid Hydrazides 

## Against Mycobacterium Tuberculosis Newcastle

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Abstract


#### Abstract

The major challenges of tuberculosis (TB) treatment are the emergence of the drug-resistant strains and the higher risk of hepatotoxicity with prolonged treatment. Therefore, efforts to effectively control TB require the discovery and development of new therapeutic options possessing new mechanisms of action. This project focused on the design and synthesis of novel agents targeting Mycobacterium tuberculosis (Mtb). We have successfully synthesised 170 hydrazides with 40 of those intermediates being converted to benzoxa-[2,1,3]diazole substituted amino acid hydrazides 65. The resulting compounds were screened against susceptible and resistant Mtb strains utilising a Resazurin Microtiter Assay (REMA). A subsequent structure activity relationship (SAR) strategy investigated the structural modification of the benzo-[2,1,3]-diazole peptidomimetics architecture 65 as the main focus of the research presented in this thesis. The findings from this SAR study indicate that an increased size of the amino acid side chain, incorporation of heavy halogens at the meta position of the aryl hydrazine, the L-configuration of the amino acid and the benzoxa-[2,1,3]-diazole moiety each play a key role in improving the antitubercular activity.




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## List of Publications arising from this research

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Abbreviations

| SAR | structure activity relationship |
| :---: | :---: |
| REMA | Resazurin Microtiter Assay |
| WHO | World Health Organisation |
| PCR | Polymerase chain reaction |
| MDR-TB | Multi drug-resistant tuberculosis |
| XDR-TB | Extensively drug-resistant tuberculosis |
| DOTS | Directly Observed Therapy Short-course |
| AG | Arabinogalactan |
| AMPs | Antimicrobial peptides |
| ATP | Adenosine triphosphate |
| AP | Atmospheric Pressure |
| Aq | Aqueous |
| Ar | Aromatic |
| BCG | Bacillus Calmette-Guérin |
| Bn | Benzyl |
| br | Broad |
| ${ }^{\circ} \mathrm{C}$ | Degrees Celsius |
| $\mathrm{CDCl}_{3}$ | Deuterated chloroform |
| Cl | Chemical Ionisation |
| cm | Centimetres |
| d | Doublet |
| DCM | Dichloromethane |
| dd | Doublet doublet |
| DMSO | Dimethyl sulfoxide |
| DMSO-d6 | Deuterated dimethyl sulfoxide |
| DNA | Deoxyribonucleic acid |
| El | Electron Ionisation |
| EMB | Ethambutol |
| eq | Equivalents |
| ESI | Electroscopy |
| EtOAc | Ethyl acetate |
| EtOH | Ethanol |
| $\mathrm{Et}_{2} \mathrm{O}$ | Diethyl ether |
| g | Grams |
| h | Hours |
| HIV | Human Immunodeficiency Virus |
| HRMS | High Resolution Mass Spectrometry |
| Hz | Hertz |
| IR | Infrared |
| INH | Isoniazid |
| LAM | Lipoarabinomannan |
| LC-MS | Liquid chromatography mass spectrometry |
| LTBI | Latent TB infection |
| FDA | Food and Drug Administration |
| $J$ | Coupling constant |
| $\mathrm{MgSO}_{4}$ | Magnesium sulphate |
| M | Molar |


| m | Multiplet |
| :---: | :---: |
| mins | Minutes |
| MIC | Minimum Inhibitory Concentration |
| Mg | Milligram |
| $\mu \mathrm{g}$ | Microgram |
| mL | Milliliter |
| MHz | Megahertz |
| mmol | Millimolar |
| m.p. | Melting Point |
| Mtb | Mycobacterium tuberculosis |
| $\mathrm{NaHCO}_{3}$ | Sodium bicarbonate |
| $\mathrm{NH}_{4} \mathrm{Cl}$ | Ammonium chloride |
| NMR | Nuclear Magnetic Resonance |
| NAG | N -acetylglucosamine |
| NAM | N -acetylmuramic acid |
| ppm | Parts per million |
| Ph | Phenyl |
| q | Quartet |
| Rf | Retention Factor |
| RIF | Rifampicin |
| RNA | Ribonucleic acid |
| Rt | Room Temperature |
| s | Singlet |
| t | Triplet |
| TB | Tuberculosis |
| THF | Tetrahydrofuran |
| DIPEA | $\mathrm{N}, \mathrm{N}$-diisopropylethylamine |
| TEA | Triethylamine |
| DCC | $N, N$ '-dicyclohexylcarbodiimide |
| HOBt | 1-hydroxybenzotriazole |
| DMTMM | 4-4,6-Dimethoxy-1,3,5-triazin-2-yl)-4-methylmorpholinium |
| HBTU | 2-(1H-benzotriazol-1-yl)-1,1,3,3- tetramethyluronium hexafluorophosphate |
| MeCN | Acetonitrile |
| ATP | Adenosine triphosphate |
| TST | Tuberculin skin test |
| TLC | Thin Layer Chromatography |
| $\delta$ | Chemical shift |
| - | Degrees |

## Originality Statement

I hereby declare that this submission is my own work and to the best of my knowledge it contains no materials previously published or written by another person, or substantial proportions of material which have been accepted for the award of any other degree or diploma at University of Newcastle or any other educational institution, except where due acknowledgment is made in the thesis. Any contribution made to the research by others, with whom I have worked at University of Newcastle or elsewhere, is explicitly acknowledged in the thesis. I also declare that the intellectual content of this thesis is the product of my own work, except to the extent that assistance from others in the project's design and conception or in style, presentation and linguistic expression is acknowledged.'

Signed: Ahmed Khalaf B. Aljohani
Date: $2^{\text {nd }}$ January 2022

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Chapter 1: Introduction

## 1 Introduction

This thesis describes research concerning i.) design of new chemical entities with promising antitubercular activity and selectivity towards $M t b^{1,2}$, ii.) the synthesis of a new library of novel benzoxa-[2,1,3]-diazole substituted amino acid hydrazides $65^{1,2}$ and iii.) structure activity relationship studies of novel benzoxa-[2,1,3]-diazole peptidomimetics $65^{1,2}$. The following section will focus on the epidemiology of the bacteria, with emphasis on background, recent epidemiological reports, resistance and treatment. Section 2 will present the successful synthesis and characterisation of novel antitubercular agents based on the benzoxa-[2,1,3]-diazole substituted amino acid hydrazides 65, with Section 3 presenting current progress towards our SAR investigations. The penultimate Section 4 will highlight the findings obtained from both Sections 2 and 3; the synthesis and SAR analysis of the resulting intermediates and final compounds. Finally, Section 5 will conclude the work presented by looking at prospects for future exploitation of the results arising from this thesis. Section 6 will detail the experimental procedures.

### 1.1 Tuberculosis epidemiology

According to the World Health Organisation's Global Tuberculosis Report 2019, Tuberculosis (TB) is recognised as a communicable disease, is responsible for a major cause of ill health and is one of the top ten causes of death worldwide from a single infectious agent. ${ }^{3}$ In 2018, there were an estimated ten million people (range 9.0-11.1 million) reported to have become infected with TB, which is equivalent to 132 cases per 100000 population, whilst ten percent of these cases were associated with other co-morbid disease such as HIV/AIDS. In addition, there are high mortality rates caused by TB with 1.2 million HIV negative patient deaths and 251,000 HIV positive patient deaths reported in $2018 .{ }^{3}$ Globally, the estimated number of new cases of TB has been relatively stable at around 10 million per year since 2014, whilst the HIV-positive TB incidence rates have increased to nearly double. Contrary to this, the mortality rates of HIV-positive TB cases has steadily declined by $64 \%$ (Table 1). ${ }^{3,4,5,6,7}$

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Table 1: Global incidence, HIV/TB co-infection and Mortality of TB in last 5 years. ${ }^{3}, 4,5,6,7$

| Year of <br> incidence | TB Incidence | HIV-pos. TB <br> Incidence | Mortality |  |
| :--- | :---: | :---: | :---: | :---: |
|  |  |  | HIV-neg. | HIV-pos. |
| $\mathbf{2 0 1 4}$ | 9.6 mil | 1.2 mil | 1.1 mil | 390,000 |
| $\mathbf{2 0 1 5}$ | 10.4 mil | 1.2 mil | 1.4 mil | 400,000 |
| $\mathbf{2 0 1 6}$ | 10.4 mil | 1.0 mil | 1.3 mil | 374,000 |
| $\mathbf{2 0 1 7}$ | 10.0 mil | 920,000 | 1.3 mil | 300,000 |
| $\mathbf{2 0 1 8}$ | 10.0 mil | 862,000 | 1.2 mil | 251,000 |

Geographically, in 2018 the WHO identified the global regions worst affected by TB as South-East Asia, Africa and the Western Pacific, with countries such as India characterised as the highest TB burden country $(27 \%, 199,000)$ followed by China (9\%, 610,000), Indonesia and the Philippines (6\%,554,000) amongst others (Table 2).

Table 2: Global incidence, HIV/TB co-infection and Mortality of TB. ${ }^{3}$

| WHO region | TB Incidence* | HIV-pos TB <br> Incidence | Mortality |  |  |
| :--- | :---: | :---: | :---: | :---: | :---: |
|  |  | \% of <br> Global <br> total |  | HIV-neg. | HIV-pos. |
| Africa | 2.45 mil | $24 \%$ | 615,000 | 397,000 | 211,000 |
| The Americas | 289,000 | $3 \%$ | 29,000 | 17,000 | 5,900 |
| Eastern <br> Mediterranean | 810,000 | $8 \%$ | 6,900 | 77,000 | 2,200 |
| Europe | 259,000 | $3 \%$ | 30,000 | 23,000 | 4,400 |
| South-East <br> Asia | 4.37 mil | $44 \%$ | 140,000 | 637,000 | 21,000 |
| Western <br> Pacific | 1.84 mil | $18 \%$ | 41,000 | 90,000 | 6,500 |
| Global total | 10 mil | $100 \%$ | 862,000 | 1.24 mil | 251,000 |

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Many studies have revealed the emergence of multi drug-resistant tuberculosis (MDR-TB) is clearly linked to mismanagement of front line treatments for drugsensitive TB. ${ }^{8,9}$ Additionally, structural weaknesses in health care systems and socioeconomic factors are associated with the sharper rise of MDR-TB, for example, repeated use of short-course TB treatments, inadequate dosing, inconsistent drug supply and poor patient adherence due to complicated regimens and prolonged treatment periods. ${ }^{9,10}$ In addition, studies have revealed the emergence of primary MDR-TB infection, arising in areas with poor TB control programmes and community transmission, which further complicates and hampers the global control of TB. 8,10

For these reasons, rapid detection and appropriate therapeutic plans are essential to save the lives of TB patients and preclude the spread of MDR-TB. One such diagnostic test is the drug susceptibility test, or phenotypic method, which is used to diagnose MDR-TB or XDR-TB. It is undertaken by culturing Mtb isolated from the sputum of an infected person in the presence of a front-line TB treatment, such as isoniazid, however, this method is time consuming and expensive. More recently, a fully automated test has been devised which utilises a polymerase chain reaction (PCR) based technique to identify gene mutations, arising from resistance markers, in Mtb cultured from an infected individual. ${ }^{10}$

With detection technology improving and becoming quicker and cheaper, the reported incidence rates have increased in countries including India, the Russian Federation and China which have contributed to an estimated 45\% of the global total MDR-TB cases in existence in 2018, with an estimated 484,000 new MDR and rifampicin resistant (RR) TB cases reported (Figure 1). ${ }^{3}$ In order to control and reduce incidence numbers, the design of an appropriate MDR-TB management plan is essential, especially in high MDR-TB burden countries. ${ }^{9}$ These treatment plans can include: utilising an assured quality treatment and optimised individualised treatment regimens based on drug susceptibility testing results and patient medical history to enhance the cure rate of MDR-TB. ${ }^{9}$


Figure 1: The global percentage of new MDR/RR-TB cases (Adapted from Global tuberculosis report, $2019{ }^{3}$ )

Alongside MDR-TB, cases are extensively drug-resistant TB (XDR-TB) which continues to add to the patient burden. Brought about predominantly by a lack of drug management, bacteria which are extensively resistant do not respond to any of the antitubercular agents, hence the treatment regimen for XDR-TB is limited. In 2019, WHO reported that XDR-TB cases totalled 13,068, arising mainly in India, China and Russia (Table 3). ${ }^{3} \mathrm{WHO}$ also declared that the treatment success rate reached by only $39 \%$ for existing patients with XDR-TB, therefore, XDR-TB is considered a growing challenge in the controlling of TB around the world. ${ }^{3}$

Table 3: Global MDR/RR-TB incidence, and XDR-TB Incidence. ${ }^{3}$

|  | MDR/RR-TB Incidence |  | XDR-TB Incidence |  |
| :--- | :---: | :---: | :---: | :---: |
| WHO region |  | \% of Global <br> Total |  | \% of Global <br> Total |
| Africa | 24,712 | $13 \%$ | 727 | $6 \%$ |
| The Americas | 4,759 | $3 \%$ | 149 | $1 \%$ |
| Eastern | 5,584 | $3 \%$ | 122 | $1 \%$ |
| Mediterranean |  |  |  |  |
| Europe | 48,739 | $26 \%$ | 7,899 | $60 \%$ |
| South-East Asia | 75,964 | $41 \%$ | 3,580 | $27 \%$ |
| Western Pacific | 27,014 | $14 \%$ | 591 | $5 \%$ |
| Global total | 186,772 | $100 \%$ | 13,068 | $100 \%$ |

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Consequently, the growing spread of XDR-TB has increased the demand even further for significant development of new treatments for TB, with new mechanisms of action.

### 1.2 Mycobacterium Tuberculosis

### 1.2.1 Life cycle of TB

In 2019, the World Health Organisation (WHO) estimated that nearly a third of the earth's population is infected with Mycobacterium tuberculosis (Mtb) based on the community spread of pulmonary TB infection, mostly in developing countries. ${ }^{3}$ Infection with Mtb usually begins when an exposed individual inhales contaminated water droplets containing Mtb which lodge in the pulmonary alveoli, where it is primarily phagocytosed by alveolar macrophages. After macrophage uptake, a long-standing Mtb - macrophage interaction is developed and leads to one of three outcomes: Mtb eradication (Route A), active infection (Route B) or dormancy (Route C) (Figure 3). ${ }^{11}$


Figure 3: The three possible outcomes of Mtb - macrophage interaction (Adapted from Russell et al. $2010{ }^{12}$ )

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Because of the broad and quick effect, an innate immune system is capable of preventing infection and microbial spread in the body. For instance, a strong and rapid pulmonary immune response is mainly responsible for the ability of the host to eradicate Mtb (Route A, Figure 3). ${ }^{12}$ During the initial interaction between Mtb and the immune cells, macrophages release pro-inflammatory cytokines to trigger recruiting additional immune cells including dendritic cells, monocytes and neutrophils into the site of infection. Then, migration of infected dendritic cells to adjacent lymph nodes, where rapid expansion occurs in order to fight the infection. ${ }^{13}$ Upon Interleukin-12 (IL-12), 18 (IL-18) and 27 (IL-27) production by infected cells, natural killer and T cells become active and directly exert cellular cytotoxicity through releasing perforin and granzyme or granulysin. They also have an indirect role to control intracellular infection by secreting interferon- $\gamma$ (IFN- $\gamma$ ) which is essential for macrophage activation. ${ }^{14}$ Once activated, they release tumour necrosis factor- $\alpha$ (TNF- $\alpha$ ) to exert their antimicrobial activity via a variety of mechanisms including phagosomal acidification, production of antimicrobial effectors (nitric oxide and reactive oxygen intermediates) and apoptosis of infected macrophages. ${ }^{13}$
Notwithstanding this, the susceptibility of patients to contracting TB with comorbidities and or who are immunocompromised with, for example HIV, is much greater in these populations when compared against the general population. This synergic effect is due predominantly to a decrease the CD4+ effector T-cells which are responsible for the antimicrobial activities of infected phagocytes. Consequently, several studies have reported the higher TB mortality rate amongst HIV-positive patients. ${ }^{15}$ Nevertheless, the emergence of active TB occurs among small groups of healthy people ( $5-10 \%$ ) due to the increase in inhibitory cytokine (Interleukin-10) levels, which alters the balance between inflammatory and immunopathological responses. Higher level of Interleukin-10 can suppress the antimicrobial mechanisms in the macrophage through inhibiting secretion of pro-inflammatory mediators (IFN- $\gamma$, TNF- $\alpha$ and IL-12) and hence, associated with uncontrolled intracellular Mtb replication. Mtb is characteristically capable of surviving and replicating within the host macrophage cells and can reach a point where the bacterial load becomes too great although all the required cellular immune components exist. At this threshold, cell death occurs and Mtb will be disseminated eventually through the respiratory tract and is characterised

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as active pulmonary TB (Route B, Figure 3). Additionally, Mtb can enter the bloodstream to infect other organs, leading to other forms of TB infection. ${ }^{16}$ Having said that, the majority of infected individuals remain healthy and harbour the infection in a latent form during their lifetime, whilst they can develop TB at some point, $90 \%$ of the cases remain dormant. Dormancy occurs when the host immune system fails to eliminate the bacteria completely, and consequently a sequence of events primes the immune system to recruit the required components to form a granuloma (Route C, Figure 3). By reducing their metabolic activity, Mtb can adapt to this intracellular environment present in the granuloma where hypoxia, acidity and low levels of nutrients exist. ${ }^{12}$ In this stage, the granuloma can segregate and prevent spreading of the infection into the airways, and hence, an equilibrium between Mtb and host contributes to establishment of the dormancy state. However, if the individual becomes immunocompromised, the equilibrium will shift towards the dormant Mtb which can regrow and then induce the granuloma to rupture, which leads to spreading viable Mtb into the whole body and lung in particular, thereby developing the reactivation. ${ }^{12}$ If untreated, half of those active TB cases die. Thus, the development of new antimycobacterial agents faces a significant challenge because of low permeability of both the mycobacterial cell wall and the latency of TB in granuloma's.

### 1.2.2 The structure of mycobacterial cell wall

Pulmonary TB infection is caused by Mtb, which is a member of the mycobacteriaceae species. Mtb is usually defined by its cell wall which has characteristics of both gram-negative and gram-positive bacteria. As grampositive bacteria contain several layers of peptidoglycan interspersed with unique teichoic acids whilst gram negative bacteria are characterised by a thin layer of peptidoglycan surrounded by an outer membrane of lipoprotein (Figure 4). ${ }^{17}$ Moreover, Mtb cell wall is considered a major determinant of virulence, and thus, it deserves more attention.

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Figure 4: Differences in cell wall structure of the gram-positive, mycobacteria and gram-negative bacteria. Arabinogalactan (AG); glycolipid (GL); lipoarabinomannan (LAM); lipoprotein (LP); lipopolysaccharide (LPS); lipoteichoic acid (LTA); mycolic acid (MA); membrane-associated protein (MAP); outer membrane (OM); peptidoglycan (PG); plasma membrane (PM); teichoic acid (TA) (Adapted from Fullam and Young, 2021. ${ }^{18}$ )

Structurally, the cell wall of mycobacteria is comprised of an asymmetric bilayer which surrounds the plasma membrane (Figure 5). The unique inner compartment of the Mtb cell wall has a characteristic hydrophobic fraction (over $60 \%$ of cell wall content as compared to $20 \%$ in gram-negative bacteria) possessing cord factor and wax-D and mycolic acids which act as stimulators to the host's immune system. The outer mycolic acids and multi-layered peptidoglycan covalently join to the arabinogalactan, a branched polysaccharide, to form the mycolic acid-arabinogalactan-peptidoglycan complex (MAPc). This complex is essential for the viability and the persistence of Mtb, particularly, in the latency stage as Mtb can survive inside of macrophages whilst also conferring distinctive immunostimulatory properties. ${ }^{19,20}$ Interestingly, the outer membrane, known as the capsule, can be impermeable to antimicrobial agents used for gram negative bacteria. In addition to saturated lipopolysaccharides, the capsule contains proteins, glycoconjugates and high content of lipoglycans including phosphatidyl-myo-inositol mannosides, lipomannan and lipoarabinomannan in particular. ${ }^{21}$


Figure 5: The cell wall structure of mycobacterial species. Phosphatidyl-myo-inositol mannosides (PIMs); lipomannan (LM); lipoarabinomannan(LAM); mannosylated lipoarabinomannan (ManLAM), peptidoglycan (PG); arabinogalactan (AG); mycolic acids (MA); trehalose monomycolate (TMM); trehalose dimycolate (TDM); diacyltrehalose (DAT); polyacyltrehalose (PAT); phthiocerol dimycocerosate (PDIM); sulfoglycolipid (SGL); mycobacterial outer membrane (MOM); arabinogalactan-peptidoglycan (AGP); mycobacterial inner membrane (MIM); diphosphatidylglycerol (DPG); phosphatidylethanolamine (PE); phosphatidylinositol (PI).

The mycobacterial peptidoglycan is made of repeating N -acetylglucosamines joined to $N$-acetylmuramic acid via peptide bonds situated on the plasma membrane. It serves to maintain cell wall rigidity, integrity and for virulence. Consequently, it is an excellent target for antibacterial drug discovery. ${ }^{22}$ Another essential component for the cell wall integrity is arabinogalactan which is composed of branched arabinans and linear galactan. Moreover, it is essential for mycobacteria replication and survival in the infected host as it links both the peptidoglycan and the mycolic acid to form an integral arrangement of the cell

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wall via a phosphoryl-N-acetyIglucosaminosyl-rhamnosyl and arabinan anchors, respectively. ${ }^{22}$
Mycolic acids are a remarkable macromolecular component of the mycobacterial cell which shield the bacteria from environmental stress such as antimicrobial pressure or adaptive immune responses. They also play a major role in regulating the permeability, acid-fast staining, viability and Mtb virulence. Mycolic acids are a strong hydrophobic component containing long chain fatty acids synthesised by two fatty acid synthase (FAS) systems which are attractive targets for the rational design of new antitubercular agents (Figure 6).


Figure 6: A general description of mycolic acid biosynthesis which can be targeted by different antitubercular agents. Isoniazid (INH) and thioamides (ETH/PTH) are InhA inhibitors; thiolactomycin (THL) is KasA inhibitor; thioacetazone (TAC) and isoxyl are dehydration step inhibitors and SQ109 inhibits the transport activity of MmpL3. (Adapted from Vilchèze, $2020{ }^{22}$ )

FAS 1 produces medium chain fatty acid (C20-C26), while FAS 2 produces a long meromycolic chain (C60-C90) with some modifications (cis/trans cyclopropanations, keto or methoxy groups) which are introduced during an elongation process. ${ }^{22}$

Phosphatidyl-myo-inositol mannosides (PIMs) are unique glycolipids which can interfere with proximal T-cell receptor signalling, thereby inhibiting CD4 ${ }^{+}$effector T-cell activation. They are not only important in cell wall permeability and integrity, but also essential in the regulation of cell septation and division. ${ }^{22}$

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Lipomannan and lipoarabinomannan are glycolipids. In addition to their role in the integrity of cell wall, they show a pathogenic role when they release in the macrophage as they exert different immunomodulatory activities including inhibiting phagosomal maturation and shifting the cytokine response from pro- to anti-inflammatory. ${ }^{24}$

The complexity of the mycobacterial cell wall structure provides for bacterial survival in harsh conditions, along with a strong impermeability to several antimicrobial agents, particularly $\beta$-lactams (penicillins and cephalosporins) due to several factors such as the presence of $\beta$-lactamase and low affinity penicillin binding proteins. ${ }^{25}$ It is also associated with mycobacterial pathogenicity and hence it remains a component of interest for developing novel antimycobacterial agents which target crucial enzymes involved in mycobacterial cell wall biosynthesis. In this context, it is comprised of three main components which are common target sites for attack by current antitubercular agents (Figure 7). ${ }^{22}$ More importantly, the cell wall inhibitors are targeting novel protein targets, including DprE1 and MmpL3 which were discovered by a drug to target screening strategy, in addition to well-known targets (InhA and peptidoglycan synthesis). ${ }^{26}$ Recently published a study highlighting the ability of the mycobacterial cell wall inhibitors such as isoniazid 1 to induce the bactericidal activity against dormant Mtb. So, the resistance may be some other mechanism. ${ }^{27}$

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Figure 7: The main components of Mtb cell wall including Phosphatidyl-myo-inositol mannosides (PIMs); lipomannan (LM); lipoarabinomannan(LAM), peptidoglycan (PG); arabinogalactan (AG); mycolic acids (MA) are targeted by antitubercular agents. (Adapted from Abrahams and Besra, $2018{ }^{22}$ )

### 1.3 Anti-tubercular Agents and Their Targets

Notwithstanding the issues arising from Mtb epidemiology, TB is recognized as a treatable infection, as according to the World Health Organisation's Global Tuberculosis Report 2019, an estimated 58 million cases (2000-2018) were treated successfully. An early effective therapeutic intervention for TB is the use of the Bacillus Calmette-Guérin (BCG) vaccine in children. This has demonstrated excellent results, however it does not provide total protection. ${ }^{3}$

### 1.3.1 Vaccination

The utilisation of vaccination to improve the protective immunity against TB has a major role to play in order to interrupt Mtb transmission and ultimately to eradicate TB. In the 19th century, the Bacillus Calmette-Guerin (BCG) vaccine which is a live attenuated form of $M$. bovis, was introduced as a part of the TB control program. ${ }^{28}$ Since the introduction of BCG vaccine, around 4 billion people around the world have been vaccinated, with an efficacy from $0-80 \%$ in children up to the age of 16 compared to adults when administered later in life. Unfortunately, this variation in BCG efficacy has emerged because of the genetic
and nutritional differences among human populations. ${ }^{28}$ Additional factors are the variation in BCG strains and prior exposure to environmental mycobacteria which could be detected using the tuberculin skin test (TST) especially in endemic areas. ${ }^{28}$ The immunisation route can also influence the protective effect of BCG and hence the mucosal route is more efficient to provide a high level of protection against disseminated and pulmonary TB with an accelerated production of IFNү, compared to intravenous route. ${ }^{29}$ However, oral BCG vaccination was initially used, until 73 accidental deaths occurred in Lübeck, Germany following orally administrated BCG vaccine which was contaminated with virulent $M t b^{30}$ so, the vaccine is now administered via the intradermal route.

Notwithstanding this, the BCG vaccine provides incomplete protection against the development of adult pulmonary TB because this age group of adults is associated with higher risk of prior exposure with environmental mycobacteria. In 2002, Brandt L. et al. demonstrated the negative impact of sensitization with environmental mycobacteria on the protective efficacy of BCG vaccine against pulmonary TB as sensitized mice with cocktails of mycobacteria initiated antimycobacterial responses which inhibit the multiplication of BCG and reduce its activity. ${ }^{31}$ Furthermore, the lack of evidence for long-term BCG immunisation could also contribute to ongoing incidence of pulmonary TB. ${ }^{32}$ Therefore, evaluating the protective efficacy of BCG vaccine against Mtb still remains a controversial issue along with the global BCG vaccination policy since the early evaluation of the BCG vaccination. ${ }^{28}$

Notwithstanding the differences in protection and routes of administration, the immunological mechanisms of parenteral BCG and poor lung protection is still under investigation to determine a new strategy to overcome the immunological gap. This gap is believed to arise from a lack of airway luminal anti-TB T cells, which can afford a delay in pulmonary T-cell responses to Mtb exposure particularly at the early stages of infection, whilst a large population of anti-TB Tcells, generated by the parenteral BCG immunization, reside in peripheral tissues. ${ }^{33}$ Consequently, there is clearly a strong demand for the development of novel and effective TB vaccines against all stages of TB infection in all age groups to reduce the global incidence of TB in general, and drug-resistant TB in particular which remains paramount to achieve the goal of the End TB Strategy by 2035.3 In order to achieve such MDR-TB elimination, the significant challenge facing TB

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vaccine developers is to design a safe and effective TB vaccine to prevent spreading the drug-resistant TB strains which relies on less effective and more toxic TB treatments. ${ }^{34}$

### 1.3.2 Anti-tubercular drugs

Consequently, treatment with small molecule drugs is necessary with front line treatments such as isoniazid 1, pyrazinamide 2, ethambutol 3 or rifampicin 4 (Figure 8).


1


2


3


4

Figure 8: Front line treatments for TB treatment.

Generally, each anti-tubercular agent has a specific target and mechanism of resistance and can be classified according to their treatment order (Table 4). ${ }^{3,35,36}$

| Anti-tubercular <br> agents | Bacterial target | Mechanism of <br> resistance | Treatment <br> order |
| :--- | :--- | :--- | :---: |
| Isoniazid (INH) 1 | Mycolic acid biosynthesis <br> inhibition | Mutations in InhA, KatG, <br> aphC and Ndh. ${ }^{37}$ |  |
| Rifampicin (RIF) 4 | RNA synthesis inhibition | Mutation in rpoB |  |
| Pyrazinamide (PAZ) <br> $\mathbf{2}$ | Membrane potential <br> disruption | Mutations in pncA | Front line |
| Ethambutol (EMB) 3 | Arabinogalactan <br> biosynthesis inhibition | Mutation in embB |  |
| Aminoglycosides | Protein synthesis inhibition | Mutation in 16S rRNA |  |
| Fluoroquinolones | DNA synthesis inhibition | Mutation in gyrA and <br> gyrB |  |
| Ethionamide (ETH) <br> $\mathbf{1 6}$ / Prothionamide <br> (PTH) $\mathbf{1 7}$ | Mycolic acid biosynthesis <br> inhibition | Mutation in ethA and <br> inhA. ${ }^{37}$ | Second line |
| D-Cycloserine (DCS) <br> $\mathbf{1 8}$ | Peptidoglycan <br> biosynthesis inhibition | Overexpression of airA |  |

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| Para-aminosalicyclic <br> acid (PAS) $\mathbf{2 0}$ | Folic acid and iron <br> metabolism inhibition | Mutation in thyA |  |
| :--- | :--- | :--- | :--- |
| Thiacetazone (TAC) | Mycolic acid <br> cyclopropanation inhibition | Mutation in ethA |  |
| Clarithromycin | Protein synthesis inhibition | Mutation in ermA, ermB <br> and ermC |  |
| Bedaquiline (BDQ) <br> $\mathbf{2 2}$ | Mycobacterial ATP <br> synthesis inhibition | Mutation in atpE and <br> rv0678 |  |
| Delamanid (DLM) 25 <br> / Pretomanid (Pa) 26 | Mycolic acid synthesis <br> inhibition | Mutation in ddn and <br> fdg1 | Third line ${ }^{39}$ |
| Clofazimine | Mycobacterial <br> DNA replication inhibition | Mutation in rv0678 |  |
|  | Linezolid (LZD) 23 | Protein synthesis inhibition | Mutation in 23S rRNA |

### 1.3.3 First-line anti-tuberculosis drugs

The most common treatment of drug-susceptible TB is with isoniazid 1 (INH, known as isonicotinic acid hydrazide), which elicits its anti-mycobacterial activity through inhibiting mycolic acid biosynthesis. This enfeebles the permeability of the mycobacterial cell wall leading to destruction of the cell integrity and ultimately its death. ${ }^{40}$ INH 1 has a bactericidal activity against rapidly dividing mycobacteria, but it is not as effective as rifampicin 4 against a non or slow growing mycobacteria, particularly those in granuloma. ${ }^{41}$ INH 1 is a prodrug which is converted by Mycobacterial catalase-peroxidase (KatG) to isonicotonic acyl radical 5 which forms a nicotinoyl-NAD adduct 6 . Consequently, disrupting fatty acid chain elongation when the active metabolite binds tightly to the enzyme enoyl acyl carrier protein reductase ( $\operatorname{lnh} A$ ) which is a precursor for mycolic acid biosynthesis (Scheme 1). ${ }^{40}$


Scheme 1: conversion of isoniazid 1 to its active metabolite 5 through KatG enzyme. ${ }^{40}$

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Rifampicin (RIF) 4 is one member of the front-line TB treatment which is a potent and broad spectrum antibacterial agent. It shows an equivalent level of antimycobacterial activity against both fast and slow growing mycobacteria through supressing the mycobacterial RNA polymerase (RNAP) which is required for transcription. RIF 4 contains naphthoquinones that attach in a pocket of RNAP $\beta$ subunit through forming hydrogen bonds between hydroxyl groups of naphthoquinone and amino acids of enzyme (Figure 9). This sterically inhibits the process of RNA elongation and thus leads to cell death. ${ }^{42}$ Due to the broad spectrum nature of rifampicin 4 , it is usually combined with other antitubercular agents to reduce the chance of resistance. For long-term TB treatment, the use of rifampicin $\mathbf{4}$ with isoniazid $\mathbf{1}$ is more common than pyrazinamide $\mathbf{2}$ due to their potential to cause severe hepatotoxicity. ${ }^{43}$


Figure 9: According to the crystal structure of rifampicin 4 in complex with the Thermus aquaticus core DNA dependent RNAP, rifampicin 4 binding involves hydrogen bonding interactions between the hydroxyl groups at $\mathrm{C}-1, \mathrm{C}-8, \mathrm{C}-21$ and $\mathrm{C}-23$ and the carbonyl oxygen of the C-25 acetoxy group with the amino acid residues Arg409, Ser411, GIn393, His406, Asp396 and Phe394. (Adapted from Campbell et al, $2001{ }^{42}$ )

Pyrazinamide (PAZ) 2, a nicotinamide analogue, is uniquely effective in treating drug susceptible TB and MDR-TB and also reduces the total duration of treatment when used in combination with rifampicin 4 and isoniazid 1.44 The clinical trials

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which were undertaken in East Africa by the British Medical Research Council tuberculosis units, demonstrated that use PAZ with rifampicin 4 showed a synergetic antimycobacterial effect and shortened TB treatment period from 12 to 9 months. Furthermore, lower doses of pyrazinamide 2 decreased the probability of hepatotoxicity. ${ }^{45}$ PAZ 2 is a prodrug and shows powerful sterilizing activity only against non-growing Mtb when it is converted to its active form 7 (pyrazinoic acid, POA) by a nicotinamidase-peroxidase enzyme known as pyrazinamidase encoded by the pncA gene in the mycolic acids (Figure 10). ${ }^{44}$ Due to its activity at acidic pH, PAZ 2 is ideal for killing intracellular Mtb that mainly reside within infected macrophages. However, it is only used in the first two months as longer use does not show additional benefit. This is presumably due to increased the acidity in the lesions during acute inflammation, which is decreased after two months treatment. ${ }^{44}$


Figure 10: The conversion of pyrazinamide 2 (PZA) to its active pyrazinoic acid 7 (POA) in acidic environment. (Adapted from Zhang et al, $2013{ }^{44}$ )

Ethambutol (EMB) 3 is a member of front-line TB treatments that reduce the global burden of TB infection caused by Mtb due to its bacteriostatic activity against growing drug susceptible and monoresistant strains. EMB 3 interferes with mycobacterial cell wall biosynthesis by inhibiting the action of arabinosyl

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transferase (EmbB) which is involved in the synthesis of arabinan for the arabinogalactan complex. EMB 3 is used in combination with front line TB treatments to improve their potency and reduce the length of TB therapy. ${ }^{46}$ Although EMB 3 is effective in preventing the emergence of drug resistance, it is not a preferred TB treatment option in young children due to its ocular toxicity. ${ }^{35}$

### 1.3.4 MDR-TB

Despite the successes of front-line TB treatments, resistance to one or more of these effective drugs (most often, isoniazid or rifampicin) is a global challenge for TB control. Therefore, the use of second-line TB treatments is an alternative option for MDR-TB cases, but this is challenging as treatments include drugs that are more toxic, expensive and require longer administration, e.g. for at least a 24month treatment course. Consequently, drug susceptibility testing (DST) for firstand second-line drugs is recommended to design the appropriate MDR-TB treatment regimen and thus improve treatment outcomes. ${ }^{10}$

### 1.3.5 Second-line antituberculosis drugs

## Aminoglycosides

Streptomycin (STR) 8, kanamycin (KAN) 9, amikacin (AMK) 10 and capreomycin (CAP) 11 are common second line TB treatments which are used to reduce mortality due to MDR-TB (Figure 11).

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8


10


9


11

Figure 11: Structures of representative aminoglycosides, including streptomycin 8, kanamycin 9, amikacin 10 and capreomycin 11.
Most are protein synthesis inhibitors, which usually act as bacteriostatic agents, but some act as bactericidal agents due to their irreversible inhibition of bacterial protein synthesis. ${ }^{47}$ Their mechanism of action, after crossing the bacterial cell envelope is to induce a ribosomal misreading on the messenger RNA template through binding to the 30 S subunit of bacterial ribosomes. The ribosomes are present in both human and bacterial cells and as a result, long term TB treatment with aminoglycosides can show ototoxicity and nephrotoxicity. ${ }^{48}$ Under anaerobic and low pH conditions, such as in the granuloma where Mtb resides in its latent state, antitubercular activity of aminoglycosides is depressed as penetration of drug to bacterial cell is reduced due to a reduction in energy-dependent transport. ${ }^{48}$

Streptomycin 8 is derived from Streptomyces griseus and effective against TB and is administered parenterally due to poor gastrointestinal tract absorption. Even though the drug susceptibility testing (DST) shows susceptibility, extensive Streptomycin 8 treatment is not recommended in order to avoid higher rates of Streptomycin 8 resistance in patients with MDR-TB.

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Kanamycin (KAN) 9 or Amikacin (AMK) 10 can be used instead as the initial choice of injectable agent because they have little cross-resistance with Streptomycin 8, less toxicity and are cheaper than Capreomycin (CAP) 12.49 Although, KAN 9 and AMK 10 have similar structures, AMK 10 is most efficacious and has a lower minimum inhibitory concentration (MIC). J. A. Dijkstra et al. (2018) reported the possible differences in MICs between AMK 10, KAN 9 and CAP 11 against wild-type Mtb which showed 2,4 and $8 \mu \mathrm{~g} / \mathrm{mL}$, respectively. ${ }^{50}$ The similarity in structure between KAN 9 and AMK 10, leads to cross-resistance and so CAP 11, a cyclic peptide, is recommended for the treatment of MDR-TB, however, CAP 11 can also have cross-resistance with amikacin/kanamycin in the presence of the rrs gene mutation. ${ }^{49}$ F. A. Sirgel et al. (2011) examined the MICs of AMK 10 and CAP 11 against Mtb isolates with the ribosomal (rrs) A1401G mutation and showed $>20$ and $15 \mu \mathrm{~g} / \mathrm{mL}$, respectively. Therefore, this gene mutation is responsible for a higher level of resistance to AMK 10, with only a minor impact on the CAP 11 MIC. ${ }^{51}$

## Fluoroquinolones

Fluoroquinolone (FQ) use is recommended by the World Health Organization for TB cases including resistance or intolerance to front-line TB treatments because of their ability to eliminate both active and persistent Mtb, particularly in the case of moxifloxacin $12 .{ }^{52}$ Moreover, the fluoroquinolones such as levofloxacin 13 have less risk of hepatotoxicity than isoniazid 1. ${ }^{53}$ The fluoroquinolones elicit antibacterial activity through binding to the bacterial topoisomerase II (DNA gyrase) and IV which are critical in DNA supercoiling process. Consequently, fluoroquinolones inhibit DNA replication and transcription, resulting in bacterial death. Fluoroquinolones have different affinity to both target enzymes. For example, levofloxacin 13 shows greater antibacterial effect on gram-positive bacteria due to their greater affinity to bacterial DNA gyrase than norfloxacin. ${ }^{47} \mathrm{By}$ binding to both enzymes, moxifloxacin 12 is advantageous in preventing the highlevel resistance if one of targeted enzymes being mutated. ${ }^{54}$ Based on results from a large clinical trial, the use of newer fluoroquinolones, such as moxifloxacin 12 and levofloxacin (L-isomer of ofloxacin) 13, are the most valuable as substitute agents for MDR-TB than their older counterparts,

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especially ciprofloxacin 14 and ofloxacin 15, because they have better pharmacokinetic and pharmacodynamic and lower minimum inhibitory concentrations (MICs) against wild-type Mtb (Figure 12). ${ }^{52}$


12
$\mathrm{MIC}=(0.5 \mu \mathrm{~g} / \mathrm{ml})$
Fourth generation


13
MIC $=(1 \mu \mathrm{~g} / \mathrm{ml})$
Third generation
'Control potency and G+ve activity'



14
MIC $=(4 \mu \mathrm{~g} / \mathrm{ml})$
Second generation


15
MIC $=(2 \mu \mathrm{~g} / \mathrm{ml})$
Second generation

Figure 12. The old and new generations of fluoroquinolones include moxifloxacin 12, levofloxacin 13, ciprofloxacin 14 and ofloxacin 15.

For instance, moxifloxacin 12 is more broad-spectrum than the earlier generation agents and effective against ofloxacin-resistant Mtb strains. It is also used for prophylaxis of those exposed to MDR-TB based on its excellent tissue penetration to alveolar macrophages. ${ }^{55}$ Furthermore, moxifloxacin 12 can be used to treat MDR-TB, if the drug susceptibility testing does not show resistance against pyrazinamide 2, fluoroquinolones or aminoglycosides, and the patient does not have TB treatment history. ${ }^{55}$ However, the use of fluoroquinolones is not recommended for children, pregnant and lactating women due to their potential for serious side effects such as cartilage toxicity and mutagenesis. ${ }^{52}$

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## Thioamides

The oral second line treatments for MDR-TB are thioamides including ethionamide 16 (ETH) and prothionamide 17 (PTH) which are structural analogues of isonicotinic acid hydrazide 1 (Figure 13).




Figure 13: All isoniazid 1, pyrazinamide 2, ethionamide 16, prothionamide 17 are derived from nicotinamide or isonicotinamide.

They exhibit bactericidal activity against growing Mtb when the pro-drugs are exposed to enzymatic activation by monooxygenase EthA. Once activated, they form covalent adducts with nicotinamide adenine dinucleotide (NAD) binding the same target of INH 1 (InhA enzyme) resulting in inhibition of mycolic acid biosynthesis. Therefore, the gene mutation in inhA, which encodes the primary target of both INH 1 and thioamides 16, 17 is responsible for potential coresistance whilst KatG mutant strains resistant to INH 1 remain to be susceptible to thioamides 16, 17 because thioamides 16, 17 require a different activator enzyme (EthA) (Figure 14). ${ }^{56}$ Based on their ability to cross the blood brain barrier, thioamides 16, 17 are also recommended to manage drug-susceptible TB meningitis. ${ }^{57}$

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Figure 14: The mechanism of resistance of thioamides 16, 17 and isoniazid 1. (Adapted from Islam et al, $2019{ }^{56}$ )

## Amino acid

D-Cycloserine 18 (DCS) is an oral second line TB treatment which is used only for MDR-TB due to its neurological toxicity. ${ }^{58}$ DCS 18 is a broad-spectrum bacteriostatic agent which targets the action of essential enzymes such as Dalanine racemase (Alr) and D-alanine:D-alanine ligase (Ddl) involved in synthesis of cell wall peptidoglycan (Figure 15).


Figure 15: D-Cycloserine 18 impairs alanine conversion leading to disrupting peptidoglycan biosynthesis. (Adapted from Gallagher et al, $2019{ }^{59}$ )

The targeting of D -alanine racemase blocks the interconversion process of L alanine to D-alanine which acts as a substrate for D-alanine ligase. Due to inability of $D$-alanine to be incorporated into the peptidoglycan, the susceptibility

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of bacteria such as methicillin-resistant strains of $S$. aureus (MRSA), to $\beta$-lactams is increased by DCS 18. ${ }^{59}$ However, mutation of alrA gene in D-alanine racemase causes resistance to DCS 18 but no cross-resistance with other antitubercular agents. ${ }^{58}$

## Sulfonamides

Para-aminosalicyclic acid 20 (PAS), a structural analogue of folate precursor (paminobenzoic acid) 19, is more potent antitubercular agent than sulfonamides 21, however, it is used only for MDR-TB due to its poor efficacy and tolerability (Figure 16). ${ }^{60}$


21
Figure 16: The chemical structure of p-aminobenzoic acid 19 and its structural analogues such as p-aminosalicyclic acid 20 and sulfonamides 21.

The bacteria cannot use an external source of folic acid and thus PAS 20 shows bacteriostatic activity via targeting the folic acid biosynthesis. PAS 20 is a prodrug which is activated by dihydropteroate synthase (DHPS) and dihydrofolate synthase (DHFS) thus interferes with dihydrofolate reductase (DHFR) enzymatic activity (Figure 17). ${ }^{60}$ Furthermore, PAS 20 may block iron uptake, thus inhibiting the synthesis of mycobactin which is a mycobacterial cell wall component. ${ }^{61}$

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PAS bioactivation

20







igure 17: Incorporation of p-aminosalicyclic acid 20 instead of p-aminobenzoic acid 19 into folate pathway and antifolate action of sulfonamides 21. (Adapted from Thiede et al, $2016{ }^{62}$ )

### 1.3.6 XDR-TB

The inappropriate/ inaccurate diagnosis of MDR-TB and use of second line TB treatment increases the risk of resistance to more antitubercular agents leading to the development of extensively drug resistant TB (XDR-TB). It is challenging because drug resistant TB requires longer treatments (over two years) and occurrence of serious adverse effects have a negative impact on patient compliance. Moreover, lack of systematic drug susceptibility tests (DST), which are not always available in developing countries, contributes to inappropriate therapy selection, and therefore poor treatment outcomes. With no new treatments for 50 years, recently a plethora of new treatments have become available which are reserved as third line treatments. Bedaquiline (BDQ) 22, linezolid (LZD) 23 and delamanid (DLM) 25 / pretomanid (Pa) 26, were recently approved in a combination regimen (B-Pa-L) for XDR-TB by the US Food and Drug Administration (FDA). According to the Nix-TB trial at three South African sites, which enrolled 109 cases with XDR-TB and who were treated with BPaL regimen for six months. After the end of treatment, a successful outcome was

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attained in 98 patients who completed 26 weeks although the majority of surviving participants around $81 \%$ suffered from a peripheral neuropathy reported on treatment. For this reason, most cases required a linezolid dose adjustment or allowable interruptions of up to 35 consecutive days but no permanently discontinued during period of treatment. ${ }^{63}$

### 1.3.7 Third-line antituberculosis drugs

## Diarylquinolines

Bedaquiline (BDQ) 22 represents the first agent of diarylquinolines to be bactericidal against both dormant and growing Mtb via inhibiting adenosine triphosphate (ATP) synthesis (Figure 18). Herein, BDQ 22 interferes with the essential enzyme for ATP synthesis, the mycobacterial ATP synthase, but not with mammalian ATP synthase, thereby resulting in disrupting the energy production and subsequent bacterial death. BDQ 22 binds to ATP synthase through the transmembrane oligomeric subunit $C$, and therefore, BDQ 22 resistance arises from the mutation in atpE gene encoding this subunit. Furthermore, resistance is enhanced with mutations in a transcriptional repressor (rv0678) resulting in the upregulation of the MmpS5-MmpL5 efflux pump which is involved in non-target-based resistance to bedaquiline. ${ }^{64}$


Figure 18: The chemical structure for bedaquiline (BDQ) 22. (Adapted from Kumar et al, 2015 ${ }^{65}$ )

## Oxazolidinones

Linezolid (LZD) 23, the only FDA-approved oxazolidinone for clinical use, has bacteriostatic activity against drug-resistant gram-positive bacteria, particularly, methicillin-resistant strains of Staphylococcus aureus (MRSA) (Figure 19).

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Furthermore, LZD 23 showed significant activity against all Mtb strains (MIC90 values $\leq 0.5 \mu \mathrm{~g} / \mathrm{mL}$ ) as it inhibits an early step of bacterial protein synthesis by binding to ribosomal 50 S subunits, surrounded by $23 S$ rRNA nucleotides. Thus, mutation of 23 S rRNA can cause LZD 23 resistance, with poor MIC90 values $\geq 16$ $\mu \mathrm{g} / \mathrm{mL}$. However, in case of the prolonged use of LZD 23, a regular dosage adjustment is required because it can inhibit the mitochondrial protein synthesis leading to serious hematologic and neurologic toxicities. ${ }^{64}$ Therefore, Pfizer has developed sutezolid 24, a thiomorpholinyl analogue of linezolid 23, which showed significant bactericidal activity in a murine model of TB and a better safety profile in comparison with linezolid 23 (Figure 19). Both linezolid 23 and sutezolid 24 showed bactericidal activity against dormant Mtb and so oxazolidinone is gaining great interest as novel antitubercular class. ${ }^{66}$


Figure 19: The chemical structure for linezolid (LZD) 23 and sutezolid 24. (Adapted from Kumar et al, $2015{ }^{65}$ )

## Nitroimidazoles

Nitroimidazoles including delamanid 25 (DLM) and pretomanid 26 (Pa), are a new class of XDR-TB treatment, with potent activity against all strains of replicating and nonreplicating Mtb but DLM 25 (nitroimidazo-oxazole) is 20 times more potent than Pa 26 (nitroimidazo-oxazine) (MIC90 values $0.006 / 0.024 \mu \mathrm{~g} / \mathrm{mL}$ and $0.015 / 0.25 \mu \mathrm{~g} / \mathrm{mL}$, respectively). ${ }^{64} \mathrm{In}$ comparison to Pa 26 , the next generation

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of nitroimidazoles such as TBA354 27, (nitroimidazo-oxazine) showed greater activity against replicating and nonreplicating Mtb ( MIC $_{90}$ values 0.006 / 0.006 $\mu \mathrm{g} / \mathrm{mL}$ ) (Figure 20). ${ }^{67}$ In 2016, TB Alliance announced discontinuation of the human clinical trials of TBA354 27 because of its toxicity and pharmacokinetic data. ${ }^{68}$


Figure 20: The chemical structure for the nitroimidazole class, which includes delamanid 25 (DLM), pretomanid 26 and TBA354 27. (Adapted from Kumar et al, $2015{ }^{65}$ )

Because of inhibition of the biosynthesis of mycolic acids, their main active metabolite (desnitro-imidazooxazole) exhibits significant bactericidal and sterilizing activity when $\mathbf{2 5 / 2 6}$ are being activated by a deazaflavin-dependent nitroreductase (ddn) (Rv3547). ${ }^{64}$ Once activated, they show also anaerobic activity against nonreplicating Mtb through intracellular reactive nitrogen species release (Figure 21). ${ }^{69}$ Therefore, the gene mutation in ddn, which encodes the activating enzyme (ddn), and in fgd1 (Rv0407) are associated with resistance to nitroimidazoles. ${ }^{64}$ Since 2013, DLM 25 was approved for clinical use by the European Medicines Agency (EMA) although the use of nitroimidazole

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antituberculosis agents for long term treatment may be associated with cardiac problems such as QT prolongation which can cause fatal arrhythmias. ${ }^{64}$


Figure 21: The anaerobic activity of nitroimidazooxazines class against nonreplicating Mtb by intracellular reactive nitrogen species release. (Adapted from Singh et al, $2008{ }^{70}$ )

### 1.4 New antimicrobial treatments

As the rise in drug-resistant infections is presented as the main challenge to current antibiotic therapies due to their ability to inactivate conventional treatments, new classes of antibiotic are required with new biological targets to offer a broad spectrum of antimicrobial activity. A selection of these new antimicrobial agents and their modes of action will be discussed in the following subsections.

### 1.4.1 Antimicrobial peptides

Antimicrobial peptides (AMPs) are derived from wide variety of living organisms which constitute an essential component of their innate immune defense and thus, called host defense peptides (HDPs). Unlike most peptide-based antibiotics of bacterial origin (polymyxins) or fungal origin ( $\beta$-lactams), the ribosomally synthesized AMPs have been found to have broad spectrum and rapid microbiocidal activity due to their versatility. ${ }^{71}$ The antimicrobial activity of these

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peptides can be affected by several factors such as sequence, charge, conformation and structure, size, hydrophobicity and amphipathic stereo geometry. ${ }^{72}$ Moreover, AMPs can overcome the bacterial resistance through targeting the bacterial cell wall and not specific receptors based on antimicrobial peptide self-association to the cell membrane. ${ }^{71}$ Herein, AMPs were found to cause osmotic lysis of the bacterial cell by forming transmembrane channels which resulted from electrostatic interaction between the bacterial, anionic membranes (lipopolysaccharide or phospholipid for gram-negative and teichoic acids for gram-positive) and cationic peptides (containing excess lysine and arginine) (Figure 22). This peptide-lipid interaction contributes to increase the selectivity of AMPs towards bacterial targets over the host cells as the differences in the lipid composition of cell membrane between bacterial and eukaryote host cell membrane represent the targets for AMPs. ${ }^{73}$


Figure 22: Damaging cell wall through forming transmembrane ion channels. (Adapted from Sato and Feix, $2006{ }^{73}$ )

Besides having immunomodulatory properties, AMPs act as immune modulators which make them especially interesting agents and it is believed that they can stimulate the innate immune response through different mechanisms including inducing pro-inflammatory cytokine production or inhibiting bacterial lipopolysaccharide (endotoxine) (Figure 23).74

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Figure 23: Both direct antimicrobial activity and immune modulatory are used by AMPs.
(Adapted from Hancock and Sahl, $2006{ }^{74}$ )

In the database, more than 2,000 AMPs with variable length and amino acid composition have been classified based on their secondary structure into four main classes: $\beta$-sheet, $\alpha$-helical, loop and extended peptides (Table 5). ${ }^{75}$

| Class | Properties ${ }^{72}$ | Examples |
| :---: | :---: | :---: |
| $\beta$-sheet peptide | Composed of at least two antiparallel $\beta$-strands stabilized by two to four interchain disulfide bridges between these strands due to containing cysteine residue. | Human $\beta$ defensin 2 |
| a-helical peptide | Representing around 30 - 50\% of all AMPs. Composed of 12 40 amino acids particularly, helix stabilising residues such as alanine, leucine, and lysine. | Magainin 2 |
| Loop peptides | Composed of 20-46 amino acids with only a single disulfide bridge. | Bactenecin ? |

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| Extended <br> peptide | Rich in specific amino acids such <br> as glycine, proline, tryptophan, <br> arginine and histidine. | Indolicidin |
| :--- | :--- | :--- |

The natural AMPs have been investigated for more than 30 years. Unfortunately, they are associated with several shortcomings including incomplete understanding of their mechanism of action, reduced activity in higher pH and the high cost of production as compared to conventional antimicrobial agents. On top of this, pharmacokinetic and pharmacodynamic issues, for instance, AMPs show poor absorption after oral administration and cannot cross the blood-brain barrier because they are large and strongly cationic peptides. Moreover, they are susceptible to the rapid proteolytic degradation in both the bloodstream and the gastrointestinal tract which is a significant factor in bioavailability issue. ${ }^{76}$
Based on the described shortcomings, the development of peptidomimetic analogues which are fragments or derivatives of natural AMPs, have been an area of great interest. They can display their advantages over their natural peptides and can be used in medicine development because of their structural simplicity with low synthetic cost and ease of optimization. ${ }^{77}$ For instance, Dipeptide $\gamma$ - $D$-glutamyl-L-tryptophan 28, $\gamma$-glutamyl derivatives, boosts immune response by stimulating T-lymphocyte and secretion of interferon- $\gamma$ and interleukin-2 to promote bacterial clearance. Based on clinical outcomes, this molecule 28 was very effective in combination with standard TB treatment, therefore, it is a prospective antitubercular agent and its future demand in the pharmaceutical industry is expected (Figure 24). ${ }^{78}$


28
Figure 24: The chemical structure of $\gamma$ - D-glutamyl-L-tryptophan $28 .{ }^{78}$

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### 1.4.2 Antimicrobial hydrazides

The continuing demand for new and effective antimicrobial agents with novel mode of action is a never-ending task to tackling the development and spread of the antimicrobial resistance. In continuation of designing of new compounds possessing antimicrobial activity based on hydrazide skeleton, such as isonicotinylhydrazide (INH) 1 which is one of the attractive backbones in antibacterial research today due to its chemical reactivity and biological activities. Based on the aforementioned facts, isoniazid 1 therapy is considered a cornerstone of TB treatment although it has two major drawbacks. Prolonged treatment with isoniazid 1 causes hepatotoxicity because it is metabolized by acetylation and dehydrazination and produces an N -acetylhydrazine metabolite which is responsible for the hepatotoxic effect. ${ }^{79}$ The second is the action of N arylaminoacetyltransferases (NATs), which are present in both mycobacterial and mammalian cells, to deactivate INH 1 by acetylation reaction at $-\mathrm{NH}_{2}$ group. So, the blockage of $-\mathrm{NH}_{2}$ group by alkyl group, as in example the N -isopropylated form of isoniazid (Iproniazid) 29, has been developed primarily as antituberculosis agent and based on its mood-elevating effect so used as antidepressant. Later, it was withdrawn due to the high hepatotoxicity. Other examples of blocking the $-\mathrm{NH}_{2}$ group include isocarboxazid 30 which exhibits less hepatotoxicity, and is thus still available for clinical use as an antidepressant (Figure 25). ${ }^{80}$


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Figure 25: The chemical structure of Iproniazid 29 and Isocarboxazid 30.

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Subsequent chemical modification of blocking N -acetylation of INH 1 was demonstrated by synthesis INH hydrazide-hydrazones 31 (Schiff's bases derivatives) that led to more effective and less hepatotoxic derivatives, such as ftivazide 32, saluzide 33 and salizide 34 (Figure 26). ${ }^{81}$


Figure 26: The synthetic route of Schiff bases of isoniazid 31 such as ftivazide 32, saluzide 33 and salizide 34 via reacting INH 1 with aromatic aldehyde. (Adapted from Villamizar-Mogotocoro et al, $2020{ }^{81}$ )

They act as an optimal transport vehicle for INH 1 moiety inside both mycobacterial and mammalian cells (Figure 27). ${ }^{81}$ However, all the efforts of redesigning INH 1, that have been undertaken, in order to develop a more effective analogue against INH resistant Mtb strains, have unfortunately met without much success. ${ }^{82}$

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Figure 27: The proposed mechanism of action of Schiff bases of isoniazid 31. (Adapted from Villamizar-Mogotocoro et al, $2020{ }^{\text {81 }}$ )

The use of the hydrazide-hydrazone skeleton 35 as a starting material has long attracted attention for new antibiotic development. Furthermore, widely used antimicrobial agents such as nitrofurazone 36, furazolidone 37 and nitrofurantoin 38 further support the interest towards the synthesis of hydrazide-hydrazones agents exhibiting wide range of antimicrobial activity (Figure 28). ${ }^{83}$




Figure 28: Examples of current antimicrobial agents containing hydrazide-hydrazone skeleton 35 such as nitrofurazone 36 , furazolidone 37 and nitrofurantoin $38 .{ }^{83}$

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In the light of literature, there were recent attempts to develop new antitubercular compounds containing hydrazide-hydrazone moiety such as 4-fluorobenzoic acid hydrazide 39, ${ }^{84}$ trans-cinnamic acid hydrazides 40, ${ }^{85}$ fluorine containing hydrazones $41{ }^{86}$ and sudoterb $42{ }^{79}$ (Figure 29).


39


41


40


42

Figure 29: Examples of hydrazide-hydrazones possessing antimycobacterial activity.

### 1.4.3 Antimicrobial aryl-oxadiazoles

Despite increased research efforts to discover new antibiotics, treatments for the resistant infections still remain a challenging task, as conventional antibiotics are often ineffective. In recent decades, literature studies reveal a vast number of nitrogen containing heterocyclic compounds that display better biological activity than non-nitrogen compounds. ${ }^{87}$ It is not surprising, therefore that the system of aryl-oxadiazoles, five membered aromatic heterocycles, have been investigated with a view to their use in the medicinal chemistry. It is known that this system of aryl-oxadiazoles contains four regioisomers including 1,2,3-, 1,2,4-, 1,3,4- or 1,2,5-oxadiazoles which have different charge distributions (Figure 30). ${ }^{88}$


1,2,3-oxadiazole


1,2,4-oxadiazole


1,3,4-oxadiazole


1,2,5-oxadiazole

Figure 30: Oxadiazole regioisomers. ${ }^{88}$

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However, 1,2,4- or 1,3,4-oxadiazoles are widely studied and exhibit wide spectrum of biological activities. Therefore, they are present in several drugs 43,
44, 45, 46 and 47 in clinical use containing oxadiazole in their backbone (Figure 31). ${ }^{89}$




Figure 31: Chemical structure of marketed medicines containing oxadiazole units. ${ }^{89}$

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Due to the ongoing problem of bacterial resistance to $\beta$-lactam antibiotics, the scoring of series of $1,2,4$-oxadiazoles against the crystal structure of a penicillinbinding protein through an in-silico analysis was carried out. They, particularly compound 48, showed antibacterial activity against gram positive bacterial strains including vancomycin-resistant, linezolid-resistant and methicillinresistant Staphylococcus aureus (Figure 27). ${ }^{90}$ In 2018, Zheng et al, reported that the 1,3,4-oxadiazoles, particularly compound 49, showed bactericidal activity against $S$. aureus by interfering with biofilm formation which is essential for staphylococcal pathogenesis. The same research group suggested the presence of the toxophoric $-\mathrm{N}=\mathrm{C}-\mathrm{O}$ - linkage group may be responsible for antibacterial activity by reacting with the nucleophilic centers of the bacterial cells (Figure 32). ${ }^{91}$


48


49

Figure 32: Compound 48 was efficacious against MRSA strain in a mouse model. ${ }^{90}$ Compound 49 demonstrated potent antibacterial activity against S. aureus. ${ }^{91}$

In connection to the antitububercular studies, the continued emergence of resistance to TB treatments has posed a challenge for treating MDR-TB and XDR-TB. Thus, the aryl-oxadiazoles have attracted many research groups to develop new TB treatments. For instance, the antitubercular activity of 2-(4-nitro-pyrrol-2-yl)-5-aryl-1,3,4- oxadiazole derivatives were screened against Mtb ( $\mathrm{H}_{37} \mathrm{Rv}$ ) and promising compound 50 showed lowest MIC value $(0.46 \mu \mathrm{M})$ which is close to INH $1(0.40 \mu \mathrm{M})$ (Figure 33). ${ }^{92}$


Figure 33: Compound $\mathbf{5 0}$ showed promising antitububercular activity. ${ }^{92}$

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In an other antitubercular activity investigation of novel 1,3,4- oxadiazoles, compounds were screened against two strains of $M t b$ ( $\mathrm{H}_{37} \mathrm{Rv}$ and isoniazidresistant) and compound 51 exhibited the highest antitubercular activity ( $0.78 \mu \mathrm{M}$ and $1.52 \mu \mathrm{M}$, respectively) (Figure 34). ${ }^{93}$


51
Figure 34: Compound 51 showed promising antitububercular activity against wild-type and isoniazid mono-resistant Mtb. ${ }^{93}$

### 1.5 Antimicrobial Resistance

With a rise in drug resistant bacteria, MDR-TB is one of more infections developed by treatment-resistant strains. It is similar to malaria, methicillinresistant S. aureus (MRSA) or vancomycin-resistant enterococci (VRE) infections, which have become serious global health threats in this century. The resistance to at least 4 of the core TB treatments brings significant challenges. Understanding these mechanisms allows for new TB treatments to be developed. ${ }^{94}$

### 1.5.1 INH resistance or resistance to front line drugs

Despite the potent antimycobacterial effect of INH 1, INH 1 resistance is considered as a surrogate marker of MDR-TB. Mtb uses several mechanisms including inactivating enzymes and the overexpression of genes which encode either the target enzyme or an enzyme that has a critical role in drug activation, to resist the action of INH 1. A primary mechanism of INH 1 resistance arises from mutation in katG gene that encodes the katG enzyme and alters its activity toward INH 1, resulting in fewer isonicotonic acyl radicals 5 being generated. Once activated, the second most common mutation occurs in the inhA gene which is associated with reducing the affinity of active metabolite 5 to the target enzyme (InhA) (Figure 35). ${ }^{95}$ A recent study reported that that low-level resistance to INH 1 is provided by gene mutations in acpM and kasA that encode two intracellular targets malonyl acyl carrier protein (AcpM) and a $\beta$-ketoacyl-ACP synthase (KasA) which are involved in process of fatty acid elongation (Figure 35). Additionally,

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active metabolite 5 has been shown to inhibit dihydrofolate reductase, a key enzyme for nucleic acid synthesis, therefore, there is correlation between INH 1 resistance and mutation in the dfrA gene, but this is still not proven. ${ }^{96}$


Figure 35: The genetic mutations are responsible for INH 1 resistance where (*) indicates to targeted genes including inhA, acpM and kasA.

Notwithstanding this, Mtb is becoming resistant to rifampicin 4 because of extensive usage of RIF 4 and its derivatives. The bacterial resistance to both RIF 4 and INH 1 has been observed among greater than 90\% of MDR-TB strains as this is responsible for worse TB treatment outcomes. Whilst not fully understood, rifampicin monooxygenase (RIFMO) could be contributing to the decreased antitubercular activity of rifampicin 4 by the N -hydroxylation of rifampicin 4 to 2 '- N -hydroxy-4-oxo-rifampicin 52 (Scheme 2). ${ }^{97}$


Scheme 2: Conversion of Rifampicin 4 to its inactive metabolite 52 through RIFMO enzyme. ${ }^{97}$

The major rifampicin-resistant strains of Mtb emerge from gene mutation in molecular target. The harbor mutations within 81 base pair of rpoB gene can

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cause a structural change in $\beta$-subunit of RNA polymerase which reduces the binding affinity to RIF 4. As a result of this, most laboratories examine the 81 base pair region to determine RIF 4 resistance by using molecular methods. ${ }^{95}$ Cross resistance occurs in conjunction with other rifamycins such as rifabutin 53 and rifapentine 54 and this is important for TB treatment in patients coinfected with HIV (Figure 36). In this case, rifabutin 53 is more preferred than RIF 4 because it is a weak inducer of the cytochrome P450 that doesn't increase the metabolism of antiretroviral therapy. ${ }^{95}$



Figure 36: Other rifamycins such as rifabutin 53 and rifapentine $54 .{ }^{95}$

In Mtb, PAZ 2 is activated by pyrazinamidase encoded by the pncA gene whilst pyrazinamidase usually loses its activity in most PAZ 2-resistant Mtb strains because of the structural gene mutation in $p n c A$. In addition to $p n c A$ regulation, pyrazinamide-resistant Mtb strains showed other resistance mechanisms including pyrazinamide uptake and pyrazinoic acid efflux. ${ }^{95}$ The resistance to PAZ 2 increases the risk of treatment failure in cases with drug-susceptible or drug-resistant TB because it provides synergistic effects when it is included in MDR-TB regimen, especially, with bedaquiline 22 and pretomanid $26 .{ }^{98}$

Not unexpectedly, Ethambutol 3 is not only part of front-line treatment, but is structurally related to the new antitubercular agent SQ 10955 (Figure 37). ${ }^{99}$ Unfortunately, several studies have shown EMB 3 resistance is commonly linked to gene mutations in embB which encodes its target enzyme. ${ }^{96}$ Interestingly, drug-resistant strains of Mtb, including EMB 3-resistant Mtb strains are susceptible to SQ109 55 with no likelihood of cross-resistance with EMB 3. Therefore, SQ109 55 is suggested to be an ideal alternative for EMB 3. ${ }^{99}$

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3


Figure 37: Process to develop new antitubercular agent (SQ109 55) based on 1,2ethylenediamine. ${ }^{99}$
The action and mechanisms of resistance to front-line TB treatments are still not fully understood, although the genetic mutations that are associated with each form of drug resistance have been detected by effective approaches such as whole-genome sequencing analysis. This is because TB treatments are usually tested against a limited number of the Mtb complex genotypes. In addition to detection of genetic mutations, further investigation about drug-resistant TB is required because other important information including the level of resistance, cross-resistance to other agents, relatedness of strains, and virulence, are increasingly being used to control the spreading of antibiotic resistance, particularly in the context of TB. ${ }^{100}$

### 1.6 Conclusion

In this chapter, facts about global TB epidemiology, themes related to Mtb, TB vaccination and treatments as well as increasing emergence of drug resistant forms of TB, especially MDR-TB and XDR-TB have been outbred. TB treatment research has been in decline since the 1960s due to introduction of INH 1 and RIF 4 until the emergence of drug-resistant TB in the end of the 20th century. Due to the increased prevalence of MDR-TB and XDR-TB, the objective of the current research is to design and synthesis novel antitubercular agents with new molecular targets to tackle the continuing global burden of this infection and the efficacy and cost limitations associated with current therapies. The hydrazide functional group compound of interest in this work could be easily modified to afford benzoxa-[2,1,3]-diazole substituted amino acid hydrazides as potential antitubercular agents.

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### 1.7 Previous work in the group

Previous studies in the group have generated a small library of diverse bioactive molecules (~100) to ascertain their utility as potential antibacterials. A Resazurin Microtiter Assay (REMA) $)^{101}$ was undertaken at a fixed concentration ( $128 \mu \mathrm{~g} / \mathrm{mL}$ ) against a range of drug susceptible bacteria including gram positive, gram negative and mycolata bacteria. ${ }^{1}$ Whilst many compounds demonstrated little activity at these high concentrations, several classes were successful at retarding bacterial growth. Analysis of the compounds revealed a variety of architectures including flavones 56, purines 57, benzo-[2,1,3]-diazoles 58, quinolines 59, indolin-2-ones 60 and thiocoumarins 61 that provided positive results at this concentration (Figure 38).


56


57


60


58


61

Figure 38: The chemical structure of flavone 56, purine 57, benzo-[2,1,3]-diazole 58, quinoline 59, indolin-2-one 60 and thiocoumarin 61.

Further analysis of the library revealed a class of molecules demonstrating activity against only mycolata bacteria. These molecules possess a core comprising a benzo-[2,1,3]-diazole that to the best of our knowledge, have been utilised as anti-cancer treatments ${ }^{102}$ and as herbicides ${ }^{103}$ in only a small number of studies. Consequently, we focused on this pharmacophore and undertook another REMA assay on a reduced set of organisms to determine the minimum inhibitory concentrations of these molecules, and to determine their utility for subsequent investigations (Table 6).

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Table 6: Minimum inhibitory concentration (MIC) results against gram positive, negative and mycolata bacteria for our library.


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The results reveal a mix of activity against the tested organisms, with simple substituted benzo-[2,1,3]-diazole compounds leading to broad organism activity and potency, mirroring the initial screening results (Entries 58a-58d, Table 6). Pleasingly, benzo-[2,1,3]-diazole molecules 62 possessing a sulfonamide peptidomimetic yielded far greater specificity and activity against mycolata bacteria including Mtb (Entry 63a, 64a-b and 65a, Table 6).

In order to explore this architecture further, subsequent experiments were undertaken to produce a variety of benzo-[2,1,3]-diazoles 65a-g and 64a-b to understand their activity against Mtb (Figure 39).



65\%
MIC $_{\text {WT. Mtb }} \mathbf{3 1 . 1 3 \mu \mathrm { M }}$


65b
MIC ${ }_{\text {WT. Mtb }} 17.25 \mu \mathrm{M}$

$65 \%$
MIC ${ }_{\text {WT. Mtb }} \mathbf{6 0 . 8 5 \mu \mathrm { M }}$


85\%
MIC $_{\text {WT. Mtb }} 59.27 \mu \mathrm{M}$


65e
85\%
MIC $_{\text {WT. Mtb }} 16.33 \mu \mathrm{M}$


65f
22\% MIC $_{\text {WT. Mtb }} 19.22 \mu \mathrm{M}$


65g
60\%
MIC ${ }_{\text {WT. Mtb }} 36.67 \mu \mathrm{M}$



64a
61\%
MIC $_{\text {WT. Mtb }} 148.4 \mu \mathrm{M}$


64b
21\%
MIC ${ }_{\text {WT. Mtb }} 44 \mu \mathrm{M}$

Figure 39: A series of benzo-[2,1,3]-diazoles 65a-g and 64a-b showed different MIC value against WT. Mtb. Where (\%) indicates to an actual yield of resulting compound.

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In this small series of compounds, we retained the hydrazine and benzo-[2,1,3]diazole component and explored the amino acid. It can be seen for compounds 65b-65d, an increase in MIC was observed, due to the increased size of the amino acid side chain. Curiously, a similar level of activity is observed with 65e as 65b, suggesting that there may be a bend or kink which can be exploited by these substrates. Notwithstanding this, retaining unsubstituted amino acids and benzo-[2,1,3]-diazole component allowed us to explore the hydrazine fragment. Interestingly, the presence of a chloro-substituent at the meta-position as seen in $65 f$ produced a lower MIC than $\mathbf{6 5 g}$ which is substituent-free suggesting that space in this fragment is available for exploitation. Finally, retaining the unsubstituted amino acid and hydrazine whilst exploring the benzo-[2,1,3]diazole indicated a further shift in MIC as replacement of benzoxa-[2,1,3]-diazoles in 64a-64b by benzothia-[2,1,3]-diazole to investigate their antimycobacterial activity showed similar or lowers levels of inhibitory activity towards Mtb (cf. 65a vs $\mathbf{6 4 a}$ ) and (cf. $\mathbf{6 5 g}$ vs $\mathbf{6 4 b}$ ). ${ }^{1}$

## Aims and Objectives

As discussed previously (cf. section 1.3.2), the requirement of highly toxic and less effective TB treatments with longer duration to treat MDR-TB or XDR-TB is a principal factor for continuous slow rise in TB infection. One of the strategies to tackle the problematic exacerbation of drug-resistant TB is an identification of new chemical entity possessing new mechanism of action which will open a new TB therapeutic avenue.
In previous work from within the group, the benzo-[2,1,3]-diazole peptidomimetics architecture 65 has demonstrated significant selectivity and mycobacterial activity. Therefore, the overall aim of this thesis is to to generate a structure activity relationship study with these substrates to further understand which components of the molecules are crucial for activity against Mtb.
Subsequent chemical investigations were undertaken to produce a variety of new library of novel benzoxa-[2,1,3]-diazole substituted amino acid hydrazides 65 to understand their activity against Mtb. We wanted to explore how the structural modification may influence the inhibition of mycobacterial growth through
screening the resulting compounds against susceptible and resistant Mtb strains utilising a Resazurin Microtiter Assay (REMA).
We approached the SAR analysis for the benzo-[2,1,3]-diazole peptidomimetics 65 to investigate the chemical structure to biological activity of this architecture further and to accelerate trends toward improved understanding of the molecular modifications which can improve the antitubercular activity and leads to a potent lead compound. The cytotoxicity of the final compounds 65 and their intermediates 67 was also examined on mammalian cell lines.

## Chapter 2: Synthesis of benzo-[2,1,3]-diazoles peptidomimetics

## 2 Synthesis of benzo-[2,1,3]-diazoles peptidomimetics

### 2.1 Introduction

In an effort to continue the previous work on benzo-[2,1,3]-diazole peptidomimetics, an SAR investigation was initiated to explore these novel molecules. After initial investigation into these novel antitubercular agents, the aim is to contribute more analogues of the lead compounds 65 to determine which parts are necessary for activity and ultimately improve their potency, affinity and reduce toxicity.

To begin with, a retrosynthetic analysis of the molecule 65 was undertaken. The initial disconnection breaks the sulfonamide bond of 65 to produce the commercially available benzoxa-[2,1,3]-diazole 66 and the amino acid hydrazide 67 or 68. Further disconnection of the amino acid hydrazide 67 or 68 at the hydrazide bond to produce commercially available aryl hydrazines 69 and N protected amino acids 70 (Figure 40).


Figure 40: Splitting the architecture of final compound 65 into its substrates.
In order to produce the target compounds, an initial peptide coupling reaction between a variety of $N$-protected amino acids 70 with a large number of aryl hydrazines 69 would give rise to hydrazides 67 and 68. Following successful

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coupling, deprotection of intermediates $\mathbf{6 7}$ or $\mathbf{6 8}$ will produce the free amine and allow for its reaction with the benzoxa-[2,1,3]-diazole sulfonyl chloride 66 to deliver the final benzoxa-[2,1,3]-diazole products 65.

Whilst the use of Boc protected amino acids 67 will predominate, there was be a requirement to use Fmoc protected amino acids 68. For when reactive side chains are present on the amino acid, an orthogonal process will be required to deprotect the backbone amine whilst having the side chain protection remain intact.

Consequently, both of the hydrazides 67 and 68 and the final benzo-[2,1,3]diazole products 65 will be screened against Mtb utilising the REMA assay to allow for comparative analysis.

### 2.1.1 Synthesis of Amino acid hydrazides

To get started, the first goal was to synthesise a library of hydrazides 67 and 68. To carry out the initial SAR investigation, we wished to explore the use of a variety of substituted aryl hydrazines 69 with the full range of amino acids $\mathbf{7 0}$. To focus the study on the amino acids 70 we chose to limit the number of aryl hydrazines 69 based on the activity shown in the previous study. ${ }^{1}$ Therefore, we looked at utilising predominantly electron withdrawing groups, donating groups and their position around the aromatic ring. Specifically, trifluoromethyl, halogens (including fluoride, chloride, bromide), nitro, nitrile, methanesulfonyl, and alkyl groups such as methyl, and isopropyl.

We began by synthesising the hydrazides 67 and 68 following the established procedure within the group. This required the coupling of amino acids 70 with aryl hydrazines 69 using standard peptide coupling reagents $N, N^{\prime}$ dicyclohexylcarbodiimide (DCC) 71 and 1-hydroxybenzotriazole HOBt 72 to afford the desired compound 67 and 68 (Scheme 3). ${ }^{1}$

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Scheme 3: Reagents and conditions: i.) DCC, HOBt, THF, 2-4 h, (47-95\%), r.t. ${ }^{1}$
Following the outlined procedure (cf. Scheme 3), the desired hydrazides 67 and 68 were synthesised successfully, however, the carbodiimides lead to the formation of a number of by-products. ${ }^{104}$ As expected, this coupling method gave rise to a very poorly water-soluble by-product (1,3-dicyclohexylurea) 73 that was difficult to separate from the product (Figure 42). ${ }^{105}$


Figure 42: DCC 71 / HOBt 72 coupling mechanism forming insoluble by-product DCU 73.
Many attempts were made to remove DCU 73, initially by quenching the reaction once it had reached completion with n-hexane and then cooling the sample overnight in fridge. The precipitated DCU 73 was then removed by filtration, ${ }^{106}$ however, some of the DCU 73 remained in the sample. Whilst it was suspected that small amounts of DCU contamination might be tolerated in the subsequent

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coupling reaction, it was decided to repeat the reaction and carry on the synthesis without rigorous purification. Unfortunately, this approach was unsuccessful with subsequent coupling to the sulfonyl chloride 66 producing a complex mixture of products. Consequently, the experiment was repeated and after extraction of the crude product and following n-hexane precipitation and filtration of DCU, flash chromatography was used to obtain a pure hydrazide. After extensive purification, the desired hydrazide 67q was isolated in 29\% yield, when DCC 71 and HOBt 72 (cf. Scheme 3) were used as amide coupling reagents (Figures 43 and 44).
${ }^{1} \mathrm{H}$-NMR spectra of crude compound



Figure 43: ${ }^{1} \mathrm{H}$ NMR spectra of crude hydrazide $\mathbf{6 7 q}$ showing the excess of HOBt 72 and DCU 73 by-product which were still shown in ${ }^{1} \mathrm{H}$ NMR spectra of filtrated hydrazide $\mathbf{6 7 q}$.

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${ }^{1} \mathrm{H}$-NMR spectra of pure compound



Figure 44: ${ }^{1} \mathrm{H}$ NMR spectra of pure hydrazide 67 q which had successfully been purified by using flash chromatography.

Synthesis of the hydrazide $\mathbf{6 7 q}$ was confirmed by ${ }^{1} \mathrm{H}$ NMR and MS with peaks at 7.11 (Arylhydrazine-H), 6.84 (Arylhydrazine-H), 6.79 (Arylhydrazine-H), 6.67 (3-Arylhydrazine-H), 3.88 (glycine- $\mathrm{CH}_{2}$ ), and 1.47 (Boc), ppm in the ${ }^{1} \mathrm{H}$ NMR spectrum for the aromatic hydrazine and amino acid components.

Whilst this method was able to provide pure products, the requirement of filtration and chromatography resulted in the preclusion of the rapid synthesis of a library of hydrazides. These steps also reduced the overall yield resulting in insufficient amounts of compound for sulfonamide coupling; thus, an alternative strategy was sought.

To circumvent this problem, we looked in the literature for an alternative coupling partner. 4-4,6-Dimethoxy-1,3,5-triazin-2-yl)-4-methyl-morpholinium (DMTMM) 74 has been highlighted as an effective activating agent for amide coupling reactions and for difficult couplings e.g. proline. ${ }^{107}$ Also, DMTMM has been shown to reduce the risk of racemisation of the chiral amino acids. ${ }^{107}$

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With this in mind, the reaction was repeated but this time the coupling of the free amine of aryl hydrazine 69 to carboxylic acid of amino acids 70 was attempted using DMTMM 74 and N,N-diisopropylethylamine (DIPEA) 75 in THF (Scheme 4). ${ }^{107}$


Scheme 4: Reagents and conditions: i.) DMTMM, DIPEA, THF, 6-8 h, (59-87\%), r.t.
$N, N$-diisopropylethylamine (DIPEA) 75 was added to deprotonate the carboxylic acid and/or neutralise the aryl hydrazine hydrochlorides 69 where the salts were present. ${ }^{107}$

Unlike DCC, this coupling method gave rise to a non-toxic triazinone by-product 76, which is insoluble in diethyl ether, thus the minor traces were removed through trituration (Figure 45).


Figure 46: DMTMM 74 / DIPEA 75 coupling mechanism forming insoluble a triazinone byproduct 76.
Although it worked very well as an amide coupling reagent in our hands, this triazine chemistry has limitations such as stability in organic solution and high

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cost. ${ }^{104,108}$ According to the qualitative stability studies on DMTMM 74, it undergoes demethylation at the morpholinium nitrogen by $50 \%$ in 15 minutes when suspended in DMF and approximately 2 hours in DMSO. While this selfimmolative degradation in chloroform would take 3 hours to be completed (Figure 47). ${ }^{108,109}$


Figure 47: Self-immolative degradation of DMTMM CI $\mathbf{7 4}$ into $\mathbf{7 4 a}$ in organic solution. ${ }^{107}$ Nevertheless, in THF it was found to be stable with only 13\% 74a detected after significant time around 13 hours. ${ }^{108,109}$ This self-immolative degradation will of course have a negative impact on the stoichiometry of the reaction. The production of chloromethane will also have ramifications for the yield and purity of the desired compound as it could methylate carboxylate substrates (giving the corresponding esters) or alkylate elsewhere in the substrate or product. ${ }^{107}$ Another drawback is that, in some cases, the excess triazinone by-product 76 needed to be removed by flash chromatography. Consequently, this method is time consuming and requires more resource to produce the valuable hydrazides. After extensive purification, the desired hydrazide 67EI was isolated in $60 \%$ yield, when DMTMM 74 (cf. Scheme 4) was used as an amide coupling reagent (Figure 48).

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${ }^{1} \mathrm{H}$ NMR spectra of crude compound


67EI

${ }^{1} \mathrm{H}$ NMR spectra of pure compound


67EI


Figure 48: Example ${ }^{1} \mathrm{H}$ NMR spectra of crude hydrazide 67 El showing the excess triazinone byproduct 76 that was successfully removed using flash chromatography.
Synthesis of the hydrazide 67EI was confirmed by ${ }^{1} \mathrm{H}$ NMR and MS with peaks at $7.33-7.30$ (phenylalanine-H), $7.23-7.20$ (phenylalanine-H), $7.03-7.00$ (Arylhydrazine-H), 6.58 - 6.55 (Arylhydrazine-H), 4.45 (phenylalanine-CH), 3.10 (phenylalaine- $\mathrm{CH}_{2}$ ), 2.86 (isopropyl- CH ), 1.44 (Boc), $1.20-1.18$ (isopropyl- $\mathrm{CH}_{3}$ ) ppm in the ${ }^{1} \mathrm{H}$ NMR for the aromatic hydrazine and amino acid components.

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Consequently, a change of coupling conditions was considered to increase the brevity of the process of hydrazide synthesis. We focused again on peptide coupling reagents and found that utilising $2-(1 H$-benzotriazol- $1-\mathrm{yl})-1,1,3,3-$ tetramethyluronium hexafluorophosphate (HBTU) 77 led to a more efficient coupling method and higher yielding synthesis, eliminating the need for flash chromatography (Scheme 5).


Scheme 5: Reagents and conditions: i.) HBTU, DIPEA, THF, 5-6 h, (47-96\%) r.t. Whilst HBTU 77 is an economical alternative coupling reagent, the tetramethyl urea (TMU) 78 by-product can be removed during work-up, is highly water soluble and has poor solubility in reduced polarity organic solvents such as diethyl ether. In addition to a quicker purification process, HBTU presents a low risk of racemisation (Figure 49). ${ }^{110}$

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Figure 49: HBTU 77 / DIPEA 75 amide coupling mechanism forming a water-soluble by-product tetramethyl urea $78 .{ }^{110}$

### 2.1.1.1 Enantiomeric purity determination

With HBTU working well to produce the hydrazides 67 and 68, the configurational integrity of these products must be determined. The duality, good yield and the preservation of optical purity is important as the biological activity of chiral compounds depend strongly on the stereochemical configurations. Recently, more than $50 \%$ of medications in clinical use are chiral compounds. ${ }^{111}$

Chirality is important as differing enantiomers of drug molecules may exhibit marked differences in pharmacodynamics and pharmacokinetics. For example, ethambutol (EMB) 3 which is a member of front-line TB treatments, shows a tuberculostatic activity via the (S,S)-isomer while its (R,R)-isomer produces optical neuritis, which does lead to blindness (Figure 50). ${ }^{111}$

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Figure 50: Concerning the absolute stereochemistry of the enantiomeric $(-)-(R, R)$ and (+)-(S,S)-ethambutol. ${ }^{112}$
Although enantiomers of chiral compounds have the same chemical structure, they exhibit different optical activity. One of these enantiomers will rotate planepolarized light counter clockwise and be laevorotatory and is by convention given a minus (-) sign, while the other which has a clockwise rotation will be dexorotatory and is given a plus sign (+). ${ }^{111}$ With the potential for unexpected racemisation of the chiral carbon of the amino acid 70 in the first step of the synthesis upon activation of the $\alpha$-carboxylic acid with HBTU, it was important to investigate the chiroptical properties of the obtained hydrazides. In order to do this, a polarimeter was used to measure the specific optical rotation ([a]D) of hydrazides 67 r with $S$ enantiomer of alanine and 67 n with $R$ enantiomer of alanine. ${ }^{111}$ Consequently, the hydrazides $\mathbf{6 7 r}$ and $\mathbf{6 7 n}$ were dissolved ( 5 mg in 1 mL of methanol) and analysed with an Optical Activity PoIAAR 2001 automatic polarimeter. The hydrazides 67 r and 67 n produced specific optical rotation ( $[\alpha]_{\mathrm{D}}$ ) of (-) 35 and (+) 33.4 respectively. This result confirms that the chiral properties of the enantiomers has not been compromised during the coupling step.

To further confirm the specific optical rotation, another hydrazide pair 67b with $S$ and 67 An with $R$ enantiomers of alanine was also performed. 67 b and 67 An were synthesized utilizing DMTMM 74 as coupling reagent with a good resistance to racemization. The [a]D values were measured at higher concentration ( $10 \mathrm{mg} / \mathrm{mL}$ ) for both returning values of (-) 47.5 and (+) 45.8 respectively, again confirming that this coupling route provides optically pure compounds (Figure 51).

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67r
(-) 0.175


67b
(-) 0.475


67n
(+) 0.167


67An
(+) 0.458

Figure 51: The corresponding hydrazides $\mathbf{6 7 r}, \mathbf{6 7 n}, \mathbf{6 7 b}$ and $\mathbf{6 7 A n}$ with their observed optical rotation values.

Taking the analysis one step further, chiral HPLC was utilized to further confirm the chiral purity of hydrazides 67 prior to their biological evaluation. Consequently, the hydrazides $67 r$ with $S$ and $67 n$ with $R$ enantiomers of alanine were ran separately through the chiral HPLC system using a Daicel Chiralpak® IB column ( $1 \mathrm{~mL} /$ minute, $94: 5: 1$ acetonitrile: methanol: $0.1 \%$ formic acid) to confirm the obtained hydrazides 67 were contained only as a single enantiomer (Figure 52 and 53).

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Figure 52: chiral HPLC of hydrazides 67 r indicating the presence the only one single peak,
Daicel Chiralpak® IB column, $4.6 \times 250 \mathrm{~mm}, 5 \mu \mathrm{~m}$ particle size, acetonitrile: methanol: $0.1 \%$ formic acid (94:5:1), $1 \mathrm{~mL} / \mathrm{min}$, UV at 260 nm .


Figure 53: chiral HPLC of hydrazides $\mathbf{6 7 n}$ indicating the presence the only one single peak, Daicel Chiralpak® IB column, $4.6 \times 250 \mathrm{~mm}, 5 \mu \mathrm{~m}$ particle size, acetonitrile: methanol: $0.1 \%$ formic acid (94:5:1), $1 \mathrm{~mL} / \mathrm{min}$, UV at 260 nm .

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With sufficient evidence to confirm that the coupling step does not racemise the chiral centre, we were fortunate enough to be able to obtain a crystal structure for compound 67Aw containing L-phenylalanine by slow evaporation from a dichloromethane solution. Analysis of the crystal structure led to the determination of the unambiguous structure of hydrazide 67Aw and assignment of absolute structure, with the flack parameter determined to be 0.008(15) (Figure 54). ${ }^{113}$


Figure 54: Crystal structure determination of non-centrosymmetric hydrazide 67 Aw with flack parameter close to zero indicating high confidence.

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### 2.1.1.2 Generation of a Library of Amino acid hydrazides

Having established coupling conditions that would allow for enantiomerically pure hydrazides to be created, subsequent studies focused on synthesis of a series of hydrazides 67 and 68 to evaluate their structure activity relationship. Initially the decision was made to vary the $L$-amino acids 70a-t whilst utilising a simple hydrazine, phenylhydrazine 69a to assess the synthesis on the full scope of substrates. Additionally, in our previous studies, hydrazide 67a was synthesised, screened and returned an MIC of $90.46 \mu \mathrm{M}$ against wildtype Mtb (Figure 55). ${ }^{1}$ During the course of our research it was observed that some of the resulting compounds had been produced as a mixture of rotamers, especially when with a Boc-Gly-OH unit. This was not investigated until the final compounds were produced and the results are presented in section 2.1.2.6.


67a
55\%
4:1
$90.46 \mu \mathrm{M}$
Figure 55: Structure of unsubstituted amino acid hydrazide 67a with a percent of actual yield, rotamer ratio and MIC value against the wild-type Mtb. ${ }^{1}$
Consequently, the first ten compounds produced were Boc-protected hydrazides 67a-j and the Fmoc-protected hydrazides 68a-j (Figure 56). Both were characterized as previously described and afforded in an acceptable yield.


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67j
$70 \%$


68d
53\%


68a
63\%


68b
63\%


68e
49\%


68f
23\%


68c 41\%


Figure 56: The corresponding Boc-protected hydrazides 67a-j and Fmoc-hydrazides 68a-j where (\%) a percent of actual yield and (-) lack of rotamer ratio.

In addition to the L-amino acids, three unnatural L-amino acids were also coupled with phenylhydrazine to assess the contribution of modified side chains in improving the potency of the resulting hydrazides $\mathbf{6 7 k}$-m (Figure 57).


Figure 57: The corresponding hydrazides $\mathbf{6 7 k}$-m containing unnatural L-amino acids where (\%) a percent of actual yield and (-) lack of rotamer ratio.

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In view of the role of the chirality in the development of antibiotics ${ }^{114}$ and in order to expand the scope of the SAR study, we produced the hydrazides $67 \mathrm{n}-\mathrm{p}$ with amino acids in D- configuration to compare their antitubercular activity to that of the hydrazides $\mathbf{6 7 b}, 67 \mathrm{~g}$ and $\mathbf{6 7 h}$ containing natural L-amino acids to determine the most active enantiomer (Figure 58).


Figure 58: The corresponding hydrazides 67n-p containing $D$-amino acids where (\%) a percent of actual yield and (-) lack of rotamer ratio.

## Electron withdrawing groups

Completion of this initial series saw attention turn to changes to the hydrazine to realise the aim of the project. We next investigated substitution of the aromatic ring with halogens, a common modification in medicinal chemistry. ${ }^{15}$ Their incorporation, especially the lighter fluorine and chlorine atoms, enhance permeability by increasing $\log P$ and allow the molecule to occupy deeper hydrophobic regions of the target protein. ${ }^{115}$

Moreover, as shown in previous work and from work ongoing within the group, the presence of chlorine in the meta position with enlarging side chains of the Lamino acid fragment, provides promising antitubercular activity (Figure 59). ${ }^{1,2}$


Figure 59: Results of synthesis of substituted amino acid hydrazides $\mathbf{6 7 q} \mathbf{q}, \mathbf{r}, \mathbf{w}$ and their MIC values. ${ }^{1,2}$
So, to expand this area and investigate the impact of chlorine at the meta position of the aryl hydrazine fragment 69, synthesis of hydrazides was expanded to

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include all L-amino acids, three selected unnatural L-amino acids, and D-amino acids (Figure 60).


67s 46\%


34\%


68\%

$67 v$
$61 \%$


67y
74\%

83\%









24\%




Figure 60: The chlorinated Boc- $N$-hydrazides 67q-z, Fmoc- $N$-hydrazides 68k-t, Boc- $N$ hydrazides 67Aa-Ac containing unnatural $L$-amino acids and Boc- $N$-hydrazides 67Ad-Af containing $D$-amino acids at the meta position where (\%) a percent of actual yield and (-) lack of rotamer ratio.

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With promising MIC results for chlorinated hydrazides 67r, w from our previous studies, exploration of mono-substituted Boc-hydrazides with halogens attached was something to be explored further. Consequently, we initiated a study to explore the 2- and 4-positions of aryl hydrazines with halogens, initially chloride, as previously known examples exhibited promising antitubercular activity (Figure 61). ${ }^{1}$


67Ag
95\%
$106.75 \mu \mathrm{M}$


67Am
24\%
4:1
$106.75 \mu \mathrm{M}$

Figure 61: Results of synthesis of substituted amino acid hydrazides 67 Ag and 67 Am and their MIC values where (\%) a percent of actual yield and (-) lack of rotamer ratio. ${ }^{1}$

Therefore, the ortho chlorinated Boc-hydrazides 67Ag-Ak and para substituted analogues 67x and 67Am-Ar were obtained in good to moderate yields (Figure 62).


Figure 62: The ortho chlorinated Boc- $N$-hydrazides $67 \mathrm{Ag}-\mathrm{Al}$ and the para chlorinated $67 \mathrm{Am}-\mathrm{Ar}$ where (\%) a percent of actual yield and (-) lack of rotamer ratio.

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In addition to mono-substitution, di-substitution of the aromatic rings was something we wished to explore. Therefore, the synthesis of di-ortho chlorinated hydrazides was achieved with a variety of amino acids (Figure 63).


52\%


67At
95\%


67Ax
34\%


67Aw
65\%


67Au
75\%


67Av
37\%

Figure 63: The di-ortho chlorinated Boc-protected hydrazides 67As-Ax where (\%) a percent of actual yield and (-) lack of rotamer ratio.
Other halogens, such as fluorine are contained within drugs such as fluoroquinolones (cf. section 1.3.5) that are of particular interest as the strong electron-withdrawing effect or isosteric similarity to a proton contribute to enhanced antibacterial activity. ${ }^{116}$ In previous work from within the group, the use of trifluoromethyl moieties demonstrated good antitubercular activity (Figure 64). ${ }^{1}$


67Bk
47\%
$12.01 \mu \mathrm{M}$


67 Br
95\%
$117.24 \mu \mathrm{M}$


67BI
92\%
$12.01 \mu \mathrm{M}$


67Bq 85\%
$214.26 \mu \mathrm{M}$

$151.14 \mu \mathrm{M}$

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Figure 64: Results of synthesis of substituted amino acid hydrazides $67 \mathrm{Bk}, 67 \mathrm{BI}, 67 \mathrm{~Bq}, 67 \mathrm{Br}$ and 67Bp and their MIC values where (\%) a percent of actual yield and (-) lack of rotamer ratio. ${ }^{1}$

Consequently, hydrazides of the remaining amino acids containing the trifluoromethyl group in the para position were synthesised for further investigation (Figure 65).


67Ay
22\%
3:1


67Bc
50\%

$67 B g$
$42 \%$
42\%


67Bm 62\%


67Az
71\%


67Bd
66\%

56\%


65\%
4:1




67Bo
44\%

Figure 65: The corresponding boc-hydrazides $67 \mathrm{Ay}-\mathrm{Bp}$ containing trifluoromethyl group on aryl hydrazine where (\%) a percent of actual yield and (-) lack of rotamer ratio.

Furthermore, like with the halogens, synthesis of disubstituted trifluoromethylated hydrazides $67 \mathrm{Bs}-\mathrm{Bx}$ at meta positions was undertaken to understand the relationship between mono and di-substitution of the aromatic ring (Figure 66).

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Figure 66: The corresponding boc-hydrazides 67Bs-Bx containing two trifluoromethyl groups on aryl hydrazine.

As previously stated, introduction of fluorine into bioactive molecules has the potential to improve their potency. ${ }^{115}$ In order to further investigate this, it was decided to synthesise the fluorinated Boc-hydrazide 67By-Cd possessing fluorine in the 2, 4- positions. Initial attempts to synthesise this in previous work utilising DMTMM as a coupling reagent yielded the product in poor yield. However, the revised synthetic route utilising HBTU was able to improve the yield significantly and produce a small library of hydrazides (Figure 67).


67By
12\%


67Bz
17\%


67Ca
66\%


67Cb
25\%


67Cc
95\%


67Cd
45\%

Figure 67: The corresponding boc-hydrazides 67By-Cd containing two fluorine atoms on aryl hydrazine.

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The success of bedaquiline 22 (cf. section 1.3.7) has generated significant interest in the use of bromine in drug molecules, therefore, it was decided to synthesise the brominated hydrazides at all three aromatic positions, mirroring the chlorine analogues. However, the presence of heavy halogen like bromine is not considered a privileged substituent in medicinal chemistry due to safety issues. ${ }^{117}$ Bromine-containing drugs, which are targeted at the lungs, have been associated with pulmonary toxicity because they have an extremely long half-life and low water solubility. Nevertheless, the brominated hydrazides $67 \mathrm{Ce}-\mathrm{Cv}$, were obtained to investigate their potency against Mtb because the potency of antitubercular drugs is often positively associated with their lipophilicity (Figure 68).


67Ce
50\%
3:1


67 Ci 53\%


67 Cm
45\%


67Cq
62\%
3:1


67Cf
84\%


67Cj
72\%


53\%


67 Cr
51\%
 67 Cg 51\%


49\%
4:1


45\%


67Cs
58\%



70\%



41\%


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67 Cu
51\%


67Cv
56\%

Figure 68: The corresponding boc-hydrazides 67Ce-Cv containing bromine at different positions.

Having sufficient coverage of the halogens, it was decided to incorporate a nitrile functionality onto the $\mathrm{C}-4$ position of the aryl hydrazine 69 . The nitrile moiety is considered a substituent of choice in SAR studies because of its polarity, directionality, and low molecular weight. As a result of these properties, introduction of a nitrile group is an important chemical motif found in drug candidates (Figure 69). ${ }^{118}$


79


80


81

Figure 69: The nitrile-containing drugs: bosutinib 79, anti-cancer; topiroxostat 80, anti-gout; MIV150 81, anti-HIV.

Moreover, as exemplified by Tong et al., the high lipophilicity of bedaquiline (BDQ) 22 may contribute to its induction of phospholipidosis, potent inhibition of hERG channel, resulting in QT prolongation and tissue overproportioned accumulation which leads to a long terminal elimination half-life. These findings suggest that less lipophilic bedaquiline analogues would be of potential interest, to improve its pharmacokinetic properties. Consequently, 6-cyano analogues 83 demonstrated a reduction in lipophilicity and potentially safer diarylquinolines, and had only modest effects on MIC values (Figure 70). ${ }^{119}$

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82

|  | $C=3-F$ | clogP | MIC |  | clogP |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{B}=\mathrm{Ph}$ | C $=3-\mathrm{F}$ | 6.22 | 0.23 |  | 4.86 | 0.69 |
| $\mathrm{B}=2-\mathrm{F}, 3-\mathrm{OMe}$ | $\mathrm{C}=3-\mathrm{OMe}$ | 6.22 | 0.10 | P lowered | 4.87 | 0.18 |
| $\mathrm{B}=2-\mathrm{F}, 3-\mathrm{OMe}$ | C $=3$-OMe | 6.00 | 0.10 |  | 4.64 | 0.09 |
| $\mathrm{B}=\mathrm{C}=2,3-\mathrm{diOMe}$ |  | 4.99 | 0.20 | d | 3.64 | 0.34 |

Figure 70: The bedaquiline analogues 82 and 6 -cyano analogues 83 with their clogP and $\mathrm{MIC}_{90}$ values. ${ }^{119}$

Encouraged by this, the para substituted nitrile hydrazides 67Cw-Db were afforded in good yields following the standard procedure to evaluate their antitubercular activity in comparison to the other electron withdrawing hydrazides (e.g. 67Bp) (Figure 71).


82 \%
5:1
67Cw


78 \%
67Cx


35 \%
67Da

52. \%

67Cy


68_\%
67Db

Figure 72: The corresponding Boc-hydrazides 67Cw-Db containing nitrile on aryl hydrazine.

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Whilst the nitrile moiety provides a polar, directional substituent with electron withdrawing properties, literature searches have shown that there is a great deal of interest in nitro compounds that have emerged as potent molecules against replicating and non-replicating Mtb. ${ }^{67,120,121}$ One example that is commonly reported in the literature is the nitroimidazoles (cf. section 1.3.7) such as delamanid 25 and pretomanid 26 that have shown potent intracellular activity in the dormancy stage due to the reductive activation of the nitro group which results in the production of nitric oxide. ${ }^{67,120,121}$ From a medicinal chemistry perspective, the structural modifications may improve activity, therefore, it was decided to synthesise hydrazides 67Dc-Dh containing a nitro group at meta position for further investigation (Figure 72).







70\%

Figure 72: The corresponding Boc-hydrazides 67Dc-Dh containing nitro on aryl hydrazine.

Along the same lines, incorporating another powerful electron withdrawing group such as a sulfonyl group has the potential to affect the antitubercular activity of these molecules. In 1999, rofecoxib 84, a common member of cyclooxygenase 2 (COX-2) selective nonsteroidal anti-inflammatory drugs (NSAID), was approved to treat chronic or acute pain conditions such as osteoarthritis, rheumatoid arthritis and migraine. Nevertheless, it was withdrawn from the market as the long term use of rofecoxib 84 increased the risk of cardiovascular thrombotic events. According to the SAR analysis for rofecoxib 84, a methylsulfonyl moiety confers the inhibitory potency and selectivity against the COX-2 enzyme (Figure 73).

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84
Figure 73: The chemical structure of rofecoxib 84 which is one of methylsulfonyl drug containing

The methylsulfonyl group had not been investigated further in previous work. Hence, the desired hydrazides 67Di-Dn were obtained in good yields utilising HBTU to couple the components to assess the effect of the presence of the methylsulfonyl moiety on MIC (Figure 74).


67Di
84\%


67Dj
99 \%
5:1


67Dm
61. \%


67Dk
59 \%


67Dn
36\%

Figure 74: The corresponding Boc-hydrazides 67Di-Dn containing methylsulfonyl on aryl hydrazine.

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## Electron donating groups

To this point, electron withdrawing components were the focus, however, we also wanted to explore the effect of electron donating groups. Our initial starting point was to consider the incorporation of a methyl group. The importance of a methylation strategy has been documented in drug design, and shown to enhance the pharmacological activity, selectivity, solubility, metabolism and pharmacokinetic/pharmacodynamic properties of drug candidates. Indeed, substitution with a single non-bioisosteric group led to a sharp increase in the potency, for example, methylated safinamide 85a, which is a monoamine oxidase type B inhibitor, resulting in a >20-fold raise of the potency and nearly 1000-fold enhancement in selectivity (Figure 75). ${ }^{122}$


Figure 75: The methylated safinamide 85a.

To study this effect, substituted hydrazides with a methyl group at all three aromatic positions were produced following the standard procedure in good yield following the standard procedure with HBTU (Figure 76).

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67Do
84\%
3:1


67Ds
68\%


67Dp
61\%


67Dt
78\%


37\%

67Dq
80\%

67Du
62\%
3:1



50\%

67Ea
72\%

67Eb
80\%
7:1


67Ec
88\%
3:1



91\%


64\%


67Dr
81\%




67Ed
69\%


67Ee
54\%


67Ef
61\%

Figure 76: The substituted boc-hydrazides 67Dn-Ef with a methyl group on aryl hydrazine.

To explore the electron donating character of other alkyl groups, attention turned to the incorporation of larger alkyl groups, like isopropyl. Incorporation of bulky groups is considered one way to enhance the selectivity of a drug to a specific target. For example, the isopropyl group of atenolol $86, \beta$-blocker medication for hypertension, allows it to be more selective toward $\beta_{1}$-receptors while the t-butyl group of salbutamol 87, bronchodilator agent, is very important for selective interaction with the $\beta_{2}$-receptor (Figure 77). ${ }^{123}$

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Figure 77: The chemical structure of atenolol 86 and salbutamol 87.
Therefore, the introduction of an isopropyl group at the para position on the aryl hydrazine 69 would be of interest. The desired hydrazides 67 Eg-El were obtained again following the standard procedure in an acceptable yield (Figure 78).


Figure 78: The corresponding boc-hydrazides 67Eg-El containing isopropyl on aryl hydrazine.
The present work in the group is focused on the determination of the fundamental structure-activity relationship (SAR) of benzo-[2,1,3]-diazoles amino acid hydrazides with the incorporation of aromatic hydrazines. ${ }^{1}$

Nevertheless, to explore the requirement for an aromatic hydrazine, the use of alkyl hydrazines to form hydrazides with amino acids was of interest. Previously the group had failed to successfully synthesise the hydrazides 89 (cf. Scheme 6) as during synthesis with DCC and HOBt it was believed that the secondary nitrogen reacted with the activated carboxylic acid due to the lack of hinderance afforded without the aromatic ring and increase nucleophilicity provided from the electron donating alkyl group.

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Nevertheless, synthesis of an alkyl hydrazide was attempted using DMTMM coupling conditions, in the belief that the fast and less sterically hindered process would allow for the hydrazide to be formed. Pleasingly, with the use of tertbutylhydrazine 88a and Boc-L-alanine 70b and DMTMM, we were able to successfully synthesise the desired hydrazide 89 in acceptable yield (Scheme 6).


Scheme 6: Reagents and conditions: i.) DMTMM, DIPEA, THF, 5-6 h, (40\%), r.t.

In a parallel thought process, we chose to use benzyl hydrazine 90 for another hydrazide to explore if the aromatic ring needs to be present close to the amine or whether it can be moved further away. For this synthesis, benzyl hydrazine 90 also coupled with Boc-L-alanine 70b and DMTMM to afford the desired hydrazide 91, except this time in a poor yield (Scheme 7).


Scheme 7: Reagents and conditions: i.) DMTMM, DIPEA, THF, 5-6 h, (27\%), r.t.

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### 2.1.2 The conversion of intermediates into the final benzoxa-[2,1,3]diazole peptidomimetic

Research thus far has focused on the synthesis of the amino acid hydrazides and now attention must turn to the synthesis of benzo-[2,1,3]-diazoles 65. To undertake this investigation, synthesis of novel benzoxa-[2,1,3]-diazoles was successfully achieved in reasonable yields via the formation of sulfonamide bonds between the synthesized $N$-protected hydrazides 67, 68, 89 or 91 and sulfonyl chloride 66. Initially, deprotection of the hydrazides $67,68,89$ or 91 allowed reaction with the sulfonyl chloride 66 to deliver the final benzoxa-[2,1,3]diazole products 65 (Scheme 8).


Scheme 8: Synthesis of benzoxa-[2,1,3]-diazole peptidomimetics 65 from either Boc/Fmoc protected amino acid hydrazides; Reagents and conditions: i) 4 M HCl in dioxane, 90 min , r.t.; (ii) DBU, octanethiol in THF, 2 h, r.t.; (iii) 66, Et3N, THF, 5-6 h, (64-85\%), r.t.; (iv) 66, DIPEA, THF, 4 h, (34-55\%), r.t.; (v) 66, pyridine, MeCN, 4 h, (39-72\%), r.t.

In this way a library of final compounds 65 could be synthesised in a straightforward manner by exploiting a key set of protection/deprotection and coupling strategies.

### 2.1.2.1 The Boc deprotection

As has been shown in previous work, 4 M HCl in dioxane has been used to deprotect the Boc amino acid hydrazides but, in previous syntheses the deprotected hydrazide had been treated as an intermediate and not isolated or purified before continuing. This made for difficult purification and separation of the impurities at the final stage and so to make a comprehensive library of compounds we needed to address this. ${ }^{1}$

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It was realized that the intermediate hydrochloride salt of the amino acid hydrazide could be isolated following dissolution in methanol and precipitation with diethyl ether as an antisolvent. ${ }^{124}$ The isolation step removed any impurities following the deprotection enabling the $N$-terminal hydrazide 92 to be utilised in the final sulfonamide coupling. An example of this is the Boc-hydrazide 67Em that was reacted with 4 M HCl in dioxane. The resulting product 92a was precipitated by using methanol and diethyl ether to deliver a pure quaternary ammonium salt (Figure 79).


92a



Figure 79: ${ }^{1} \mathrm{H}$ NMR spectra of N -hydrazide 67Em (top) showed the absence of a Boc group which had successfully been removed by using 4 M HCl in dioxane whilst, ${ }^{1} \mathrm{H}$ NMR spectra of liquid fraction (bottom) showed the impurities which had successfully been removed by using the precipitation process.

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### 2.1.2 2 The Fmoc deprotection

Alongside the Boc protected hydrazides, we also made Fmoc (9-fluorenylmethoxycarbony-) protected hydrazides. Fmoc groups are base-labile and acid-stable; thus it is frequently used in an orthogonal protection strategy for amino acids that have reactive side chains requiring protection through subsequent synthetic steps.

Consequently, Fmoc-protected hydrazides 68 were used in a trial deprotection reaction to generate free $N$-terminal hydrazide 93 by stirring with $20 \%$ piperidine 94 in THF overnight. Unfortunately, analysis of the crude reaction mixture revealed the presence of an undesired product 96 that arose through reaction of the desired product 93 with the benzofulvene by-product 95 (Figure 80).


Figure 80: Fmoc deprotection mechanism using piperidine 94.
To explore this further and attempt to isolate the free amine, other bases where investigated. Firstly, 4-methylpiperidine was utilised in THF to afford the free amine of hydrazide 93. ${ }^{125}$ Although this basic deprotection condition was successful (loss of Fmoc observed), the benzofulvene by-product 95 was the primary product.

Consequently, it was clear that the standard deprotection method was not going to enable the production of the desired product and so further investigation was required.

Upon reviewing the literature, a paper by Sheppeck et al. demonstrated the effective use of the base 1,8-Diazabicyclo[5.4.0]undec-7-ene (DBU) in a catalytic

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quantity with a high equivalence of octanethiol as a Michael donor to outcompete the reaction of the amine nucleophile for the benzofulvene by-product 95 would produce the by-product 97 and the desired free amine in a solution phase reaction (Scheme 9). ${ }^{126}$


Scheme 9: (i.) DBU, octanthiol in THF, 2 h, r.t.
Consequently, we set about repeating this study and to our delight, we were able to isolate the free amine of the Fmoc protected amino acid hydrazides. One example of this is the Fmoc-protected hydrazide 68Ab that was treated with a catalytic amount of DBU and octanethiol in THF for 2 hours under $\mathrm{N}_{2}$ and then precipitated with methanol and diethyl ether to afford the free $N$-terminal hydrazide 93a. The analysis of the ${ }^{1} \mathrm{H}$ NMR spectra of the precipitate showed a detectable amount of the octanethiol alongside the desired free $N$-hydrazide 93a (Figure 81).


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Figure 81: ${ }^{1} \mathrm{H}$ NMR spectra of $N$-protected hydrazide 68Ab showed the presence of Fmoc group which had successfully been removed by using DBU/octanethiol in THF.

Unfortunately, we were not able to purify away the octanethiol completely from the reaction product and so in this instance we opted to carry on with the synthesis of benzoxa-[2,1,3]-diazole peptidomimetics carrying this impurity through.

### 2.1.2.3 The sulfonamide bond formation

Sulfonamide bond formation is one of the most used transformations in the synthesis of pharmaceutical compounds displaying a wide range of biological activities (Figure 82). For example, glibenclamide 98 has found use as a hypoglycaemic agent, E7070 99 as an anticancer agent, amprenavir 100 is used in HIV therapy and furosemide 101 as a diuretic. ${ }^{127,128}$


98


99


101

Figure 82: The chemical structure of therapeutics that display a wide range of biological activities.

The sulfonamide bond greater resistance to metabolic catabolism due to its structural similarity to the peptide bond in enzymatic hydrolysis, and its more acidic hydrogen giving rise to the possibility of a stronger H -bonding are some of the reasons why these functional groups appear in drug molecules. ${ }^{129}$ For these reasons, there are a plethora of methods for sulfonamide bond formation. ${ }^{130}$ The

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 most frequently used procedures involve the reaction of sulfonyl chlorides with primary or secondary amines.
### 2.1.2.4 Generation of a Library of Benzoxa-[2,1,3]-diazole

## peptidomimetics

To develop of our novel benzoxa-[2,1,3]-diazoles 65, we were able to utilise coupling conditions described in previous work. We reasoned the use of tertiary amines such as triethylamine or diisopropylethylamine would be required to neutralise the HCl salt of the hydrazide hydrochloride and the hydrochloride formed during the reaction. ${ }^{1}$ Therefore, substituted $N$-hydrazides 92 or 93 and sulfonyl chlorides 66 where mixed in the presence of 3 equivalents of triethylamine in THF to produce the desired products 65 (Scheme 10). ${ }^{1}$


Scheme 10: Reagents and conditions: i.) Et3N, THF, 4-5 h, (26-97\%),

With the synthesis outlined, we began by deprotecting the protected N hydrazides produced previously and then reacted these with the corresponding benzoxadiazole sulphonylchloride to afford the desired compound $65 \mathrm{~h}-\mathrm{Al}$ in fair to very good yields (Figure 83).

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65h
48\%
4:1


65k
26\%

$65 n$
$54 \%$

$65 q$
$67 \%$



67\%


65i
68\%
5:1


651
48\%


650
97\%


65r
86\%


5r


65j
76\%
4:1


65m
86\%


65p
79\%

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65t
62\%


65w
72\%

$65 z$
56\%


65Ac
72\%


65Ad
66\%


65v
53\%


65y
69\%


65Ab
61\%


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65Af
68\%


65Ai 59\%


65Ag
64\%



65Aj
68\%


65Ah
45\%


65Ak
54\%

65AI
61\%

Figure 83: The resulting sulfonamidopeptides $\mathbf{6 5 h}$-Al which had been successfully synthesized in presence of triethylamine with very good yields.

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In attempts to improve the yield, other tertiary amines such as diisopropylethylamine (DIPEA) were examined. Unfortunately, using DIPEA in place of triethylamine saw no improvement on the efficiency of the system, and in fact the yield of the desired compounds 65Am-Ao was somewhat disappointing (Figure 84). As compared to (cf. Figure 83), low yields reported for these compounds 65Am-Ao, necessitating switch to alternative bases such as pyridine.


65Am 14\% 3:1


65An
26\%


65Ao
50\%

Figure 84: The resulting sulfonamidopeptides 65Am-Ao which had successfully been synthesized in presence of DIPEA 75 with low yields.

Less frequently seen is the replacement of the tertiary amine with pyridine 102 (cf. section 2.1.2.3). In the optimized reaction condition utilising HBTU, the reactions were attempted in the presence of a slight excess of pyridine 102. ${ }^{128,132}$ Pleasingly, this work demonstrated that pyridine was successful in producing benzoxa-[2,1,3]-diazoles 65Ap-Ar in acceptable yields, however, not better than with triethylamine (Figure 85).

65Ap
6:1

65Aq
54\%

65Ar
72\%

Figure 85: The resulting sulfonamidopeptides 65Ap-Ar that were successfully synthesized in presence of pyridine $\mathbf{1 0 2}$ with acceptable yields.

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In summary, this work has identified a convenient procedure for the synthesis of benzoxa-[2,1,3]-diazoles 65 from substituted $N$-hydrazides 92 or 93 . Our studies show that the conversion to benzoxa-[2,1,3]-diazoles 65 was accomplished in a lower yield with DIPEA 75 and pyridine 102, than with triethylamine. So, optimisation of the reaction base or solvent failed to improve the overall yield.

### 2.1.2.5 Enantiomeric purity determination

As this synthesis utilises a base, racemisation of the chiral centre at the amino acid is a possibility and needed to be checked. Therefore, to determine if this was the case, we utilised the same resolution techniques (described previously) to determine the enantiomeric purity of our novel benzo-[2,1,3]-diazoles 65. Chiral HPLC (Chiralcel-OD, MeCN: $0.1 \%$ formic acid/MeOH: $0.1 \%$ formic acid) produced the electronic circular dichroism (ECD) spectra of the corresponding enantiomers. In the case of synthesized $L$-benzo-[2,1,3]-diazoles 65 and $D$ -benzo-[2,1,3]-diazoles 65 m we confirmed that the ECD spectra of each contained only a single enantiomer (Figure 86 and 87).


Figure 86: chiral HPLC of $L$-benzo-[2,1,3]-diazole 65 indicating the presence of a single peak, Daicel Chiralpak® IB column, $4.6 \times 250 \mathrm{~mm}, 5 \mu \mathrm{~m}$ particle size, acetonitrile: methanol: $0.1 \%$ formic acid (94:5:1), $1 \mathrm{~mL} / \mathrm{min}$, UV at 282 nm .

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Figure 87: chiral HPLC of $D$-benzo-[2,1,3]-diazole 65 m indicating the presence of a single peak, Daicel Chiralpak® IB column, $4.6 \times 250 \mathrm{~mm}, 5 \mu \mathrm{~m}$ particle size, acetonitrile: methanol: $0.1 \%$ formic acid (94:5:1), $1 \mathrm{~mL} / \mathrm{min}$, UV at 282 nm .

### 2.1.2.6 Determination of the rotameric mixture of benzoxa-[2,1,3]diazole peptidomimetics

As alluded to earlier, in some cases we observed the presence of another compound in the ${ }^{1} \mathrm{H}$ NMR which we surmised was a rotamer. To determine if this was the case, we undertook a variable temperature ${ }^{1} \mathrm{H}$ NMR study of the benzoxa-[2,1,3]-diazole 65g to determine this. Dissolving 65g in DMSO-d6 and beginning with a room temperature experiment, we then raised the temperature of the probe in $10^{\circ} \mathrm{C}$ increments and once stable, an ${ }^{1} \mathrm{H}$ NMR was recorded. Measurements were taken up to $90^{\circ} \mathrm{C}$ and then the sample returned to room temperature where a final ${ }^{1} \mathrm{H}$ NMR spectrum was taken for analysis (Figure 88).

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Figure 88: Variable temperature ${ }^{1} \mathrm{H}$ NMR of benzoxa-[2,1,3]-diazole $\mathbf{6 5 g}$
The result of the experiment determined that the second signals in the ${ }^{1} \mathrm{H}$ NMR are from a rotamer of the compound. The ratio of the two compounds at the beginning of the experiment is $4: 1$ and then as the temperature was raised through $50{ }^{\circ} \mathrm{C}$ the signals begin to coalesce (nothing changed before this temperatre) until we get to $90^{\circ} \mathrm{C}$ where this is complete. Pleasingly when the sample was returned to room temperature the ${ }^{1} \mathrm{H}$ NMR produced was in complete corroboration with the NMR before the experiments were undertaken and the rotamer ratio returns to 4:1.

In order to determine the conformation of the major rotamer of our benzoxa-[2,1,3]-diazoles we initiated a study utilising unassigned nuclear Overhauser effect spectroscopy (NOESY) experiment. This experiment records the throughspace dipolar interactions between protons nearby in three-dimensional (3D) space. To determine the configuration, we utilised the benzoxa-[2,1,3]-diazole peptidomimetic 65 to determine which hydrazine NH proton coupled to the $\mathrm{CH}_{2}$ protons and ultimately determine the cis/trans conformation of the major rotamer (Scheme 11).

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Scheme 11: The possible orientations of $\mathbf{6 5 f}$ around the amide bond to arise trans and cis rotamers.

The NOESY spectra showed the $\mathrm{CH}_{2}$ peak at 3.95 ppm (highlighted with blue) coupled to the amide proton (highlighted with purple) NH at 9.79 ppm (Figure 89).


Figure 89: 2D NOESY of $\mathbf{6 5 f}$ displaying the through space coupling of hydrogens to verify trans is present as the major rotamer.

Therefore, ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ analysis confirmed the trans orientation of 65 f as the major rotamer as would be expected from the steric repulsion from the aryl hydrazine and the benzoxa-[2,1,3]-diazole. Consequently, knowing we have a mixture of compounds will have an impact on activity as one of the rotamers is likely to have the correct orientation for stronger interactions with the target.

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## 3 Structure-activity-relationship studies of benzo-[2,1,3]-diazoles peptidomimetics

Benzoxa-[2,1,3]-diazole peptidomimetics 65 were designed as promising antitubercular agents. Previously, a limited number of simple substituted benzo-[2,1,3]-diazole peptidomimetics 65a-g have been shown to exhibit promising antitubercular activity and selectivity against mycolata over Gram-positive and Gram-negative bacteria (cf. section 1.7). ${ }^{1}$ Based on this preliminary study, further investigation into the antitubercular activity of benzo-[2,1,3]-diazole peptidomimetics 65 is warranted. Therefore, the substituted benzoxa-[2,1,3]diazole peptidomimetics 65 were chosen as the partner to amino acid hydrazides 67, 68, 89 and 91 for further investigation, via an SAR study to understand the importance of amino acids 70 and the aryl hydrazine 69 on antitubercular activity. To undertake this, the methods of synthesis outlined (cf. section 2.1.1) in this study provided the capability to produce a variety of hydrazides $67,68,89$ and 91 and some of these intermediates were converted to the benzoxa-[2,1,3]diazole peptidomimetics 65h-AI (cf. section 2.1.2). In vitro screening was undertaken to evaluate the antitubercular potency of both intermediates and final compounds against wild-type Mtb, isoniazid-resistant Mtb strains and rifampicin-resistant Mtb via a REMA assay.

### 3.1.1 Antimycobacterial susceptibility testing

The colorimetric redox indicator technique has been developed and endorsed by the WHO to test drug susceptibility in non-reference laboratories. ${ }^{131}$ The Resazurin microtiter assay (REMA) is a colorimetric drug-susceptibility method that uses a redox indicator, Resazurin 105. Palomino et al. (2002) demonstrated that REMA is a simple, rapid and inexpensive technique to determine minimal inhibitory concentrations (MIC) of bioactive molecules. ${ }^{132}$ This colorimetric assay works through irreversible reduction of resazurin (purple) 105 to resofurin (pink) indicating mycobacterial growth (Figure 90).

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Figure 90: Irreversible reduction of resazurin 105 to resorufin 106 due to mycobacterial growth. Further reduction may be observed which shows the formation of dihydroresorufin 107.

The redox potential of resorufin can be observed visually for colour changes and read by a spectrophotometer at an excitation of 530 nm with an emission of 590 nm . In brief, the Mtb strain is diluted with Middlebrook 7H9GC medium and added to 96 well plates along with compounds that have been serially diluted across the plate. The plates are incubated at $37^{\circ} \mathrm{C}$ for seven days. Then, the resazurin 105 is added into the plates and tests are read $\sim 48$ hours later by observing the color change (Figure 91).


Figure 91: Example of REMA plate set up.
The minimum inhibitory concentration (MIC) is defined as the lowest concentration of a drug which has no colour change for the resazurin (purple) while the maintenance of a dark blue colour is indicative of the ability of compounds to provide 95-100\% inhibition in mycobacterial growth. Therefore, we can see that compound 2 has a minimum inhibitory concentration of $1.95 \mu \mathrm{~g} / \mathrm{mL}$ (Figure 91). ${ }^{131}$

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### 3.1.2 The SAR exploration of amino acid hydrazide

The first part of this project was successful, to generate a library of substituted and unsubstituted hydrazides with good yields. Modification of the amino acids or aryl hydrazines within the hydrazides 67, 68, 89 and 91 would provide the opportunity to explore their structure activity relationships against Mtb strains. Consequenly, to assess their anti-tubercular activity we chose to use wild-type Mtb, isoniazid-resistant Mtb and rifampicin-resistant Mtb in a REMA assay to allow for a comparative analysis of their minimum inhibitory concentrations.

To begin it was envisaged that utilising the unsubstituted aryl hydrazides coupled to the amino acids would allow for the comparison of amino acid side chains and their antitubercular activity. It was demonstrated that increasing the side chain of the amino acid would increase the $\log P$ and enhance membrane permeability but also provide some insight into the optimum structure required for antitubercular activity. Therefore, the antimycobacterial activity of unsubstituted Boc- $N$-hydrazides 67a-j and Fmoc- $N$-hydrazides 68a-j was assessed in vitro against wild-type Mtb, isoniazid-resistant Mtb strain and rifampicin-resistant Mtb at concentrations of $128-0.125 \mu \mathrm{~g} / \mathrm{mL}$, and converted to $\mu \mathrm{M}$ (Appendix, Table A1).

Gratifyingly, the biological analysis provided evidence for the importance of the size of side chain as the MIC result of the unsubstituted hydrazide 67a with no side chain against the susceptible Mtb was disappointing, with no activity against the more-resistant strains. It was observed that the small side chain of the unsubstituted hydrazide 67b improved the potency two-fold compound to 67a against the wild-type Mtb, but still with no observable activity against the monoresistant strains. The continued increase in the size of side chain of the unsubstituted Boc- $N$-hydrazides 67a-h indicated that an improved cell wall penetration can be attributed to the large side chain, owing to the increase in lipophilicity. It is very clear that the linear and larger hydrophobic side chains were important for the antitubercular activity and in line with what we postulated. As a result, the most potent unsubstituted hydrazide was found to be 67 h which contained an L-phenylalanine (Figure 92).

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67a
$241.22 \mu \mathrm{M}$ wt. Mtb
$-\mu \mathrm{M} \mathrm{INH}^{\mathrm{R}} \mathrm{Mtb}$


67d
$12.44 \mu \mathrm{M}$ wt. Mtb
$199.12 \mu \mathrm{M} \mathrm{INH}^{\mathrm{R}}$ Mtb


67b
$114.56 \mu \mathrm{M}$ wt. Mtb

- $\mu \mathrm{M} \mathrm{INH}^{\mathrm{R}}$ Mtb


67g
$5.89 \mu \mathrm{M}$ wt. Mtb
$188.54 \mu \mathrm{M} \mathrm{INH}^{\mathrm{R}}$ Mtb


67c
$13.01 \mu \mathrm{M}$ wt. Mtb
$208.20 \mu \mathrm{M} \mathrm{NH}^{\mathrm{R}}$ Mtb


67h
$2.81 \mu \mathrm{M}$ wt. Mtb
$90.03 \mu \mathrm{M} \mathrm{NH}^{\mathrm{R}}$ Mtb

Figure 92: The improvement of MIC results against the wild-type Mtb and INH ${ }^{R}$ Mtb as increase in the side chain of the unsubstituted hydrazides 67a-h.

To gain more insight into the role of the size of side chain in this regard, $L$-histidine or L-tyrosine with protected side chain groups were utilised, a significantly diminished antitubercular activity was seen. We believe these results from the incorporation of a much larger side chain hindering the molecules binding (cf. 67h vs 67i) and (cf. 67h vs 67j) (Appendix, Table A1).

A discouraging observation was that while 67 h showed a high level of potency against susceptible Mtb comparable to the current TB treatment on the market (Ethambutol 3 MIC $4 \mu \mathrm{M}$ and Ethionamide 16 MIC $6 \mu \mathrm{M}$ ), ${ }^{2}$ the unsubstituted Fmoc- $N$-hydrazides 68a-j produced a limited antitubercular activity against the wild-type and resistant Mtb strains. Slightly more promising results were observed that also supported our believe in the critical role of the linear and larger hydrophobic side chain (cf. 68a vs 68b) and (cf. 68c vs 68d) (Appendix, Table A1).

Also, further variations we wanted to investigate were whether replacing the $L$ amino acid with unnatural $L$-amino acids or $D$-amino acids affected antitubercular activity. These variations could elucidate some key features including the modified side chains and configuration that may improve the antimycobacterial performance and tackling of the antimycobacterial resistance (Appendix, Table A2).

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Owing to the investigation of structural modification of the amino acids, the dipeptide of glycine (glycylglycine) 67k was utilised to lengthen the hydrazide in an attempt to explore the proximity requirements of the aryl group. Regrettably, 67k lost all its ability to inhibit the growth of Mtb (cf. 67a vs 67k Table A2). This suggests that the target site requires a single amino acid to be present.

As shown in Figure 92, the promising MIC results of the unsubstituted hydrazides 67a-h increases with the size of side chain. We explored substitution with homophenylalanine ( $\mathbf{6 7 m}$ ). In view of the lengthening of the alkyl chain, further distance between phenyl and chiral carbon doesn't have a positive impact on MIC result (cf. 67h vs 67m) (Appendix, Table A2).

Antimicrobial peptides (cf. section 1.4.1) have been reported in the literature using both $L$ - and $D$ - amino acids that showed a similar antibacterial activity against Gram-positive and Gram-negative bacteria. ${ }^{133}$ Furthermore, there is limited data on their antitubercular activity. For these reasons, strong interest lies in the synthesis and evaluation of hydrazides that contain both $L$ - and $D$ - amino acids against susceptible and resistant Mtb strains. To investigate the ability of configuration to affect activity, the poor MIC results of the hydrazides 67n-p suggest a negative impact of $D$-configuration on the antimycobacterial activity as no observable activity against all Mtb strains as compared to the hydrazides with L- configuration (Figure 93). Thereby it is perhaps unsurprising that optimal activity requires an L-configuration.

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67b
$114.56 \mu \mathrm{M}$ wt. Mtb $-\mu \mathrm{M} \mathrm{INH}^{\mathrm{R}}$ Mtb


67n
$-\mu \mathrm{M}$ wt. Mtb
$-\mu \mathrm{M} \mathrm{INH}^{\mathrm{R}}$ Mtb


67g
$5.89 \mu \mathrm{M}$ wt. Mtb
$188.54 \mu \mathrm{M} \mathrm{NH}^{\mathrm{R}}$ Mtb


670
$-\mu \mathrm{M}$ wt. Mtb
$-\mu \mathrm{M} \mathrm{INH}^{\mathrm{R}}$ Mtb


67h
$2.81 \mu \mathrm{M}$ wt. Mtb
$90.03 \mu \mathrm{M} \mathrm{INH}{ }^{\mathrm{R}}$ Mtb


Figure 93: The negative impact of $D$-configuration on the antimycobacterial activity; where a (-) indicates to a lack of inhibition activity.

To support the investigation discussed above, the substituted Boc- $N$-hydrazides 67q-Af and Fmoc-N-hydrazides 68k-t were obtained from utilizing the same amino acid as seen in (Appendix, Table A1 and A2), but in this case the coupling reaction was carried out with a 3-chlorophenylhydrazine to confirm the effect of size of side chain and configuration of amino acid on MIC results. We also wanted to examine whether the involvement of substituents in aryl hydrazines affected their potency. Therefore, MIC results of the antimycobacterial susceptibility testing of the substituted Boc- $N$-hydrazides 67q-Af and Fmoc- $N$-hydrazides 68ku were obtained (Appendix, Table A3).

The initial investigation confirmed the existence of a direct association between the size of linear hydrophobic side chains of the amino acid and a good MIC result. This being so, the MIC results of the substituted hydrazides against the wild-type Mtb strain were improved as an increase of the length of hydrophobic side chain (Figure 94). Considering above the MIC results of unsubstituted hydrazides 67a, b, g against the wild-type Mtb strain, a clear improvement in potency was observed as compared to meta chlorinated hydrazides $\mathbf{6 7 q}, \mathbf{r}, \mathbf{w}$.

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67q
$13.34 \mu \mathrm{M}$ wt. Mtb


67r
$3.19 \mu \mathrm{M}$ wt. Mtb


67w
$2.67 \mu \mathrm{M}$ wt. Mtb

Figure 94: The improvement of MIC results against the wild-type Mtb as increase in the side chain of the substituted hydrazides $\mathbf{6 7 q} \mathbf{q}, \mathbf{r}, \mathbf{w}$.

By assessing the impact of the configuration, a specific replacement of $L$-amino acid with $D$-amino acid component also impaired antimycobacterial activity of the substituted hydrazides 67 Ad-Af (cf. 67Ad vs 67r), (cf. 67Ae vs 67w) and (cf. 67Af vs 67x) (Appendix, Table A3). Unlike the unsubstituted hydrazides 67n-p previously discussed (cf. Figure 93), the substituted hydrazides 67Ad-Af showed better activity than unsubstituted hydrazides 67n-p. It is believed that the unwanted effect of $D$-configuration on MIC could be improved by incorporation a substituent on aryl hydrazine 69.
in order to expand the scope of the positive effect of a substituent on the potency was demonstrated by replacement of phenylhydrazine (Appendix, Table A1 and A2). with 3-chlorophenylhydrazine in the structures (Appendix, Table A3) which resulted in an improved antitubercular activity. The substituted hydrazides showed a potent activity as the amino acid fragment remained constant with introducing the chlorine group; this substitution led to a vast improvement in MIC results when compared to unsubstituted hydrazides (Figure 95).

67e
$-\mu \mathrm{M}$ wt. Mtb

68g
$-\mu \mathrm{M}$ wt. Mtb

67m
$86.61 \mu \mathrm{M}$ wt. Mtb

67u
$44.96 \mu \mathrm{M}$ wt. Mtb

$10.88{ }_{6}{ }^{1} \mathrm{R}_{\mathrm{w}} \mathrm{w}_{\mathrm{wt}}$ Mtb

67Ac
$23.40 \mu \mathrm{M}$ wt. Mtb

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Figure 95: The positive impact of substituent on the antimycobacterial activity.
As observed in (Appendix, Table A3), the chlorinated hydrazide 67 w showed higher activity against the wild-type Mtb. We explored the impact of the size of the carbamate group on the antimycobacterial activity; an introduction of the Fmoc group instead of the Boc group was utilized. This alteration resulted in reduced potency (cf. 67w vs 68u).

Consequently, this study employed a similar feature comparison to attempt to understand the key structural features including, the size of side chains, configuration, the incorporation of substituents and the size of carbamate group that has been shown to have a significant effect on the antitubercular activity of these intermediates. More specifically, the findings from the initial screening revealed that the incorporation of the synthetic amino acids with side chains or with D-configuration, the unsubstituted aryl hydrazine and the bulky carbamate group were explored to improve the potency, but they showed an undesired influence on the MIC results. This being so, it was decided to focus on the six $L$ amino acids that showed promising MIC values and coupled with different substituted aryl hydrazines to investigate the most active hydrazide, and to explore the impact of the nature of substituent and its position on the antitubercular activity of intermediates.

Due to the promising MIC results of the chlorinated hydrazides $\mathbf{6 7 q} \mathbf{- z}$, it was decided to conduct a simple comparison of the structures of hydrazides $\mathbf{6 7 \mathrm { Ag } - \mathrm { Ar }}$ which revealed a couple of key differences in activity: i) the increase of the linear lipophilic side chain; ii) the position of the chlorine group. Therefore, the MIC results of in vitro antitubercular screening of the chlorinated Boc- $N$-hydrazides 67Ag-Ar at different positions on aryl hydrazines were obtained (Appendix, Table A4).

In search of more potent antitubercular agents, MIC results (Appendix, Table A4) confirmed the critical role of the size of side chain as the ortho chlorinated hydrazides $67 \mathrm{Ag}-\mathrm{Al}$ and para position $67 \mathrm{Am}-\mathrm{Ar}$ with large side chains providing much improved inhibition capability compared to those with no side chain (cf. 67 Ag vs 67 Ak ) and (cf. 67 Am vs 67 Aq ). The para chlorinated hydrazide 67An, with the smallest side chain, showed the lowest MIC against the wild-type Mtb

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$(1.59 \mu \mathrm{M})$. It is clear from the three examples given in (Figure 96) that the trend is that meta is the preferred position for activity against the wild-type Mtb as compared to other positions.



67Ak
$21.40 \mu \mathrm{M}$ wt. Mtb


67AI
$164.15 \mu \mathrm{M}$ wt. Mtb


67q
$13.34 \mu \mathrm{M}$ wt. Mtb

67w
$2.67 \mu \mathrm{M}$ wt. Mtb


67x
$5.13 \mu \mathrm{M}$ wt. Mtb


67Am
$213.51 \mu \mathrm{M}$ wt. Mtb


67Aq
$21.40 \mu \mathrm{M}$ wt. Mtb


67Ar
$41.04 \mu \mathrm{M}$ wt. Mtb

Figure 96: The chlorinated hydrazides at meta position showed much better antitubercular activity against the wild-type Mtb than other positions.

Although the chlorinated Boc- $N$-hydrazides at the meta position have been highly effective against the wild-type Mtb, they are less effective against the resistant Mtb strains than the para chlorinated Boc- $N$-hydrazides (Figure 97).

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67Ai
$187.22 \mu \mathrm{M} \mathrm{INH}{ }^{\mathrm{R}}$ Mtb
$187.22 \mu$ M RIF $^{\mathrm{R}}$ Mtb


67Ak
$171.17 \mu \mathrm{M} \mathrm{INH}^{\mathrm{R}}$ Mtb
$171.17 \mu \mathrm{M}$ RIF $^{\mathrm{R}}$ Mtb


67q

- $\mu \mathrm{M}$ INH ${ }^{\mathrm{R}}$ Mtb
- $\mu \mathrm{M}$ RIFR Mtb


67s
$93.61 \mu \mathrm{M} \mathrm{INH}{ }^{\mathrm{R}}$ Mtb
$93.61 \mu \mathrm{M}$ RIF $^{\mathrm{R}}$ Mtb


67w
$42.79 \mu \mathrm{M} \mathrm{INH}^{\mathrm{R}}$ Mtb - $\mu \mathrm{M} \mathrm{RIF}^{R}$ Mtb


67Am
$13.34 \mu \mathrm{M} \mathrm{INH}^{\mathrm{R}}$ Mtb
$53.38 \mu \mathrm{M} \mathrm{RIF}^{\mathrm{R}}$ Mtb


67Ao
$46.81 \mu \mathrm{M} \mathrm{INH}^{\mathrm{R}}$ Mtb
$93.61 \mu \mathrm{M} \mathrm{RIF}{ }^{\mathrm{R}}$ Mtb


67Aq
$21.40 \mu \mathrm{M} \mathrm{INH}^{\mathrm{R}}$ Mtb
$85.58 \mu \mathrm{M} \mathrm{RIF}^{\mathrm{R}}$ Mtb

Figure 97: The chlorinated hydrazides at para position showed much better antitubercular activity against the resistant Mtb than other positions.

We decided to investigate the incorporation of two chlorine groups. For this reason, the di-ortho chlorinated Boc- N -hydrazides $67 \mathrm{As}-\mathrm{Ax}$ were obtained and screened to examine their antimycobacterial activity and compared to the mono chlorinated Boc- $N$-hydrazides (Appendix, Table A5).

In comparison, MIC data analysis was carried out on mono or di- chlorinated BocN -hydrazides and it was found that the di-chlorinated Boc- N -hydrazides provided a better potency against the wild-type Mtb than ortho or para chlorinated Boc- $N$ hydrazides but not greater than the meta chlorinated Boc- $N$-hydrazides (Figure 98).

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67Ag
$106.75 \mu \mathrm{M}$ wt. Mtb


67Ak
$21.40 \mu \mathrm{M}$ wt. Mtb



67AI
$164.15 \mu \mathrm{M}$ wt. Mtb


67As
$23.94 \mu \mathrm{M}$ wt. Mtb


67Aw
$19.59 \mu \mathrm{M}$ wt. Mtb


67Ax
$18.85 \mu \mathrm{M}$ wt. Mtb


67q
$13.34 \mu \mathrm{M}$ wt. Mtb


67w
$2.67 \mu \mathrm{M}$ wt. Mtb


67x
$5.13 \mu \mathrm{M}$ wt. Mtb


67Am
$213.51 \mu \mathrm{M}$ wt. $M t b$


67Aq
$21.40 \mu \mathrm{M}$ wt. Mtb


67 Ar
$41.04 \mu \mathrm{M}$ wt. Mtb

Figure 98: The MIC comparison between the di-chlorinated hydrazides and mono-chlorinated hydrazides.

The di-chlorinated Boc- $N$-hydrazides exhibited reduced activity against resistant strains as compared to the mono-chlorinated Boc- $N$-hydrazides. This led us to further investigations with other substituents at different positions.

To achieve this, exchanging the chlorine moiety for a trifluoromethyl group was assessed. The introduction of the trifluoromethyl group into medicinal agents has been increasingly popular and served as a promising strategy in lead optimization to enhance drug potency. It is certainly true that fluorine has desirable properties, such as enhancing binding energy to target proteins, small atomic size, moderate electron withdrawing effect and lipophilicity. ${ }^{134}$ Therefore, the trifluoromethylated Boc- $N$-hydrazides 67Ay-Bp were screened to investigate the impact of the nature of substituent on MIC (Appendix, Table A6).

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No significant improvements were observed in drug potency based on the alteration of -Cl to $-\mathrm{CF}_{3}$ (Figure 99).


67r
$3.19 \mu \mathrm{M}$ wt. Mtb


67w
$2.67 \mu \mathrm{M}$ wt. Mtb


67x
$5.13 \mu \mathrm{M}$ wt. $M t b$


67Az

- $\mu \mathrm{M}$ wt. Mtb


67Bc
$78.54 \mu \mathrm{M}$ wt. Mtb


67Bd
$-\mu \mathrm{M}$ wt. Mtb


67Bf
$92.13 \mu \mathrm{M}$ wt. Mtb


67Bi
$-\mu \mathrm{M}$ wt. Mtb


67Bj
$37.79 \mu \mathrm{M}$ wt. Mtb


67BI
$46.06 \mu \mathrm{M}$ wt. Mtb


67Bo
$39.27 \mu \mathrm{M}$ wt. Mtb


67Bp
$151.14 \mu \mathrm{M}$ wt. Mtb

Figure 99: The negative impact of replacement of -Cl to $-\mathrm{CF}_{3}$ on MIC results against the susceptible Mtb.

Moreover, the position of the trifluoromethyl group on the aryl hydrazine was examined and it did not ensure an improvement of antitubercular activity on average (Appendix, Table A6). Overall, the introduction of trifluoromethyl group has no advantage on potency as compared to the chlorine group.

As it appeared that the MIC data of the trifluoromethylated Boc- $N$-hydrazides 67Ay-Bp was disappointing, therefore, the involvement of two trifluoromethyl groups was also examined. For this reason, the di-meta trifluoromethylated Boc-$N$-hydrazides 67Bs-Bx were obtained and screened to examine their antimycobacterial activity as compared to the mono trifluoromethylated Boc- N hydrazides 67Ay-Bp (Appendix, Table A7).

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Unfortunately, this failed to provide any increase in activity (Appendix, Table A7). We are still inspired by the continuous success of fluorinated drugs as 13 new fluoro-pharmaceuticals were approved by the FDA in 2019. ${ }^{135}$ Owing to this, the incorporation of other fluorinated functional groups, such as a fluorine atom was explored. ${ }^{135}$ therefore, the activity of di-fluorinated hydrazides 67By-Cd were investigated (Appendix, Table A8). Given this precedent, it was demonstrated that the di-fluorinated Boc-N-hydrazide 67By-Cd showed no more activity than the di-trifluoromethylated Boc- $N$-hydrazides $67 \mathrm{Bs}-\mathrm{Bx}$.

Common to all TB treatments, high lipophilicity is positively associated with enhanced potency against Mtb strains (cf. section 2.1.1.2). Because of this, introduction of a bulky halogen substituent such as bromine that could be an invaluable substituent to improve the MIC of these intermediates. For this reason, the brominated hydrazides, $67 \mathrm{Ce}-\mathrm{Cv}$, were obtained and screened to compare their potency against Mtb strains with the previous substituents such as chlorine, trifluoromethyl and fluorine and to investigate the effect of the electronic nature of the substituent on antimycobacterial activity (Appendix, Table A9).

The promising change observed relates to the introduction of a bromine group which has shed some light onto the utility of the bulky substituents. The MIC results showed that the brominated hydrazides $67 \mathrm{Ce}-\mathrm{Cv}$, particularly, at the ortho and para positions containing the higher lipophilic substituent exhibited excellent efficacy against the wild-type Mtb strain (Figure 100). In addition to the positive correlation between potency (MIC) toward Mtb strains and drug lipophilicity, the MIC data demonstrated that changing the substituent on aryl hydrazine as a useful tool in the search for potentially potent hydrazides (Appendix, Table A1 A9).

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67Cf
$44.66 \mu \mathrm{M}$ wt. Mtb


67Ah
$50.99 \mu \mathrm{M}$ wt. Mtb


67Bf
$92.13 \mu \mathrm{M}$ wt. Mtb


67Ai
$187.22 \mu \mathrm{M}$ wt. Mtb


67Cq
$23.24 \mu \mathrm{M}$ wt. Mtb


67 Cv
$36.84 \mu \mathrm{M}$ wt. Mtb


67Am
$213.51 \mu \mathrm{M}$ wt. Mtb


67 Ar
$41.04 \mu \mathrm{M}$ wt. Mtb

Figure 100: The positive impact of incorporation - Br on MIC results against the susceptible Mtb.

Meanwhile, the brominated hydrazides $\mathbf{6 7 C e}-\mathrm{Cv}$ were shown to have greater activity against the isoniazid resistant Mtb strains over other substituted hydrazides. This observed potency was comparable to that of ethambutol 3 and ethionamide 16 (Figure 101). ${ }^{2}$


Figure 101: The MIC results of the brominated hydrazides against the resistant INH Mtb as compared to current TB treatments.

To confirm the correlation between drug potency and lipophilicity, the substitution of a bulky lipophilic group (bromine) with a less lipophilic substituent such as a nitrile group was proposed. For this reason, the inhibiting capability of cyanated hydrazides 67Cw-Db was investigated (Appendix, Table A10).

With the MIC analysis in mind, a brief search in the literature suggests that this substitution reduces lipophilicity as mentioned previously (cf. section 2.1.1.2). ${ }^{119}$

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As shown in (Appendix, Table A10) the cyanation of hydrazides does not seem to be a way of improving drug potency in the TB treatments.

Contrary to this, the antitubercular activity of the hydrazides 67Dc-Dh containing a nitro group, that is also a less lipophilic substituent, was evaluated against replicating Mtb strains. Moreover, the nitro moiety plays a vital role in anaerobic activity of nitroimidazoles including delamanid 25 and pretomanid 26 (cf. section 1.3.7). For this reason, the obtained hydrazides 67Dc-Dh were screened with the MIC results of in vitro antitubercular screening revealing a significant decrease in activity against all replicating Mtb strains (Appendix, Table A11).

Also involvement of the less lipophilic methylsulfonyl group in the hydrazides had not been investigated further in previous work. This group was also incorporated into hydrazides 67Di-Dn at the para position (Appendix, Table A12).

Unfortunately, MIC results were also disappointing raising concerns about retained antitubercular activity of hydrazides with incorporating a less lipophilic substituent such as a nitrile, nitro or methylsulfonyl (Appendix, Table A10-A12). For this reason, interest lies in the MIC investigation of hydrazides with higher lipophilic substituents.

As a consequence of these observations, the methylated hydrazides 67Do-Ef would be of interest to enhance the potency of intermediates (Appendix, Table A13). A small alkyl substituent has the potential to improve potency, when compared to the other less lipophilic groups. During the MIC analysis, it was observed that the para methylated hydrazides 67 Eb and 67 Ee were able to induce a very potent antitubercular activity although the meta methylated hydrazides 67Du-Dz showed, in general, better inhibitory activity than other positions. Nevertheless, it is important to highlight that this substituent does not provide better MIC results than chlorinated hydrazides (Figure 102).

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67r
$3.19 \mu \mathrm{M}$ wt. Mtb


67t
$89.92 \mu \mathrm{M}$ wt. Mtb


67w
$2.67 \mu \mathrm{M}$ wt. Mtb


67x
$5.13 \mu \mathrm{M}$ wt. Mtb


67CI
$44.66 \mu \mathrm{M}$ wt. Mtb


67Cn
$159.87 \mu \mathrm{M}$ wt. Mtb


67Co
$19.98 \mu \mathrm{M}$ wt. Mtb


67Cp
$73.68 \mu \mathrm{M}$ wt. Mtb


67Dv
$109.08 \mu \mathrm{M}$ wt. Mtb


67Dx
$95.39 \mu \mathrm{M}$ wt. Mtb



67Dz
$86.61 \mu \mathrm{M}$ wt. Mtb

Figure 102: No positive impact of incorporating $-\mathrm{CH}_{3}$ instead of -Cl on MIC results against the susceptible Mtb.

It is important to highlight that the antitubercular activity of the methylated hydrazides (Appendix, Table A13) against the resistant strains was not as good as the chlorinated hydrazides (Appendix, Table A3 - A4) and the brominated hydrazides (Appendix, Table A9).

For the purpose of this SAR study, the range of the substituted hydrazides was expanded, across a number of modified aryl hydrazines, seeking more bulky alternatives to the methyl group that would provide analogues with greater potency. We chose the isopropyl group as it features enhanced cell penetration and has the potential to occupy deeper hydrophobic regions of the drug target. Owing to this, the hydrazides 67 Eg -El containing isopropyl groups at the para position were obtained and screened (Appendix, Table A14).

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The MIC results demonstrate that incorporation of a larger alkyl substituent slightly improves inhibitory activity as compared to the methylated Boc- N hydrazides (cf. 67Eg vs 67Ea), (cf. 67Ei vs 67Ec) and (cf. 67El vs 67Ef) (Appendix, Table A14).

Now it is worthy of saying that in this SAR work, the use of aryl hydrazines is associated with higher potency, which can be modulated by exchanging the aryl hydrazine with other units, such as alkyl hydrazine, benzyl hydrazine, benzyl amine or 3-chloro benzylamine. Owing to this, it could provide access to a range of structurally diverse amino acid hydrazides and examine if and how these changes affected the pharmacological and physicochemical properties of the resulting compounds (Figure 103).


89
$-\mu \mathrm{M}$ wt. Mtb
$-\mu \mathrm{M} \mathrm{INHR}^{\text {R }}$ Mtb
$-\mu \mathrm{M}$ RIFR Mtb


91
$-\mu \mathrm{M}$ wt. Mtb
$-\mu \mathrm{M} \operatorname{INH} \mathrm{R}^{\text {Mth }} \mathrm{M}$
$-\mu \mathrm{M} \mathrm{RIFR}^{\text {M }}$ (b


114
$-\mu \mathrm{M}$ wt. Mtb
$-\mu \mathrm{M} \mathrm{INH}^{\mathrm{R}}$ Mtb
$-\mu \mathrm{M}$ RIFR $^{\text {Mtb }}$


114a

- $\mu \mathrm{M}$ wt. Mtb
$-\mu \mathrm{M} \operatorname{INH}^{\mathrm{R}}$ Mtb
$-\mu \mathrm{M}$ RIFR ${ }^{\text {Mtb }}$

Figure 103: The negative impact of exchanging of the aryl hydrazine $\mathbf{6 9}$ with alkyl hydrazine 88, benzyl hydrazine 90 , benzyl amine or 3-chloro benzylamine on antitubercular activity.

Unfortunately, this modulation led to the loss of antimycobaterial activity of the amino acid hydrazides. The involvement of alkyl hydrazine 89 and benzyl hydrazine 91 proved ineffective as no activity was observed. Moreover, the importance of the hydrazide moiety was confirmed as the incorporation of benzyl amine 114 failed to show any activity against all Mtb strains and even with 3chloro benzylamine 114a. For these reasons, the use of alkyl hydrazines, benzyl hydrazines, benzyl amine or 3-chloro benzylamine does not seem to be a way of enhancing drug potency. Therefore, aryl hydrazines would remain constant to preserve the hydrazide pharmacophore and the results have demonstrated that substitution can be achieved at three sites around these molecules (Figure 104).

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Figure 104: The core building block structure of hydrazide.
The data collected so far allowed for an SAR study to investigate the most active components of the protected hydrazides. Depending on their structural composition, these have been shown to possess varying degrees of potency towards susceptible and resistant Mtb strains. The lack of activity related to the alkyl hydrazine and the benzyl hydrazine indicate the aryl hydrazine is essential for activity. Taken together, it is possible that the substantial changes observed in MIC results relate to the amino acid itself, most likely due to the presence of some bioactive functional groups including phenol 70j indole 70s, thioether 70g sulfhydryl 70p, guanidinyl 70t and imidazole 70i. Therefore, the MIC results confirmed that the single protected amino acid did not interfere with mycobacterial growth (Appendix, Table A15).

The screening profile of amino acids revealed that the protected amino acids showed no activity but have the potential to be a novel component when to designing novel TB treatments. Based on these investigations, the promising antitubercular activity came from the hydrazide core, that are designed to act as potential antimycobacterial intermediates.

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### 3.1.3 The SAR Exploration of Benzoxa-[2,1,3]-diazole

Promising antitubercular activity has been achieved screening many hydrazides, and there are very active examples discussed throughout this report. Due to their efficacy, the substituted Boc- $N$-hydrazide, mainly with higher lipophilic substituents, still inspire drug design. The conversion of these intermediates into the final compounds 65 could improve their potency and binding affinity as the benzoxa-[2,1,3]-diazole core appears to be essential for activity and selectivity (cf. section 1.7). ${ }^{1}$ Thus, it would be expected that the designed benzoxa-[2,1,3]diazole peptidomimetics $\mathbf{6 5}$ tend to better occupy the active target site, including a deeper pocket.

Thereby, this conjugation provides a good opportunity for enhancing the potency of potential drug candidates. Therefore, the efficacy of the designed benzoxa-[2,1,3]-diazole peptidomimetics 65 required further assessment.

As a result, the final benzoxa-[2,1,3]-diazoles 65h-Ar were screened in the same way against the drug-susceptible, mono-resistant and double-resistant Mtb strains (Table 7).

Table 7: The MIC results of the antimycobacterial susceptibility testing of the benzoxa-[2,1,3]diazole peptidomimetics 65h-Ar; where a (-) indicates to a lack of inhibition activity.

|  |  | MIC ( $\mu \mathrm{M}$ ) |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| Entry | Structure | WT. Mtb <br> $\mathrm{mc}^{27902}$ | INH. resistant Mtb $\mathrm{mc}^{2} 8245$ | INH./ RIF. resistant $M t b$ $\mathrm{mc}^{2} 8250$ | INH./ RIF. resistant Mtb $\mathrm{mc}^{2} 8258$ |
| 1 |  <br> 65h | 34.73 | 34.73 | 34.73 | 34.73 |
| 2 |  <br> $65 i$ | 34.73 | 34.73 | 34.73 | 34.73 |

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| 3 |  <br> 65j | 69.46 | 69.46 | 69.46 | 69.46 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 4 |  | 161.69 | 161.69 | 161.69 | 161.69 |
| 5 |  <br> 651 | 74.37 | 74.37 | 148.75 | 74.37 |
| 6 |  | - | - | - | - |
| 7 |  | 74.37 | 74.37 | 74.37 | 74.37 |
| 8 |  <br> 650 | - | - | - | - |

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| 9 |  | 67.41 | 67.41 | 67.41 | 67.41 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 10 |  | 67.41 | 67.41 | 67.41 | 67.41 |
| 11 |  | 78.08 | 156.15 | 78.08 | 78.08 |
| 12 |  <br> 65s | 73.08 | 146.15 | 146.15 | 146.15 |
| 13 |  <br> 65t | 76.04 | 76.04 | 76.04 | 76.04 |
| 14 |  <br> 65u | 72.59 | 72.59 | 72.59 | 72.59 |

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| 15 |  | 140.37 | - | - | - |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 16 |  <br> 65w | - | - | - | - |
| 17 |  | 122.15 | 122.15 | 122.15 | 122.15 |
| 18 |  | 65.26 | 65.26 | 65.26 | 65.26 |
| 19 |  <br> $65 z$ | 65.26 | 32.63 | 65.26 | 65.26 |

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| 20 |  | 150.99 | - | - | - |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 21 |  | 130.12 | 130.12 | 130.12 | 130.12 |
| 22 |  <br> 65Ac | 69.82 | 69.82 | 69.82 | 69.82 |
| 23 |  | 69.82 | 69.82 | 69.82 | 69.82 |
| 24 |  | - | - | - | - |

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(s)

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| 30 |  | 135.50 | 67.75 | 135.50 | 135.50 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 31 |  | 123.94 | 123.94 | 123.94 | 123.94 |
| 32 |  | 74.37 | 74.37 | 37.19 | 37.19 |
| 33 |  | 67.41 | 67.41 | 33.70 | 33.70 |
| 34 |  <br> 65Ao | - | - | - | - |

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| 35 |  | - | - |  | - |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 36 |  | - | - | - | - |
| 37 |  | - | - | - | - |

In most cases, introducing the benzoxa-[2,1,3]-diazole moiety demonstrated the expected increment in antitubercular activity against the wild-type and particularly mono-resistant and double resistant Mtb as compared to their intermediates (cf. $65 u$ vs 67 Dd ), (cf. 65 z vs 67 Ae ), (cf. 65Ab vs 67 Bm ), (cf. 65Ac vs 67 Ai ), (cf. 65Ad vs 67s), (cf. 65Ag vs 67Af) and (cf. 65Ai vs 67Bn).

The biological data agreed with the results (discussed in section 3.1.2) as consequence of the promising change in MIC results that were observed following introduction of the substituted aryl hydrazine instead of phenylhydrazine. For example, using different halogen substituents on the aryl hydrazine showed a significant effect on MIC (cf. 65p vs 65k), (cf. 65y vs 65v), (cf. 65Ad vs 65Aa), (cf. 65Ag vs 65Ae) and (cf. 65Ak vs 65Aj). Likewise, changing the position of the substituent, particularly, at the meta position resulted in a noticeable decrease in MIC (cf. 65i vs 65j).

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In light of these results, we examined the activities of designed benzoxa-[2,1,3]diazole peptidomimetics bearing di-halogenated aryl hydrazines. Regrettably, the involvement of di-chlorinated aryl hydrazine 650, 65Ap or di-fluorinated aryl hydrazine 65Aq led to a significant loss of activity to inhibit the growth of Mtb. This finding raises concerns about the feasibility of addition more than a substituent on aryl hydrazine. In particular, we were interested to show how presence of a double substituent on aryl hydrazine may also reduce the antimycobacterial activity of both intermediates (cf. Table A5, Table A7 and Table A8) and final compounds 650, 65Ap and 65Aq. Consequently, this leads to the belief that the target site requires a single substituent to be present.

The effect of the linear and larger hydrophobic side chains on the antitubercular activity is somewhat conflicting. In the results (cf. Figure 92 and 94), the adaptation of the amino acid generally resulted in a decrease in MIC as a consequence of enlarging the side chain. Owing to this, an increase in the length and size of side chain gave more promising MIC results of the chlorinated final compounds (Figure 105). There is also agreement with the earlier findings (cf. Figure 95). It was found that involvement of homophenylalanine 65Ah did not show any improvement in MIC result (cf. 65Af vs 65Ah).


Figure 105: The improvement of MIC results against the wild-type Mtb as increase in the side chain of the chlorinated final compounds.

Despite their structural similarities, the MIC data of the final compounds bearing bromine substituent, have an advantage in potency as compared to the chlorine group (cf. 65An vs 65Am), (cf. 65p vs 65I) and (cf. 65q vs 65n). However, in this case, the fluorinated compounds $65 x$ and 65 Ab provided a poorer activity against all Mtb strains. Taken together, this leads to the conclusion that the involvement

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of high lipophilic substituents could be desirable to increase the potency of benzoxa-[2,1,3]-diazole peptidomimetics.

As indicated in Table 7, the poor MIC results of final compounds 65I, 65w and 65Ae also demonstrated the negative impact of D-configuration on the antimycobacterial activity with no observable activity against all Mtb strains.

The substitutions to the aryl hydrazine 69 were applied using different electron donating groups such as alkyl groups, and electron withdrawing groups such as halogens to investigate the effect these substitutions have on the biological activity of benzoxa-[2,1,3]-diazole peptidomimetics 65 against the wild-type, mono-resistant and double-resistant Mtb. As seen in table 7, compounds 65n and 65q having 4-chlorophenylhydrazine and 4-bromophenylhydrazine in their structures were determined to have significantly higher antitubercular potential against all Mtb strains than compound 65 r and 65 s having 4methylophenylhydrazine and 4-isopropylphenylhydrazine (cf. 65n vs 65r) and (cf. $65 q$ vs 65 s). However, in the compounds $65 t$ and $65 u$ containing strong electron withdrawing groups namely cyano and nitro substituent, no meaningful changes were observed in MIC results most probably because of their reduced lipophilicity.

To investigate the ability of exchanging the aryl hydrazine 69 with alkyl hydrazine 88 to affect the activity, this modulation proved ineffective as no activity was observed (cf. 65k vs 65Ao).

According to the obtained MIC data (Table 7), the final compound 65Af possessed significant antitubercular activity against the susceptible, monoresistant and double-resistant Mtb strains with a MIC value of $31.60 \mu \mathrm{M}$

Finally, the designed benzoxa-[2,1,3]-diazole peptidomimetics 65, having large linear side chain amino acids, L-configuration and halogenated aryl hydrazine at meta position, gave rise to the most potent structures.

## Chapter 4: Conclusion

## 4 Conclusion

The continuing demand for effective TB treatments with novel modes of action is particularly awaited in order to combat the development and spread of the multiand extensively drug-resistant Mtb strains. Owing to this, attention turned to a drug - to - target strategy because it may shorten the time scale for TB drug discovery. The focus of this strategy is to assess new chemical entities to identify the most potent and selective inhibitor to Mtb growth. ${ }^{136}$
Our previous studies demonstrated that benzo-[2,1,3]-diazole peptidomimetics 65 have excellent activity and selectivity against Mtb. ${ }^{1}$ Based on these findings, the present work was mainly focused on the determination of the fundamental structure-activity relationship of this architecture 65 as promising antitubercular agents. Throughout this study, a rational series of derivatives based around the benzo-[2,1,3]-diazole peptidomimetics 65 was synthesised in an attempt to investigate antimycobacterial activity against wild-type, INH-resistant, and INH/RIF-resistant Mtb strains through a Resazurin Microtiter Assay (REMA). A number of conclusions could be derived from this work. Firstly, the section of chemistry explored in this work provides a good framework for the amide and sulfonamide bond formations to generate a new library of intermediates 67, 68, 89 and 91 and final compounds 65. With many problems to contend with, the synthesis of hydrazides highlighted that a change of amide coupling conditions was important. Replacement of DCC /HOBt with alternative coupling partners including DMTMM and HBTU proved successful. The revised synthetic route employing HBTU was quicker, higher yielding and more reliable for the synthesis of hydrazides. Once synthesised, the enantiomeric purity of resulting intermediates 67 was determined through a variety of techniques, including polarimetry, chiral HPLC, and the Flack parameter of a single crystal structure.

The final step involved the Boc or Fmoc deprotection of intermediates to yield the free amine hydrazide 92 and 93 to allow for their reaction with the benzoxa-[2,1,3]-diazole sulfonyl chloride 66 to deliver the final benzoxa-[2,1,3]-diazole peptidomimetic 65. The enantiomeric purity of the final compounds 65 was also confirmed by chiral HPLC and the major rotamer of our benzoxa-[2,1,3]-diazoles (trans) was determined utilising a nuclear Overhauser effect spectroscopy

## Chapter 4 : Conclusion

(NOESY) experiment. During the course of the MIC analysis, it was observed that the conjugation of the hydrazides 67, 68 and 89 with benzoxa-[2,1,3]-diazole moiety 66 provided a compound with good level of activity against susceptible, mono-resistant and double-resistant Mtb strains although it appears to have observable cytotoxicity. The findings from this SAR study indicated that an increased the size of side chain of L-amino acids $\mathbf{7 0}$ showed much better antitubercular activity against Mtb. Alkyl hydrazine 88, benzyl hydrazine 90, larger carbamate group (Fmoc group), less lipophilic substituents on aryl hydrazine 69 (nitro, nitrile and methylsulfonyl), the di-substituted aryl hydrazine 69 and D-configuration did not show any improvement on MIC results. Whilst, the size of side chain (linear unreactive side chains), incorporation of heavy halogens (bromide and chloride), L-configuration, mono-substitution at meta position and the benzoxa-[2,1,3]-diazole moiety 66 play a key role in improving the antimycobacterial activity.

The final conclusion drawn was that the structural modification of this architecture 65 at three sites (the hydrazine 69, amino acid 70, and the benzoxa-[2,1,3]-diazole 66) would considerably improve the antitubercular activity and reduce the cellular cytotoxicity of the promising compounds whilst not affecting their mycobacterial selectivity, making them attractive architectures for further exploitation as novel antitubercular agents (Figure 106).

* Replacement of hydrazide with amide will abolish activity


Figure 106: SAR conclusions structural modification of the benzoxa-[2,1,3]-diazole peptidomimetics architecture 65 is to develop a novel TB treatment.

## Chapter 5: Future work

## 5 Future work and additional studies

### 5.1 The incorporation of the functionalized amino acids

Pleased by the initial results but with no time remaining, a fruitful strategy for investigating the efficacy of incorporation of the functionalized amino acids was proposed. This strategy started with azido-modified unnatural amino acids as only the three derivatives Boc-3-azido-L-alanine 70Ab, Boc-6-azido-L-norleucine 70Ac and Boc-4-azido-L-phenylalanine 70Ad (Figure 107) are available commercially.


70Ab


70Ac


70Ad

Figure 107: The chemical structures of azido-modified unnatural amino acids.

The synthetic approach that was followed to obtain the hydrazides 67Em-Eo containing azido derivatives of amino acids would again be undertaken to investigate the efficacy of more azido derivatives (Scheme 4) ${ }^{107}$.


Scheme 4: Reagents and conditions: i.) DMTMM, DIPEA, THF, 6-8 h, (59-87\%), r.t.

Due to impurities and rapid decomposition, particularly with the alkyl azide derivatives, this methodology is time-consuming and limited to the aryl azide derivatives which led to difficulties in obtaining the pure hydrazides $\mathbf{6 7}$ containing the alkyl azide residues. As a result of this, it was decided to focus on coupling the Boc-4-azido-L-phenylalanine 70Ad with three different aryl hydrazines 69 to afford the hydrazides 67Em-Eo (Figure 108).

## Chapter 5: Future work



67Em
77\%


67En
49\%


67Eo
23\%

Figure 108: The chemical structures of the hydrazides 67Em-Eo containing azido derivatives of amino acids.

To undertake this investigation, the impact of incorporation of the azido derivatives of amino acids was evaluated through screening these resulting hydrazides 67Em-Eo against the susceptible and resistant Mtb strains (Figure 109).


67Em
$10.09 \mu \mathrm{M}$ wt. $\mathrm{Mtb}_{\mathrm{tb}}$
$161.43 \mu \mathrm{M} \mathrm{INH}{ }^{\mathrm{R}}$ Mtb
$80.72 \mu \mathrm{M} \mathrm{RIF}^{\mathrm{R}}$ Mtb


67En
$1.16 \mu \mathrm{M}$ wt. Mtb
$37.13 \mu \mathrm{M} \mathrm{INH}^{\mathrm{R}}$ Mtb
$37.13 \mu \mathrm{M} \mathrm{RIFR}$ Rtb


67Eo
$18.98 \mu \mathrm{M}$ wt $M_{M+b}{ }^{2}$
$-\mu \mathrm{M}$ INHR ${ }^{\text {Mtb }}{ }^{\text {b }}$

- $\mu \mathrm{M}$ RIFR Mtb

Figure 109: The positive impact of incorporation of the azido derivatives of amino acids on antitubercular activity.

Based on the promising results, introducing Boc-4-azido-L-phenylalanine 70Ad demonstrated a further decrease in MIC. In comparison, the hydrazides 67EmEo, azido-containing derivatives exhibited the best antitubercular activity as the resulting hydrazides (discussed in section 3.1.2) were less potent against all the Mtb strains.

In fact, when the azido group is attached to the side chain of amino acids, the drug affinity increased as well as the potency greatly enhanced particularly, in the case of cyanated Boc-N-hydrazides (cf. 67Db vs 67Eo).

In regard to the conversion of these intermediates into the final compounds 65, the hydrazide 67En, which possesses the lowest MIC value, was converted into

## Chapter 5: Future work

the final benzo-[2,1,3]-diazole 65As. Due to instability of the azide group in the strong acidic conditions, this step consumed a long time to accomplish even with different acids in different concentration (Table 8).

| Entry | Condition | Yield \% |
| :---: | :--- | :---: |
| $\mathbf{1}$ | 4 M HCl in dioxan, 30 minutes at rt. | - |
| $\mathbf{2}$ | $25 \% \mathrm{v} / \mathrm{v}$ TFA in DCM, 2 h at rt. | - |
| $\mathbf{3}$ | $50 \% \mathrm{v} / \mathrm{v} \mathrm{TFA} \mathrm{in} \mathrm{DCM}$,30 minutes at rt. | - |
| $\mathbf{4}$ | 1 M HCl in H H O, 2 h at rt. | - |
| $\mathbf{5}$ | 2 eq. v/v 4 M HCl in dioxan in Et $2 \mathrm{O}, 18 \mathrm{~h}$ at rt. | - |
| $\mathbf{6}$ | $15 \% \mathrm{v} / \mathrm{v}$ aq. HCl in DCM, 1 h at rt. | $95 \%$ |

Following this is the deprotection of intermediate, which will allow reacting with sulfonyl chloride 66 to deliver the final benzo-[2,1,3]-diazole 65As (Scheme 12).


Scheme 12: Reagents and conditions: i.) pyridine, $\mathrm{MeCN}, 4-5 \mathrm{~h}$, (39-72\%), r.t.
The antitubercular activity of benzo-[2,1,3]-diazole 65As was examined against all Mtb strains (Figure 110).


65As $58.46 \mu \mathrm{M}$ wt. Mtb
35\% $29.23 \mu \mathrm{M} \quad \mathrm{NH}^{\mathrm{R}} \quad \mathrm{Mtb}$ $58.46 \mu \mathrm{M}$ INH/RIFR ${ }^{\text {Mtb }}$ $29.23 \mu \mathrm{M} \mathrm{INH/RIF}{ }^{R}$ Mtb

## Chapter 5: Future work

Figure 110: The yield and MIC results of the benzo-[2,1,3]-diazoles 65As containing azido derivatives of amino acids.

## The incorporation of the functionalized amino acids for future work

As mentioned previously, there is a positive correlation the lipophilicity and potency of the resulting compounds, therefore, the natural amino acid fragment had been changed to a functionalized unnatural amino acid. Out of the amino acid derivatives used it appears that those with Boc-4-azido-L-phenylalanine 70Ad proved to be more potent, hence why it may be worth exploring more phenyl substituted amino acids to carry forward to investigate a new lead compound. As a result of this SAR study, the future aim of this project will continually evaluate the impact of the functionalized amino acids, therefore, there are a number of functionalized derivatives of amino acids such as Boc-4-bromo-L-phenylalanine 70Ae, Boc-4-chloro-L-phenylalanine 70Af, Boc-4-fluoro-L-phenylalanine 70Ag (Figure 111). All of this information points to the conclusion that more functionalized derivatives of amino acids need more study alongside changing the substituted aryl hydrazine.


70Ae


70Af


70Ag

Figure 111: The chemical structure of functionalized derivatives of amino acids $\mathbf{7 0 A e} \mathbf{- A g}$

## Chapter 5: Future work

### 5.2 Modification of the benzoxa-[2,1,3]-diazole moiety to produce a better inhibitor effect for future work

It is also possible to incorporate new substituted sulfonyl chloride moieties 69b-e (Figure 112) by changing the substituents in selected positions. This strategy could improve the potency of a new series of modified benzo-[2,1,3]-diazole peptidomimetics.

69b

69e

69c

$69 f$

69g

Figure 112: The chemical structure of new substituted sulfonyl chloride moieties 69b-g
Once the final sulfonamide coupling is successfully achieved, the strategy will be to evaluate the pharmacological effect of the new series of modified benzo-[2,1,3]-diazole against the susceptible and resistant Mtb strains. The SAR investigation of these derivatives would be useful as it would uncover the role of the substituent at different position in 69. Further synthesis and MIC analysis in the future would determine which substituted sulfonyl chloride moiety 69 has a better activity against all Mtb strains.

## Chapter 5: Future work

### 5.3 Proteomic investigation for future work

The major limitation of this explorative study was that a mechanism underlying the inhibitory effect of the compounds being tested was not investigated. Future work would need to address this shortfall. In an attempt to identify the drug target, the designing of chemical probes related to hydrazides 67 and benzo-[2,1,3]diazole derivatives 65 is a future aim of this project. This will help us to understand the toxic and medicinal properties of these bioactive compounds. In proteomic studies, there are two valuable techniques for studying the interaction of drugs with their target proteins. These include photoaffinity labelling and activity-based protein profiling (ABPP).

### 5.3.1 Photoaffinity probe design

The most common photoactivatable probes utilized in photoaffinity labelling (PAL) studies are benzophenone 108, aryl azide 109 and alkyl diazirine 110 which can be activated by UV light (Figure 113).


Figure 113: The mode of action of different UV-photoactivating groups. ${ }^{137}$

These covalently crosslink to the drug target protein which permits the use of analytical tools such as electrophoresis and MS to investigate the structure of protein receptor-drug complexes. ${ }^{138}$ This study begun with incorporating the azido-modified unnatural amino acids 70Ab-Ad (discussed in section 5.1) In the PAL study, we present a significantly improved procedure for the synthesis of hydrazide 67Em-Eo and benzo-[2,1,3]-diazole 65As probes containing the aryl azide group, which was incubated with all Mtb strains and then bound specifically

## Chapter 5: Future work

to the pocket of receptor due to their affinity. Although azides have been widely used in nitrene transfer reactions, the photolysis of these azide groups may produce imines as byproducts. The exposure of these probes to light and heat could induce this decomposition. Furthermore, these probes can react with other reagents and displace the azide group to form new substituted molecules due to their similar behaviour to halogens. ${ }^{139}$ We can therefore conclude that these azido-modified unnatural amino acid 70Ab-Ad showed poor stability and very restricted suitability for in vivo application. Another conclusion which can be drawn from these data is that the approach of activity-based protein profiling (ABPP) could be useful for protein target identification purpose.

### 5.3.2 Biotinylated probe design

In ABPP experiments biotinylated probes (biotin-tagged) are incubated with substances such as avidin, streptavidin or neutravidin beads, followed by the digestion step of enriched proteins with trypsin and LC-MS/MS analysis (Figure 114). ${ }^{140}$



Figure114: Incubation process of biotinylated probes in activity-based protein profiling (ABPP). ${ }^{140}$

### 5.3.2.1 Synthesis of biotinylated hydrazide probe

Herein, we describe the design and synthesis of a biotinylated affinity probe 108 which is composed of the hydrazide component 67 of interest tethered via a lipophilic linker 109 to a biotin-tag. This could potentially be used for identifying the molecular target of the active compound 67.

## Chapter 5: Future work

We look at a four-step synthesis to derive the biotinylated hydrazide probe 108 which will be incubated with Mtb cells and interact with target proteins in their native environment inside a living cell. ${ }^{141}$ This synthesis process will be accomplished via multiple peptide coupling reactions between the lead hydrazide 67 with Boc-6-aminohexanoic acid 109 to give rise to intermediate 110. Following this is the deprotection of intermediate 110 which will allow reacting with biotinNHS 111 to deliver the biotinylated hydrazide probe 108 (Scheme 13).


Scheme 13: Reagents and conditions (a.) HBTU, DIPEA, THF, 6-7 h, r.t. (b.i.) 4 M HCl in dioxane, 1 h, r.t.; (b.ii.) 111, DIPEA, THF, 6-7 h, r.t.

### 5.3.2.2 Synthesis of biotinylated benzoxa-[2,1,3]-diazole probe

As indicated above, the attachment of biotin to the lead benzo-[2,1,3]-diazole 65 provides a powerful tool in proteomic which is capable of identifying the targets of benzo-[2,1,3]-diazole peptidomimetics 65. Here, we present a synthetic approach to design the biotinylated probe 112 which is composed of the bioactive benzo-[2,1,3]-diazole 65 of interest tethered via a lipophilic linker 109 to a biotintag. This could potentially be used for identifying the molecular target of the active compound 65 (Scheme 14).

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Scheme 14: (a.) DIPEA, THF, r.t. (b.i.) 4 M HCl in dioxane, 1 h, r.t; (b.ii.). 111, DIPEA, THF, r.t.

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### 5.4 In vitro cytotoxicity evaluation

As stated already in the first paper which was published by Brown et al, ${ }^{1}$ the MTT assay was applied within the group to determine the cytotoxicity of a limited examples of the intermediates 67 and the final benzo-[2,1,3]-diazole peptidomimetics 65 against the murine macrophage cell line RAW264.7 (Figure 115).

67Bk
0\%

67BI
0\%



65b
$>90 \%$


65d
$>90 \%$


Figure 115: The percentage of cytotoxicity of a limited examples of the intermediates 67 and the final benzo-[2,1,3]-diazole peptidomimetics 65 against the murine macrophage cell line.

## In vitro hepatotoxicity evaluation for future work

One of the main issues facing the long-term TB treatments is the rise in incidence of hepatotoxicity as a result of the lack of alternative TB treatments (discussed in section 1.3.3). However, drug-induced hepatotoxicity is difficult to predict and remains a significant cause of drug development failures. Predictive toxicology screening assays for identifying biomarkers and providing mechanistic evaluations enable identification of potential liabilities earlier in the lead optimization stage and can result in recommendations to mitigate these risks. In addition to that, the potential hepatotoxicity is, on the one hand, an important determinant of TB treatment pharmacology, leading to efficacy or failure of drug candidate during drug development. The hepatocytes

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have a critical role in xenobiotics metabolism, treatments, which induces a severe hepatotoxicity, are frequently withdrawn from the market. For this reason, one of the future aims is to assess their cellular cytotoxicity against the rat $\mathrm{B}-13$ hepatocyte progenitor cells via MTT assay. ${ }^{142}$ As mentioned previously, it would be ideal to develop new antitubercular agents with novel mechanism of action that are effective against both the susceptible and resistant Mtb strains and possess less toxicity potential.

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## 6 Experimental section

### 6.1 General Experimental Information

### 6.1.1 Analysis

All reactions were carried out under a nitrogen atmosphere in glassware dried in the oven overnight ( $>75{ }^{\circ} \mathrm{C}$ ) or infrequently by a heat-gun. All solvents and reagents used as supplied direct from the chemical supplier unless stated otherwise. When utilising mixtures of solvents, the ratios refer to the volumes used. The use of 40-60 petrol refers to the fraction of alkanes boiling between 40 and $60^{\circ} \mathrm{C}$ and was redistilled.

Analytical thin layer chromatography (TLC) was completed for all reactions using aluminium plates (silica gel $60 \AA \mathrm{~F}_{254}$ ) and visualised by UV radiation at 254 nm , or via staining with ninhydrin. All flash chromatography was executed using 12 g pre-packed silica gel $40-63 \mu 60 \AA$ columns.
All ${ }^{1} \mathrm{H},{ }^{13} \mathrm{C}$, and ${ }^{19} \mathrm{~F}$ NMR spectra were recorded in deuterated solvents $\mathrm{CDCl}_{3}$ or DMSO-d 6 on a Bruker Advance III HD 700 MHz , Jeol Lambda 500 MHz , Jeol ECS-400 MHz or Bruker Advance III 300 MHz NMR spectrometer instruments. Spectra analysis reported as the following; chemical shift $\delta$ (ppm) (number of equivalent protons, multiplicity, coupling constant $J(\mathrm{~Hz})$, assignment). All quoted chemical shift values are in parts per million relative to the standard tetramethylsilane ( $\delta \mathrm{H}=0.00 \mathrm{ppm}$ ) with all coupling constants stated to the nearest 1 Hz . Analysis and assignment of spectra supported by COSY, HSQC and HMBC experiments. Internal reference used dependent on solvent present using residual protic peaks of $\mathrm{CHCl}_{3}$ to be $\delta \mathrm{H}=7.26$ or DMSO as $\delta \mathrm{H}=2.50$. Similarly, with ${ }^{13} \mathrm{C}$ NMR the internal reference was taken as the central resonance of $\mathrm{CDCl}_{3}$ with $\delta \mathrm{C}=77.0 \mathrm{ppm}$ or DMSO with a value of $\delta \mathrm{C}=39.5 \mathrm{ppm}$.
Melting points were measured using Stuart SMP3 $_{3}$ melting point apparatus. Infrared (IR) spectra were recorded on either a Varian Scimitar 800 FT-IR spectrometer or a Perkin Elmer Spectrum Two FT-IR spectrometer. High resolution mass spectra (HRMS) were provided by the Mass Spectrometry Service (University of Durham and University of Teesside). Specific optical

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rotation ([a]D) was measured using an Optical Activity PoIAAR 2001 automatic polarimeter.

### 6.1.2 Standard procedures

### 6.1.2.1 $\quad$ Synthesis of $N$-protected Hydrazides - General Procedure

Under a nitrogen atmosphere $N$-protected amino acid (1 equiv.) dissolved in THF ( 5 mL ) at room temperature and treated with HOBt (2 equiv.) followed by DCC (1.2 equiv.). After stirring for 10 minutes at room temperature addition of N -aryl hydrazine ( 1.2 equiv.) continued stirring for another 2 hours at room temperature. Reaction mixed with ethyl acetate ( 8 mL ) and distilled water ( 5 mL ). After separation of the two phases the organic layer washed again with distilled water ( $5 \mathrm{~mL} \times 3 \mathrm{~mL}$ ) followed by a wash with sat. aq. $\mathrm{NH}_{4} \mathrm{Cl}(6 \mathrm{~mL})$ then sat. aq. $\mathrm{NaHCO}_{3}(6 \mathrm{~mL})$ followed by brine ( 8 mL ). Organic layer dried over $\mathrm{MgSO}_{4}$, filtered, evaporated and dried in vacuo achieving the desired N-Boc hydrazides 67.

## tert-Butyl

(2-oxo-2-(2-(4-(trifluoromethyl)phenyl)hydrazineyl)ethyl)
carbamate 67Bk


Following the general procedure outlined, $N$-Boc-glycine ( $0.50 \mathrm{~g}, 2.90 \mathrm{mmol}$ ) and 4-trifluorophenylhydrazine hydrochloride ( $0.61 \mathrm{~g}, 3.50 \mathrm{mmol}$ ) were transformed following flash chromatography ( $\mathrm{DCM} / \mathrm{EtOH} / \mathrm{NH}_{3}$ 200:8:1) into the title compound which was isolated as a white solid ( $0.45 \mathrm{~g}, 47 \%$ ); m.p. $185-187^{\circ} \mathrm{C}$; $V_{\max } 3370$ (NH), 3278 (NH), 3234 (NH), 3108, 3058, 2996, 1649 (C=O), 1615, 1518 (C=O), 1330, 1245, 1156, 1107, 1052, 828, $559 \mathrm{~cm}^{-1}$; ठн ( 700 MHz , DMSO-d 6 ) 9.78 (1H, s, Ar-NHNH), 8.32 (1H, s, Ar-NHNH), 7.40 ( $2 \mathrm{H}, \mathrm{d}, J$ 9, Ar-H), 7.07 (1H, t, J 6, BocNH), 6.78 (2H, d, J 9, Ar-H), 3.60 ( $2 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{NHCH}_{2}$ ), 1.37 ( $9 \mathrm{H}, \mathrm{s}$, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{COC}(\mathrm{O}) \mathrm{NH}\right) ; \quad$ бс (176 MHz, DMSO-d6) 170.2 (CONH), 156.6 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{COC}(\mathrm{O}) \mathrm{NH}\right), 153.1$ ( $\mathrm{Ar}-\mathrm{C}$ ), 126.7 ( $\mathrm{Ar}-\mathrm{C}$ ), 126.4 ( $\mathrm{Ar}-\mathrm{C}$ ), 118.82 ( $\mathrm{Ar}-\mathrm{C}$ ), 118.63 ( $\mathrm{Ar}-\mathrm{C}$ ), 112.1 ( $\mathrm{Ar}-\mathrm{C}), 78.8\left(\mathrm{NHCH}_{2}\right), 42.7\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 28.8\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$;

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$\delta_{\text {F }}(658 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d} 6)-59.30 ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 356\left(\mathrm{MNa}^{+}\right), 689\left(2 \mathrm{M}+\mathrm{Na}^{+}\right) ;$HRMS $\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MH}^{+}, 334.1372\left(\mathrm{C}_{14} \mathrm{H}_{19} \mathrm{~F}_{3} \mathrm{~N}_{3} \mathrm{O}_{3}\right.$ requires 334.1373).

## tert-Butyl (S)-(1-oxo-3-phenyl-1-(2-(4-(trifluoromethyl)phenyl)hydrazineyl) propan-2-yl)carbamate 67Bp



Following the general procedure outlined, $N$-Boc-L-phenylalanine ( $0.25 \mathrm{~g}, 0.94$ mmol ) and 4-trifluorophenylhydrazine hydrochloride ( $0.30 \mathrm{~g}, 1.75 \mathrm{mmol}$ ) were transformed following flash chromatography into the title compound as a goldenbrown solid ( $0.330 \mathrm{~g}, 83 \%$ ); Rf 0.39 (DCM/EtOH/NH3 [200:8:1]); m.p. 157 - 159 ${ }^{\circ} \mathrm{C}$; $V_{\max } 3324$ (N-H), 3274 (N-H), 2931, 2851, 1686 (C=O), 1660 (C=O), 1520, 1328, 1156, 1103, 1067, 835, 716, $515 \mathrm{~cm}^{-1}$; ठH ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 7.93 (1H, bs, NH), 7.39 (2H, d, J 8, Ar-H), 7.36 - 7.32 (3H, m, Ar-H), 7.26 - 7.22 (2H, m, Ar$H$ ), $6.61(2 \mathrm{H}, \mathrm{d}, \mathrm{J} 8, \mathrm{Ar}-H), 6.18(1 \mathrm{H}, \mathrm{bs}, \mathrm{NH}), 5.09(1 \mathrm{H}, \mathrm{bd}, J 7, \mathrm{NH}), 4.46(1 \mathrm{H}, \mathrm{q}$, J 8, CH2CHCO), 3.12 ( $2 \mathrm{H}, \mathrm{d}, \mathrm{J} 8, \mathrm{CH}_{2} \mathrm{CHCO}$ ), 1.47 ( $9 \mathrm{H}, \mathrm{s}, \mathrm{C}(\mathrm{CH} 3) 3$ ); סc (176 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) 171.6$ (NHCHCO), 150.3 (NHCOO), 135.96 (Ar-C), 129.3 (Ar-C), 128.9 (Ar-C), 127.2 (Ar-C), 126.4 (Ar-C), 126.4 (Ar-C), 125.2 (Ar-C), 123.6 (CF3), 112.7 (Ar-C), 80.9 ( $\left.\left(\mathrm{CH}_{3}\right) 3\right), 54.5$ ( NHCHCO ), 33.5 ( $\left.\mathrm{CH}_{2} \mathrm{CHCO}\right), 28.3$ (C(CH3)3); $\delta \mathrm{F}(376 \mathrm{MHz}, \mathrm{CDCl} 3)-61.56(3 \mathrm{~F}, \mathrm{~s}, \mathrm{CF} 3) ; ~ m / z\left(\mathrm{ES}^{+}\right) 424\left(\mathrm{MH}^{+}\right), 446$ $\left(\mathrm{MNa}^{+}\right), 869\left(2 \mathrm{M}+\mathrm{Na}^{+}\right)$; HRMS $\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MH}^{+}, 424.1847\left(\mathrm{C}_{2} 1 \mathrm{H}_{25} \mathrm{~F}_{3} \mathrm{~N}_{3} \mathrm{O}_{3}\right.$ requires 424.1843 ).

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### 6.1.2.2 Synthesis of $N$-protected Hydrazides - General Procedure

Under a nitrogen atmosphere $N$-protected amino acid (1 equiv.) dissolved in THF ( 5 mL ) at room temperature and treated with DIPEA ( 1.5 equiv.) followed by DMTMM (1.2 equiv.). After stirring for 10 minutes at room temperature addition of N-Aryl / alkyl or benzyl hydrazine (1.1 equiv.) continued stirring for another 5 hours at room temperature. Reaction mixed with ethyl acetate ( 8 mL ) and distilled water ( 5 mL ). After separation of the two phases the organic layer washed again with distilled water ( $5 \mathrm{~mL} \times 3 \mathrm{~mL}$ ) followed by a wash with sat. aq. $\mathrm{NH}_{4} \mathrm{Cl}(6 \mathrm{~mL})$ then sat. aq. $\mathrm{NaHCO}_{3}(6 \mathrm{~mL})$ followed by brine ( 8 mL ). Organic layer dried over $\mathrm{MgSO}_{4}$, filtered, evaporated and dried in vacuo achieving the desired N -Boc hydrazides 67, 89 or 91.

## tert-Butyl (2-oxo-2-(2-phenylhydrazineyl)ethyl)carbamate 67a



Following the general procedure outlined, $N$-Boc-glycine ( $0.20 \mathrm{~g}, 1.14 \mathrm{mmol}$ ) and phenylhydrazine ( $0.14 \mathrm{~mL}, 1.37 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{CHCl}_{3}$ into the title compound which was isolated as a dark brown gummy solid ( $0.17 \mathrm{~g}, 55 \%$ ) as a mixture of rotamers [4:1]; $\mathrm{Rf}_{\mathrm{f}} 0.44$ (DCM/EtOH/NH 3 200:8:1); m.p. $86-88^{\circ} \mathrm{C}$; $v_{\max } 3281$ (N-H), 2977, 2929 (C-H), 1678 (C=O), 1602 (C=O), 1495, 1366, 1248, 1160, 750, $692 \mathrm{~cm}^{-1}$; all data provided for the major rotamer; бн ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.73 (1H, bs, CONHNH), 7.18 - 7.14 (2H, m, Ar-H), 6.85 (1H, m, Ar-H), $6.75-6.74$ (2H, d, J 8, Ar-H), 5.64 (1H, bs , BocNHCH2), 3.80 (2H, d, J 6, BocNHCH2), 1.43 (9H, s, ( $\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}$ ); $\delta c\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 170.3$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.4 ( $\mathrm{C}=\mathrm{ONH}$ ), 147.7 (ipso-Ar-C), 129.15 (Ar-C), 129.12 (Ar-C), 120.9 ( $\mathrm{Ar}-\mathrm{C}$ ), 113.45 ( $\mathrm{Ar}-\mathrm{C}), 113.29(\mathrm{Ar}-\mathrm{C}), 80.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 43.1\left(\mathrm{BocNHCH}_{2}\right)$, $\left.28.1\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 266\left(\mathrm{MH}^{+}\right) ; \operatorname{HRMS}\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MH}^{+}, 266.1515$ ( $\mathrm{C}_{13} \mathrm{H}_{20} \mathrm{~N}_{3} \mathrm{O}_{3}$ requires 266.1505).

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## tert-Butyl (S)-(1-oxo-1-(2-phenylhydrazineyl)propan-2-yl)carbamate 67b



Following the general procedure outlined, $N$-Boc- $L$-alanine ( $0.20 \mathrm{~g}, 1.06 \mathrm{mmol}$ ) and phenylhydrazine ( $0.13 \mathrm{~mL}, 1.27 \mathrm{mmol}$ ) were transformed following work up with EtOAc into the title compound which was isolated as a light orange solid ( 0.09 g, 33\%); Rf 0.26 (DCM/EtOH/NH $400: 8: 1$ ); m.p. $129-132{ }^{\circ} \mathrm{C} ; v_{\max } 3432$, 3288, 3026 (N-H), 2982(C-H), 1688 (C=O), 1601 (C=O), 1492, 1454, 1369, 1243, $1155,755,701 \mathrm{~cm}^{-1}$; ठн ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.78 (1H, bs, CONHNH), $7.20-7.14$ (2H, m, Ar-H), 6.85 (1H, t, J 7, Ar-H), 6.81-6.71 (2H, m, Ar-H), 5.40 (1H, bs, BocNHCH), 4.31 (1H, m, BocNHCH), 1.43 ( $9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}$ ), 1.33 (3H, d, J 6, $\mathrm{NHCHCH}_{3}$ ); ठс ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 173.2 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.8 ( $\mathrm{C}=\mathrm{ONH}$ ), 147.8 (ipso-Ar-C), 129.3 (Ar-C), 129.1 (Ar-C), 120.9 (Ar-C), 115.5 (Ar-C), 113.4 (Ar-C), $80.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 48.6$ (BocNHCH), $28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 18.1\left(\mathrm{NHCHCH}_{3}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right)$ $280\left(\mathrm{MH}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\mathrm{MH}^{+}, 280.1674\left(\mathrm{C}_{14} \mathrm{H}_{22} \mathrm{~N}_{3} \mathrm{O}_{3}\right.$ requires 280.1661).

## tert-Butyl

 carbamate 67c(S)-(3-methyl-1-oxo-1-(2-phenylhydrazineyl)butan-2-yl)


Following the general procedure outlined, $N$-Boc- $L$-valine ( $0.20 \mathrm{~g}, 0.92 \mathrm{mmol}$ ) and phenylhydrazine ( $0.10 \mathrm{~mL}, 1.01 \mathrm{mmol}$ ) were transformed following flash column chromatography (DCM/EtOH/NH3 200:8:1) into the title compound which was isolated as a reddish brown oil ( $0.07 \mathrm{~g}, 24 \%$ ) as a mixture of rotamers [4:1]; $\mathrm{R}_{\mathrm{f}}$ 0.44 (DCM/EtOH/NH3 200:8:1); $v_{\max }$ 3341, 3259 (N-H), 2960, 2925 (C-H), 1660 (C=O), 1521, 1307, 1250, 1168, $689 \mathrm{~cm}^{-1}$; all data provided for the major rotamer; $\delta_{\mathrm{H}}\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.00(1 \mathrm{H}, \mathrm{bs}, \mathrm{CONHNH}), 7.23-7.20(2 \mathrm{H}, \mathrm{t}, \mathrm{J} 7, \mathrm{Ar}-\mathrm{H}), 6.90$ (1H, t, J 7, Ar-H), $6.85-6.83$ (2H, d, J 7, Ar-H), 5.04 (1H, bd, J 9, BocNH), 3.96 (1H, m, BocNHCH), 2.18 (1H, m, $\left.\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.47\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.00(3 \mathrm{H}$, d, J 6, CHCH (CH3 $)_{2}$ ), 0.98 (3H, d, J 6, CHCH(CH3 $)_{2}$ ); סc ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 171.9 (C=ONHNH), 155.9 (C=ONH), 147.7 (ipso-Ar-C), 129.0 (Ar-C), 128.7 (Ar-C),

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121.9 (Ar-C), 113.6 (Ar-C), 113.0 (Ar-C), 80.4 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 58.9$ (BocNHCH), 31.3 $\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 18.9\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 18.1\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; m / z$ $\left(\mathrm{ES}^{+}\right) 308\left(\mathrm{MH}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\mathrm{MH}^{+}, 308.1965\left(\mathrm{C}_{16} \mathrm{H}_{26} \mathrm{~N}_{3} \mathrm{O}_{3}\right.$ requires 308.1974).
tert-Butyl (S)-(4-methyl-1-oxo-1-(2-phenylhydrazinyl)pentan-2-yl)carbamate 67d


Following the general procedure outlined, $N$-Boc-L-leucine ( $0.20 \mathrm{~g}, 0.87 \mathrm{mmol}$ ) and phenylhydrazine ( $0.09 \mathrm{~mL}, 0.95 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a light orange solid (0.05 g, 19\%) Rf 0.32 (n-hex/EtOAc 3:1); m.p. $110-114{ }^{\circ} \mathrm{C}$; $v_{\max } 3338,3275(\mathrm{~N}-\mathrm{H})$, 2963 (C-H), 1672 (C=O), 1520, 1367, 1257, 1167, 851, $694 \mathrm{~cm}^{-1}$; ठн ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) 8.20(1 \mathrm{H}, \mathrm{bs}, \mathrm{CONHNH}), 7.23-7.19(2 \mathrm{H}, \mathrm{t}, \mathrm{J} 7, \mathrm{Ar}-\mathrm{H}), 6.90(1 \mathrm{H}, \mathrm{t}, \mathrm{J} 7$, Ar-H), $6.84(2 \mathrm{H}, \mathrm{d}, J 7$, Ar-H), 4.91 (1H, bd, J8, BocNH), 4.20 (1H, m, BocNHCH), $1.77-1.67\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CHHCH}\left(\mathrm{CH}_{3}\right)_{2}, \mathrm{CHHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.55\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHHCH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $1.47\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.97\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 0.95(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6$, $\left.\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ;$ бс (400 MHz, $\left.\mathrm{CDCl}_{3}\right) 129.2$ ( $\mathrm{Ar}-\mathrm{C}$ ), 121.5 ( $\mathrm{Ar}-\mathrm{C}$ ), 113.8 ( $\mathrm{Ar}-\mathrm{C}$ ), $51.2(\mathrm{BocNHCH}), 40.2\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 24.7\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.8\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.1\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 322\left(\mathrm{MH}^{+}\right) ; ~ H R M S ~(E S+)$ Found $\mathrm{MH}^{+}, 322.2122\left(\mathrm{C}_{17} \mathrm{H}_{28} \mathrm{~N}_{3} \mathrm{O}_{3}\right.$ requires 322.2131).
tert-Butyl (S)-(4-(methylthio)-1-oxo-1-(2-phenylhydrazinyl)butan-2-yl) carbamate 67 g


Following the general procedure outlined, $N$-Boc-L-methionine $(0.20 \mathrm{~g}, 0.80$ mmol ) and phenylhydrazine ( $0.09 \mathrm{~mL}, 0.88 \mathrm{mmol}$ ) were transformed following

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trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a white solid ( $0.06 \mathrm{~g}, 23 \%$ ); Rf 0.38 (DCM/EtOH/NH3 200:8:1); m.p. 106-109 ${ }^{\circ} \mathrm{C}$; $v_{\max } 3316$, 3258 (N-H), 2972 (C-H), 1685, 1656 (C=O), 1522, 1495, 1308, 1249, 1166, 1055, 764, $689 \mathrm{~cm}^{-1}$; ठн ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.67 (1H, bs, CONHNH), $7.20-7.16(2 \mathrm{H}, \mathrm{t}$, J 7, Ar-H), 6.88 (1H, t, J 7, Ar-H), 6.81 - 6.79 (2H, d, J 8, Ar-H), 5.39 (1H, bd, J 8, BocNHCH), 4.41 (1H, m, BocNHCH), 2.52 (2H, t, J 7, CHCH2CH2S), $2.16-$ $2.04\left(4 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}, \mathrm{CHCHHCH}_{2} \mathrm{~S}\right), 1.93(1 \mathrm{H}, \mathrm{m}, \mathrm{CHCHHCH} 2 \mathrm{~S}),, 1.45$ (9H, s, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta \mathrm{c}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 171.9$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.9 ( $\mathrm{C}=\mathrm{ONH}$ ), 147.5 (ipso-Ar-C), 129.5 (Ar-C), 129.2 (Ar-C), 121.4 (Ar-C), 113.7 (Ar-C), 112.9 (Ar-C), $80.6 \quad\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 52.0 \quad(\mathrm{BocNHCH}), 31.2 \quad\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 30.2$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 28.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $15.4\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right) ; m / z\left(\mathrm{ES}^{+}\right) 340\left(\mathrm{MH}^{+}\right)$; HRMS (ES') Found $\mathrm{MH}^{+}, 340.1690\left(\mathrm{C}_{16} \mathrm{H}_{25} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}\right.$ requires 340.1695$)$.

## tert-Butyl

carbamate 67h
(S)-(1-oxo-3-phenyl-1-(2-phenylhydrazinyl)propan-2-yl)


Following the general procedure outlined, $N$-Boc-L-phenylalanine ( $0.20 \mathrm{~g}, 0.75$ mmol ) and phenylhydrazine ( $0.09 \mathrm{~mL}, 0.91 \mathrm{mmol}$ ) were transformed following flash column chromatography (DCM/EtOH/ $\mathrm{NH}_{3} 400: 8: 1$ ) into the title compound which was isolated as a brown gummy solid ( $0.11 \mathrm{~g}, 50 \%$ ) as a mixture of rotamers [5:1]; Rf 0.31 (DCM/EtOH/NH3 400:8:1); $v_{\max } 3262$ (N-H), 2978, 1669 (C=O), 1602 (C=O), 1494, 1455, 1366, 1249, 1162, 749, $692 \mathrm{~cm}^{-1}$; all data provided for the major rotamer $\delta н\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.45$ (1H, bs, CONHNH), 7.28 - 7.24 (3H, m, Ar-H), 7.18-7.15 (3H, m, Ar-H), 7.12-7.09 (2H, t, J 7, Ar-H), 6.85-6.82 (1H, m, Ar-H), 6.53-6.51 (1H, d, J 7, Ar-H), 5.37 (1H, bs, BocNHCH), 4.54 (1H, m, BocNHCH), 3.03 (2H, m, Ar-CH2), 1.41 ( $9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}$ ); $\delta \mathrm{c}(500$ MHz, $\mathrm{CDCl}_{3}$ ) 171.8 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.7 (C=ONH), 147.5 (ipso-Ar-C), 136.3 (ipso-Ar-C), 129.5 (Ar-C), 129.4 (Ar-C), 129.1 (Ar-C), 128.7 (Ar-C), 126.9 (Ar-C), 120.9 (Ar-C), 115.4 (Ar-C), 113.5 (Ar-C), 112.9 (Ar-C), 80.5 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 54.4$

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( BocNHCH ), $38.4\left(\mathrm{Ar}^{-} \mathrm{CH}_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 356\left(\mathrm{MH}^{+}\right)$; HRMS (ES $\left.{ }^{+}\right)$ Found $\mathrm{MH}^{+}, 356.1977\left(\mathrm{C}_{20} \mathrm{H}_{26} \mathrm{~N}_{3} \mathrm{O}_{3}\right.$ requires 356.1974$)$.
tert-Butyl (2-(2-(2-chlorophenyl)hydrazineyl)-2-oxoethyl)carbamate 67Ag


Following the general procedure outlined, $N$-Boc-glycine ( $0.25 \mathrm{~g}, 1.43 \mathrm{mmol}$ ) and 2-chlorophenylhydrazine hydrochloride ( $0.31 \mathrm{~g}, 1.71 \mathrm{mmol}$ ) were transformed following work up with EtOAc into the title compound which was isolated as a brown gum ( $0.24 \mathrm{~g}, 56 \%$ ) as a [4:1] rotameric mixture; Rf: 0.21 (DCM/EtOH/NH3 200:8:1). m.p 113-119${ }^{\circ} \mathrm{C}$. $U_{\max } 3367$ (NH) 3271 (NH), 1677 (C=O), 1586, 1494, 1253, 1366, 1248, 1155, 1042 $\mathrm{cm}^{-1}$; NMR data given for the major rotamer ठн $^{\text {( } 700}$ MHz, DMSO-d6) 1.40 (9H, s, C( $\left.\mathrm{CH}_{3}\right)_{3}$ ), 3.64 (2H, d, J 7, NHCH2), 5.30 (1H, bs, NH), 6.75 (1H, t, Ar-H), 6.84 (1H, d, J 14 Ar-H), 7.12 (1H, dd, J 7, J 21, Ar-H), 7.28 (1H, d, J 7); ठc (700 MHz, DMSO-d6) 169.9 (C=O), 156.4 (C=O), 145.1 (ipso-Ar-C), 129.9 (Ar-C), 128.1, (Ar-C), 120.3 (Ar-C), 117.5 (Ar-C-Cl), 113.6 (Ar-C), $78.6\left(\mathrm{O}-\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 42.5\left(\mathrm{NHCH}_{2}\right), 28.7\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right) ; m / z\left(\mathrm{ES}^{+}\right) 300\left(\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 302$ $\left.\left(\left[{ }^{37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 322\left(\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right), 324\left({ }^{37} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\mathrm{MH}^{+}$, $300.1115\left(\mathrm{C}_{13} \mathrm{H}_{19} \mathrm{~N}_{3} \mathrm{O}_{3}{ }^{35} \mathrm{CI}\right.$ requires 300.1115$)$.
tert-Butyl (2-(2-(2,6-dichlorophenyl)hydrazinyl)-2-oxoethyl)carbamate 67As


Following the general procedure outlined, $N$-Boc-glycine ( $0.20 \mathrm{~g}, 1.14 \mathrm{mmol}$ ) and 2,6-dichlorophenylhydrazine hydrochloride ( $0.29 \mathrm{~g}, 1.37 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a white solid ( $0.19 \mathrm{~g}, 52 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.38$ (DCM/EtOH/ $\mathrm{NH}_{3} 200: 6: 1$ ); m.p. $115-118{ }^{\circ} \mathrm{C}$; $v_{\max } 3468,3309,3209(\mathrm{~N}-\mathrm{H}), 1721,1673$ (C=O), 1505, 1448, 1361, 1266, 1169, 1054, 868, $765 \mathrm{~cm}^{-1}$; ठн ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.43 (1H, bs, CONHNH), 7.25-7.23 (2H, d, J 7, Ar-H), 6.93 (1H, t, J 7, Ar-H), 6.76 (1H, bd, J 4, Ar-NH), 5.13 (1H, bs,

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$\mathrm{BocNHCH}_{2}$ ), $3.84\left(2 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{BocNHCH}_{2}\right), 1.45\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta \mathrm{c}(100 \mathrm{MHz}$, $\mathrm{CDCl}_{3}$ ) 168.7 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 140.8 ( $\mathrm{ipso}-\mathrm{Ar}-\mathrm{C}$ ), 128.9 ( $\mathrm{Ar}-\mathrm{C}$ ), 128.8 ( $\mathrm{Ar}-\mathrm{C}$ ), 126.1 (ipso-Ar-C), 125.1 (ipso-Ar-C), 124.2 (Ar-C), $80.7\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 43.1\left(\mathrm{BocNHCH}_{2}\right)$, $28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; \quad \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 356\left(\left[{ }^{35,35} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right), 358\left(\left[{ }^{35,37} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right), 360$ ( $\left.\left[{ }^{37,37} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right) ;$HRMS (ES $\left.{ }^{+}\right)$Found $\quad\left[{ }^{35,35} \mathrm{CI}\right] \mathrm{MNa}^{+}, \quad 356.0540$ $\left(\mathrm{C}_{13} \mathrm{H}_{17}{ }^{35,35} \mathrm{CIN}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 356.0545 ).

## tert-Butyl (2-(2-(3-chlorophenyl)hydrazinyl)-2-oxoethyl)carbamate 67q



Following the general procedure outlined, $N$-Boc-glycine ( $0.23 \mathrm{~g}, 1.32 \mathrm{mmol}$ ) and 3-chlorophenylhydrazine hydrochloride ( $0.28 \mathrm{~g}, 1.58 \mathrm{mmol}$ ) were transformed following work up with EtOAc into the title compound which was isolated as a pale yellow solid ( $0.39 \mathrm{~g}, 99 \%$ ); as a [3:1] rotameric mixture; $\mathrm{Rf}_{\mathrm{f}} 0.39$ (DCM/EtOH/NH3 200:8:1); mp $96-104^{\circ} \mathrm{C}$; $v \max 3344,3269$ (NH), 3078 (Ar-CH), 2980, 2933 (CH), 1712, 1659 (C=O), 1513 (C=C), 1367 (C-N), 1154 (C-O) $\mathrm{cm}^{-1}$; NMR data given for major rotamer $\delta_{\mathrm{H}}\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.43(1 \mathrm{H}, \mathrm{s}, \mathrm{ArNH}), 7.11$ (1H, t, J 7, ArCH ), 6.84 (1H, dd, J 7, 1, Ar-CH), 6.79 (1H, t, J 1, Ar-CH), 6.67 (1H, dd, J 7, 2, Ar-CH), 6.22 (1H, s, CONHNH) 5.38 (1H, s, NHCH2), 3.88 (1H, d, J 5, NHCH2), 1.47 (9H, s, C(CH3) $)^{2}$; ठc (700 MHz, CDCl 3 ) 169.9 (COO), 156.7 (CHC=O) 148.9 (1-Ar-CNH), 135.1 (3-Ar-CCI), 130.2 (5-Ar-CH), 121.3 (6-Ar-CH), 113.4 (2-Ar-CH), $111.8(4-\mathrm{Ar}-\mathrm{CH}), 55.9\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 43.6\left(\mathrm{NHCH}_{2}\right), 28.3\left(\mathrm{C}\left(\mathrm{CH}_{3}\right) 3\right) ; m / z\left(\mathrm{ES}^{+}\right) 322$ $\left(\mathrm{MNa}^{+}\right), 300\left(\mathrm{MH}^{+}\right), 200\left(\mathrm{M}-\mathrm{Boc}^{+}\right) ; \operatorname{HRMS}\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MH}^{+}, 300.1115$ $\left(\mathrm{C}_{13} \mathrm{H}_{19}{ }^{35} \mathrm{CIN}_{3} \mathrm{O}_{3}\right.$ requires 300.1120 ).
tert-Butyl-(2-(2-(4-chlorophenyl)hydrazinyl)-2-oxoethyl)carbamate 67Am


Following the general procedure outlined, $N$-Boc-glycine ( $0.20 \mathrm{~g}, 1.14 \mathrm{mmol}$ ) and 4-chlorophenylhydrazine hydrochloride ( $0.25 \mathrm{~g}, 1.37 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a

## Chapter 6: Experimental section

white solid ( $0.08 \mathrm{~g}, 24 \%$ ) as a mixture of rotamers [4:1]; $\mathrm{Rf}_{\mathrm{f}} 0.30$ (DCM/EtOH/NH3 200:8:1); m.p. $97-99^{\circ} \mathrm{C}$; $v_{\max } 3273,3119$ (N-H), 2979 (C-H), 1656 (C=O), 1527, 1489, 1363, 1241, 1159, $815 \mathrm{~cm}^{-1}$; NMR data given for major rotamer $\delta_{H}(400$ MHz, CDCl3) 8.45 (1H, bs, CONHNH), $7.15-7.13$ (2H, d, J 8, Ar-H), 6.73-6.71 (2H, d, J 8, Ar-H), 6.25 (1H, bs, Ar-NH), 5.41 (1H, bs, BocNHCH2), 3.86 (2H, d, J 5, BocNHCH2), 1.46 (9H, s, ( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; ठс ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 170.1 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.5 (C=ONH), 146.3 (ipso-Ar-C), 129.4 (Ar-C), 129.1 (Ar-C), 125.9 (ipso-Ar-C), 114.7 ( $\mathrm{Ar}-\mathrm{C}), 113.8(\mathrm{Ar}-\mathrm{C}), 80.9\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 43.3\left(\mathrm{BocNHCH}_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 322\left(\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right)$, 324 ( $\left.{ }^{37} \mathrm{Cl}\right] \mathrm{MNa}^{+}$); HRMS (ES ${ }^{+}$) Found $\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}$, $322.0932\left(\mathrm{C}_{13} \mathrm{H}_{18} \mathrm{~N}_{3} \mathrm{O}_{3}{ }^{35} \mathrm{CINa}\right.$ requires 322.0934).
tert-Butyl-(2-(2-(2,4-difluorophenyl)hydrazinyl)-2-oxoethyl)carbamate 67By


Following the general procedure outlined, $N$-Boc-glycine ( $0.20 \mathrm{~g}, 1.14 \mathrm{mmol}$ ) and 2,4-difluorophenylhydrazine hydrochloride ( $0.25 \mathrm{~g}, 1.37 \mathrm{mmol}$ ) were transformed following flash column chromatography (DCM/MeOH 9:1) into the title compound which was isolated as a brown oil ( $0.04 \mathrm{~g}, 12 \%$ ) as a mixture of rotamers [4:1]; Rf 0.38 (DCM/EtOH/NH3 200:8:1); $v_{\max } 3281$ (N-H), 2980 (C-H), 1855, 1679, 1611 (C=O), 1504, 1368, 1256, 1162, 1143, 960, $848 \mathrm{~cm}^{-1}$; all data provided for the major rotamer; $\delta \mathrm{H}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.50(1 \mathrm{H}, \mathrm{bs}, \mathrm{NH}), 6.86-6.71(3 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-$ H), 5.39 ( 1 H , bs, BocNHCH2), 3.86 ( $2 \mathrm{H}, \mathrm{d}, \mathrm{J} 5$, BocNHCH $)_{2} 1.44$ ( $9 \mathrm{H}, \mathrm{s}$, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta_{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 170.1$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.8 ( $\mathrm{C}=\mathrm{ONH}$ ), 132.2 (ipso-Ar-C), 132.1 (ipso-Ar-C), 114.3 (Ar-C), 110.9 (Ar-C), 104.2 (Ar-C), 80.8 $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 43.2\left(\mathrm{BocNHCH}_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; \delta_{\mathrm{F}}\left(282 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$-120.4 (ArF), -128.9 (Ar-F); m/z (ES ${ }^{+}$) $324\left(\mathrm{MNa}^{+}\right) ;$HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}, 324.1165$ $\left(\mathrm{C}_{13} \mathrm{H}_{17} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~F}_{2} \mathrm{Na}\right.$ requires 324.1136).

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## tert-Butyl <br> (S)-(1-(2-(2-chlorophenyl)hydrazineyl)-1-oxopropan-2-yl) carbamate 67Ai



Following the general procedure outlined, N -Boc- L -alanine ( $0.20 \mathrm{~g}, 1.06 \mathrm{mmol}$ ) and 2-chlorophenylhydrazine hydrochloride ( $0.23 \mathrm{~g}, 1.27 \mathrm{mmol}$ ) were transformed following work up with EtOAc into the title compound which was isolated as a yellow solid ( $0.30 \mathrm{~g}, 91 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.56$ (DCM/EtOH/NH3 100:8:1); m.p. 146-149 ${ }^{\circ} \mathrm{C}$; $v_{\max } 3453,3284$ (N-H), 2976 (C-H), 1693 (C=O), 1675 (C=O), 1492, 1455, 1371, 1247, 1160, 753, $658 \mathrm{~cm}^{1}$; $\delta \mathrm{H}\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.36(1 \mathrm{H}, \mathrm{bs}$, CONHNH), 7.26 (1H, dd, J 8, 2, Ar-H), 7.12 (1H, dt, J7, 2, Ar-H), 6.85 (1H, dd, J 8, 2, Ar-H), 6.81 (1H, dt, J7, 2, Ar-H), 6.42 (1H, bs, Ar-NH), 5.05 (1H, bs, BocNHCH), 4.29 (1H, m, BocNHCH), 1.46 ( $\left.9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$ 1.41 (3H, d, J7, $\mathrm{NHCHCH}_{3}$ ); $\delta \mathrm{c}(700 \mathrm{MHz}$, $\mathrm{CDCl}_{3}$ ) 172.5 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.9 ( $\mathrm{C}=\mathrm{ONH}$ ), 143.7 (ipso-Ar-C), 129.5 (Ar-C), 127.6 (Ar-C), 121.4 (Ar-C), 119.6 (Ar-C), 113.5 ( $\mathrm{Ar}-\mathrm{C}$ ), 80.7 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 48.6$ (BocNHCH), $28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 17.5\left(\mathrm{NHCHCH}_{3}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 314\left(\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 316$ $\left(\left[{ }^{37} \mathrm{CI}\right] \mathrm{MH}^{+}\right)$; HRMS (ES $)$Found $\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}, 314.1285\left(\mathrm{C}_{14} \mathrm{H}_{21}{ }^{35} \mathrm{CIN}_{3} \mathrm{O}_{3}\right.$ requires 314.1271).
tert-Butyl (S)-(1-(2-(3-chlorophenyl)hydrazineyl)-1-oxopropan-2-yl)
carbamate 67r


Following the general procedure outlined, N -Boc- L -alanine ( $0.25 \mathrm{~g}, 1.32 \mathrm{mmol}$ ) and 3-chlorophenylhydrazine hydrochloride ( $0.28 \mathrm{~g}, 1.56 \mathrm{mmol})$ were transformed following work up with EtOAc into the title compound which was isolated as a pale yellow solid ( $0.39 \mathrm{~g}, 94 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.19$ (DCM:EtOH:NH3 200:8:1); mp 125 - $127^{\circ} \mathrm{C}$; $v_{\max } 3428$, 3278 (NH), 3078 (Ar-CH), 2979, 2933 (CH), 1678 (C=O), 1504, 1367, $1158 \mathrm{~cm}^{-1} ; \delta \mathrm{H}\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.52$ (1H, bs, CONHNH),

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7.12 (1H, t, J 7, Ar-H), 6.85 (1H, d, J 7, Ar-H), 6.80 (1H, s, Ar-H), 6.70 (1H, d, J 7, Ar-H), $5.15\left(1 \mathrm{H}, \mathrm{s}, \mathrm{NHCH}\left(\mathrm{CH}_{3}\right)\right), 4.30\left(1 \mathrm{H}, \mathrm{m}, \mathrm{NHCHCH}_{3}\right), 1.48(9 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 1.40\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 7, \mathrm{NHCH}\left(\mathrm{CH}_{3}\right)\right) ; \delta \mathrm{c}\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 172.9(\mathrm{C=O}), 155.9$ ( $\mathrm{CHC}=\mathrm{O}$ ) 149.1 ( $\mathrm{Ar}-\mathrm{C}$ ), 135.1 ( $\mathrm{Ar}-\mathrm{C}$ ), 130.2 ( $\mathrm{Ar}-\mathrm{C}$ ), 121.1 ( $\mathrm{Ar}-\mathrm{CCl}$ ), 113.4 ( $\mathrm{Ar}-\mathrm{C}$ ), 111.7 ( $\mathrm{Ar}-\mathrm{C}), 80.8\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 48.6\left(\mathrm{NHCH}\left(\mathrm{CH}_{3}\right)\right), 28.3\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 17.5$ $\left(\mathrm{NHCHCH}_{3}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 314\left(\mathrm{MH}^{+}\right), 258$ (M-tert-butyl$\left.{ }^{+}\right) ;$HRMS (ES $\left.{ }^{+}\right)$Found $\mathrm{MH}^{+}$, $314.1259\left(\mathrm{C}_{14} \mathrm{H}_{21}{ }^{35} \mathrm{CIN}_{3} \mathrm{O}_{3}\right.$ requires 314.1271$)$.
tert-butyl
(S)-(1-(2-(4-chlorophenyl)hydrazineyl)-1-oxopropan-2-yl)
carbamate 67An


Following the general procedure outlined, N -Boc-L-alanine ( $0.10 \mathrm{~g}, 0.53 \mathrm{mmol}$ ) and 4-chlorophenylhydrazine hydrochloride ( $0.10 \mathrm{~g}, 0.53 \mathrm{mmol}$ ) were transformed following work up with EtOAc into the title compound which was isolated as an orange solid ( $0.08 \mathrm{~g}, 46 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.23$ (DCM:EtOH: $\mathrm{NH}_{3}$ 200:8:1); mp $122-124{ }^{\circ} \mathrm{C}$; $v_{\max } 3428,3278(\mathrm{NH}), 3078$ (Ar-CH), 2979, 2933 (CH), 1678 (C=O), 1504, 1367, $1158 \mathrm{~cm}^{-1} ; \delta$ ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.68 (1H, bs, CONHNH), 7.11 7.09 (2H, m, Ar-H), 6.69 - 6.67 (2H, m, Ar-H), 6.80 (1H, s, Ar-H), 6.70 (1H, d, J 7, Ar-H), 5.27 ( $1 \mathrm{H}, \mathrm{bs}, \mathrm{NHCHCH}_{3}$ ), $4.28\left(1 \mathrm{H}, \mathrm{m}, \mathrm{NHCHCH}_{3}\right), 1.44(9 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 1.35\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 7, \mathrm{NHCHCH}_{3}\right) ; \delta c\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 173.3(\mathrm{C}=\mathrm{O}), 155.9$ ( $\mathrm{CHC}=\mathrm{O}$ ) 146.6 (ipso-Ar-C), 129.1 (Ar-C), 125.8 ( $\mathrm{Ar}-\mathrm{CCl}$ ), 114.7 (Ar-C), 80.8
 $\left.\left({ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}\right)$, $\left.316 \quad\left({ }^{37} \mathrm{C}\right] \mathrm{MH}^{+}\right)$; HRMS (ES $)$Found $\left[{ }^{35} \mathrm{CI}\right] \mathrm{MH}^{+}, 314.1193$ $\left(\mathrm{C}_{14} \mathrm{H}_{21}{ }^{35} \mathrm{ClN}_{3} \mathrm{O}_{3}\right.$ requires 314.1181 ).

## Chapter 6: Experimental section

## tert-Butyl (S)-(1-(2-(2,6-dichlorophenyl)hydrazinyl)-1-oxopropan-2-yl)

 carbamate 67At

Following the general procedure outlined, N -Boc- L -alanine ( $0.20 \mathrm{~g}, 1.06 \mathrm{mmol}$ ) and 2,6-dichlorophenylhydrazine hydrochloride ( $0.27 \mathrm{~g}, 1.27 \mathrm{mmol}$ ) were transformed following work up with EtOAc into the title compound which was isolated as a golden brown solid ( $0.35 \mathrm{~g}, 95 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.48$ ( $\mathrm{DCM} / E t O H / \mathrm{NH}_{3} 200: 8: 1$ ); m.p. 98-101 ${ }^{\circ} \mathrm{C}$; $v_{\max } 3335,3304$ (N-H), 2983 (C-H), 16786, 1674 (C=O), 1521, 1449, 1317, 1248, 1164, 1162, 766, $627 \mathrm{~cm}^{-1}$; бн ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.68 (1H, bs, CONHNH), 7.26 - 7.24 (2H, d, J 8, Ar-H), 6.92 (1H, t, J 8, Ar-H), 6.74 (1H, bs, ArNH), 5.05 (1H, d, J 7, BocNHCH), 4.29 (1H, m, BocNHCH), 1.46 (9H, s, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.37\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 7, \mathrm{NHCHCH}_{3}\right)$; ठc ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 171.8 (C=ONHNH), 155.6 (C=ONH), 140.9 (ipso-Ar-C), 128.9 (Ar-C), 125.9 (ipso-Ar-C), 124.0 (Ar-C), $80.5\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 48.4(\mathrm{BocNHCH}), 28.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 17.7\left(\mathrm{NHCHCH}_{3}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right)$ $\left.\left.348\left(\left[{ }^{35,35} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 350\left({ }^{35,37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 352\left({ }^{37,37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right)$; HRMS (ES $)$Found $\left[{ }^{35,35} \mathrm{Cl}\right] \mathrm{MH}^{+}, 348.0865\left(\mathrm{C}_{14} \mathrm{H}_{20}{ }^{35,35} \mathrm{ClN}_{3} \mathrm{O}_{3}\right.$ requires 348.0882$)$.
tert-Butyl
carbamate 67Cf


Following the general procedure outlined, $N$-Boc- $L$-alanine ( $0.20 \mathrm{~g}, 1.06 \mathrm{mmol}$ ) and 2-bromophenylhydrazine hydrochloride ( $0.28 \mathrm{~g}, 1.27 \mathrm{mmol}$ ) were transformed following work up with EtOAc into the title compound which was isolated as a light brown solid ( $0.32 \mathrm{~g}, 84 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.5$ (DCM/EtOH/NH3 200:8:1); m.p. 112-114 ${ }^{\circ} \mathrm{C}$; $v_{\max } 3286$ (N-H), 2983 (C-H), 1672 (C=O), 1491, 1453, 1366, 1247, 1160, $749,647 \mathrm{~cm}^{-1}$; ठн ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.66 (1H, bs, CONHNH), 7.41 (1H, dd, J 8, 1, Ar-H), 7.14 (1H, m, Ar-H), 6.83 (1H, dd, J 8, 1, Ar-H), 6.73 (1H,

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m, Ar-H), 6.39 (1H, d, J 3 Ar-NH), 5.24 (1H, d, J 6, BocNHCH), 4.33 (1H, m, BocNHCH), $1.44\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) 1.39\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 7, \mathrm{NHCHCH}_{3}\right)$; $\delta \mathrm{c}(700 \mathrm{MHz}$, $\mathrm{CDCl}_{3}$ ) 172.8 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.9 ( $\mathrm{C}=\mathrm{ONH}$ ), 144.6 (ipso-Ar-C), 132.6 (Ar-C), 128.3 (Ar-C), 121.8 ( $\mathrm{Ar}-\mathrm{C}$ ), 113.7 ( $\mathrm{Ar}-\mathrm{C}$ ), 109.1 ( $\mathrm{Ar}-\mathrm{C}$ ), 80.6 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 48.6$ (BocNHCH), $\left.28.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 17.8\left(\mathrm{NHCHCH}_{3}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 358\left({ }^{79} \mathrm{Br}\right] \mathrm{MH}^{+}\right), 360$ ( $\left.{ }^{81} \mathrm{Br}\right] \mathrm{MH}^{+}$); HRMS (ES ${ }^{+}$) Found $\left[{ }^{79} \mathrm{Br}\right] \mathrm{MH}^{+}, 358.0769\left(\mathrm{C}_{14} \mathrm{H}_{21}{ }^{79} \mathrm{BrN}_{3} \mathrm{O}_{3}\right.$ requires 358.0766).

## tert-Butyl

(S)-(1-(2-(2,4-difluorophenyl)hydrazinyl)-1-oxopropan-2-yl)
carbamate 67Bz


Following the general procedure outlined, $N$-Boc-L-alanine ( $0.20 \mathrm{~g}, 1.06 \mathrm{mmol}$ ) and 2,4-difluorophenylhydrazine hydrochloride ( $0.22 \mathrm{~g}, 1.27 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as an orange gummy solid ( $0.06 \mathrm{~g}, 17 \%$ ); Rf 0.29 (DCM/EtOH/NH3 400:8:1); m.p. $91-94^{\circ} \mathrm{C} ; v_{\max } 3215$ (N-H), 2989 (C-H), 1686, 1668 (C=O), 1508, 1319, 1252, 1161, 1135, 963, 853, $669 \mathrm{~cm}^{-1}$; $\delta_{\mathrm{H}}\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.24$ (1H, bs, CONHNH), 6.84 (1H, m, Ar-H), 6.80 (1H, m, Ar-H), 6.73 (1H, t, J 7, Ar-H), 6.12 (1H, bs, Ar-NH), 4.97 (1H, bd, J7, BocNHCH), 4.23 (1H, m, BocNHCH), 1.47 ( 9 H , s, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.40\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 7, \mathrm{NHCHCH}_{3}\right) ; \delta \mathrm{c}\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 171.2$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.9 ( $\mathrm{C}=\mathrm{ONH}$ ), 132.3 ( $\mathrm{Ar}-\mathrm{C}$ ), 115.4 ( $\mathrm{Ar}-\mathrm{C}$ ), 110.8 ( $\mathrm{Ar}-\mathrm{C}$ ), 103.9 (Ar-C), $80.8\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 48.7$ (BocNHCH), $28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 21.1\left(\mathrm{NHCHCH}_{3}\right)$; $\delta_{F}\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$-120.7 (Ar-F), -129.1 (Ar-F); m/z (ES $\left.{ }^{+}\right) 338$ (MNa+); HRMS $\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}, 338.1300\left(\mathrm{C}_{14} \mathrm{H}_{19} \mathrm{~F}_{2} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 338.1292).

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tert-Butyl
carbamate 67Cx
(S)-(1-(2-(4-cyanophenyl)hydrazinyl)-1-oxopropan-2-yl)


Following the general procedure outlined, $N$-Boc-L-alanine ( $0.20 \mathrm{~g}, 1.06 \mathrm{mmol}$ ) and 4-cyanophenylhydrazine hydrochloride ( $0.22 \mathrm{~g}, 1.27 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow solid ( $0.25 \mathrm{~g}, 78 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.21$ ( $\mathrm{DCM} / E t O H / \mathrm{NH}_{3}$ 200:8:1); m.p. 115-121 ${ }^{\circ} \mathrm{C} ; v_{\max } 3282(\mathrm{~N}-\mathrm{H}), 2220(\mathrm{C} \equiv \mathrm{N}), 16780,1603(\mathrm{C}=\mathrm{O}), 1508,1318$, 1250, 1164, 1166, 830, $743,537 \mathrm{~cm}^{-1}$; $\delta \mathrm{H}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.40(1 \mathrm{H}, \mathrm{bs}$,
 (1H, bd, J3, Ar-NH), 5.00 (1H, d, J7, BocNHCH), 4.25 (1H, m, BocNHCH), 1.47 (9H, s, ( $\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}$ ), $1.42\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 7, \mathrm{NHCHCH}_{3}\right)$; $\delta c\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 172.9$ ( $C=O N H N H$ ), 153.9 ( $C=O N H$ ), 151.4 (ipso-Ar-C), 133.6 (Ar-C), 119.5 ( $\mathrm{C} \equiv \mathrm{N}$ ), 112.9 (Ar-C), 111.4 (Ar-C), 103.3 (Ar-C), 81.1 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 48.7$ (BocNHCH), 28.3 $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 17.1\left(\mathrm{NHCHCH}_{3}\right) ; m / z\left(\mathrm{ES}^{+}\right) 305\left(\mathrm{MH}^{+}\right) ;$HRMS (ES+$)$Found $\mathrm{MH}^{+}$, $305.1617\left(\mathrm{C}_{15} \mathrm{H}_{21} \mathrm{~N}_{4} \mathrm{O}_{3}\right.$ requires 305.1614).
tert-Butyl (S)-(1-(2-(4-(methylsulfonyl)phenyl)hydrazinyl)-1-oxopropan-2-yl) carbamate 67Dj


Following the general procedure outlined, N -Boc-L-alanine ( $0.20 \mathrm{~g}, 1.06 \mathrm{mmol}$ ) and 4-(methylsulphonyl)phenylhydrazine hydrochloride ( $0.26 \mathrm{~g}, 1.12 \mathrm{mmol}$ ) were transformed following work up with EtOAc into the title compound which was isolated as a light yellow solid ( $0.38 \mathrm{~g}, 99 \%$ ) as a mixture of rotamers [5:1]; $\mathrm{R}_{\mathrm{f}}$ 0.59 (DCM/EtOH/NH3 200:8:1); m.p. 126-130 ${ }^{\circ} \mathrm{C}$; $v_{\max } 3326(\mathrm{~N}-\mathrm{H}), 2976(\mathrm{C}-\mathrm{H})$, 1718, 1688 (C=O), 1513, 1371. 1145, 829, 775, $516 \mathrm{~cm}^{-1}$; NMR data given for the major rotamer $\delta$ н $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.46(1 \mathrm{H}$, bs, CONHNH$), 7.74-7.72(2 \mathrm{H}$, dd, J6, 1, Ar-H), $6.88-6.86$ (2H, dd, J6, 1, Ar-H), 5.02 (1H, bs, BocNHCH), 4.26

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(1H, m, BocNHCH), $4.26\left(3 \mathrm{H}, \mathrm{s}, \mathrm{SO}_{2} \mathrm{CH}_{3}\right), 1.48\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.43(3 \mathrm{H}, \mathrm{d}$, $\left.J 7, \mathrm{NHCHCH}_{3}\right) ; ~ \delta c\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 172.9$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 152.4 ( $\left.\mathrm{C}=\mathrm{ONH}\right), 131.7$ (ipso-Ar-C), 129.3 (Ar-C), 129.2 (Ar-C), $81.1\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 48.8$ (BocNHCH), 44.9 $\left(\mathrm{SO}_{2} \mathrm{CH}_{3}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 17.1\left(\mathrm{NHCHCH}_{3}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 358\left(\mathrm{MH}^{+}\right)$; HRMS (ES$\left.{ }^{+}\right)$ Found $\mathrm{MH}^{+}, 358.1450\left(\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{~N}_{3} \mathrm{O}_{5} \mathrm{~S}\right.$ requires 358.1437).

## tert-Butyl (S)-(1-oxo-1-(2-(p-tolyl)hydrazinyl)propan-2-yl)carbamate 67Eb



Following the general procedure outlined, $N$-Boc-L-alanine ( $0.20 \mathrm{~g}, 1.06 \mathrm{mmol}$ ) and 4-methylphenylhydrazine hydrochloride ( $0.20 \mathrm{~g}, 1.27 \mathrm{mmol}$ ) were transformed following work up with EtOAc into the title compound which was isolated as a yellowish orange solid ( $0.25 \mathrm{~g}, 80 \%$ ) as a mixture of rotamers [7:1]; Rf 0.5 (DCM/EtOH/NH3 200:8:1); m.p. 112-114 ${ }^{\circ} \mathrm{C}$; $v_{\max } 3213$ (N-H), 2987 (C-H), 1686 (C=O), 1664 (C=O), 1513, 1321, 1250, 1163, 805, $641 \mathrm{~cm}^{-1}$; $\delta$ ( 700 MHz , $\left.\mathrm{CDCl}_{3}\right) 8.50(1 \mathrm{H}, \mathrm{bs}, \mathrm{CONHNH}), 7.00-6.98(2 \mathrm{H}, \mathrm{d}, \mathrm{J} 8, \mathrm{Ar}-\mathrm{H}), 7.71-6.69(2 \mathrm{H}$, d, J 8, Ar-H), 6.09 (1H, bs, Ar-NH), 5.23 (1H, bs, BocNHCH), 4.28 (1H, m, BocNHCH), $2.24\left(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{CH}_{3}\right), 1.45\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.36(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6$, $\mathrm{NHCHCH}_{3}$ ); $\delta c\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 172.8$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.8 ( $\mathrm{C}=\mathrm{ONH}$ ), 145.4 (ipso-Ar-C), 130.7 (Ar-C), 129.9 (Ar-C), 129.6 (Ar-C), 113.7 (Ar-C), 112.9 (Ar-C), $80.5\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 48.6$ (BocNHCH$), 28.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 20.6$ ( $\left.\mathrm{Ar}-\mathrm{CH}_{3}\right), 17.9$ $\left(\mathrm{NHCHCH}_{3}\right) ; ~ m / z ~\left(E S^{+}\right) 294\left(\mathrm{MH}^{+}\right) ; ~ H R M S ~\left(E S^{+}\right)$Found $\mathrm{MH}^{+}, 294.1829$ ( $\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{~N}_{3} \mathrm{O}_{3}$ requires 294.1818).

## tert-Butyl <br> (S)-(1-(2-(4-isopropylphenyl)hydrazinyl)-1-oxopropan-2-yl) carbamate 67Eh



Following the general procedure outlined, N -Boc-L-alanine ( $0.20 \mathrm{~g}, 1.06 \mathrm{mmol}$ ) and 4-isopropylphenylhydrazine hydrochloride ( $0.20 \mathrm{~g}, 1.27 \mathrm{mmol}$ ) were transformed following flash column chromatography (DCM/EtOH 100:6) into the

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title compound which was isolated as a brown oil ( $0.11 \mathrm{~g}, 33 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.46$ (DCM/EtOH/NH3 200:8:1); $v_{\max } 3288$ (N-H), 2959 (C-H), 1669, 1615 (C=O), 1512, 1365, 1248, 1052, 827, 754, $539 \mathrm{~cm}^{-1}$; ठн ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.51 (1H, bs, CONHNH), 7.05 - 7.03 (2H, d, J 7, Ar-H), 6.74 - 6.72 (2H, d, J7, Ar-H), 5.21 (1H, bs, BocNHCH), 4.29 (1H, m, BocNHCH), 2.81 (1H, m, CH(CH3)2), 1.45 ( $9 \mathrm{H}, \mathrm{s}$, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.37\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 7, \mathrm{NHCHCH}_{3}\right), 1.19-1.18\left(6 \mathrm{H}, \mathrm{d}, \mathrm{J} 7, \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ;$ бc (400 MHz, CDCl 3 ) 172.8 ( $\mathrm{C=ONHNH}$ ), 155.7 ( $\mathrm{C}=\mathrm{ONH}$ ), 145.7 (ipso-Ar-C), 141.7 (ipso-Ar-C), 127.3 (Ar-C), 127.0 (Ar-C), 113.7 (Ar-C), 113.1 (Ar-C), 80.5 $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 48.7 \quad(\mathrm{BocNHCH}), 33.3 \quad\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.4 \quad\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 24.2$ $\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 24.0\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 17.9\left(\mathrm{NHCHCH}_{3}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 322\left(\mathrm{MH}^{+}\right) ;$HRMS ( $\mathrm{ES}^{+}$) Found $\mathrm{MH}^{+}, 322.2136\left(\mathrm{C}_{17} \mathrm{H}_{28} \mathrm{~N}_{3} \mathrm{O}_{3}\right.$ requires 322.2131).

## tert-Butyl (S)-(1-oxo-1-(2-(4-(trifluoromethyl)phenyl)hydrazineyl)propan-2-

## yl)carbamate 67BI



Following the general procedure outlined, $N$-Boc-L-alanine ( $0.25 \mathrm{~g}, 1.32 \mathrm{mmol}$ ) and 4-trifluorophenyl hydrazine hydrochloride ( $0.30 \mathrm{~g}, 1.75 \mathrm{mmol}$ ) were transformed following flash chromatography (DCM/EtOH/NH $200: 8: 1$ ) into the title compound as a dark brown solid ( $0.43 \mathrm{~g}, 93 \%$ ); Rf 0.37 (DCM/EtOH/NH [200:8:1]); m.p. $126-131^{\circ} \mathrm{C}$; $v_{\max } 3321$ (N-H), 2929, 2851, 1683 (C=O), 1664 (C=O), 1617, 1524, 1322, 1157, 1101, 1066, 830, $641 \mathrm{~cm}^{-1}$; ठH ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.50 (1H, bs, NH), 7.46 (2H, d, J 8, Ar-H), 6.87 (2H, d, J 8, Ar-H), 6.34 (1H, bs, NH), 5.07 (1H, bd, J 6, NH), 4.40 (1H, m, CH3CH), 1.50 ( $\left.9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right) 3\right), 1.43$ (3H, d, J 7, CH3CH); סC (176 MHz, CDCl3) 171.7 (NHCHCO), 158.7 (NHCOO), 129.3 (Ar-C), 128.9 (Ar-C), 127.2 (Ar-C), 126.4 (CF3), 112.8 (Ar-C), 80.9 (C(CH3)3), 54.6 ( NHCHCO ), $33.7\left(\mathrm{CHCH}_{3}\right), 28.3(\mathrm{C}(\mathrm{CH} 3) 3)$; $\delta \mathrm{F}(376 \mathrm{MHz}$, CDCl3) -61.55 (3F, s, CF3); m/z (ES ${ }^{+}$) $348\left(\mathrm{MH}^{+}\right), 370\left(\mathrm{MNa}^{+}\right), 717\left(2 \mathrm{M}+\mathrm{Na}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}, 370.1352\left(\mathrm{C}_{15} \mathrm{H}_{20} \mathrm{~F} 3 \mathrm{~N} 3 \mathrm{O} 3 \mathrm{Na}\right.$ requires 370.1349).

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## tert-Butyl

(S)-(1-(2-(2,4-bis(trifluoromethyl)phenyl)hydrazinyl)-1-oxopropan-2-yl)carbamate 67Bt


Following the general procedure outlined, $N$-Boc-L-alanine ( $0.20 \mathrm{~g}, 1.06 \mathrm{mmol}$ ) and (3,5-bis(trifluoromethyl)phenyl)hydrazine hydrochloride ( $0.27 \mathrm{~g}, 0.95 \mathrm{mmol}$ ) were transformed following flash chromatography (DCM/EtOH/ $\mathrm{NH}_{3} 400: 8: 1$ ) into the title compound as a golden white solid ( $0.03 \mathrm{~g}, 6 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.26$ (DCM/EtOH/NH3 400:8:1); m.p. $174-177{ }^{\circ} \mathrm{C}$; $v_{\max } 3331,3267(\mathrm{~N}-\mathrm{H}), 2995(\mathrm{C}-\mathrm{H}), 1669,1623$ (C=O), 1522, 1285, 1161, 1123, 1086, 878, $682 \mathrm{~cm}^{-1}$; ठн (700 MHz, CDCI3) 8.44 (1H, bs, CONHNH), 7.36 (1H, s, Ar-H), 7.21 (2H, s, Ar-H), 6.39 (1H, bs, Ar-NH), 4.94 (1H, bd, J 7, BocNHCH), 4.27 (1H, m, BocNHCH), 1.47 ( $\left.9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $1.42\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 7, \mathrm{NHCHCH}_{3}\right) ; \delta c\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 173.1$ ( $\left.\mathrm{C}=\mathrm{ONHNH}\right), 156.1$ ( $C=O N H$ ), 149.2 (ipso-Ar-C), 132.6 (1C, q, J 33, Ar-CF3), 123.3 (1C, q, J 272, ipso-Ar-CCF3), 114.4 (Ar-C), 112.9 ( $\mathrm{Ar}-\mathrm{C}$ ), 81.2 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 48.7$ ( BocNHCH ), $28.2\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 16.8\left(\mathrm{NHCHCH}_{3}\right) ; \delta \mathrm{F}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)-63.1\left(\mathrm{CF}_{3}\right) ; ~ m / z\left(\mathrm{ES}^{-}\right)$ 414 (MH-); HRMS (ES-) Found (MH $\left.{ }^{-}\right), 414.1251\left(\mathrm{C}_{16} \mathrm{H}_{18} \mathrm{~F}_{6} \mathrm{~N}_{3} \mathrm{O}_{3}\right.$ requires 414.1252).

## tert-Butyl (S)-(1-(2-(3-chlorophenyl)hydrazinyl)-3-methyl-1-oxobutan-2-yl) carbamate 67s



Following the general procedure outlined, $N$-Boc- $L$-valine ( $0.30 \mathrm{~g}, 1.38 \mathrm{mmol}$ ) and 3-chlorophenylhydrazine hydrochloride ( $0.27 \mathrm{~g}, 1.52 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound as a white solid ( 0.22 g , 46\%); Rf 0.61 (DCM/EtOH/NH3 200:8:1); m.p. $150-153{ }^{\circ} \mathrm{C}$; $v_{\max } 3323,3259$ (NH), 2969 (C-H), 1685, 1663 (C=O), 1522, 1468, 1306, 1246, 1168, 858, $680 \mathrm{~cm}^{-}$ ${ }^{1}$; $\delta н\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.49$ (1H, bs, CONHNH), 7.11 (1H, t, J 8, Ar-H), $6.84-$

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6.80 (2H, m, Ar-H), 6.69 (1H, dd, J 8, 2, Ar-H), 6.27 (1H, bd, J 3, Ar-NH), 5.19 (1H, bd, J 9, BocNHCH), 4.00 (1H, t, J 8, BocNHCH), 2.12 (1H, m, CHCH(CH3)2), $1.45\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.99-0.96\left(6 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right)$; $\delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ 172.2 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.2 ( $\mathrm{C}=\mathrm{ONH}$ ), 149.2 (ipso-Ar-C), 134.9 (ipso-Ar-C), 130.2 (Ar-C), 121.1 (Ar-C), 113.6 (Ar-C), 111.7 ( $\mathrm{Ar}-\mathrm{C}$ ), $80.5\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 58.8$ ( BocNHCH ), $30.5\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 19.4\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 18.2$ $\left.\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 364\left({ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right)$, 366 ( $\left.{ }^{37} \mathrm{Cl}\right] \mathrm{MNa}^{+}$); HRMS (ES ${ }^{+}$) Found $\left[{ }^{35} \mathrm{CI}\right] \mathrm{MNa}^{+}, 364.1401\left(\mathrm{C}_{16} \mathrm{H}_{24}{ }^{35} \mathrm{CIN}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 364.1404).

## tert-Butyl-(S)-(1-(2-(4-chlorophenyl)hydrazinyl)-3-methyl-1-oxobutan-2yl)carbamate 67 Ao



Following the general procedure outlined, $N$-Boc- $L$-valine ( $0.20 \mathrm{~g}, 0.92 \mathrm{mmol}$ ) and 4-chlorophenylhydrazine hydrochloride ( $0.14 \mathrm{~g}, 1.01 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound as a yellow solid ( 0.12 g , $38 \%) ; \mathrm{R}_{\mathrm{f}} 0.32$ (DCM/EtOH/NH3 200:8:1); m.p. $146-150{ }^{\circ} \mathrm{C}$; $v_{\max } 3325,3281$ (NH), 2967, 2870 (C-H), 1685, 1649 (C=O), 1522, 1487, 1382, 1245, 1167, 829, $643 \mathrm{~cm}^{-1}$; ठн ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.35 (1H, bs, CONHNH), 7.15-7.13 (2H, d, J 8, Ar-H), 6.75 - 6.73 (2H, d, J 8, Ar-H), 6.19 (1H, bd, J 3, Ar-NH), 5.15 (1H, bd, J 8, BocNHCH), 3.97 (1H, t, J 8, BocNHCH), 2.14 (1H, m, CHCH(CH3)2), 1.45 (9H, s, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.98-0.97\left(6 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ;$ бc (100 MHz, CDCl 3 ) 172.2 ( $C=O N H N H$ ), 156.2 ( $C=O N H$ ), 146.5 (ipso-Ar-C), 129.3 (Ar-C), 129.0 (Ar-C), 125.9 (ipso-Ar-C), 114.8 (Ar-C), 114.3 (Ar-C), 80.5 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 58.8$ (BocNHCH), $30.4\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 19.4\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 18.2\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right)$; $m / z\left(\mathrm{ES}^{+}\right) 364\left(\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right), 366\left(\left[{ }^{37} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}$, $364.1412\left(\mathrm{C}_{16} \mathrm{H}_{24}{ }^{35} \mathrm{CIN}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 364.1404$)$.

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## tert-Butyl (S)-(1-(2-(2,6-dichlorophenyl)hydrazinyl)-3-methyl-1-oxobutan-2-

 $y l) c a r b a m a t e ~ 67 A u$

Following the general procedure outlined, $N$-Boc- $L$-valine ( $0.20 \mathrm{~g}, 0.92 \mathrm{mmol}$ ) and 2,6-dichlorophenylhydrazine hydrochloride ( $0.24 \mathrm{~g}, 1.11 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound as a light brown solid ( 0.26 g, $75 \%$ ); Rf 0.43 (DCM/EtOH/NH3 200:6:1); m.p. $123-126{ }^{\circ} \mathrm{C}$; $v_{\max } 3328(\mathrm{~N}-\mathrm{H})$, 2975 (C-H), 1689, 1670 (C=O), 1521, 1449, 1300, 1244, 1168, 766, $654 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.45 (1H, bs, CONHNH), $7.25-7.22$ (2H, d, J 7, Ar-H), 6.93 (1H, t, J 7, Ar-H), 6.77 (1H, bd, J 4, Ar-NH), 5.07 (1H, bd, J 8, BocNHCH), 3.96 (1H, t, J 7, BocNHCH), 2.13 (1H, m, CHCH(CH3 $)_{2}$ ), 1.43 (9H, s, ( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.91$ - 0.86 ( $\left.6 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right)$; бc ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 171.1 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.8 (C=ONH), 140.9 (ipso-Ar-C), 129.0 (Ar-C), 128.7 (Ar-C), 126.1 (ipso-Ar-C), 124.1 (Ar-C), $80.2\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 58.3$ (BocNHCH), $30.7\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $19.2\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $17.8\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 398\left(\left[{ }^{35,35} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right), 400$ ( $\left[^{35,37} \mathrm{CI}\right] \mathrm{MNa}^{+}$), $402\left(\left[{ }^{37,37} \mathrm{CI}\right] \mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\left[{ }^{35,35} \mathrm{CI}\right] \mathrm{MNa}^{+}, 398.1015$ ( $\mathrm{C}_{16} \mathrm{H}_{23}{ }^{35,35} \mathrm{CIN}_{3} \mathrm{O}_{3} \mathrm{Na}$ requires 398.1014 ).

## tert-Butyl-(S)-(1-(2-(2,4-difluorophenyl)hydrazinyl)-3-methyl-1-oxobutan-2yl)carbamate 67 Ca



Following the general procedure outlined, $N$-Boc- $L$-valine ( $0.20 \mathrm{~g}, 0.92 \mathrm{mmol}$ ) and 2,4- difluorophenylhydrazine hydrochloride ( $0.20 \mathrm{~g}, 1.11 \mathrm{mmol}$ ) were transformed following work up with EtOAc into the title compound as an orange solid ( 0.21 g , $66 \%) ; \mathrm{R}_{\mathrm{f}} 0.52$ (DCM/EtOH/NH3 200:8:1); m.p. $125-128{ }^{\circ} \mathrm{C}$; $v_{\max } 3453,3273$ ( $\mathrm{N}-$ H), 2975 (C-H), 1689, 1613 (C=O), 1501, 1371, 1258, 1162, 958, $846 \mathrm{~cm}^{-1}$; ठн (400 MHz, DMSO-d6) 9.82 (1H, bd, J 2, CONHNH), 7.62 (1H, bs, Ar-NH), 7.10 (1H, dt, J 9, 2, Ar-H), 6.89-6.76 (2H, m, Ar-H), 3.73 (1H, dd, J 8, BocNHCH),

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$1.95\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.40\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.89-0.87(6 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right)$; $\delta c\left(100 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d}_{6}\right) 172.1$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.1 ( $\mathrm{C}=\mathrm{ONH}$ ), 134.2 (ipso-Ar-C), 114.7 (Ar-C), 111.0 (Ar-C), 110.8 (ipso-Ar-C), 104.3 (Ar-C), 104.1 (ipso-Ar-C), 78.6 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 59.4$ (BocNHCH), $30.1\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.6$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 19.7\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 19.2\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta$ ( 282 MHz , DMSO-d6) 124.1 (Ar-F), -128.4 (Ar-F); m/z (ES $\left.{ }^{+}\right) 366\left(\mathrm{MNa}^{+}\right) ;$HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}$, $366.1589\left(\mathrm{C}_{16} \mathrm{H}_{23} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~F}_{2} \mathrm{Na}\right.$ requires 366.1605).
tert-Butyl (S)-(1-(2-(2-bromophenyl)hydrazineyl)-3-methyl-1-oxobutan-2-yl) carbamate 67 Cg


Following the general procedure outlined, $N$-Boc- $L$-valine ( $0.20 \mathrm{~g}, 0.92 \mathrm{mmol}$ ) and 2-bromophenylhydrazine hydrochloride ( $0.23 \mathrm{~g}, 1.02 \mathrm{mmol}$ ) were transformed following flash chromatography (DCM/EtOH/NH3 [200:8:1] into the title compound as an orange solid ( $0.18 \mathrm{~g}, 51 \%$ ); Rf 0.52 (DCM/EtOH/NH3 200:8:1); m.p. 120-125 ${ }^{\circ} \mathrm{C}$; $v_{\max } 3318(\mathrm{~N}-\mathrm{H}), 2960$, 2870, 1678 (C=O), 1650 (C=O), 1525, 1481, 1301,1170, $744 \mathrm{~cm}^{-1}$; $\delta_{\mathrm{H}}\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.68$ (1H, m, CONHNH), 7.47 (1H, d, J 8, Ar-H), 7.14 (1H, t, J 8, Ar-H), 6.86 (1H, d, J 8, Ar-H), 6.75 (1H, t, J 8, Ar-H), 6.46 (1H, bd, J 4, Ar-NH), 5.30 (1H, bs, BocNHCH), 4.05 (1H, t, J 8, BocNHCH), $2.13\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.45\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.01(6 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right)$; $\delta \mathrm{c}\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 172.1$ ( $\left.\mathrm{C}=\mathrm{ONHNH}\right)$, 156.2 ( $\left.\mathrm{C}=\mathrm{ONH}\right), 144.7$ (ipso-Ar-C), 132.6 (Ar-C), 128.3 (Ar-C), 121.8 (ipso-Ar-C), 113.8 (Ar-C), 109.1 (Ar-C), $80.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 58.8(\mathrm{BocNHCH}), 30.5\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $\left.19.4\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 19.3\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; m / z\left(\mathrm{ES}^{+}\right) 386\left({ }^{[79} \mathrm{Br}\right] \mathrm{MH}^{+}\right), 388\left({ }^{81} \mathrm{Br}\right]$ $\left.\mathrm{MH}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\left[{ }^{79} \mathrm{Br}\right] \mathrm{MH}^{+}, 386.1107\left(\mathrm{C}_{16} \mathrm{H}_{24}{ }^{79} \mathrm{BrN}_{3} \mathrm{O}_{3}\right.$ requires 386.1079).

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tert-Butyl (S)-(1-(2-(4-chlorophenyl)hydrazinyl)-4-methyl-1-oxopentan-2-yl) carbamate 67Ap


Following the general procedure outlined, $N$-Boc-L-leucine ( $0.20 \mathrm{~g}, 0.87 \mathrm{mmol}$ ) and 4-chlorophenylhydrazine hydrochloride $(0.15 \mathrm{~g}, 1.04 \mathrm{mmol})$ were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a white solid ( $0.02 \mathrm{~g}, 8 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.58$ ( $\mathrm{DCM} / E t O H / \mathrm{NH}_{3}$ 200:8:1); m.p. 96 - $99^{\circ} \mathrm{C}$; $V_{\max } 3307$ (N-H), 2959 (C-H), 1652 (C=O), 1519, 1491, 1366, 1235, 1166, $822 \mathrm{~cm}^{-1}$; ठн (400 MHz, CDCl3) 8.23 (1H, bs, CONHNH), 7.16 - 7.14 (2H, d, J 8, Ar-H), 6.75 - 6.73 (2H, d, J 8, Ar-H), 6.10 (1H, bs, CONHNH), 4.93 (1H, bd, J 8, BocNHCH), $4.18(1 \mathrm{H}, \mathrm{m}, \mathrm{BocNHCH}), 1.72-1.66\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CHHCH}\left(\mathrm{CH}_{3}\right)_{2}\right.$, $\left.\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.54\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.47\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.97-0.95$ $\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) 0.94-0.92\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$; $\delta \mathrm{c}(100 \mathrm{MHz}$, $\mathrm{CDCl}_{3}$ ) 172.6 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 146.5 (ipso-Ar-C), 129.1 (Ar-C), 125.9 (ipso-Ar-C), 114.8 ( $\mathrm{Ar}-\mathrm{C}$ ), $80.7\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, 51.4 ( BocNHCH$), 40.2\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.3$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $24.7\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.8\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.1\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; m / z$ (ES $\left.\left.{ }^{+}\right) 356\left({ }^{[35} \mathrm{Cl}\right] \mathrm{MH}^{+}\right)$, $\left.358\left({ }^{37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}, 356.1694$ $\left(\mathrm{C}_{17} \mathrm{H}_{27} \mathrm{~N}_{3} \mathrm{O}_{3}{ }^{35} \mathrm{Cl}\right.$ requires 356.1741$)$.
tert-Butyl (S)-(1-(2-(3-chlorophenyl)hydrazinyl)-4-(methylthio)-1-oxobutan-

## 2-yl)carbamate 67w



Following the general procedure outlined, $N$-Boc-L-methionine ( $0.20 \mathrm{~g}, 0.80$ mmol ) and 3-chlorophenylhydrazine hydrochloride ( $0.13 \mathrm{~g}, 0.88 \mathrm{mmol}$ ) were transformed following flash chromatography (DCM/EtOH/NH3 400:8:1) and

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(DCM/EtOH/NH3 200:8:1) into the title compound which was isolated as a pale yellow solid ( $0.16 \mathrm{~g}, 53 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.30$ (DCM/EtOH/ $\mathrm{NH}_{3} 200: 8: 1$ ); ); m.p. $94-9{ }^{\circ} \mathrm{C}$; $V_{\max } 3331,3272$ (N-H), 2985 (C-H), 1683, 1658 (C=O), 1521, 1163, $768 \mathrm{~cm}^{-1}$; $\delta_{H}$ (700 MHz, CDCl 3 ) 8.73 (1H, bs, CONHNH), 7.10 (1H, t, J 8, Ar-H ), 6.85 (1H, d, J 8, Ar-H ), 6.80 (1H, s, Ar-H), 6.67 (1H, d, J 8, Ar-H), 6.34 (1H, bs, Ar-NH), 5.42 (1H, bd, J8, BocNHCH), 4.43 (1H, q, J 7, BocNHCH), 2.57 (2H, t, J7, CH2CH2S), 2.13 - $2.09\left(5 \mathrm{H}, \mathrm{m}, \mathrm{CHHCH} \mathrm{C}_{2} \mathrm{~S}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}, \mathrm{CHHCH} \mathrm{C}_{2} \mathrm{~S}\right), 1.47(9 \mathrm{H}, \mathrm{s}$, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta с\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 172.2$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.9 ( $\mathrm{C}=\mathrm{ONH}$ ), 149.0 (ipso-Ar-C), 134.9 (Ar-C), 130.2 (Ar-C), 121.0 ( $\mathrm{Ar}-\mathrm{C}$ ), 113.4 ( $\mathrm{Ar}-\mathrm{C}$ ), 111.6 ( $\mathrm{Ar}-\mathrm{C}$ ), $80.8\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 51.9(\mathrm{BocNHCH}), 31.0\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 30.2\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right)$, $28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 15.5\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right) ; m / z\left(\mathrm{ES}^{+}\right) 374\left(\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 376\left(\left[{ }^{37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right)$, $396\left(\left[{ }^{35} \mathrm{CI}\right] \mathrm{MNa}^{+}\right)$, $\left.398\left({ }^{37} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) found ( $\left.{ }^{35} \mathrm{CI}\right] \mathrm{MNa}^{+}$), 396.1141 $\left(\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}^{35} \mathrm{CINa}\right.$ requires 396.1125$)$.

## tert-Butyl (S)-(1-(2-(4-chlorophenyl)hydrazineyl)-4-(methylthio)-1-oxobutan-2-yl)carbamate 67 Ar



Following the general procedure outlined, N-Boc-L-methionine ( $0.20 \mathrm{~g}, 0.80$ mmol ) and 4-chlorophenylhydrazine hydrochloride ( $0.13 \mathrm{~g}, 0.88 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as an orange gum ( $0.07 \mathrm{~g}, 23 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.35$ ( $\mathrm{DCM} / \mathrm{EtOH} / \mathrm{NH}_{3} 200: 8: 1$ ); $v_{\text {max }}$ 3313, 3271 (N-H), 2977 (C-H), 1657 (C=O), 1519, 1489, 1305, 1249, 1159, 821 $\mathrm{cm}^{-1}$; ठн ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.58 (1H, bs, CONHNH), $7.15-7.13$ ( $2 \mathrm{H}, \mathrm{d}, \mathrm{J} 8$, ArH), $6.74-6.72$ (2H, d, J 8, Ar-H), 6.24 (1H, bs, Ar-NH), 5.34 (1H, bs, BocNHCH), 4.39 (1H, m, BocNHCH), 2.57 (2H, m, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~S}$ ), 2.13-2.06 (4H, m, CHHCH2S, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}$ ), $1.97(1 \mathrm{H}, \mathrm{m}, \mathrm{CHHCH} 2 \mathrm{~S}), 1.45\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta \mathrm{c}(100 \mathrm{MHz}$, $\mathrm{CDCl}_{3}$ ) 172.0 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.9 ( $\mathrm{C}=\mathrm{ONH}$ ), 146.4 (ipso-Ar-C), 129.4 (Ar-C), 129.1 (Ar-C), 125.9 (ipso-Ar-C), 114.8 (Ar-C), 114.2 ( $\mathrm{Ar}-\mathrm{C}$ ), $80.8\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $52.0(\mathrm{BocNHCH}), 30.9\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 30.2\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right)$, $28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$,

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$15.4\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 396\left(\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right), 398\left(\left[{ }^{37} \mathrm{CI}\right] \mathrm{MNa}^{+}\right) ;$HRMS (ES $\left.{ }^{+}\right)$ Found $\left[{ }^{35} \mathrm{CI}\right] \mathrm{MNa}^{+}, 396.1140\left(\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}^{35} \mathrm{CINa}\right.$ requires 396.1125$)$.

## tert-Butyl <br> (S)-(1-(2-(2,6-dichlorophenyl)hydrazinyl)-4-(methylthio)-1-oxobutan-2-yl)carbamate 67Aw



Following the general procedure outlined, $N$-Boc-L-methionine $(0.20 \mathrm{~g}, 0.80$ mmol ) and 2,6-dichlorophenylhydrazine hydrochloride ( $0.19 \mathrm{~g}, 0.88 \mathrm{mmol}$ ) were transformed following flash column chromatography ( $n$-hex/EtOAc 3:1) into the title compound which was isolated as a yellow gum ( $0.21 \mathrm{~g}, 65 \%$ ); Rf 0.40 ( n hex/EtOAc 3:1); m.p. $89-91^{\circ} \mathrm{C}$; $V_{\max } 3342,3286(\mathrm{~N}-\mathrm{H}), 2998,2967,2914(\mathrm{C}-\mathrm{H})$, 1673 (C=O), 1519, 1308, 1256, 1167, $767 \mathrm{~cm}^{-1}$; $\delta$ н ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $8.73(1 \mathrm{H}$, bs, CONHNH), 7.25 (1H, d, J 8, Ar-H), 6.90 (1H, t, J 8, Ar-H), 6.75 (1H, d, J 4, Ar-H), 5.26 (1H, bd, J 8, BocNHCH), 4.40 (1H, m, BocNHCH), $2.50(2 \mathrm{H}, \mathrm{t}, \mathrm{J} 8$, $\mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{~S}$ ), 2.14 - $2.03\left(4 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}, \mathrm{CHCHHCH}_{2} \mathrm{~S}\right), 1.91(1 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CHCHHCH}_{2} \mathrm{~S}\right), 1.44\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; ~ \delta c\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 170.9$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.7 ( $\mathrm{C}=\mathrm{ONH}$ ), 140.8 (ipso-Ar-C), 130.2 (Ar-C), 128.9 (Ar-C), 128.5 (Ar-C), 128.0 (Ar-C), 126.0 (Ar-C), 124.1 (Ar-C), 80.5 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 51.7$ (BocNHCH), 31.4 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 30.0\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 15.3\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right) ; \mathrm{m} / \mathrm{z}$ (ES ${ }^{+}$) $408\left(\left[{ }^{35,35} \mathrm{CI}\right] \mathrm{MH}^{+}\right), 410\left(\left[{ }^{35,37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 412\left(\left[{ }^{37,37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right)$; HRMS (ES $\left.{ }^{+}\right)$ Found $\left[{ }^{35,35} \mathrm{Cl}\right] \mathrm{MH}^{+}, 408.0916\left(\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}^{35,35} \mathrm{Cl}\right.$ requires 408.0915$)$.
tert-Butyl-(S)-(1-(2-(2,4-difluorophenyl)hydrazinyl)-4-(methylthio)-1-oxobutan-2-yl)carbamate 67Cc


Following the general procedure outlined, $N$-Boc-L-methionine $(0.20 \mathrm{~g}, 0.80$ mmol ) and 2,4-difluorophenylhydrazine hydrochloride ( $0.16 \mathrm{~g}, 0.88 \mathrm{mmol}$ ) were

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transformed following flash column chromatography ( $n$-hex/EtOAc 3:1) into the title compound which was isolated as an orange gummy solid ( $0.29 \mathrm{~g}, 95 \%$ ); $\mathrm{R}_{\mathrm{f}}$ 0.43 (DCM/EtOH/NH3 200:8:1); m.p. $92-95^{\circ} \mathrm{C}$; $v_{\max } 3317$ (N-H), 2979 (C-H), 1665, 1611 (C=O), 1507, 1369, 1254, 1164, 958, 834 cm$^{-1} ; \delta_{\text {н }}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ 8.40 (1H, bs, CONHNH), 6.81 - 6.69 (2H, m, Ar-H), 6.65 (1H, t, J 8, Ar-H), 5.18 (1H, bs, BocNHCH), 4.33 (1H, q, J 7, BocNHCH), 2.51 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{~S}$ ), 2.08 - $2.02\left(4 \mathrm{H}, \mathrm{m}, \mathrm{CHCHHCH}_{2} \mathrm{~S}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 1.93(1 \mathrm{H}, \mathrm{m}, \mathrm{CHCHHCH} 2 \mathrm{~S})$, 1.39 ( $\left.9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 171.9$ ( $\left.\mathrm{C}=\mathrm{ONHNH}\right), 155.9$ (C=ONH), 149.9 (ipso-Ar-C), 132.3 (ipso-Ar-C), 115.4 (Ar-C), 111.0 (Ar-C), 110.9 (ipso-Ar-C), $104.2 \quad(\mathrm{Ar}-\mathrm{C}), \quad 80.8 \quad\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), \quad 51.9 \quad(\mathrm{BocNHCH}), 30.8$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 30.2\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 15.4\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right) ; \delta{ }_{F}$ (282 MHz, CDCl3) -120.5 (Ar-F), -129.0 (Ar-F); m/z (ES ${ }^{+}$) 398 (MNa+); HRMS $\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}, 398.1348\left(\mathrm{C}_{16} \mathrm{H}_{23} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~F}_{2} \mathrm{SNa}\right.$ requires 398.1326).

## tert-Butyl (S)-(1-(2-(4-cyanophenyl)hydrazinyl)-4-(methylthio)-1-oxobutan-2-yl)carbamate 67Da



Following the general procedure outlined, $N$-Boc-L-methionine $(0.20 \mathrm{~g}, 0.80$ mmol ) and 4-cyanophenylhydrazine hydrochloride ( $0.15 \mathrm{~g}, 0.88 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow gummy solid ( $0.10 \mathrm{~g}, 35 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.41$ ( $\mathrm{DCM} / E t O H / \mathrm{NH}_{3} 200: 8: 1$ ); m.p. 57-60 ${ }^{\circ} \mathrm{C}$; $V_{\max } 3264$ (N-H), 2979, 2218 (C-H), 2112 (C=N), 1673, 1605 (C=O), 1505, 1366. 1249, 1161, 1047, 831, $545 \mathrm{~cm}^{-1}$; ठн ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.39 (1H, bs, CONHNH), 7.49 - 7.47 (2H, d, J 8, Ar-H), $6.83-6.81$ (2H, d, J 8, Ar-H), 6.38 (1H, bs, Ar-NH), 5.19 (1H, bd, J 7, BocNHCH), 4.37 (1H, q, J7, BocNHCH), $2.59\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right), 2.14-2.09\left(4 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}, \mathrm{CHCHHCH}_{2} \mathrm{~S}\right)$, $2.00(1 \mathrm{H}, \mathrm{m}, \mathrm{CHCHHCH} 2 \mathrm{~S}), 1.48\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta \mathrm{c}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 171.9$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.1 ( $\mathrm{C}=\mathrm{ONH}$ ), 151.3 (Ar-C), 133.7 (ipso-Ar-C), 119.4 ( $C=N$ ), 113.1 (Ar-C), 103.5 (Ar-C), $81.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 51.9$ (BocNHCH), 30.3 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 30.2\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 15.4\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right) ; \mathrm{m} / \mathrm{z}$

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$\left(\mathrm{ES}^{+}\right) 365\left(\mathrm{MH}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\mathrm{MH}^{+}, 365.1653\left(\mathrm{C}_{17} \mathrm{H}_{25} \mathrm{~N}_{4} \mathrm{O}_{3} \mathrm{~S}\right.$ requires 365.1647).
tert-Butyl (S)-(4-(methylthio)-1-oxo-1-(2-(4-(trifluoromethyl)phenyl) hydrazineyl)butan-2-yl)carbamate 67Bo


Following the general procedure outlined, $N$-Boc-L-methionine $(0.20 \mathrm{~g}, 0.80$ mmol ) and 4-trifluoromethylphenylhydrazine hydrochloride ( $0.19 \mathrm{~g}, 0.88 \mathrm{mmol}$ ) were transformed following flash column chromatography (DCM/EtOH/NH3 200:6:1) into the title compound which was isolated as a pale yellow solid ( 0.14 g, 44\%); Rf 0.32 (DCM/EtOH/NH3 200:6:1); m.p. $102-108{ }^{\circ} \mathrm{C}$; $v_{\max } 3321$ (N-H), 2984 (C-H), 1680 (C=O), 1662(C=O) 1515, 1327, 1306, 1156, 1102, 830, 589 $\mathrm{cm}^{-1}$; ठн (700 MHz, CDCl3) 8.82 (1H, bs, CONHNH), 7.41 (2H, d, J 8, Ar-H), 6.81 (2H, d, J 8, Ar-H), 6.53 (1H, bs, Ar-NH), 5.43 (1H, bd, J 8, BocNHCH), 4.44 (1H, q, J 7, BocNHCH), $2.56\left(2 \mathrm{H}, \mathrm{t}, \mathrm{J} 7, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right), 2.09-2.14(5 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right)$, $1.49\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta c\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 172.4$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.1 ( $\mathrm{C}=\mathrm{ONH}$ ), 150.5 (ipso-Ar-C), 133.0 (Ar-CF3), 126.3 (1C, t, J 5, ipso-Ar-CCF 3 ), 125.2 (Ar-C), 120.0 (Ar-C), 123.7 (Ar-C), 112.7 (Ar-C), 80.9 $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 51.9(\mathrm{BocNHCH}), 30.9\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 30.2\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 28.3$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 15.4\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right) ; \delta_{\mathrm{F}}\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)-61.8\left(\mathrm{CF}_{3}\right) ; m / z\left(\mathrm{ES}^{+}\right)$ $408\left(\mathrm{MH}^{+}\right)$; HRMS $\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MH}^{+}, 408.1573\left(\mathrm{C}_{17} \mathrm{H}_{25} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~F}_{3} \mathrm{~S}\right.$ requires 408.1569)

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## tert-Butyl (S)-(4-(methylthio)-1-oxo-1-(2-(p-tolyl)hydrazineyl)butan-2-yl) carbamate 67Ee



Following the general procedure outlined, $N$-Boc-L-methionine ( $0.20 \mathrm{~g}, 0.80$ mmol ) and 4-methylphenylhydrazine hydrochloride ( $0.19 \mathrm{~g}, 0.88 \mathrm{mmol}$ ) were transformed following flash column chromatography (DCM/EtOH/NH3 200:6:1) into the title compound which was isolated as a reddish brown solid ( $0.16 \mathrm{~g}, 54 \%$ ); Rf 0.26 (DCM/EtOH/NH3 200:6:1); m.p. $96-99^{\circ} \mathrm{C}$; $v_{\max } 3324$ (N-H); 3201 (N-H), 2967 (C-H), 2919, 1682 (C=O), 1657 (C=O), 1513, 1290, 1162, $641 \mathrm{~cm}^{-1}$; ठн $^{(500}$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.60$ (1H, bs, CONHNH), 7.02 (2H, d, J 8, Ar-H), 6.73 (2H, d, J 8, Ar-H), 6.12 (1H, bs, Ar-NH), 5.41 (1H, bs, BocNHCH), 4.39 (1H, m, BocNHCH), $2.54\left(2 \mathrm{H}, \mathrm{t}, \mathrm{J} 7, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right), 2.26\left(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{CH}_{3}\right), 2.09-2.11(5 \mathrm{H}, \mathrm{m}$, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~S}$ ), 1.48 ( $9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}$ ); ठс ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 171.8 ( $C=O N H N H$ ), 155.8 ( $C=O N H$ ), 144.4 (ipso-Ar-C), 130.6 (Ar-C), 129.7 (Ar-C), 113.8 (Ar-C), $80.6\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 52.0(\mathrm{BocNHCH}), 31.3\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 30.2$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 28.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 20.6\left(\mathrm{Ar}-\mathrm{CH}_{3}\right), 15.4\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right)$ $354\left(\mathrm{MH}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\mathrm{MH}^{+}, 354.1860\left(\mathrm{C}_{17} \mathrm{H}_{28} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}\right.$ requires 354.1851).
tert-Butyl (S)-(1-(2-(3-chlorophenyl)hydrazinyl)-1-oxo-3-phenylpropan-2-yl) carbamate 67x


Following the general procedure outlined, $N$-Boc-L-phenylalanine $(0.20 \mathrm{~g}, 0.75$ mmol ) and 3-chlorophenylhydrazine hydrochloride ( $0.16 \mathrm{~g}, 0.91 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a light brown solid ( $0.10 \mathrm{~g}, 34 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.79$ (DCM/EtOH/NH3 200:8:1]); m.p. 148-150 ${ }^{\circ} \mathrm{C}$; $v_{\max } 3339,3238(\mathrm{~N}-\mathrm{H}), 2986,1694$ (C=O), 1674 (C=O), 1596, 1525, 1495, 1492, 1249, 1047, $862 \mathrm{~cm}^{-1}$; $\delta \mathrm{H}\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.83$ (1H, bs, NH),
$7.34-7.31$ (2H, m, CH2Ar-H), 7.29 (1H, m, Ar-H), 7.22 - 7.20 (2H, m, Ar-H), 7.06 (1H, t, J 8, Ar-H), 6.83 (1H, d, J 8, NHAr-H), 6.65 (1H, s, Ar-H), 6.48 (1H, d, J 8, Ar-H), 5.05 (1H, s, BocNHCH), 4.42 (1H, dt, J 8, BocNHCH), 3.09 ( $2 \mathrm{H}, \mathrm{d}, \mathrm{J} 7$, $\left.\mathrm{CH}_{2} \mathrm{Ph}\right), 1.44\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta \mathrm{c}\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 171.6(\mathrm{C}=\mathrm{ONHNH}), 148.8$ ( $\mathrm{C}=\mathrm{ONH}$ ), 136.0 ( $\mathrm{Ar}-\mathrm{C}$ ), 134.9 ( $\mathrm{Ar}-\mathrm{C}$ ), 130.1 ( $\mathrm{Ar}-\mathrm{C}$ ), 129.3 ( $\mathrm{Ar}-\mathrm{C}$ ), 128.9 ( $\mathrm{Ar}-\mathrm{C}$ ), 127.3 (Ar-C), 121.2 (Ar-C), 113.4 (Ar-C), 111.7 (Ar-C), $\left.81.0\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 54.5$ (BocNHCH), $\left.37.8\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 28.3\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 390\left(\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 392$ $\left(\left[{ }^{37} \mathrm{CI}\right] \mathrm{MH}^{+}\right)$; HRMS $\left(\mathrm{ES}^{+}\right)$Found $\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}, 390.1579\left(\mathrm{C}_{20} \mathrm{H}_{25}{ }^{35} \mathrm{CIN}_{3} \mathrm{O}_{3}\right.$ requires 390.1584).

## tert-Butyl-(S)-(1-(2-(4-chlorophenyl)hydrazinyl)-1-oxo-3-phenylpropan-2yl)carbamate 67 Ar



Following the general procedure outlined, $N$-Boc-L-phenylalanine ( $0.20 \mathrm{~g}, 0.75$ mmol ) and 4-chlorophenylhydrazine hydrochloride ( $0.16 \mathrm{~g}, 0.91 \mathrm{mmol}$ ) were transformed following work up with EtOAc into the title compound which was isolated as a yellow solid ( $0.27 \mathrm{~g}, 91 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.31$ ( $\mathrm{DCM} / E t O H / \mathrm{NH}_{3}$ 200:8:1); m.p. 139-141 ${ }^{\circ} \mathrm{C} ; v_{\max } 3328,3246$ (N-H), 2976 (C-H), 1662 (C=O), 1518, 1489, 1296, 1165, $622 \mathrm{~cm}^{-1}$; ठн ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.02 (1H, bs, CONHNH), $7.31-7.28(3 \mathrm{H}$, m, Ar-H), 7.21 - 7.19 (2H, d, J7, Ar-H), 7.08 - 7.06 (2H, d, J 8, Ar-H), $6.47-6.45$ (2H, d, J 8, Ar-H), 6.03(1H, bs, Ar-NH), 5.16 (1H, bd, J 8, BocNHCH), 4.45 (1H, m, BocNHCH), 3.08 (2H, d, J 7, Ar-CH2), $1.43\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta c(100 \mathrm{MHz}$, $\mathrm{CDCl}_{3}$ ) 171.6 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.6 ( $\mathrm{C}=\mathrm{ONH}$ ), 146.1 (ipso-Ar-C), 136.1 (ipso-ArC), 129.5 (Ar-C), 129.4 (Ar-C), 129.3 ( $\mathrm{Ar}-\mathrm{C}$ ), 128.9 ( $\mathrm{Ar}-\mathrm{C}$ ), 128.8 ( $\mathrm{Ar}-\mathrm{C}$ ), 128.7 ( $\mathrm{Ar}-\mathrm{C}$ ), 128.6 ( $\mathrm{Ar}-\mathrm{C}$ ), 127.2 ( $\mathrm{Ar}-\mathrm{C}$ ), 125.9 (ipso-Ar-C), 114.8 ( $\mathrm{Ar}-\mathrm{C}$ ), 80.8 $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 54.5(\mathrm{BocNHCH}), 37.9\left(\mathrm{Ar}-\mathrm{CH}_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 412$ $\left.\left(\left[{ }^{35} \mathrm{CI}\right] \mathrm{MNa}^{+}\right), 414\left({ }^{37} \mathrm{C}\right] \mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\left.{ }^{35} \mathrm{CI}\right] \mathrm{MNa}^{+}, 412.1395$ ( $\mathrm{C}_{20} \mathrm{H}_{24} \mathrm{~N}_{3} \mathrm{O}_{3}{ }^{35} \mathrm{CINa}$ requires 412.1404 ).

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## tert-Butyl (S)-(1-(2-(2,6-dichlorophenyl)hydrazinyl)-1-oxo-3-phenylpropan-

## 2-yl)carbamate 67Ax



Following the general procedure outlined, $N$-Boc-L-phenylalanine ( $0.20 \mathrm{~g}, 0.75$ mmol ) and 2,6-dichlorophenylhydrazine hydrochloride ( $0.19 \mathrm{~g}, 0.91 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a pale yellow solid ( $0.11 \mathrm{~g}, 34 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.27$ (DCM/EtOH/NH3 200:8:1); m.p. $147-150^{\circ} \mathrm{C}$; $v_{\max } 3337$, 3303 (N-H), 2967 (C-H), 1691, 1669 (C=O), 1525, $1495,1454,1292,1248,1168,764,698 \mathrm{~cm}^{-1}$; $\delta_{H}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.17(1 \mathrm{H}, \mathrm{bs}$, CONHNH), 7.23 - 7.21 (5H, m, Ar-H), 7.12-7.10 (2H, m, Ar-H), 6.92 (1H, t, J 8, Ar-H), 6.75 (1H, bd, Ar-NH), 4.91 (1H, bs, BocNHCH), 4.39 (1H, m, BocNHCH), $3.03\left(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{CH}_{2}\right), 1.40\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta с\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 170.3$ (C=ONHNH), 155.4 ( $\mathrm{C}=\mathrm{ONH}$ ), 140.7 (ipso-Ar-C), 136.1 (ipso-Ar-C), 129.2 (Ar-C), 128.8 (Ar-C), 128.7 (Ar-C), 127.3 (Ar-C), 126.3 (Ar-C), 124.2 (Ar-C), 80.6 $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 54.3(\mathrm{BocNHCH}), 37.9\left(\mathrm{Ar}-\mathrm{CH}_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; m / z\left(\mathrm{ES}^{+}\right) 424$ ( $\left.{ }^{35,35} \mathrm{CI}\right] \mathrm{MH}^{+}$), 426 ( $\left[^{35,37} \mathrm{Cl}\right] \mathrm{MH}^{+}$), 428 ( $\left.{ }^{37,37} \mathrm{Cl}\right] \mathrm{MH}^{+}$); HRMS (ES ${ }^{+}$) Found $\left[{ }^{35,35} \mathrm{Cl}\right] \mathrm{MH}^{+}, 424.1199\left(\mathrm{C}_{20} \mathrm{H}_{24} \mathrm{~N}_{3} \mathrm{O}_{3}{ }^{35,35} \mathrm{C}\right.$ requires 424.1195).

## tert-Butyl (S)-(1-(2-(2,4-difluorophenyl)hydrazinyl)-1-oxo-3-phenylpropan-2yl)carbamate 67Cd



Following the general procedure outlined, $N$-Boc-L-phenylalanine ( $0.20 \mathrm{~g}, 0.75$ mmol ) and 2,4-difluorophenylhydrazine hydrochloride ( $0.16 \mathrm{~g}, 0.91 \mathrm{mmol}$ ) were transformed following flash column chromatography (DCM/EtOH/NH3 400:8:1) into the title compound which was isolated as a light orange solid ( $0.13 \mathrm{~g}, 45 \%$ ); $\mathrm{R}_{\mathrm{f}} 0.35$ (DCM/EtOH/ $\mathrm{NH}_{3} 400: 8: 1$ ); m.p. 132 - $138{ }^{\circ} \mathrm{C}$; $v_{\max } 3325$ (N-H), 2974 (C-

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H), 1667, 1523, 1367, 1291, 1166, 960, 850, $703 \mathrm{~cm}^{-1} ;$ бн ( $\left.500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.84$ (1H, bs, CONHNH), $7.33-7.28$ (3H, m, Ar-H), 7.21-7.20 (2H, d, J 6, Ar-H), 6.78 (1H, dt, J 8, 2, Ar-H), 6.61 (1H, t, J8, Ar-H), 6.40 (1H, m, Ar-H), 6.03 (1H, bs, ArNH), 5.07 (1H, bs, BocNHCH), 4.42 (1H, q, J7, BocNHCH), 3.09 ( $2 \mathrm{H}, \mathrm{d}, ~ J 7, \mathrm{Ar}-$ $\mathrm{CH}_{2}$ ), $1.43\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta \mathrm{c}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 171.5$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.6 (C=ONH), 136.0 (ipso-Ar-C), 131.9 (ipso-Ar-C), 129.3 (Ar-C), 128.9 (Ar-C), 127.2 (Ar-C), 115.4 (Ar-C), 110.9 (Ar-C, dd, J 22, 4), 103.8 (Ar-C, dd, J 26, 22), 80.8 $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 54.6$ (BocNHCH), $37.8\left(\mathrm{Ar}-\mathrm{CH}_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta \mathrm{F}(500 \mathrm{MHz}$, $\mathrm{CDCl}_{3}$ ) -120.7 (Ar-F), -129.2 (Ar-F); m/z (ES ${ }^{+}$) $392\left(\mathrm{MH}^{+}\right) ;$HRMS (ES ${ }^{+}$) Found $\mathrm{MH}^{+}, 392.1785\left(\mathrm{C}_{20} \mathrm{H}_{24} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~F}_{2}\right.$ requires 392.1786).

## tert-Butyl (S)-(1-(2-(4-isopropylphenyl)hydrazineyl)-1-oxo-3-phenylpropan-2-yl)carbamate 67EI



Following the general procedure outlined, $N$-Boc-L-phenylalanine ( $0.30 \mathrm{~g}, 1.13$ mmol ) and 4-isopropylphenylhydrazine hydrochloride ( $0.23 \mathrm{~g}, 1.24 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a light brown solid ( $0.27 \mathrm{~g}, 60 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.51$ (DCM/EtOH/NH3 200:8:1); m.p. $141-143^{\circ} \mathrm{C}$; $v_{\max } 3319$ (N-H), 2962 (C-H), 1688, 1656 (C=O), 1516, 1498, 1365, 1252, 1171, $825 \mathrm{~cm}^{-1}$; $\delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.72$ (1H, bs, CONHNH), 7.33 - 7.30 (3H, m, Ar-H), 7.23 - 7.20 (2H, dd, J 7, 1, Ar-H), 7.03 - 7.00 (2H, d, J 8, Ar-H), $6.58-6.55(2 H, d, J 8, A r-H), 5.04(1 H, b s, B o c N H C H), 4.45(1 H, q, J 7$, BocNHCH), 3.10 (2H, d, J 7, Ar-CH2), $2.86\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.44(9 \mathrm{H}, \mathrm{s}$, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.20-1.18\left(6 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 171.6$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.6 ( $\mathrm{C}=\mathrm{ONH}$ ), 145.4 (ipso-Ar-C), 136.4 (ipso-Ar-C), 129.4 (Ar-C), 128.7 ( $\mathrm{Ar}-\mathrm{C}$ ), 128.6 ( $\mathrm{Ar}-\mathrm{C}$ ), 128.4 ( $\mathrm{Ar}-\mathrm{C}$ ), 127.0 ( $\mathrm{Ar}-\mathrm{C}$ ), 126.9 ( $\mathrm{Ar}-\mathrm{C}$ ), 113.7 ( $\mathrm{Ar}-$ C), $80.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 53.5(\mathrm{BocNHCH}), 38.4\left(\mathrm{Ar}-\mathrm{CH}_{2}\right), 33.3\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.3$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $24.2\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 420\left(\mathrm{MNa}^{+}\right) ;$HRMS $\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}$, $420.2262\left(\mathrm{C}_{23} \mathrm{H}_{31} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 420.2258$)$.

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tert-Butyl
((2S,3S)-1-(2-(3-chlorophenyl)hydrazineyl)-3-methyl-1-oxopentan-2-yl)carbamate 67u


Following the general procedure outlined, $N$-Boc- $L$-isoleucine ( $0.20 \mathrm{~g}, 0.87 \mathrm{mmol}$ ) and 3-chlorophenylhydrazine hydrochloride ( $0.19 \mathrm{~g}, 1.04 \mathrm{mmol}$ ) were transformed following work up with EtOAc into the title compound which was isolated as a pale yellow solid ( $0.21 \mathrm{~g}, 68 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.61$ (DCM/EtOH/NH3 200:8:1); m.p. $117-119{ }^{\circ} \mathrm{C}$; $v_{\max } 3326,3272$ (N-H), 2964 (C-H), 1682, 1657 (C=O), 1519, 1470, 1243, 1168, $775 \mathrm{~cm}^{-1}$; ठн ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.43 (1H, bs, CONHNH), 7.09 (1H, t, J 7, Ar-H), 6.84 (1H, d, J 7, Ar-H), 6.80 (1H, d, J 2, Ar-H), 6.69 (1H, d, J 7, Ar-H), 5.16 (1H, bd, J 8, BocNHCH), 4.01 (1H, t, J 8, BocNHCH), 1.88 (1H, m, $\mathrm{NHCHCH}), 1.61\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHCHHCH}_{3}\right), 1.45\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.16(1 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CHHCH}_{3}\right), 0.96\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 0.92(3 \mathrm{H}, \mathrm{t}, \mathrm{J} 7$, $\left.\mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$; $\delta с$ ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 172.3 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.1 ( $\mathrm{C}=\mathrm{ONH}$ ), 149.2 (ipso-Ar-C), 135.0 (ipso-Ar-C), 130.2 (Ar-C), 121.1 (Ar-C), 113.6 (Ar-C), 111.8 ( $\mathrm{Ar}-\mathrm{C}$ ), $80.6\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 57.8$ (BocNHCH), 36.6 (NHCHCH), 28.3 $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), \quad 24.8 \quad\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{2} \mathrm{CH}_{3}\right), \quad 15.6 \quad\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{2} \mathrm{CH}_{3}\right), \quad 11.1$ $\left.\left.\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 378\left({ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right), 380\left({ }^{37} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right) ;$HRMS (ES $\left.{ }^{+}\right)$ Found $\left[{ }^{35} \mathrm{CI}\right] \mathrm{MNa}^{+}, 378.1539\left(\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{~N}_{3} \mathrm{O}_{3}{ }^{35} \mathrm{CINa}\right.$ requires 378.1560).
(9H-fluoren-9-yl)methyl tert-Butyl (6-(2-(3-chlorophenyl)hydrazinyl)-6-oxohexane-1,5-diyl)(S)-dicarbamate 68r


Following the general procedure outlined, $N$-Fmoc-N6-(tert-butoxycarbonyl)-Llysine ( $0.20 \mathrm{~g}, 0.43 \mathrm{mmol}$ ) and 3-chlorophenylhydrazine hydrochloride ( 0.09 g , 0.51 mmol ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a pale yellow solid ( $0.24 \mathrm{~g}, 93 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.50$ ( $n-$

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hexane/EtOAc 4:1); m.p. $145-148^{\circ} \mathrm{C}$; $v_{\max } 3288(\mathrm{~N}-\mathrm{H}), 2939(\mathrm{C}-\mathrm{H}), 1683,1647$ (C=O), 1532, 1478, 1271, 1250, 1165, $737 \mathrm{~cm}^{-1}$; ठн ( $399 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.39 (1H, bs, CONHNH), $7.76-7.74$ (2H, d, J 7, Ar-H), $7.57-7.55$ (2H, d, J 7, Ar-H), 7.41 - 7.36 (2H, t, J 7, Ar-H), 7.31 - 7.28 (2H, d, J 7, Ar-H), 7.10 (1H, t, J 8, Ar-H), $6.84-6.77$ (2H, m, Ar-H), 6.66 (1H, d, J 8, Ar-H), 6.18 (1H, bs, Ar-NH), 5.64 (1H, bs, FmocNHCH), 4.67 (1H, m, $\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}$-Fluorene), $4.47\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}-\right.$ Fluorene), 4.22 - $4.17\left(2 \mathrm{H}, \mathrm{m}, \mathrm{FmocNHCH}, \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene), $3.10(2 \mathrm{H}, \mathrm{m}$, $\mathrm{BocNHCH}_{2}$ ), 1.86 (1H, m, FmocNHCHCHH), 1.72 - 1.60 ( 3H, m, FmocNHCHCCHH, $\mathrm{BocNHCH}_{2} \mathrm{CH}_{2}$ ), $1.50\left(2 \mathrm{H}, \mathrm{m}, \mathrm{NHCHCH}_{2} \mathrm{CH}_{2}\right), 1.42(9 \mathrm{H}, \mathrm{s}$, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; бс (100 MHz, CDCl 3 ) 172.0 (C=ONHNH), 156.4 (C=ONH), 149.1 (ipso-Ar-C), 143.6 (ipso-Ar-C), 141.3 (ipso-Ar-C), 134.9 (ipso-Ar-C), 130.2 (Ar-C), 127.8 (Ar-C), 127.1 (Ar-C), 124.9 (Ar-C), 121.1 (Ar-C), 120.0 (Ar-C), 113.5 (ArC), 111.7 (Ar-C), $79.6 \quad\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 67.2 \quad\left(\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene), 53.1 (FmocNHCH), 47.2 (C-Fluorene), $39.8\left(\mathrm{BocNHCH}_{2}\right), 31.4$ ( $\mathrm{FmocNHCHCH}_{2}$ ), $29.6\left(\mathrm{BocNHCH}_{2} \mathrm{CH}_{2}\right), 28.6\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 22.5\left(\mathrm{NHCHCH}_{2} \mathrm{CH}_{2}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 615$ $\left(\left[{ }^{35} \mathrm{C}\right] \mathrm{MNa}^{+}\right)$, 617 ( $\left.\left.{ }^{37} \mathrm{C}\right]\right] \mathrm{MNa}^{+}$); HRMS (ES ${ }^{+}$) Found $\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}, 615.2349$ $\left(\mathrm{C}_{32} \mathrm{H}_{37}{ }^{35} \mathrm{CIN}_{4} \mathrm{O}_{5} \mathrm{Na}\right.$ requires 615.2350$)$.

## tert-Butyl(S)-4-((( 9 H -fluoren-9-yl)methoxy)carbonyl)amino)-5-(2-(3-chlorophenyl)hydrazinyl)-5-oxopentanoate 68m



Following the general procedure outlined, $N$-Fmoc-O-(tert-butyl)-L-aspartic acid ( $0.30 \mathrm{~g}, 0.73 \mathrm{mmol}$ ) and 3-chlorophenylhydrazine hydrochloride ( $0.14 \mathrm{~g}, 0.80$ $\mathrm{mmol})$ were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a white solid ( $0.22 \mathrm{~g}, 57 \%$ ); Rf 0.72 ( n -hex/EtOAc 4:1); m.p. 131-133 ${ }^{\circ} \mathrm{C} ;$ v $\max 3284$ (N-H), 1692, 1667 (C=O), 1543, 1478, 1269, 1165, 1049, 737 cm-1; бн (300 MHz, DMSO-d6) 9.94 (1H, bd, J 2, CONHNH), 8.05 (1H, bd, J 2, CONHCH), 7.91 - 7.89 (2H, d, J 7, Ar-H), 7.75 - 7.71 (2H, m, Ar-H), 7.45 7.40 (2H, td, J 7,1, Ar-H), 7.36 - 7.30 (2H, td, J 7, 1, Ar-H), 7.13 (1H, t, J 7, Ar-

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$H), 6.85(1 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-H), 6.76(1 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-H), 6.71(1 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-H), 4.49(1 \mathrm{H}, \mathrm{m}$, FmocNHCH), $4.35\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene), $4.28\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}-\right.$ Fluorene), $2.77\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHCHHCO}_{2}\right), 2.59(1 \mathrm{H}, \mathrm{m}, \mathrm{CHCHHCO} 2), 1.41(9 \mathrm{H}, \mathrm{s}$, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta с$ ( 100 MHz , DMSO-d6) 170.8 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.5 ( $\left.\mathrm{C}=\mathrm{ONH}\right), 148.9$ (ipso-Ar-C), 143.6 (ipso-Ar-C), 141.4 (ipso-Ar-C), 135.0 (ipso-Ar-C), 130.2 (Ar-C), 127.9 ( $\mathrm{Ar}-\mathrm{C}$ ), 127.2 ( $\mathrm{Ar}-\mathrm{C}$ ), 124.9 ( $\mathrm{Ar}-\mathrm{C}$ ), 121.2 ( $\mathrm{Ar}-\mathrm{C}$ ), 120.1 ( $\mathrm{Ar}-\mathrm{C}$ ), 113.5 ( $\mathrm{Ar}-$ C), 111.8 ( $\mathrm{Ar}-\mathrm{C}$ ), $82.3 \quad\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 67.3 \quad\left(\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene), 50.2 (FmocNHCH), 47.2 (C-Fluorene), $36.9\left(\mathrm{CHCH}_{2} \mathrm{CO}_{2}\right), 28.1\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right)$ $\left.536\left(\left[{ }^{35} \mathrm{CI}\right] \mathrm{MH}^{+}\right), 538\left({ }^{[37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right)$, $\left.558\left({ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right), 560\left(\left[{ }^{37} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right) ;$HRMS (ES $\left.{ }^{+}\right)$ Found $\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}, 536.1978\left(\mathrm{C}_{29} \mathrm{H}_{32}{ }^{35} \mathrm{CIN}_{3} \mathrm{O}_{5}\right.$ requires 536.1947).
(9H-fluoren-9-yl)methyl (S)-(3-(tert-butoxy)-1-(2-(3-chlorophenyl)hydrazinyl) -1-oxopropan-2-yl)carbamate 68k


Following the general procedure outlined, N -Fmoc- O -(tert-butyl)-L-serine ( 0.30 g , 0.78 mmol ) and 3-chlorophenylhydrazine hydrochloride ( $0.15 \mathrm{~g}, 0.86 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as an orange solid ( $0.29 \mathrm{~g}, 72 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.83$ ( $n$-hexane/EtOAc 4:1); m.p. 141-144 ${ }^{\circ} \mathrm{C}$; $V_{\max } 3325,3273(\mathrm{~N}-\mathrm{H}), 1687,1667$ (C=O), 1531, 1296, 1035, 756, $737 \mathrm{~cm}^{-1}$; ठн (399 MHz, CDCl $)^{2} 8.12$ (1H, bs, CONHNH), $7.77-7.75$ (2H, d, J 7, Ar-H), $7.59-7.58$ (2H, d, J 7, Ar-H), $7.43-7.37$ (2H, q, J7, 5, Ar-H), $7.33-7.29$ (2H, m, Ar-H), 7.12 (1H, t, J 7, Ar-H), $6.87-6.84(2 H, m, A r-H), 6.73$ (1H, d, J 7, Ar-H), 6.10 (1H, bs, Ar-NH), 5.68 (1H, bd, J 5, FmocNHCH), 4.49 ( $2 \mathrm{H}, \mathrm{m}$, $\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}$-Fluorene), 4.36 (1H, m, FmocNHCH), 4.23 (1H, t, J 6, $\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}-$ Fluorene), 3.84 (1H, m, FmocNHCHCHH), 3.49 (1H, m, FmocNHCHCHH), 1.23 (9H, s, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; \delta c\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 170.7(\mathrm{C}=\mathrm{ONHNH}), 156.1$ ( $\mathrm{C}=\mathrm{ONH}$ ), 148.9 (ipso-Ar-C), 143.7 (ipso-Ar-C), 141.4 (ipso-Ar-C), 135.1 (ipso-Ar-C), 130.2 (Ar-C), 127.8 (Ar-C), 127.1 (Ar-C), 124.9 (Ar-C), 121.2 (Ar-C), 120.1 (Ar-C), 113.5 (Ar-C), 111.7 (Ar-C), $74.6\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 67.2\left(\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene), 61.5 (FmocNHCHCH2), 54.1 (FmocNHCH), 47.2 (C-Fluorene), 27.5 ((CH33)3CO); m/z

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$\left(E S^{+}\right) 530\left(\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right)$, $\left.532\left({ }^{37} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}$, $530.1819\left(\mathrm{C}_{28} \mathrm{H}_{30} \mathrm{~N}_{3} \mathrm{O}_{4}{ }^{35} \mathrm{CINa}\right.$ requires 530.1823$)$.
(9H-fluoren-9-yl)methyl (S)-(1-(2-(3-chlorophenyl)hydrazinyl)-1,5-dioxo-5-(tritylamino)pentan-2-yl)carbamate 68q


Following the general procedure outlined, $N$-Fmoc- $N$-(trityl)-L-glutamine ( 0.30 g , 0.49 mmol ) and 3-chlorophenylhydrazine hydrochloride ( $0.09 \mathrm{~g}, 0.54 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow solid ( $0.31 \mathrm{~g}, 85 \%$ ); Rf 0.69 ( n -hex/EtOAc 4:1); m.p. 115 $118{ }^{\circ} \mathrm{C}$; $v$ max 3246, 3062 (N-H), 1659 (C=O), 1598, 1489, 1231, 1035, 740, 698 cm-1; $\boldsymbol{\delta}_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.56$ (1H, bs, CONHNH), 7.73 - 7.71 (2H, d, J7, ArH), $7.54-7.52$ (3H, d, J 7, Ar-H), 7.38 - 7.33 (3H, t, J 7, Ar-H), $7.27-7.18$ (15H, m, Ar-H), 7.04 (1H, m, Ar-H), 6.79 (1H, d, J 7, Ar-H), 6.66 (1H, s, Ar-H), 6.53 (1H, d, J7, Ar-H), $5.98(1 \mathrm{H}, \mathrm{bd}, J 7, \mathrm{FmocNHCH}), 4.39\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene), 4.18 - $4.04\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene, FmocNHCH), $2.49(2 \mathrm{H}, \mathrm{m}$, $\mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{CO}$ ), 2.03 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{CHCHHCH}_{2} \mathrm{CO}$ ), 1.92 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{CHCHHCH}_{2} \mathrm{CO}$ ); $\delta c$ ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 171.9 ( $\mathrm{C}=\mathrm{ONH}($ trityl) $)$, 171.7 ( $\left.\mathrm{C}=\mathrm{ONHNH}\right), 156.5$ ( $\mathrm{C}=\mathrm{ONH}$ ), 149.0 (ipso-Ar-C), 144.3 (ipso-Ar-C), 143.8 (ipso-Ar-C), 143.7 (ipso-Ar-C), 141.3 (ipso-Ar-C), 141.2 (ipso-Ar-C), 134.9 (ipso-Ar-C), 130.2 (Ar-C), 128.7 (Ar-C), 128.0 ( $\mathrm{Ar}-\mathrm{C}$ ), 127.8 (Ar-C), 127.2 (Ar-C), 127.1 (Ar-C), 125.1 (Ar-C), 120.9 (ArC), 120.0 (Ar-C), 113.4 (Ar-C), 111.6 (Ar-C), 70.7 (C-(trityl), $67.1\left(\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}-\right.$ Fluorene), 52.7 (FmocNHCH), 47.2 (C-Fluorene), $33.3\left(\mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{CO}\right), 29.2$ $\left(\mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{CO}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 757\left(\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right)$, 759 ( $\left.{ }^{37} \mathrm{Cl}\right] \mathrm{MNa}^{+}$); HRMS (ES $\left.{ }^{+}\right)$ Found $\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}, 757.2456\left(\mathrm{C}_{45} \mathrm{H}_{39}{ }^{35} \mathrm{ClN}_{4} \mathrm{O}_{4} \mathrm{Na}\right.$ requires 757.2552).

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## tert-Butyl (S)-2-(2-(3-chlorophenyl)hydrazine-1-carbonyl)pyrrolidine-1carboxylate 67v



Following the general procedure outlined, $N$-Boc-L-proline ( $0.30 \mathrm{~g}, 1.39 \mathrm{mmol}$ ) and 3-chlorophenylhydrazine hydrochloride ( $0.27 \mathrm{~g}, 1.53 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a white solid ( $0.29 \mathrm{~g}, 61 \%$ ); $\mathrm{R}_{\mathrm{f}} 0.75$ (n-hexane/EtOAc 4:1); m.p. 158 $160{ }^{\circ} \mathrm{C} ; \quad v_{\max } 3282(\mathrm{~N}-\mathrm{H})$, 2977, 2874 (C-H), 1697, 1661 (C=O), 1598, 1406, 1395, 1162, 1133, 780, $770 \mathrm{~cm}^{-1}$; $\delta$ н ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.84 (1H, bs, CONHNH), 7.14 (1H, t, J 8, Ar-H), 6.85 (1H, d, J 8, Ar-H), 6.80 (1H, s, Ar-H), 6.71 (1H, dd, J 8, 2, Ar-H), 6.12 (1H, bs, Ar-NH), 4.39 (1H, d, J 6, CH2BocNCH), 3.45 (2H, m, $\mathrm{CH}_{2} \mathrm{BocNCH}$ ), $2.38\left(1 \mathrm{H}, \mathrm{m}, \mathrm{BocNCHCH}_{2}\right), 1.94\left(3 \mathrm{H}, \mathrm{m}, \mathrm{BocNCHCH}_{2}\right.$, $\left.\mathrm{CHCH}_{2} \mathrm{CH}_{2}\right), 1.52\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) \delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 172.1$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 149.4 (ipso-Ar-C), 135.1 (ipso-Ar-C), 130.2 (Ar-C), 120.9 (Ar-C), 113.4 (Ar-C), 111.8 (Ar-C), $81.0\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 58.4\left(\mathrm{CH}_{2} \mathrm{BocNCH}\right), 47.2\left(\mathrm{CH}_{2} \mathrm{BocNCH}\right) 28.5$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 27.6\left(\mathrm{BocNCHCH}_{2}\right), 24.8\left(\mathrm{CHCH}_{2} \mathrm{CH}_{2}\right) ; m / z\left(\mathrm{ES}^{+}\right) 362\left(\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right)$, $364\left(\left[{ }^{37} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right)$; HRMS (ES+$)$Found $\left[{ }^{35} \mathrm{CI}\right] \mathrm{MNa}^{+}, 362.1243\left(\mathrm{C}_{16} \mathrm{H}_{22}{ }^{35} \mathrm{CIN}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 362.1247 ).

## tert-Butyl(S)-(3-(4-(benzyloxy)phenyl)-1-(2-(3-chlorophenyl)hydrazinyl)-1-oxopropan-2-yl)carbamate $\mathbf{6 7 z}$



Following the general procedure outlined, N -Boc-O-benzyl-L-tyrosine ( 0.20 g , 0.54 mmol ) and 3-chlorophenylhydrazine hydrochloride ( $0.16 \mathrm{~g}, 0.65 \mathrm{mmol}$ ) were transformed following trituration with DCM into the title compound which was isolated as a white solid ( $0.22 \mathrm{~g}, 83 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.52$ (DCM/EtOH/NH3 200:8:1); m.p. 133-135 ${ }^{\circ} \mathrm{C}$; $v_{\max } 3347,3243$ (N-H), 2985 (C-H), 1676 (C=O), 1518, 1238, 1167, 1015, $748 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 7.91 (1H, bs, CONHNH), $7.45-7.33$ (5H,

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m, Ar-H), $7.13-7.10$ (2H, d, J 8, Ar-H), 7.07 (1H, t, J 8, Ar-H), $6.94-6.91$ (2H, d, J 8, Ar-H), 6.84 (1H, d, J 8, Ar-H), 6.66 (1H, s, Ar-H), 6.46 (1H, d, J 8, Ar-H), 6.04 (1H, bs, CONHNH), 5.09 - 5.02 (3H, m, BocNHCH, Ar-CH2O), 4.42 (1H, q, $J 7$, BocNHCH), 3.04 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{CH}_{2} \mathrm{CH}$ ), 1.44 ( $9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}$ ); $\delta c(75 \mathrm{MHz}$, $\mathrm{CDCl}_{3}$ ) 171.8 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 157.9 ( $\mathrm{C}=\mathrm{ONH}$ ), 148.8 ( $\mathrm{ipso}-\mathrm{Ar}-\mathrm{C}$ ), 130.0 ( $\mathrm{Ar}-\mathrm{C}$ ), 130.2 (Ar-C), 128.2 (Ar-C), 128.1 (Ar-C), 127.5 (Ar-C), 121.2 (Ar-C), 115.2 (ArC), 113.4 ( $\mathrm{Ar}-\mathrm{C}$ ), 111.8 ( $\mathrm{Ar}-\mathrm{C}$ ), $80.8\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 70.0\left(\mathrm{Ar}^{2} \mathrm{CH}_{2} \mathrm{O}\right), 54.6$ (BocNHCH), $36.9\left(\mathrm{Ar}-\mathrm{CH}_{2} \mathrm{CH}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 518\left(\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right), 520$ ( $\left[{ }^{37} \mathrm{Cl}\right] \mathrm{MNa}^{+}$); HRMS (ES ${ }^{+}$) Found $\left[{ }^{35} \mathrm{CI}\right] \mathrm{MNa}^{+}$, $518.1838\left(\mathrm{C}_{27} \mathrm{H}_{30} \mathrm{~N}_{3} \mathrm{O}_{4}{ }^{35} \mathrm{CINa}\right.$ requires 518.1823).
tert-Butyl (2-((2-(2-(3-chlorophenyl)hydrazinyl)-2-oxoethyl)amino)-2oxoethyl)carbamate 67Aa


Following the general procedure outlined, $N$-Boc-glycylglycine ( $0.30 \mathrm{~g}, 1.29 \mathrm{mmol}$ ) and 3-chlorophenylhydrazine hydrochloride ( $0.25 \mathrm{~g}, 1.42 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a light brown solid ( $0.11 \mathrm{~g}, 24 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.17$ ( $n$-hexane/EtOAc 4:1); m.p. 174-179 ${ }^{\circ} \mathrm{C} ; \quad v_{\max } 3314,3274$ (N-H), 3054, 2983 (C-H), 1653 (C=O), 1597, 1548, 1307, 1235, 1169, 947, $664 \mathrm{~cm}^{-1}$; бн $^{(399 \mathrm{MHz}, ~ D M S O-d 6)} 9.65$ (1H, bs, CONHNH), 8.22 (1H, m, CONHCH2), 7.15 (1H, t, J 7, Ar-H), 7.05 (1H, t, J 6, Ar$H$ ), $6.72(1 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{H}), 6.70(1 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}), 3.82\left(2 \mathrm{H}, \mathrm{t}, \mathrm{J} 6, \mathrm{CONHCH}_{2}\right), 3.62(1 \mathrm{H}$, t, J 6, BocNHCH2), 1.39 (9H, s, ( $\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}$ ); סc (100 MHz, DMSO-d6) 170.4 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 169.3 ( $\mathrm{C}=\mathrm{ONH}$ ), 156.4 ( $\mathrm{C}=\mathrm{ONH}$ ), 151.3 (ipso-Ar-C), 133.9 (ipso-Ar-C), 130.7 (Ar-C), 118.3 (Ar-C), 111.9 (Ar-C), 111.3 (Ar-C), $78.7\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $43.9\left(\mathrm{BocNHCH}_{2}\right), 41.5\left(\mathrm{CONHCH}_{2}\right), 28.7\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; \mathrm{m} / \mathrm{z} \quad\left(\mathrm{ES}^{+}\right) 379$ ( $\left[^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}$), 381 ( $\left[{ }^{37} \mathrm{Cl}\right] \mathrm{MNa}^{+}$); HRMS (ES ${ }^{+}$) Found $\left[{ }^{35} \mathrm{CI}\right] \mathrm{MNa}^{+}, 379.1150$ ( $\mathrm{C}_{15} \mathrm{H}_{21}{ }^{35} \mathrm{CIN}_{4} \mathrm{O}_{4} \mathrm{Na}$ requires 379.1149 ).

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tert-Butyl (S)-(1-(2-benzylhydrazinyl)-1-oxopropan-2-yl)carbamate 91


Following the general procedure outlined, $N$-Boc-L-alanine ( $0.20 \mathrm{~g}, 1.06 \mathrm{mmol}$ ) and benzylhydrazine hydrochloride ( $0.25 \mathrm{~g}, 1.27 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a light yellow solid ( 0.084 g, 27\%); Rf 0.43 (DCM/EtOH/NH3 100:8:1); m.p. $112-114{ }^{\circ} \mathrm{C}$; $v_{\max } 3324,3000(\mathrm{~N}-\mathrm{H}), 2911$ (C-H), 1689 (C=O), 1651 (C=O), 1523, 1491, 1249, 1162, 1053, $746,701 \mathrm{~cm}^{-1}$; $\delta_{\mathrm{H}}\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.58$ (1H, bs, NH), $7.35-7.32$ (5H, m, Ar-H), 7.28 (1H, m, Ar-H), 4.89 (1H, bs, BocNHCH), 4.09 (1H, m, BocNHCH), 3.97 (2H, s, $\left.\mathrm{CH}_{2} \mathrm{Ar}\right), 1.42$ ( $\left.9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) 1.33$ (3H, d, J 7, $\mathrm{NHCHCH}_{3}$ ); ठс (700 MHz, CDCl3) 172.1 (C=ONHNH), 155.4 (C=ONH), 137.3 (ipso-Ar-C), 128.9 (Ar-C), 128.8 (Ar-C), 128.5 (Ar-C), 127.6 (Ar-C), 80.3 $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 55.8\left(\mathrm{CH}_{2} \mathrm{Ar}\right), 48.8(\mathrm{BocNHCH}), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 18.1\left(\mathrm{NHCHCH}_{3}\right)$; $m / z\left(\mathrm{ES}^{+}\right) 294\left(\mathrm{MH}^{+}\right)$; HRMS (ES $)$Found $\mathrm{MH}^{+}, 294.1819\left(\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{~N}_{3} \mathrm{O}_{3}\right.$ requires 294.1818).
tert-Butyl (S)-(1-(2-(tert-butyl)hydrazinyl)-1-oxopropan-2-yl)carbamate 89


Following the general procedure outlined, $N$-Boc-L-alanine ( $0.20 \mathrm{~g}, 1.06 \mathrm{mmol}$ ) and tert-butylhydrazine hydrochloride ( $0.16 \mathrm{~g}, 1.27 \mathrm{mmol}$ ) were transformed following work up with EtOAc into the title compound which was isolated as a white solid ( $0.11 \mathrm{~g}, 40 \%$ ); m.p. $133-135{ }^{\circ} \mathrm{C}$; $v_{\max } 3380(\mathrm{~N}-\mathrm{H}), 2973(\mathrm{C}-\mathrm{H}), 1681$ (C=O), 1642 (C=O), 1530, 1365, 1248, 1168, 1044, 828, 764, $634 \mathrm{~cm}^{-1}$; $\delta_{H}(500$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.55$ (1H, bs, NH), 4.95 (1H, bs, NH), 4.17 (1H, m, BocNHCH), 1.47 (9H, s, ( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.40\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 8, \mathrm{NHCHCH}_{3}\right), 1.10\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CNH}\right)$; סc (500 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) 171.4$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.6 ( $\mathrm{C}=\mathrm{ONH}$ ), 80.4 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 55.2$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CNH}\right), 48.9$ (BocNHCH$), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 27.1 \quad\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CNH}\right), 17.7$ $\left(\mathrm{NHCHCH}_{3}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 260\left(\mathrm{MH}^{+}\right) ; \operatorname{HRMS}\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MH}^{+}, 260.1985$ $\left(\mathrm{C}_{12} \mathrm{H}_{26} \mathrm{~N}_{3} \mathrm{O}_{3}\right.$ requires 260. 1974).

## Chapter 6: Experimental section

## tert-Butyl <br> (S)-(3-(4-azidophenyl)-1-(2-(3-chlorophenyl)hydrazinyl)-1-

 oxopropan-2-yl)carbamate 67En

Following the general procedure outlined, N -Boc-L-4-azidophenylalanine ( 0.20 g , 0.65 mmol ) and 3-chlorophenylhydrazine hydrochloride ( $0.14 \mathrm{~g}, 0.78 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a pale yellow solid ( $0.14 \mathrm{~g}, 49 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.35$ (DCM/EtOH/NH3 200:8:1); m.p. 153-156 ${ }^{\circ} \mathrm{C}$; $v_{\max } 3326,3242(\mathrm{~N}-\mathrm{H}), 2099\left(\mathrm{~N}_{3}\right), 1672$ (C=O), 1595, 1529, 1272, 1168, 1050, 861, $771 \mathrm{~cm}^{-1}$; ठн ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 7.96 (1H, bs, NH), 7.21 7.20 (2H, d, J 8, Ar-H), 7.10 (1H, t, J 8, Ar-H), 7.01 - 7.00 (2H, d, J 8, Ar-H), 6.88 (1H, d, J 8, Ar-H), 6.64 (1H, s, Ar-H), 6.54 (1H, d, J 8, Ar-H), 6.07 (1H, bs, ArNH), 5.08 (1H, bd, J 8, BocNHCH), 4.43 (1H, q, J 7, BocNHCH), 3.09 (2H, m, Ar$\mathrm{CH}_{2}$ ), 1.47 (9H, s, ( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta c\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 171.4$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.4 (C=ONH), 148.8 (ipso-Ar-C), 139.1 (ipso-Ar-C), 135.0 (ipso-Ar-C), 132.7 (ipso-Ar-C), 130.7 (Ar-C), 130.2 ( $\mathrm{Ar}-\mathrm{C}$ ), 121.3 ( $\mathrm{Ar}-\mathrm{C}$ ), 119.5 ( $\mathrm{Ar}-\mathrm{C}$ ), 119.1 ( $\mathrm{Ar}-\mathrm{C}$ ), 113.4 (Ar-C), 111.8 (Ar-C), 110.9 (Ar-C), $81.0\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 54.4$ (BocNHCH), 37.0 (Ar$\left.\mathrm{CH}_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 431\left(\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 433\left(\left[{ }^{37} \mathrm{CI}\right] \mathrm{MH}^{+}\right) ;$HRMS (ES $)$ Found $\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}, 431.1592\left(\mathrm{C}_{20} \mathrm{H}_{24} \mathrm{~N}_{6} \mathrm{O}_{3}{ }^{35} \mathrm{Cl}\right.$ requires 431.1598).
tert-Butyl (S)-(3-(4-azidophenyl)-1-(2-(4-cyanophenyl)hydrazinyl)-1-oxopropan-2-yl)carbamate 67Eo


Following the general procedure outlined, N -Boc-L-4-azidophenylalanine ( 0.20 g , 0.65 mmol ) and 4-cyanophenylhydrazine hydrochloride ( $0.13 \mathrm{~g}, 0.78 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a pale yellow solid ( $0.06 \mathrm{~g}, 23 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.31$ (DCM/EtOH/ $\mathrm{NH}_{3}$ 200:8:1); m.p. $171-174{ }^{\circ} \mathrm{C}$; $v_{\max } 3302(\mathrm{~N}-\mathrm{H})$, 2218 (CN), 2104 ( $\mathrm{N}_{3}$ ), 1689, 1655 (C=O),

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1507, 1368, 1293, 1163, 1050, 832, $542 \mathrm{~cm}^{-1}$; $\delta_{н}\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.83$ (1H, bs, NH), 7.46 - 7.44 (2H, dd, J 8, 1, Ar-H), 7.22-7.20 (2H, dd, J 8, 1, Ar-H), 7.00-6.99 (2H, dd, J 8, 2, Ar-H), 6.65-6.64 (2H, d, J 8, Ar-H), 6.15 (1H, bs, Ar-NH), 4.96 (1H, bs, BocNHCH), 4.36 (1H, m, BocNHCH), 3.11 (1H, m, Ar-CH2), 3.06 (1H, m, Ar$\mathrm{CH}_{2}$ ), 1.46 ( $\left.9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta \mathrm{c}\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 171.4$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 151.0 (C=ONH), 143.1 (ipso-Ar-C), 134.3 (ipso-Ar-C), 133.6 (Ar-C), 133.4 (Ar-C), 130.8 (Ar-C), 129.3 (Ar-C), 119.4 (Ar-C), 119.3 (Ar-C), 116.3 (Ar-C), 113.1 (Ar-C),. 4 $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 54.3(\mathrm{BocNHCH}), 36.6\left(\mathrm{Ar}-\mathrm{CH}_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 422$ $\left(\mathrm{MH}^{+}\right)$; HRMS $\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MH}^{+}, 422.1948\left(\mathrm{C}_{21} \mathrm{H}_{24} \mathrm{~N}_{7} \mathrm{O}_{3}\right.$ requires 422.1941).

## tert-Butyl (S)-(3-(4-azidophenyl)-1-oxo-1-(2-phenylhydrazinyl)propan-2-yl) carbamate 67Em



Following the general procedure outlined, N -Boc-L-4-azidophenylalanine ( 0.20 g , $0.65 \mathrm{mmol})$ and phenylhydrazine ( $0.07 \mathrm{~mL}, 0.72 \mathrm{mmol}$ ) were transformed following flash column chromatography (DCM/EtOH/ $\mathrm{NH}_{3}$ 120:6:1) into the title compound which was isolated as a brown gum ( $0.20 \mathrm{~g}, 77 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.34$ (DCM/EtOH/NH3 200:8:1); m.p. $92-95{ }^{\circ} \mathrm{C} ; v_{\max } 3300(\mathrm{~N}-\mathrm{H}), 2926(\mathrm{C}-\mathrm{H}), 2111$ ( $\mathrm{N}_{3}$ ), 1685, 1659 (C=O), 1505, 1366, 1289, 1166, 1048, 752, $691 \mathrm{~cm}^{-1}$; $\delta_{H}(700$ MHz, CDCl3) 8.09 (1H, bs, NH), 7.18 - 7.14 (3H, m, Ar-H), $6.94-6.93$ (2H, d, J 8, Ar-H), 6.89 - 6.87 (2H, m, Ar-H), 6.60 - 6.59 (2H, d, J 8, Ar-H), 5.15 (1H, bd, J 8, BocNHCH), 4.43 (1H, m, BocNHCH), 3.08 (1H, m, Ar-CH2), 3.02 (1H, m, Ar$\mathrm{CH}_{2}$ ), 1.44 (9H, s, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta c\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 171.3$ ( $\left.\mathrm{C}=\mathrm{ONHNH}\right), 155.6$ (C=ONH), 147.4 (ipso-Ar-C), 138.9 (ipso-Ar-C), 132.9 (ipso-Ar-C), 130.8 (Ar-C), 129.1 (Ar-C), 121.3 (Ar-C), 119.3 (Ar-C), 119.0 (Ar-C), 115.4 (Ar-C), 113.5 (ArC), 112.8 ( $\mathrm{Ar}-\mathrm{C}$ ), $80.8\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 54.4$ ( BocNHCH ), $37.4\left(\mathrm{Ar}-\mathrm{CH}_{2}\right), 28.3$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 397\left(\mathrm{MH}^{+}\right) ; \operatorname{HRMS}\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MH}^{+}, 397.1971$ ( $\mathrm{C}_{20} \mathrm{H}_{25} \mathrm{~N}_{6} \mathrm{O}_{3}$ requires 397.1988 ).

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### 6.1.2.3 Synthesis of $N$-protected Hydrazides - General Procedure

Under a nitrogen atmosphere $N$-protected amino acid (1 equiv.) dissolved in THF ( 5 mL ) at room temperature and treated with DIPEA (1.2 equiv.) followed by HBTU (1.2 equiv.). After stirring for 10 minutes at room temperature addition of N -aryl hydrazine ( 1.1 equiv.) continued stirring for another 5 hours at room temperature. Reaction mixed with diethyl ether ( 8 mL ) and distilled water ( 5 mL ). After separation of the two phases the organic layer washed again with distilled water ( $5 \mathrm{~mL} \times 3 \mathrm{~mL}$ ) followed by a wash with sat. aq. $\mathrm{NH}_{4} \mathrm{Cl}(6 \mathrm{~mL})$ then sat. aq. $\mathrm{NaHCO}_{3}(6 \mathrm{~mL})$ followed by brine ( 8 mL ). Organic layer dried over $\mathrm{MgSO}_{4}$, filtered, evaporated and dried in vacuo achieving the desired $N$-Boc hydrazides 67 or N Fmoc hydrazides 68.

## tert-Butyl (S)-4-((( 9 H -fluoren-9-yl)methoxy)carbonyl)amino)-5-oxo-5-(2-

 phenylhydrazineyl)pentanoate 68d

Following the general procedure outlined, N -Fmoc-O-(tert-butyl)-L-glutamic acid $(0.40 \mathrm{~g}, 0.94 \mathrm{mmol})$ and phenylhydrazine ( $0.10 \mathrm{~mL}, 1.03 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a light brown solid ( $0.26 \mathrm{~g}, 53 \%$ ); Rf 0.46 (DCM/EtOH/ $\mathrm{NH}_{3} 400: 6: 1$ ); m.p. $113-116{ }^{\circ} \mathrm{C}$; $v \max 3287$ (N-H), 2969 (C-H), 1691, 1658 (C=O), 1539, 1497, 1257, 1152, 1085, 738, $692 \mathrm{~cm}-1$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.38 (1H, bs, CONHNH), 7.79 - 7.76 (2H, d, J 7, Ar-H), 7.60 - 7.58 (2H, d, J 7, Ar-H), 7.44 - 7.39 (2H, t, J 7, Ar-H), 7.34 7.31 (2H, t, J 7, 1, Ar-H), $7.24-7.19$ (2H, t, J 7, Ar-H), 6.93 (1H, t, J 7, Ar-H), 6.83 - 8.81 (1H, d, J 7, Ar-H), 5.85 (1H, bd, J 7, FmocNHCH), 4.46 ( $2 \mathrm{H}, \mathrm{m}$, $\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}$-Fluorene), $4.39\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene), $4.25(1 \mathrm{H}, \mathrm{m}$, FmocNHCH), $2.50\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2}\right), 2.19\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHCHHCH}_{2} \mathrm{CO}_{2}\right), 2.06$ ( $1 \mathrm{H}, \mathrm{m}, \mathrm{CHCHHCH} 2 \mathrm{CO}_{2}$ ), $1.49\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 172.8$ ( $\left.\mathrm{C}=\mathrm{OO}\left(\mathrm{CH}_{3}\right)_{3}\right), 171.5$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.8 ( $\mathrm{C}=\mathrm{ONH}$ ), 147.6 (ipso-Ar-C), 143.7

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(ipso-Ar-C), 141.3 (ipso-Ar-C), 129.2 (Ar-C), 127.8 (Ar-C), 127.1 (Ar-C), 125.0 (Ar-C), 121.3 (Ar-C), 120.0 (Ar-C), 113.6 (Ar-C), 81.3 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 67.2$ $\left(\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene), $53.1 \quad(\mathrm{FmocNHCH}), \quad 47.1$ (C-Fluorene), 31.6 $\left(\mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2}\right), 28.1\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 27.2\left(\mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 538$ ( $\mathrm{MNa}^{+}$); $\mathrm{HRMS}\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}, 538.2327\left(\mathrm{C}_{30} \mathrm{H}_{33} \mathrm{~N}_{3} \mathrm{O}_{5} \mathrm{Na}\right.$ requires 538.2312).

## tert-Butyl <br> ((2S,3S)-3-methyl-1-oxo-1-(2-phenylhydrazineyl)pentan-2-yl)

 carbamate 67e

Following the general procedure outlined, $N$-Boc-L-isoleucine ( $0.30 \mathrm{~g}, 1.29 \mathrm{mmol}$ ) and phenylhydrazine ( $0.14 \mathrm{~g}, 1.43 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a brown solid $(0.39 \mathrm{~g}$, 93\%); Rf 0.43 (DCM/EtOH/NH3 200:6:1); m.p. $142-145{ }^{\circ} \mathrm{C}$; $v_{\max } 3320,3270$ (NH), 2962 (C-H), 1686, 1658 (C=O), 1522, 1494, 1292, 1170, 769, $692 \mathrm{~cm}^{-1}$; бн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.46 (1H, bs, CONHNH), 7.20 - 7.15 (2H, t, J 7, Ar-H), 6.89 (1H, t, J 7, Ar-H), $6.82-6.78$ (2H, m, Ar-H), 6.20 (1H, bs, Ar-NH), 5.22 (1H, bd, J 9, BocNHCH), 4.05 (1H, t, J 8, BocNHCH), 1.89 (1H, m, NHCHCH), 1.55 (1H, $\left.\mathrm{m}, \mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CHHCH}_{3}\right), 1.45\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.18\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CHHCH}_{3}\right)$, $0.94-0.86\left(6 \mathrm{H}, \mathrm{m}, \mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{2} \mathrm{CH}_{3}, \mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$; ठc (100 MHz, CDCl 3$)$ 172.1 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.1 ( $\mathrm{C}=\mathrm{ONH}$ ), 147.8 (ipso-Ar-C), 129.1 (Ar-C), 121.1 (ArC), 113.6 ( $\mathrm{Ar}-\mathrm{C}$ ), 80.3 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 57.8$ (BocNHCH), 36.7 (NHCHCH), 28.3 $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), \quad 24.8 \quad\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{2} \mathrm{CH}_{3}\right), \quad 15.6 \quad\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{2} \mathrm{CH}_{3}\right), \quad 11.1$ $\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; m / z\left(\mathrm{ES}^{+}\right) 344\left(\mathrm{MNa}^{+}\right) ;$HRMS $\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}, 344.1966$ $\left(\mathrm{C}_{17} \mathrm{H}_{27} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 344.1945).

## Chapter 6: Experimental section

(9H-fluoren-9-yl)methyl tert-Butyl (6-oxo-6-(2-phenylhydrazineyl)hexane-1,5-diyl)(S)-dicarbamate 68h


Following the general procedure outlined, N-Fmoc-N6-(tert-butoxycarbonyl)-Llysine ( $0.40 \mathrm{~g}, 0.85 \mathrm{mmol}$ ) and phenylhydrazine ( $0.10 \mathrm{~g}, 0.94 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a light brown solid ( $0.24 \mathrm{~g}, 51 \%$ ); $\mathrm{R}_{\mathrm{f}} 0.37$ (DCM/EtOH/ $\mathrm{NH}_{3}$ 200:6:1); m.p. 125-128 ${ }^{\circ} \mathrm{C}$; $V_{\max } 3222,3300(\mathrm{~N}-\mathrm{H})$, 1687, 1655 (C=O), 1527, 1487, 1362, 1252, 1159, 764, $685 \mathrm{~cm}^{-1}$; ठн ( $399 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.34 (1H, bs, CONHNH), 7.76 - 7.74 (2H, d, J 7, Ar-H), $7.57-7.55$ (2H, d, J 7, Ar-H), 7.41 - 7.36 (2H, t, J 7, Ar-H), $7.31-7.28(2 H, d, J 7$, Ar-H), $7.20-7.15(2 H, t, J 7$, Ar-H), $6.89(1 H, t, J$ 7, Ar-H), 6.79 - 6.76 (2H, d, J7, Ar-H), 5.63 (1H, bd, J7, FmocNHCH), 4.64 (1H, $\mathrm{m}, \mathrm{BocNHCH} 2), 4.45\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene), $4.25-4.17(2 \mathrm{H}, \mathrm{m}$, FmocNHCH, $\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}$-Fluorene), 3.09 ( $2 \mathrm{H}, \mathrm{m}$, $\mathrm{BocNHCH}_{2}$ ), 1.71 (1H, m, FmocNHCHCHH), 1.63 ( 1H, m, FmocNHCHCHH), 1.51 (2H, m, BocNHCH2CH2), 1.43 (9H, s, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; бс (100 MHz, CDCl3) 171.9 (C=ONHNH), 156.3 (C=ONH), 147.7 (ipso-Ar-C), 143.7 (ipso-Ar-C), 141.3 (ipso-Ar-C), 129.2 (Ar-C), 127.8 (Ar-C), 127.1 (Ar-C), 125.0 (Ar-C), 121.3 (Ar-C), 120.0 (Ar-C), 113.6 (ArC), $79.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 67.1\left(\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene), 53.4 (FmocNHCH), 47.2 (CFluorene), $39.7\left(\mathrm{BocNHCH}_{2}\right)$, $31.6\left(\mathrm{FmocNHCHCH}_{2}\right)$, $29.6\left(\mathrm{BocNHCH}_{2} \mathrm{CH}_{2}\right)$, $28.5\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 22.5\left(\mathrm{NHCHCH}_{2} \mathrm{CH}_{2}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 581\left(\mathrm{MNa}^{+}\right) ;$HRMS (ES$\left.{ }^{+}\right)$ Found $\mathrm{MNa}^{+}, 581.2764\left(\mathrm{C}_{32} \mathrm{H}_{38} \mathrm{~N}_{4} \mathrm{O}_{5} \mathrm{Na}\right.$ requires 581.2762).

## Chapter 6: Experimental section

## tert-Butyl(S)-3-((((9H-fluoren-9-yl)methoxy)carbonyl)amino)-4-oxo-4-(2phenylhydrazineyl)butanoate 68c



Following the general procedure outlined, N -Fmoc-O-(tert-butyl)-L-aspartic acid $(0.40 \mathrm{~g}, 0.97 \mathrm{mmol})$ and phenylhydrazine $(0.11 \mathrm{~mL}, 1.07 \mathrm{mmol})$ were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a light brown solid ( $0.20 \mathrm{~g}, 41 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.46$ (DCM/EtOH/ $\mathrm{NH}_{3} 200: 6: 1$ ); m.p. $158-161^{\circ} \mathrm{C}$; vmax 3287 (N-H), 1690, 1656 (C=O), 1534, 1496, 1266, 1159, 1046, 738, 692 cm-1; бн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ), 8.26 (1H, bs, CONHNH), 7.76 - 7.74 (2H, d, J7, ArH), 7.58 - 7.55 (2H, d, J 7, Ar-H), 7.42 - 7.35 (2H, td, J 7, 3, Ar-H), 7.32 - 7.26 (2H, m, Ar-H), 7.20-7.15 (2H, t, J 7, Ar-H), 6.89 (1H, t, J 7, Ar-H), 6.80-6.77 (1H, d, J 7, Ar-H), 6.02 (1H, m, FmocNHCH), 4.62 (1H, m, FmocNHCH), 4.53 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}$-Fluorene), $4.22\left(1 \mathrm{H}, \mathrm{t}, \mathrm{J} 6, \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene), $2.92(1 \mathrm{H}$, dd, J 16, 4, CHCHHCO2), $2.70\left(1 \mathrm{H}, \mathrm{dd}, \mathrm{J} 16,6, \mathrm{CHCHHCO}_{2}\right), 1.44(9 \mathrm{H}, \mathrm{s}$, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; ~ \delta c\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 170.9(\mathrm{C}=\mathrm{O}), 170.7$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.2 (C=ONH), 147.6 (ipso-Ar-C), 143.6 (ipso-Ar-C), 143.6 (ipso-Ar-C), 143.5 (ipso-Ar-C), 141.4 (ipso-Ar-C), 129.2 (Ar-C), 127.9 (Ar-C), 127.8 (Ar-C), 127.2 (Ar-C), 127.1 (Ar-C), 124.9 (Ar-C), 121.3 (Ar-C), 120.1 (Ar-C), 120.0 (Ar-C), 113.6 (ArC), $82.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 67.3\left(\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}-\right.$ Fluorene $), 50.2$ (FmocNHCH), 47.2 (CFluorene), $37.1\left(\mathrm{CHCH}_{2} \mathrm{CO}_{2}\right)$, $28.1\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; m / z\left(\mathrm{ES}^{+}\right) 524\left(\mathrm{MNa}^{+}\right), 1003$ $\left(2 \mathrm{M}+\mathrm{H}^{+}\right), 1025\left(2 \mathrm{M}+\mathrm{Na}^{+}\right)$; HRMS $\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}, 524.2174\left(\mathrm{C}_{29} \mathrm{H}_{31} \mathrm{~N}_{3} \mathrm{O}_{5} \mathrm{Na}\right.$ requires 524.2156).

## Chapter 6: Experimental section

(9H-fluoren-9-yl)methyl (S)-(3-(tert-butoxy)-1-oxo-1-(2-phenylhydrazineyl)propan-2-yl)carbamate 68a


Following the general procedure outlined, $N$-Fmoc-O-(tert-butyl)-L-serine ( 0.40 g , $1.04 \mathrm{mmol})$ and phenylhydrazine ( $0.11 \mathrm{~mL}, 1.15 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a light brown solid ( $0.31 \mathrm{~g}, 63 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.51$ (DCM/EtOH/ $\mathrm{NH}_{3} 200: 6: 1$ ); m.p. $140-142{ }^{\circ} \mathrm{C}$; $V_{\max } 3388$ (N-H), 2979 (C-H), 1693, 1651 (C=O), 1543, 1233, 11193, 1032, 756, $738 \mathrm{~cm}^{-1}$; ठн (300 MHz, DMSO-d6) 9.81 (1H, bd, J2, CONHNH), 7.91 - 7.89 (2H, d, J 7, Ar-H), 7.77 - 7.76 (2H, m, Ar-H), 7.45-7.40 (2H, t, J7, Ar-H), 7.36-7.31 (2H, t, J 7, Ar-H), 7.13 - 7.09 (2H, t, J 7, Ar-H), 6.80 - 6.77 (2H, d, J 7, Ar-H), $6.71(1 \mathrm{H}, \mathrm{t}, \mathrm{J} 7, \mathrm{Ar}-\mathrm{H}), 4.31-4.21\left(4 \mathrm{H}, \mathrm{m}, \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}-\right.$ Fluorene, FmocNHCH , $\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}$-Fluorene), 3.55 (2H, m, FmocNHCHCH2), 1.18 (9H, s, ( $\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}$ ); ठc (100 MHz, DMSO-d6) 170.5 ( $C=O N H N H$ ), 156.3 ( $C=O N H$ ), 149.5 (ipso-Ar-C), 144.4 (ipso-Ar-C), 144.2 (ipso-Ar-C), 141.2 (ipso-Ar-C), 128.9 (Ar-C), 128.1 (ArC), 127.5 (Ar-C), 125.8 (Ar-C), 120.6 (Ar-C), 118.8 (Ar-C), 112.8 (Ar-C), 73.4 $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), \quad 66.4 \quad\left(\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene), 62.2 ( $\mathrm{FmocNHCHCH}_{2}$ ), 54.4 (FmocNHCH), 47.1 (C-Fluorene), 27.7 ((CH3)3CO); m/z (ES ${ }^{+}$) 496 (MNa ${ }^{+}$); HRMS $\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}, 496.2253\left(\mathrm{C}_{28} \mathrm{H}_{31} \mathrm{~N}_{3} \mathrm{O}_{4} \mathrm{Na}\right.$ requires 496.2207).

## Chapter 6: Experimental section

(9H-fluoren-9-yl)methyl (S)-(1,5-dioxo-1-(2-phenylhydrazineyl)-5-
(tritylamino)pentan-2-yl)carbamate $\mathbf{6 8 g}$


Following the general procedure outlined, $N$-Fmoc- $N$-(trityl)-L-glutamine ( 0.40 g , 0.66 mmol ) and phenylhydrazine ( $0.07 \mathrm{~g}, 0.72 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a light brown solid ( $0.29 \mathrm{~g}, 65 \%$ ); Rf 0.40 (DCM/EtOH/NH3 200:6:1); m.p. $126-129{ }^{\circ} \mathrm{C}$; $v_{\max }$ 3247 (N-H), 1717, 1659 (C=O), 1533, 1495, 1230, 1046, 744, 699 cm-1; סн (300 MHz, DMSO-d6) 9.75 (1H, bd, J 2, CONHNH), 8.61 (1H, bs, CONH-trityl), 7.92 7.89 (2H, d, J 7, Ar-H), 7.77 - 7.72 (3H, m, Ar-H), $7.45-7.40$ (2H, t, J 7, Ar-H), $7.34-7.18$ (15H, m, Ar-H), $7.13-7.08$ (2H, t, J 7, Ar-H), $6.72-6.69$ (3H, d, J 8, Ar-H), $4.34\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene), $4.27\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene), $4.08(1 \mathrm{H}, \mathrm{m}, \mathrm{FmocNHCH}), 2.45\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{CO}\right), 1.95(2 \mathrm{H}, \mathrm{m}$, $\mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{CO}$ ); $\delta с$ ( 100 MHz , DMSO-d6) 172.1 ( $\mathrm{C}=\mathrm{ONH}($ trityl)), 171.8 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.5 ( $\mathrm{C}=\mathrm{ONH}$ ), 149.7 (ipso-Ar-C), 145.4 (ipso-Ar-C), 144.4 (ipso-Ar-C), 144.3 (ipso-Ar-C), 141.2 (ipso-Ar-C), 129.1 (Ar-C), 128.9 (Ar-C), 128.1 (ArC), 127.9 (Ar-C), 127.6 (Ar-C), 126.8 (Ar-C), 125.8 (Ar-C), 120.6 (Ar-C), 118.9 (Ar-C), 112.6 (Ar-C), 69.7 (C-(trityl), $66.2\left(\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene), 53.7 (FmocNHCH), 47.2 (C-Fluorene), $33.3\left(\mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{CO}\right), 28.3\left(\mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{CO}\right)$; $\mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 723\left(\mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}$, $723.2944\left(\mathrm{C}_{45} \mathrm{H}_{40} \mathrm{~N}_{4} \mathrm{O}_{4} \mathrm{Na}\right.$ requires 723.2942 ).

## Chapter 6: Experimental section

(9H-fluoren-9-yl)methyl dihydrobenzofuran-5-yl)sulfonyl)guanidino)-1-(2-phenylhydrazineyl) pentan-2-yl)carbamate 68j


Following the general procedure outlined, $\quad N$-Fmoc- $N$-((2,2,4,6,7-pentamethyldihydro-benzofuran-5-sulfonyl)-L-arginine ( $0.40 \mathrm{~g}, 0.62 \mathrm{mmol}$ ) and phenylhydrazine ( $0.07 \mathrm{~mL}, 0.68 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a white solid ( $0.14 \mathrm{~g}, 30 \%$ ); Rf 0.20 (DCM/EtOH/NH3 200:6:1); m.p. $137-140{ }^{\circ} \mathrm{C}$; vmax 3319, 1727, 1664 (C=O), 1584, 1517, 1249, 1102, 1065, 735, 619 cm-1; бн (300 MHz, DMSO-d6) 9.82 (1H, bd, J 2, CONHNH), 7.91 - 7.89 (2H, d, J 7, Ar-H), $7.77-7.74$ (2H, m, Ar-H), $7.45-7.40$ (2H, t, J 7, Ar-H), $7.35-7.31$ (2H, d, J 7, Ar-H), $7.14-7.09$ (2H, t, J 7, Ar-H), 6.74 - 6.71 (2H, d, J 8, Ar-H), 6.46 (1H, bs, Ar-NH), 4.33 (2H, $\mathrm{m}, \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}$-Fluorene), 4.26 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}$-Fluorene), 4.13 ( $1 \mathrm{H}, \mathrm{m}$, FmocNHCH), 3.12 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{NHCNHNH}$ ), 2.94 ( $2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2}$-benzofuran), 2.53 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}$-benzofuran), $2.46\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right.$-benzofuran), $2.02\left(3 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3}-\right.$ benzofuran), $1.70(2 \mathrm{H}, \mathrm{m}, \mathrm{FmocNHCHCH} 2), 1.54\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}_{2} \mathrm{CH}_{2}\right), 1.40(6 \mathrm{H}$, $\mathrm{s},\left(\mathrm{CH}_{3}\right)_{2}$-benzofuran); $\delta c$ ( $100 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d} 6$ ) 172.2 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 157.9 (C=NHNH) 156.6 (C=ONH), 149.7 (ipso-Ar-C), 144.4 (ipso-Ar-C), 144.2 (ipso-ArC), 141.2 (ipso-Ar-C), 137.8 (ipso-Ar-C), 134.7 (ipso-Ar-C), 131.9 (ipso-Ar-C), 129.1 (Ar-C), 128.8 (Ar-C), 128.1 (Ar-C), 127.6 (Ar-C), 125.7 (Ar-C), 124.8 (ArC), 120.6 (Ar-C), 118.9 (Ar-C), 116.8 (ipso-Ar-C), 112.7 (Ar-C), 86.8 (Cbenzofuran), $66.2\left(\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene), 53.5 ( FmocNHCH ), 47.2 (C-Fluorene), 42.9 ( $\mathrm{CH}_{2}$-benzofuran), $40.8\left(\mathrm{CH}_{2} \mathrm{NHCNHNH}\right), 29.6$ ( $\mathrm{FmocNHCHCH}_{2}$ ), 28.8

## Chapter 6: Experimental section

$\left(\mathrm{CH}_{3}\right)_{2}$-benzofuran), 25,6 $\left(\mathrm{CHCH}_{2} \mathrm{CH}_{2}\right)$, $19.5\left(\mathrm{CH}_{3}\right.$-benzofuran), $18.1\left(\mathrm{CH}_{3}-\right.$ benzofuran), 12.8 ( $\mathrm{CH}_{3}$-benzofuran); $\mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 739\left(\mathrm{MH}^{+}\right)$, 761 ( $\mathrm{MNa}{ }^{+}$); HRMS (ES ${ }^{+}$) Found $\mathrm{MH}^{+}, 739.3268\left(\mathrm{C}_{40} \mathrm{H}_{47} \mathrm{~N}_{6} \mathrm{O}_{6} \mathrm{~S}\right.$ requires 739.3272).
(9H-fluoren-9-yl)methyl
(S)-(1-oxo-1-(2-phenylhydrazineyl)-3-(tritylthio) propan-2-yl)carbamate 68f



Following the general procedure outlined, N-Fmoc-S-(trityl)-L-cysteine ( 0.40 g , $0.68 \mathrm{mmol})$ and phenylhydrazine ( $0.07 \mathrm{~mL}, 0.75 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a brown oil ( $0.11 \mathrm{~g}, 23 \%$ ); Rf 0.35 (DCM/EtOH/NH3 200:6:1); vmax 3234, 3054,
 MHz, DMSO-d6) 9.92 (1H, bd, J 2, CONHNH), 8.60 (1H, bs, Ar-NH), 8.05 (1H, bd, J 2, FmocNHCH), 7.92 - 7.90 (2H, d, J 7, Ar-H), $7.76-7.73$ (3H, dd, J 7, 2, Ar-H), 7.43 - 7.39 (3H, q, J 7, Ar-H), $7.24-7.19$ (15H, m, Ar-H), 7.13 (1H, t, J 7, Ar-H), 6.73 - 6.70 (2H, m, Ar-H), $6.69-6.63(2 H, m, A r-H), 4.43(2 H, m$, $\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}$-Fluorene), $4.29-4.22\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene, FmocNHCH), 2.83 (1H, m, CHCHHS-(trityl)), 2.59 (1H, d, J 4, CHCHHS-(trityl)); סc (100 MHz, DMSO-d6) 171.8 ( $C=O N H N H$ ), 156.3 ( $C=O N H$ ), 151.3 (ipso-Ar-C), 145.2 (ipso-Ar-C), 144.4 (ipso-Ar-C), 144.2 (ipso-Ar-C), 141.2 (ipso-Ar-C), 133.9 (ipso-Ar-C), 130.7 ( $\mathrm{Ar}-\mathrm{C}$ ), 129.1 ( $\mathrm{Ar}-\mathrm{C}$ ), 128.1 ( $\mathrm{Ar}-\mathrm{C}$ ), 127.9 ( $\mathrm{Ar}-\mathrm{C}$ ), 127.8 ( $\mathrm{Ar}-\mathrm{C}$ ), 126.8 ( $\mathrm{Ar}-$ C), 125.8 (Ar-C), 125.7 (Ar-C), 120.6 (Ar-C), 118.3 (Ar-C), 111.9 (ipso-Ar-C), 111.2 (Ar-C), 69.9 (C-(trityl), $66.4\left(\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene), 51.4 (FmocNHCH), 47.2 (C-Fluorene), 39.2 ( $\mathrm{CHCH}_{2} \mathrm{~S}$-(trityl)); m/z (ES+ 698 (MNa+); HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}, 698.2353\left(\mathrm{C}_{43} \mathrm{H}_{37} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{SNa}\right.$ requires 698.2448).

## Chapter 6: Experimental section

## tert-Butyl (S)-(3-(1-benzyl-1H-imidazol-5-yl)-1-oxo-1-(2-phenylhydrazineyl)

 propan-2-yl)carbamate 67i

Following the general procedure outlined, $N$-Fmoc- $N$-(im)-benzyl-L-histidine $(0.40 \mathrm{~g}, 1.16 \mathrm{mmol})$ and phenylhydrazine ( $0.13 \mathrm{~mL}, 1.27 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow gummy solid ( $0.36 \mathrm{~g}, 70 \%$ ); Rf 0.20 (DCM/EtOH/NH3 200:6:1); vmax 3274 (N-H), 2977, 2930 (C-H), 1676, 1602 (C=O), 14953, 1365, 1248, 1163, 1019, 751, 693 cm-1; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 9.30 (1H, bs, CONHNH), 7.81 (1H, bs, Ar-NH), 7.38 - 7.34 (4H, m, Ar-H), 7.18 - 7.14 (3H, m, Ar-H), 7.12 (1H, s, Ar-H), 6.86 (1H, t, J 7, Ar-H), 6.79 (1H, s, Ar-H), 6.65 (1H, d, J 7, Ar-H), 6.32 (1H, bd, J7, BocNHCH), 5.06 (2H, s, Ar-CH2lm), 4.63 (1H, m, BocNHCH), 3.12 (2H, m, CHCH 2 Im ), 1.45 (9H, s, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; \delta c\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 171.6$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 148. (ipso-Ar-C), 136.3 (ipso-Ar-C), 135.2 (ipso-Ar-C), 129.1 (Ar-C), 129.0 (Ar-C), 128.6 (Ar-C), 128.3 (Ar-C), 127.7 (Ar-C), 120.8 (Ar-C), 117.9 ( $\mathrm{Ar}-\mathrm{C}$ ), 113.5 ( $\mathrm{Ar}-\mathrm{C}$ ), 80.1 $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 51.4$ (Ar-CH2Im), $30.0\left(\mathrm{CHCH}_{2} \mathrm{Im}\right), 28.4$ ((CH3)3CO); m/z (ES+) 436 $\left(\mathrm{MH}^{+}\right), 458\left(\mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\mathrm{MH}^{+}, 436.2355\left(\mathrm{C}_{24} \mathrm{H}_{29} \mathrm{~N}_{5} \mathrm{O}_{3}\right.$ requires 436.2343).

## Chapter 6: Experimental section

## tert-Butyl (S)-3-(2-((() $9 \mathrm{H}-\mathrm{fluoren}-9-\mathrm{yl}) m e t h o x y)$ carbonyl)amino)-3-oxo-3-(2-

 phenylhydrazineyl)propyl)-1H-indole-1-carboxylate 68i

Following the general procedure outlined, $N$-Fmoc- $N$-(tert-butyl)-L-tryptophan ( $0.40 \mathrm{~g}, 0.76 \mathrm{mmol}$ ) and phenylhydrazine ( $0.08 \mathrm{~mL}, 0.84 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a white solid ( 0.31 g, 67\%); Rf 0.62 (DCM/EtOH/NH $200: 6: 1$ ); m.p. $163-166{ }^{\circ} \mathrm{C} ; ~ v m a x$ 3277 (N-H), 2980 (C-H), 1732, 1706, 1687 (C=O), 1455, 1372, 1255, 1154, 1085, 747 cm-1; бн (300 MHz, DMSO-d6) 10.00 (1H, bd, J 2, CONHNH), 8.08 (1H, bd, J 8, FmocNHCH), 7.89 - 7.87 (3H, d, J7, Ar-H), 7.83 (1H, d, J7, Ar-H), 7.77 (1H, d, J 2, Ar-H), 7.67 - 7.63 (3H, t, J 6, Ar-H), 7.42 - 7.35 (3H, m, Ar-H), 7.31 - 7.23 (2H, m, Ar-H), 7.09-7. 04 (2H, t, J 7, Ar-H), 6.69 (1H, t, J 7, Ar-H), 6.61-6.59 (2H, d, J 7, Ar-H), 4.54 (1H, m, FmocNHCH), $4.23-4.18$ (3H, m, $\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}-$ Fluorene, $\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}$-Fluorene), 3.19 (1H, dd, J 14, 5, FmocNHCHCHH), 3.07 (1H, dd, J 14, 9, FmocNHCHCHH), 1.59 (9H, s, (CH3)3CO); סc (100 MHz, DMSOd6) 171.6 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.4 ( $\mathrm{C=ONH}$ ), 149.5 (ipso-Ar-C), 144.3 (ipso-Ar-C), 144.1 (ipso-Ar-C), 141.1 (ipso-Ar-C), 135.2 (ipso-Ar-C), 130.6 (Ar-C), 129.4 (ArC), 129.0 (Ar-C), 128.1 (Ar-C), 127.8 (Ar-C), 127.5 (Ar-C), 125.7 (Ar-C), 124.9 (Ar-C), 124.8 (Ar-C), 122.9 (Ar-C), 121.8 (Ar-C), 120.6 (Ar-C), 120.1 (Ar-C), 118.9 (Ar-C), 116.9 (Ar-C), 115.1 (Ar-C), 112.6 (Ar-C), (Ar-C), 84.0 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 66.3$ $\left(\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene), 53.6 (FmocNHCH), 47.0 (C-Fluorene), $28.1\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $m / z\left(\mathrm{ES}^{+}\right) 639\left(\mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}, 639.2583\left(\mathrm{C}_{37} \mathrm{H}_{36} \mathrm{~N}_{4} \mathrm{O}_{5} \mathrm{Na}\right.$ requires 639.2503).

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(9H-fluoren-9-yl)methyl ((2S,3S)-3-(tert-butoxy)-1-oxo-1-(2-phenylhydrazineyl)butan-2-yl)carbamate 68b


Following the general procedure outlined, N-Fmoc-O-(tert-butyl)-L-threonine ( $0.40 \mathrm{~g}, 1.01 \mathrm{mmol}$ ) and phenylhydrazine ( $0.09 \mathrm{~mL}, 1.11 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a reddish brown gummy solid ( $0.31 \mathrm{~g}, 63 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.48$ ( $\mathrm{DCM} / E t O H / \mathrm{NH}_{3} 200: 6: 1$ ); $v_{\text {max }}$ 3317, 3252 (N-H), 2979 (C-H), 1692, 1652 (C=O), 1535, 1496, 1249, 1181, 1068, 757, $736 \mathrm{~cm}^{-1}$; $\delta_{H}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.59(1 \mathrm{H}, \mathrm{bs}, \mathrm{CONHNH}), 7.80-7.78(2 \mathrm{H}, \mathrm{d}$, J 7, Ar-H), $7.74-7.71$ (2H, d, J 7, Ar-H), $7.45-7.40$ (2H, t, J 7, Ar-H), $7.35-$ 7.31 (3H, t, J 7, Ar-H), 6.95 - 6.91 (1H, m, Ar-H), 5.99 (1H, bd, J 6, FmocNHCH), 4.47 (2H, d, J 7, $\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}$-Fluorene), 4.36 (1H, m, FmocNHCH), 4.28 - 4.21 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{FmocNHCHCH}, \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}$-Fluorene), $1.35\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.14(3 \mathrm{H}$, d, J 6, NHCHCHCH3); $\delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 169.8$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.1 ( $\mathrm{C}=\mathrm{ONH}$ ), 147.8 (ipso-Ar-C), 143.8 (ipso-Ar-C), 143.8 (ipso-Ar-C), 143.8 (ipso-Ar-C), 141.4 (ipso-Ar-C), 141.3 (ipso-Ar-C), 129.2 (Ar-C), 127.8 (Ar-C), 127.1 (Ar-C), 125.1 (Ar-C), 121.4 (Ar-C), 120.0 (Ar-C), 113.8 ( $\mathrm{Ar}-\mathrm{C}$ ), 75.9 ((CH3)3 CO ), 67.1 ( $\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}-$ Fluorene), 66.6 (FmocNHCHCH), 58.2 (FmocNHCH), 47.2 (CFluorene), $28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) 17.2\left(\mathrm{NHCHCHCH}_{3}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 510\left(\mathrm{MNa}^{+}\right) ;$HRMS $\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}, 510.2423\left(\mathrm{C}_{29} \mathrm{H}_{33} \mathrm{~N}_{3} \mathrm{O}_{4} \mathrm{Na}\right.$ requires 510.2421).

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## tert-Butyl (S)-2-(2-phenylhydrazine-1-carbonyl)pyrrolidine-1-carboxylate 67f



Following the general procedure outlined, $N$-Boc-L-proline ( $0.40 \mathrm{~g}, 1.86 \mathrm{mmol}$ ) and phenylhydrazine ( $0.20 \mathrm{~mL}, 2.04 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a light brown solid (0.29 g, 52\%); Rf 0.40 (DCM/EtOH/NH3 200:6:1); m.p. $151-153^{\circ} \mathrm{C}$; $v_{\max } 3274(\mathrm{~N}-\mathrm{H})$, 2972, 2874 (C-H), 1665, 1603 (C=O), 1538, 1404, 1398, 1162, 1134, 756, 698 $\mathrm{cm}^{-1}$; $\delta \mathrm{H}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.69$ (1H, bs, CONHNH), $7.17-7.12$ ( $2 \mathrm{H}, \mathrm{t}, \mathrm{J} 7$, ArH), 6.84 (1H, d, J 7, Ar-H), 6.77 - 6.74 (2H, dd, J 8, 1, Ar-H), 6.02 (1H, bs, ArNH), 4.31 (1H, m, CH2BocNCH), 3.41 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{BocNCH}$ ), 2.31 ( $1 \mathrm{H}, \mathrm{m}$, BocNCHCHH), 1.87 (3H, m, BocNCHCHH, $\mathrm{CHCH}_{2} \mathrm{CH}_{2}$ ), $1.44\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; ठс (100 MHz, CDCl3) 148.0 (ipso-Ar-C), 129.1 (Ar-C), 120.0 (Ar-C), 113.6 (Ar-C), $80.5\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 58.4\left(\mathrm{CH}_{2} \mathrm{BocNCH}\right), 47.2\left(\mathrm{CH}_{2} \mathrm{BocNCH}\right) 28.5\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 27.3$ $\left(\mathrm{BocNCHCH}_{2}\right), 24.6\left(\mathrm{CHCH}_{2} \mathrm{CH}_{2}\right) ; m / z\left(\mathrm{ES}^{+}\right) 328\left(\mathrm{MNa}^{+}\right) ; \mathrm{HRMS}\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}$, $328.1659\left(\mathrm{C}_{16} \mathrm{H}_{23} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 328.1632).
(9H-fluoren-9-yl)methyl
(S)-(1,4-dioxo-1-(2-phenylhydrazineyl)-4-(tritylamino)butan-2-yl)carbamate 68e


Following the general procedure outlined, $N$-Fmoc- $N$-(trityl)-L-asparagine ( 0.40 g , 0.67 mmol ) and phenylhydrazine ( $0.07 \mathrm{~mL}, 0.74 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a light brown solid ( $0.23 \mathrm{~g}, 49 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.40$ (DCM/EtOH/NH3 200:6:1); m.p. $144-146{ }^{\circ} \mathrm{C}$; $v \max 3292$ (N-H), 1696, 1655 (C=O), 1528, 1493, 1448, 1259, 1085, 839, 738 cm-1; $\delta_{H}(300 \mathrm{MHz}, ~ D M S O-d 6) 9.83$ (1H, bd, J 2, CONHNH), 8.59 (1H, bs,

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CONH-trityl), 7.92 - 7.90 (2H, d, J7, Ar-H), 7.77 - 7.71 (4H, m, Ar-H), 7.46 - 7.39 (2H, q, J 7, Ar-H), $7.25-7.19$ (15H, m, Ar-H), $7.13-7.09$ (2H, t, J 7, Ar-H), 6.73 - $6.66(3 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}), 4.45\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene $), 4.31-4.22(2 \mathrm{H}, \mathrm{m}$, FmocNHCH, $\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}$-Fluorene), 2.84 (1H, m, FmocNHCHCHH), $2.62(1 \mathrm{H}$, dd, J 14, 4, FmocNHCHCHH); ठc (100 MHz, DMSO-d6) 171.7 (C=ONHNH), 169.1 (C=NH-trityl), 156.2 ( $C=O N H$ ), 149.7 (ipso-Ar-C), 144.3 (ipso-Ar-C), 144.2 (ipso-Ar-C), 141.2 (ipso-Ar-C), 141.2 (ipso-Ar-C), 129.1 (Ar-C), 128.1 (Ar-C), 127.6 (Ar-C), 127.6 (Ar-C), 127.6 (Ar-C), 126.8 (Ar-C), 125.8 (Ar-C), 125.7 (ArC), 120.6 (Ar-C), 118.9 (Ar-C), 112.6 (Ar-C), 69.9 (C-(trityl), $66.4\left(\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}-\right.$ Fluorene), 51.5 (FmocNHCH), 47.2 (C-Fluorene) 39.2 ( $\mathrm{FmocNHCHCH}_{2}$ ); m/z $\left(\mathrm{ES}^{+}\right) 687\left(\mathrm{MH}^{+}\right), 709\left(\mathrm{MNa}^{+}\right) ; \mathrm{HRMS}\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}, 709.2783$ $\left(\mathrm{C}_{44} \mathrm{H}_{38} \mathrm{~N}_{4} \mathrm{O}_{4} \mathrm{Na}\right.$ requires 709.2785).

## tert-Butyl (S)-(3-(4-(benzyloxy)phenyl)-1-oxo-1-(2-phenylhydrazineyl) propan-2-yl)carbamate 67j



Following the general procedure outlined, N -Boc-O-benzyl-L-tyrosine ( 0.40 g , $1.08 \mathrm{mmol})$ and phenylhydrazine ( $0.12 \mathrm{~mL}, 1.19 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a light brown solid ( $0.35 \mathrm{~g}, 70 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.37$ (DCM/EtOH/NH3 200:6:1); m.p. $130-133{ }^{\circ} \mathrm{C}$; $V_{\max } 3339,3229$ (N-H), 1689, 1665 (C=O), 1521, 1489, 1247, 1164, 1044, 731, $695 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 7.75 (1H, bs, CONHNH), 7.46 - 7.44 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-$ H), 7.42 (1H, m, Ar-H), $7.39-7.35$ (2H, m, Ar-H), $7.17-7.14$ (4H, m, Ar-H), 6.96 - 6.86 (4H, m, Ar-H), 6.65 (1H, d, J 7, Ar-H), 5.07 (2H, m, Ar-CH2O), 4.44 (1H, q, $J 7$, BocNHCH), 3.07 (2H, dd, J 7, 3, Ar-CH2CH), 1.47 (9H, s, ( $\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}$ ); ठc (100 MHz, CDCl3) 171.4 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 158.0 ( $\mathrm{C}=\mathrm{ONH}$ ), 147.4 (ipso-Ar-C), 136.9 (ipso-Ar-C), 130.5 (Ar-C), 129.1 (Ar-C), 128.6 (Ar-C), 128.0 (Ar-C), 127.5 (Ar-C), 121.3 ( $\mathrm{Ar}-\mathrm{C}$ ), 115.2 ( $\mathrm{Ar}-\mathrm{C}), 113.3(\mathrm{Ar}-\mathrm{C}), 70.1\left(\mathrm{Ar}-\mathrm{CH}_{2} \mathrm{O}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right)$ $462\left(\mathrm{MH}^{+}\right), 484\left(\mathrm{MNa}^{+}\right)$; HRMS $\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MH}^{+}, 462.2396\left(\mathrm{C}_{27} \mathrm{H}_{32} \mathrm{~N}_{3} \mathrm{O}_{4}\right.$ requires 462.2387).

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## tert-Butyl (2-oxo-2-((2-oxo-2-(2-phenylhydrazineyl)ethyl)amino)ethyl) carbamate 67k



Following the general procedure outlined, $N$-Boc-glycylglycine ( $0.40 \mathrm{~g}, 1.72 \mathrm{mmol}$ ) and phenylhydrazine ( $0.19 \mathrm{~mL}, 1.89 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a white solid ( $0.38 \mathrm{~g}, 69 \%$ ) as a mixture of rotamers [5:1]; Rf 0.11 (DCM/EtOH/NH3 200:6:1); m.p. 157-160 ${ }^{\circ}$ C; $V_{\max }$ 3296, 3235 (N-H), 2975 (C-H), 1662, 1646 (C=O), 1552, 1496, 1309, 1234, 1168, 946, $753 \mathrm{~cm}^{-1}$; all data provided for the major rotamer; $\delta \mathrm{H}(300 \mathrm{MHz}$, DMSO-d6) 9.59 (1H, bd, J 2, CONHNH), 8.17 (1H, bt, J 6, CONHCH 2 ), 7.70 (1H, bd, J 2, BocNHCH2), $7.15-7.10(2 H, t, J 7, A r-H), 6.73-6.70(3 H, d d, J 8,2$, Ar-H), 3.82 (2H, d, J 5, CONHCH2), 3.62 ( $2 \mathrm{H}, \mathrm{d}, \mathrm{J} 5$, BocNHCH2), 1.38 ( $9 \mathrm{H}, \mathrm{s}$, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; ठс ( 100 MHz , DMSO-d6) 170.4 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 169.2 ( $\left.\mathrm{C}=\mathrm{ONH}\right), 156.3$ (C=ONH), 149.6 (ipso-Ar-C), 129.1 (Ar-C), 118.9 (Ar-C), 112.7 (Ar-C), 78.6 $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 43.9\left(\mathrm{BocNHCH}_{2}\right), 41.5\left(\mathrm{CONHCH}_{2}\right), 28.7\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right)$ $323\left(\mathrm{MH}^{+}\right), 345\left(\mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}, 345.1548\left(\mathrm{C}_{15} \mathrm{H}_{22} \mathrm{~N}_{4} \mathrm{O}_{4} \mathrm{Na}\right.$ requires 345.1533 ).
tert-Butyl (S)-(2-oxo-1-phenyl-2-(2-phenylhydrazineyl)ethyl)carbamate 67I


Following the general procedure outlined, $N$-Boc-L-phenylglycine ( $0.40 \mathrm{~g}, 1.59$ mmol ) and phenylhydrazine ( $0.17 \mathrm{~mL}, 1.75 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a white solid ( 0.31 g, $57 \%$ ); Rf 0.38 (DCM/EtOH/NH $300: 6: 1$ ); m.p. $155-158{ }^{\circ} \mathrm{C}$; $v m a x 3327$, 3273 (N-H), 17717, 1662 (C=O), 1508, 1497, 1369, 1256, 1159, 1027, 753, 697 cm-1; ठн (300 MHz, DMSO-d6) 8.23 (1H, bs, CONHNH), 7.41 - 7.35 (4H, m, ArH), 7.26 (1H, m, Ar-H), 7.17 - 7.11 (2H, m, Ar-H), 7.05 (1H, t, J 7, Ar-H), 6.64 $6.60(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}), 5.41$ (1H, m, CONHCH), $1.43\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; ठc ( 100 MHz , DMSO-d6) 170.8 ( $C=O N H N H$ ), 155.4 ( $C=O N H$ ), 147.5 (ipso-Ar-C), 146.2 (ipso-

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Ar-C), 137.4 (ipso-Ar-C), 129.2 (Ar-C), 129.1 (Ar-C), 129.1 (Ar-C), 128.8 (Ar-C), 128.6 (Ar-C), 127.9 (Ar-C), 127.2 (Ar-C), 113.4 ( $\mathrm{Ar}-\mathrm{C}$ ), 112.9 (Ar-C), 80.6 $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 57.1$ (BocNHCH), $28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 364\left(\mathrm{MNa}^{+}\right) ;$HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}, 364.1649\left(\mathrm{C}_{19} \mathrm{H}_{23} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 364.1632).
tert-Butyl (S)-(1-oxo-4-phenyl-1-(2-phenylhydrazineyl)butan-2-yl)carbamate 67m


Following the general procedure outlined, $N$-Boc-L-homophenylalanine ( 0.40 g , 1.43 mmol ) and phenylhydrazine ( $0.16 \mathrm{~mL}, 1.58 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a pale yellow solid ( $0.42 \mathrm{~g}, 79 \%$ ); Rf 0.38 (DCM/EtOH/NH3 200:6:1); m.p. $129-132{ }^{\circ} \mathrm{C}$; $v_{\max } 3292$ (N-H), 1678, 1665 (C=O), 1518, 1495, 1366, 1278, 1164, 748, $693 \mathrm{~cm}-$ 1; бн (300 MHz, CDCl3) 8.44 (1H, bs, CONHNH), 7.40 (1H, m, Ar-NH), 7.19 $7.16(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}), 7.13(1 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-H), 7.10-7.04(4 \mathrm{H}, \mathrm{m}, \operatorname{Ar}-H), 6.80(1 \mathrm{H}, \mathrm{t}, \mathrm{J}$ 7, Ar-H), 6.71 - 6.68 (2H, dd, J 8, 1, Ar-H), 5.22 (1H, bd, J 8, CONHCH), 4.18 (1H, m, CONHCH), 2.61 ( $2 \mathrm{H}, \mathrm{t}, \mathrm{J} 7, \mathrm{CH}_{2} \mathrm{CH}_{2}-\mathrm{Ar}$ ), 2.11 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{CONHCHCHH}$ ), 1.90 (1H, m, CONHCHCHH), 1.38 ( $\left.9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 172.3$ (C=ONHNH), 155.9 (C=ONH), 147.7 (ipso-Ar-C), 140.7 (ipso-Ar-C), 129.6 (Ar-C), 129.4 (Ar-C), 129.1 (Ar-C), 128.6 (Ar-C), 128.4 (Ar-C), 128.3 (Ar-C), 126.2 (ArC), 121.1 (Ar-C), 115.4 (Ar-C), 113.5 (Ar-C), 112.9 (Ar-C), 80.6 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 52.6$ (BocNHCH), $33.6\left(\mathrm{CONHCHCH}_{2}\right), 31.8\left(\mathrm{CH}_{2} \mathrm{CH}_{2}-\mathrm{Ar}\right), 28.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right)$ $392\left(\mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}, 392.1964\left(\mathrm{C}_{21} \mathrm{H}_{27} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 392.1945).

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tert-Butyl (R)-(1-oxo-1-(2-phenylhydrazineyl)propan-2-yl)carbamate 67n


Following the general procedure outlined, $N$-Boc-D-alanine ( $0.40 \mathrm{~g}, 2.11 \mathrm{mmol}$ ) and phenylhydrazine ( $0.23 \mathrm{~mL}, 2.33 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a light brown solid (0.29 g, 49\%); Rf 0.32 (DCM/EtOH/NH3 200:6:1); m.p. $82-85^{\circ} \mathrm{C} ;$ vmax 3433, 3289 ( $\mathrm{N}-$ H), 1689, 1674 (C=O), 1539, 1493, 1370, 1244, 1156, 1022, 763, 702 cm-1; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.21 (1H, bs, CONHNH), $7.24-7.19$ (3H, m, Ar-H), 6.92 (1H, t, J 7, 1, Ar-H), $6.83-6.80$ (1H, m, Ar-H), 5.02 (1H, bd, J7, CONHCH), 4.31 (1H, m, CONHCH), $1.47\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.41\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J} 7, \mathrm{CONHCHCH}_{3}\right)$; $\delta_{\mathrm{c}}(100$ $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 173.1 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.8 ( $\mathrm{C}=\mathrm{ONH}$ ), 147.8 (ipso-Ar-C), 129.1 (ArC), 121.0 (Ar-C), 115.5 (Ar-C), 1113.5 (Ar-C), 112.9 ( $\mathrm{Ar}-\mathrm{C}), 80.4$ (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 48.7$ ( BocNHCH ), $28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) 18.0\left(\mathrm{CONHCHCH}_{3}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 302\left(\mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}, 302.1489\left(\mathrm{C}_{14} \mathrm{H}_{21} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 302.1475).
tert-Butyl $\quad(R)$-(4-(methylthio)-1-oxo-1-(2-phenylhydrazineyl)butan-2yl)carbamate 670


Following the general procedure outlined, $N$-Boc-D-methionine ( $0.40 \mathrm{~g}, 1.60$ mmol ) and phenylhydrazine ( $0.17 \mathrm{~mL}, 1.77 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow solid ( $0.17 \mathrm{~g}, 31 \%$ ); $\mathrm{R}_{\mathrm{f}} 0.38$ (DCM/EtOH/ $\mathrm{NH}_{3}$ 200:6:1); m.p. $119-122^{\circ} \mathrm{C}$; $v \max 3329$, 3258 (N-H), 1685, 1655 (C=O), 1523, 1495, 1308, 1249, 1166, 1055, 764, 689 cm-1; $\delta_{\text {н }}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.70$ (1H, bs, CONHNH), 7.13 - 7.09 (2H, t, J7, ArH), 6.82 (1H, t, J 7, Ar-H), 6.71 - 6.69 (2H, d, J 8, Ar-H), 6.18 (1H, bd, J 3, NHAr), 5.42 (1H, bd, J 5, CONHCH), 4.37 (1H, q, J 7, CONHCH), $2.46(2 \mathrm{H}, \mathrm{t}, J 7$, $\left.\mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right), 2.09\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right), 1.99\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 1.89$ $\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right), 1.37\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 172.1$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.9 ( $\mathrm{C}=\mathrm{ONH}$ ), 147.7 (ipso-Ar-C), 129.2 ( $\mathrm{Ar}-\mathrm{C}$ ), 121.1 (Ar-C),

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113.5 (Ar-C), 112.9 (Ar-C), $80.6 \quad\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 52.0$ (BocNHCH), 31.4 $\left(\mathrm{CONHCHCH}_{2}\right), 30.1\left(\mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right), 28.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 15.4\left(\mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right)$; $m / z\left(\mathrm{ES}^{+}\right) 362\left(\mathrm{MNa}^{+}\right) ;$HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}$, $362.1536\left(\mathrm{C}_{16} \mathrm{H}_{25} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{SNa}\right.$ requires 362.1509 ).

## tert-Butyl $y l$ carbamate 67p

(R)-(1-oxo-3-phenyl-1-(2-phenylhydrazineyl)propan-2-


Following the general procedure outlined, $N$-Boc-D-phenylalanine ( $0.40 \mathrm{~g}, 1.51$ mmol ) and phenylhydrazine ( $0.16 \mathrm{~mL}, 1.66 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a brown gum ( 0.14 g, 25 \%); Rf 0.35 (DCM/EtOH/NH3 200:6:1); vmax 3314 (N-H), 1688, 1667 (C=O), 1521, 1496, 1366, 1251, 1167, 749, 694 cm-1; бн (300 MHz, $\left.\mathrm{CDCl}_{3}\right) 8.05$ (1H, bs, CONHNH), 7.22 - 7.18 (3H, m, Ar-H), 7.12 - 7.09 (2H, m, Ar-H), 7.08 $-7.03(2 \mathrm{H}, \mathrm{t}, \mathrm{J} 7, \operatorname{Ar}-H), 6.78(1 \mathrm{H}, \mathrm{t}, J 7$, Ar-H), $6.49-6.46(2 \mathrm{H}, \mathrm{d}, ~ J 7$, Ar-H), 5.16 (1H, bd, J 8, CONHCH), 4.42 (1H,m, CONHCH), 3.00 (1H, dd, J 7, 2, $\left.\mathrm{CONHCHCH}_{2}\right), 1.35\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; ~ \delta c ~\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 171.6$ (C=ONHNH), 155.6 (C=ONH), 147.5 (ipso-Ar-C), 136.3 (ipso-Ar-C), 129.4 (Ar-C), 129.1 (Ar-C), 128.8 (Ar-C), 127.0 (Ar-C), 121.0 (Ar-C), 113.5 (Ar-C), 112.9 (Ar-C), 80.6 $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 54.5(\mathrm{BocNHCH}), 38.2\left(\mathrm{CONHCHCH}_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right)$ $378\left(\mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}, 378.1811\left(\mathrm{C}_{20} \mathrm{H}_{25} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 378.1788).
tert-Butyl (2-(2-(2-bromophenyl)hydrazineyl)-2-oxoethyl)carbamate 67Ce


Following the general procedure outlined, $N$-Boc-glycine ( $0.40 \mathrm{~g}, 2.28 \mathrm{mmol}$ ) and 2-bromophenylhydrazine hydrochloride ( $0.56 \mathrm{~g}, 2.51 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow oil ( $0.39 \mathrm{~g}, 50 \%$ ) as a mixture of rotamers [3:1]; Rf 0.46 (DCM/EtOH/NH3 200:6:1); $V_{\max } 3345$ (N-H), 2979 (C-H), 1703, 1653 (C=O), 1595, 1465, 1367, 1277, 1155,

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1023, $750 \mathrm{~cm}^{-1}$; all data provided for the major rotamer; $\boldsymbol{\delta H}^{( } 300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $8.52(1 \mathrm{H}, \mathrm{bs}, \mathrm{CONHNH}), 7.43$ (1H, m, Ar-H), 7.16 (1H, td, J 8,1, Ar-H), 6.85 (1H, m, Ar-H), 6.77 ( $1 \mathrm{H}, \mathrm{td}, ~ J 7,1, ~ A r-H), 6.39(1 \mathrm{H}, \mathrm{bd}, \mathrm{J} 3$, Ar-NH), 5.45 (1H, bs, $\mathrm{BocNHCH}_{2}$ ), $3.88\left(2 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{BocNHCH}\right.$ ), $1.45\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; ठc ( 100 MHz , $\mathrm{CDCl}_{3}$ ) 169.8 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.5 ( $\mathrm{C}=\mathrm{ONH}$ ), 144.5 ( $\mathrm{ipso}-\mathrm{Ar}-\mathrm{C}$ ), 132.6 ( $\mathrm{Ar}-\mathrm{C}$ ), 128.3 (Ar-C), 121.9 (Ar-C), 113.7 (Ar-C), 109.0 (ipso-Ar-C), 80.8 (( $\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}$ ), 43.3 ( $\mathrm{BocNHCH}_{2}$ ), $28.4\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}$ ); $\left.m / z\left(\mathrm{ES}^{+}\right) 366\left({ }^{[99} \mathrm{Br}\right] \mathrm{MNa}{ }^{+}\right), 368\left({ }^{81} \mathrm{Br}\right]$ $\mathrm{MNa}^{+}$), 709 ([ $\left.{ }^{79,79} \mathrm{Br}\right] 2 \mathrm{M}+\mathrm{Na}^{+}$), 711 ([ $\left.{ }^{79,81} \mathrm{Br}\right] 2 \mathrm{M}+\mathrm{Na}^{+}$), 713 ( $\left.{ }^{81,81} \mathrm{Br}\right] 2 \mathrm{M}+\mathrm{Na}^{+}$); HRMS (ES ${ }^{+}$) Found $\left[{ }^{79} \mathrm{Br}\right] \mathrm{MH}^{+}, 366.0438\left(\mathrm{C}_{13} \mathrm{H}_{18}{ }^{79} \mathrm{BrN}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 366.0424).
tert-Butyl (2-(2-(3-bromophenyl)hydrazineyl)-2-oxoethyl)carbamate 67Ck


Following the general procedure outlined, $N$-Boc-glycine ( $0.40 \mathrm{~g}, 2.28 \mathrm{mmol}$ ) and 3-bromophenylhydrazine hydrochloride ( $0.56 \mathrm{~g}, 2.51 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a brown oil ( $0.39 \mathrm{~g}, 49 \%$ ) as a mixture of rotamers [4:1]; $\mathrm{Rf}_{\mathrm{f}} 0.32$ (DCM/EtOH/NH3 200:6:1); $v_{\max } 3276$ (N-H), 2979 (C-H), 1676 (C=O), 1595, 1474, 1367, 1249, 1157, 1047, 849, $768 \mathrm{~cm}^{-1}$; all data provided for the major rotamer; $\delta_{H}(300 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 8.63(1 \mathrm{H}, \mathrm{bs}, \mathrm{CONHNH}), 7.01(1 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{H}), 6.98(1 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}), 6.92(1 \mathrm{H}$, m, Ar-H), 6.70 (1H, m, Ar-H), 6.40 (1H, bs, Ar-NH), 5.55 (1H, bs, BocNHCH2), $3.84\left(2 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{BocNHCH}_{2}\right), 1.45\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 170.5$ (C=ONHNH), 156.5 (C=ONH), 149.1 (ipso-Ar-C), 130.5 (Ar-C), 123.8 (ipso-Ar-C), 123.1 (Ar-C), 116.0 (Ar-C), $112.0(\mathrm{Ar}-\mathrm{C}), 80.8\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 43.2\left(\mathrm{BocNHCH}_{2}\right)$, $\left.\left.28.3\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 366\left({ }^{79} \mathrm{Br}\right] \mathrm{MNa}{ }^{+}\right), 368$ ([ $\left.{ }^{81} \mathrm{Br}\right] \mathrm{MNa}^{+}$), 709 ( $\left.{ }^{799,79} \mathrm{Br}\right]$ $2 \mathrm{M}+\mathrm{Na}^{+}$), $\left.\left.711\left({ }^{[79,81} \mathrm{Br}\right] 2 \mathrm{M}+\mathrm{Na}^{+}\right), 713\left({ }^{81,81} \mathrm{Br}\right] 2 \mathrm{M}+\mathrm{Na}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\left.{ }^{79}{ }^{3} \mathrm{Br}\right] \mathrm{MH}^{+}, 366.0452\left(\mathrm{C}_{13} \mathrm{H}_{18}{ }^{79} \mathrm{BrN}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 366.0423$)$.

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tert-Butyl carbamate 67Di
(2-(2-(4-(methylsulfonyl)phenyl)hydrazineyl)-2-oxoethyl)


Following the general procedure outlined, $N$-Boc-glycine ( $0.20 \mathrm{~g}, 1.14 \mathrm{mmol}$ ) and 4-(methylsulphonyl)phenylhydrazine hydrochloride ( $0.28 \mathrm{~g}, 1.26 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a light brown oil ( $0.33 \mathrm{~g}, 84 \%$ ); $\mathrm{R}_{\mathrm{f}} 0.44$ (DCM/EtOH/NH3 200:6:1); m.p. 158-161 ${ }^{\circ} \mathrm{C}$; $V_{\max } 3359,3291$ (N-H), 1678, 1600 (C=O), 1506, 1285, 1139, 960, $773 \mathrm{~cm}^{-1}$; $\delta_{\mathrm{H}}(300 \mathrm{MHz}$, DMSO-d6) 9.89 (1H, bs, CONHNH), 8.59 (1H, bs, ArNH), $7.64-7.61(2 H, d, J 8, A r-H), 7.15\left(1 H, b t, J 6, \operatorname{BocNHCH}_{2}\right), 6.84-6.81$ (2H, d, J 8, Ar-H), $3.63\left(2 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{BocNHCH}_{2}\right), 3.08\left(3 \mathrm{H}, \mathrm{s}, \mathrm{SO}_{2} \mathrm{CH}_{3}\right), 1.40(9 \mathrm{H}$, s, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 169.9$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.4 ( $\left.\mathrm{C}=\mathrm{ONH}\right), 153.9$ (ipso-Ar-C), 129.5 (Ar-C), 128.9 (Ar-C), 111.6 (Ar-C), 78.6 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 44.7$ $\left.\left(\mathrm{SO}_{2} \mathrm{CH}_{3}\right), 42.5\left(\mathrm{BocNHCH}_{2}\right), 28.7\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 366$ (MNah+$) ; ~ H R M S$ $\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}, 366.1105\left(\mathrm{C}_{14} \mathrm{H}_{21} \mathrm{~N}_{3} \mathrm{O}_{5} \mathrm{~S}\right.$ requires 366.1094).
tert-Butyl (2-oxo-2-(2-(p-tolyl)hydrazineyl)ethyl)carbamate 67Ea


Following the general procedure outlined, $N$-Boc-glycine ( $0.30 \mathrm{~g}, 1.71 \mathrm{mmol}$ ) and 4-methylphenylhydrazine hydrochloride ( $0.29 \mathrm{~g}, 1.89 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a brown gum ( $0.35 \mathrm{~g}, 72 \%$ ) as a mixture of rotamers [3:1]; Rf 0.35 (DCM/EtOH/NH3 200:6:1); $V_{\max } 3289$ (N-H), 2979, 2929 (C-H), 1676, 1615 (C=O), 1511, 1366, 1247, 1161, $730 \mathrm{~cm}^{-1}$; all data provided for the major rotamer; $\delta_{H}(300 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 8.36(1 \mathrm{H}, \mathrm{bs}, \mathrm{CONHNH}), 6.93-6.91(2 \mathrm{H}, \mathrm{d}, \mathrm{J} 8, \mathrm{Ar}-\mathrm{H}), 6.64-6.61(2 \mathrm{H}$, d, J 8, Ar-H), 5.79 (1H, bs, Ar-NH), 5.37 (1H, bs, BocNHCH2), 3.76 (2H, d, J 5, $\mathrm{BocNHCH}_{2}$ ), 2.17 (3H, s, $\mathrm{Ar}^{2} \mathrm{CH}_{3}$ ), 1.38 ( $\left.9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ 169.9 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.4 ( $\mathrm{C}=\mathrm{ONH}$ ), 145.3 (ipso-Ar-C), 129.9 ( $\mathrm{Ar}-\mathrm{C}$ ), 129.7 ( $\mathrm{Ar}-$ C), 113.7 (Ar-C), 112.8 (Ar-C), $80.6\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 43.2\left(\mathrm{BocNHCH}_{2}\right), 28.3$

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$\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 20.6\left(\mathrm{Ar}^{-} \mathrm{CH}_{3}\right) ; m / z\left(\mathrm{ES}^{+}\right) 302\left(\mathrm{MNa}^{+}\right) ;$HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}$, $302.1492\left(\mathrm{C}_{14} \mathrm{H}_{21} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 302.1481).

## tert-Butyl (2-(2-(4-isopropylphenyl)hydrazineyl)-2-oxoethyl)carbamate 67Eg



Following the general procedure outlined, $N$-Boc-glycine ( $0.40 \mathrm{~g}, 2.28 \mathrm{mmol}$ ) and 4-isopropylphenylhydrazine hydrochloride ( $0.47 \mathrm{~g}, 2.51 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a brown solid ( $0.34 \mathrm{~g}, 48 \%$ ) as a mixture of rotamers [2:1]; Rf 0.37 (DCM/EtOH/NH3 200:6:1); m.p. $156-159{ }^{\circ} \mathrm{C}$; $v_{\max } 3357,3263$ (N-H), 2958(C-H), 1650, 1614 (C=O), 1512, 1473, 1367, 1243, 1166, 1051, $814 \mathrm{~cm}^{-1}$; all data provided for the major rotamer; $\delta_{н}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.24$ (1H, bs, CONHNH), 7.09 - 7.06 (2H, d, J 8, Ar-H), 6.77 - 6.74 (2H, d, J 8, Ar-H), 6.07 (1H, bs, Ar-NH), 5.31 (1H, bt, J 6, BocNHCH2), 3.86 (2H, d, J 5, BocNHCH2), 2.85 (1H, m, Ar-CH(CH3 $)_{2}$ ), 1.46 (9H, s, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.20-1.18\left(6 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{Ar}-\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ 169.7 (C=ONHNH), 145.5 (ipso-Ar-C), 141.9 (ipso-Ar-C), 127.4 (Ar-C), 127.1 (ArC), 113.8 ( $\mathrm{Ar}-\mathrm{C}$ ), $112.8(\mathrm{Ar}-\mathrm{C}), 80.7\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 43.2\left(\mathrm{BocNHCH}_{2}\right), 33.3$ ( $\mathrm{Ar}-$ $\left.\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $\left.28.4\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $24.1\left(\mathrm{Ar-CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; m / z\left(\mathrm{ES}^{+}\right) 330\left(\mathrm{MNa}{ }^{+}\right) ;$HRMS $\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}, 330.1803\left(\mathrm{C}_{16} \mathrm{H}_{25} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 330.1788).

## tert-Butyl (2-(2-(4-cyanophenyl)hydrazineyl)-2-oxoethyl)carbamate 67Cw



Following the general procedure outlined, $N$-Boc-glycine ( $0.40 \mathrm{~g}, 2.28 \mathrm{mmol}$ ) and 4-cyanophenylhydrazine hydrochloride ( $0.43 \mathrm{~g}, 2.51 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a brown solid ( $0.54 \mathrm{~g}, 82 \%$ ) as a mixture of rotamers [5:1]; Rf 0.72 ( $n$-hex/EtOAc 2:1); m.p. 113 - $116{ }^{\circ} \mathrm{C}$; $v_{\max } 3331,3276$ (N-H), 2219 (C=N), 1683, 1605 (C=O), 1519, 1250, 1163, $833 \mathrm{~cm}^{-1}$; all data provided for the major rotamer; $\delta_{H}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ 8.33 (1H, bs, CONHNH), 7.49 - 7.47 (2H, d, J 8, Ar-H), 6.84 - 6.81 (2H, d, J 8,

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Ar-H), 6.45 (1H, bd, J 2, Ar-NH), 5.29 (1H, bt, J 6, BocNHCH $)$, 3.90 (2H, d, J 6, $\left.\mathrm{BocNHCH}_{2}\right), 1.46\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 170.1$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 151.3 (C=ONH), 133.7 (Ar-C), 119.5 ( $C \equiv N$ ), 112.9 (Ar-C), 103.4 (ipso-Ar-C), 81.3 $\left.\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 43.6\left(\mathrm{BocNHCH}_{2}\right), 28.3\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 313\left(\mathrm{MNa}^{+}\right) ;$HRMS $\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}, 313.1285\left(\mathrm{C}_{14} \mathrm{H}_{18} \mathrm{~N}_{4} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 313.1271).

## tert-Butyl (2-(2-(3,5-bis(trifluoromethyl)phenyl)hydrazineyl)-2-oxoethyl) carbamate 67Bs



Following the general procedure outlined, $N$-Boc-glycine ( $0.40 \mathrm{~g}, 2.28 \mathrm{mmol}$ ) and 3,5-bis(trifluoromethyl)phenyl)hydrazine hydrochloride ( $0.70 \mathrm{~g}, 2.51 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a brown solid ( $0.52 \mathrm{~g}, 56 \%$ ); Rf 0.56 ( $n$-hex/EtOAc 4:1); m.p. 111 $114{ }^{\circ} \mathrm{C}$; $V_{\max } 3342,3237$ (N-H), 1686, 1678 (C=O), 1524, 1387, 1280, 1124, 879 $\mathrm{cm}^{-1}$; ठн $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.49(1 \mathrm{H}, \mathrm{bs}, \mathrm{CONHNH}), 7.35(1 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{H}), 7.20(2 \mathrm{H}$, s, Ar-H), 6.58 (1H, bs, Ar-NH), 5.30 (1H, bs, BocNHCH2), 3.92 (2H, d, J 6, $\left.\mathrm{BocNHCH}_{2}\right), 1.47\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta с\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 170.4$ (C=ONHNH), 149.9 (C=ONH), 132.7 (ipso-Ar-C), 125.1 (ipso-Ar-C), 121.4 (CF3), 114.3 (Ar-C), 112.9 ( $\left.\mathrm{Ar}-\mathrm{C}), 81.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 43.5\left(\mathrm{BocNHCH}_{2}\right), 28.2\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; ~ \delta ғ(300 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right)$-63.2 $\left(\mathrm{CF}_{3}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 424\left(\mathrm{MNa}^{+}\right) ;$HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}, 424.1069$ $\left(\mathrm{C}_{15} \mathrm{H}_{17} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~F}_{6} \mathrm{Na}\right.$ requires 424.1066).

## tert-Butyl (2-oxo-2-(2-(o-tolyl)hydrazineyl)ethyl)carbamate 67Do



Following the general procedure outlined, $N$-Boc-glycine ( $0.40 \mathrm{~g}, 2.28 \mathrm{mmol}$ ) and 2-methylphenylhydrazine hydrochloride ( $0.39 \mathrm{~g}, 2.51 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a brown solid ( $0.53 \mathrm{~g}, 84 \%$ ) as a mixture of rotamers [3:1]; Rf 0.56 ( $n$-hex/EtOAc 4:1); m.p.

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$90-92^{\circ} \mathrm{C} ; V_{\max } 3299(\mathrm{~N}-\mathrm{H}), 1672$ (C=O), 1529, 1368, 1286, 1161, $751 \mathrm{~cm}^{-1}$; all data provided for the major rotamer; $\delta_{H}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.33(1 \mathrm{H}, \mathrm{bs}$, CONHNH), 7.09 - 7.04 (2H, m, Ar-H), 6.83 - 6.78 (2H, m, Ar-H), 5.77 (1H, bs, Ar-NH), 5.38 (1H, bt, , J 5, BocNHCH2), 3.88 (2H, d, J 5, BocNHCH2), 2.21 (3H, s, Ar- $\mathrm{CH}_{3}$ ), 1.46 ( $\left.9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; бc (100 MHz, CDCl3) 169.6 ( $\mathrm{C=ONHNH}$ ), 155.9 (C=ONH), 145.3 (ipso-Ar-C), 130.8 (Ar-C), 126.9 (Ar-C), 121.0 (Ar-C), 112.1 (Ar-C), $\left.80.9\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 43.4\left(\mathrm{BocNHCH}_{2}\right), 28.3\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 17.0\left(\mathrm{Ar}-\mathrm{CH}_{3}\right)$; $m / z\left(\mathrm{ES}^{+}\right) 302\left(\mathrm{MNa}^{+}\right) ;$HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}$, $302.1493\left(\mathrm{C}_{14} \mathrm{H}_{2} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 302.1475 ).
tert-Butyl (2-oxo-2-(2-(m-tolyl)hydrazineyl)ethyl)carbamatecarbamate 67Du


Following the general procedure outlined, $N$-Boc-glycine ( $0.40 \mathrm{~g}, 2.28 \mathrm{mmol}$ ) and 3-methylphenylhydrazine hydrochloride ( $0.39 \mathrm{~g}, 2.51 \mathrm{mmol}$ ) were transformed following flash column chromatography (DCM/EtOH/ $\mathrm{NH}_{3}$ 200:6:1) into the title compound which was isolated as a brown oil ( $0.33 \mathrm{~g}, 51 \%$ ) as a mixture of rotamers [3:1]; Rf 0.37 (DCM/EtOH/NH3 200:6:1); $V_{\max } 3276$ (N-H), 2927 (C-H), 1656, 1606 (C=O), 1524, 1498, 1367, 1249, 1161, $777 \mathrm{~cm}^{-1}$; all data provided for the major rotamer; $\delta \mathrm{H}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.48(1 \mathrm{H}, \mathrm{bs}, \mathrm{CONHNH}), 7.10(1 \mathrm{H}, \mathrm{t}, \mathrm{J}$ 7, $\operatorname{Ar}-H$ ), 6.70 (1H, d, J 7, Ar-H), 6.59 (1H, s, Ar-H), 6.57 (1H, m, Ar-H), 6.16 (1H, bd, J 3, Ar-NH), 5.49 (1H, bs, BocNHCH2), 3.84 (2H, d, J 5, BocNHCH2), 2.25 (3H, s, Ar-CH3), 1.45 ( $\left.9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 170.0$ (C=ONHNH), 156.4 (C=ONH), 147.7 (ipso-Ar-C), 130.1 (Ar-C), 129.0 (Ar-C), 122.0 (Ar-C), 114.2 ( $\mathrm{Ar}-\mathrm{C}$ ), 110.6 ( $\mathrm{Ar}-\mathrm{C}$ ), $80.5\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 43.2\left(\mathrm{BocNHCH}_{2}\right)$, $\left.28.3\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 21.5\left(\mathrm{Ar}-\mathrm{CH}_{3}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 581\left(2 \mathrm{M}+\mathrm{Na}^{+}\right) ; \mathrm{HRMS}\left(\mathrm{ES}^{+}\right)$Found $2 \mathrm{M}+\mathrm{Na}^{+}, 581.3073\left(\mathrm{C}_{28} \mathrm{H}_{42} \mathrm{~N}_{6} \mathrm{O}_{6} \mathrm{Na}\right.$ requires 581.3058).

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tert-Butyl (2-(2-(4-bromophenyl)hydrazineyl)-2-oxoethyl)carbamate 67Cq


Following the general procedure outlined, $N$-Boc-glycine ( $0.40 \mathrm{~g}, 2.28 \mathrm{mmol}$ ) and 4-bromophenylhydrazine hydrochloride ( $0.56 \mathrm{~g}, 2.51 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a brown oil ( $0.49 \mathrm{~g}, 62 \%$ ) as a mixture of rotamers [3:1]; $\mathrm{Rf}_{\mathrm{f}} 0.51$ (DCM/EtOH/NH3 200:6:1); $V_{\max } 3369,3265$ (N-H), 2985 (C-H), 1698, 1652 (C=O), 1525, 1488, 1369, 1242, 1159, $812 \mathrm{~cm}^{-1}$; all data provided for the major rotamer; $\delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ 8.57 (1H, bs, CONHNH), 7.27 - 7.24 (2H, d, J 8, Ar-H), 6.66 - 6.63 (2H, d, J 8, Ar-H), 5.40 (1H, bs, BocNHCH2), 3.84 ( $2 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{BocNHCH}_{2}$ ), 1.45 ( $9 \mathrm{H}, \mathrm{s}$, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta с\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 170.3$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.5 ( $\mathrm{C}=\mathrm{ONH}$ ), 146.8 (ipso-Ar-C), 132.3 (Ar-C), 131.9 (ipso-Ar-C), 115.1 (Ar-C), 113.0 (Ar-C), 80.8 $\left.\left.\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 43.2\left(\mathrm{BocNHCH}_{2}\right), 28.3\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 366\left({ }^{79} \mathrm{Br}\right] \mathrm{MNa}^{+}\right), 368$ ( $\left.{ }^{81} \mathrm{Br}\right] \mathrm{MNa}^{+}$); HRMS (ES ${ }^{+}$) Found $\left.{ }^{79} \mathrm{Br}\right] \mathrm{MNa}^{+}, 366.0434\left(\mathrm{C}_{13} \mathrm{H}_{18}{ }^{79} \mathrm{BrN}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 366.0424 ).

## tert-Butyl carbamate 67Ay <br> (2-oxo-2-(2-(2-(trifluoromethyl)phenyl)hydrazineyl)ethyl)



Following the general procedure outlined, $N$-Boc-glycine ( $0.40 \mathrm{~g}, 2.28 \mathrm{mmol}$ ) and 2-trifluoromethylphenylhydrazine hydrochloride ( $0.53 \mathrm{~g}, 2.51 \mathrm{mmol}$ ) were transformed following trituration with DCM into the title compound which was isolated as a pale yellow solid ( $0.17 \mathrm{~g}, 22 \%$ ) as a mixture of rotamers [3:1]; $\mathrm{R}_{\mathrm{f}}$ 0.46 (n-hex/EtOAc 4:1); m.p. $123-125^{\circ} \mathrm{C}$; $v_{\max } 3335,3194$ (N-H), 1703, 1654 ( $\mathrm{C}=\mathrm{O}$ ), $1535,1322,1276,1101,763 \mathrm{~cm}^{-1}$; all data provided for the major rotamer; $\delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.23(1 \mathrm{H}, \mathrm{bs}, \mathrm{CONHNH}), 7.52$ (1H, d, J 7, Ar-H), 7.41 (1H, t, J , Ar-H), 7.02 (1H, d, J 8, Ar-H), 6.99 (1H, t, J 7, Ar-H), 6.52 (1H, bs, Ar-NH), 5.26 (1H, bt, J 6, BocNHCH2), 3.92 (2H, d, J 6, BocNHCH2), 1.47 (9H, s, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta с$ ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 169.7 (C=ONHNH), 145.20 (ipso-Ar-C), 133.1

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(Ar-C), 126.6 (Ar-C), 120.3 (Ar-C), 113.5 (Ar-C), 80.9 ((CH3)3CO), 43.4 $\left.\left(\mathrm{BocNHCH}_{2}\right), 28.3\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta_{F}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$-61.8 $\left(\mathrm{CF}_{3}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 356$ $\left(\mathrm{MNa}^{+}\right)$; HRMS $\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}, 356.1208\left(\mathrm{C}_{14} \mathrm{H}_{18} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~F}_{3} \mathrm{Na}\right.$ requires 356.1193).
tert-Butyl
carbamate 67Be

## (2-0xo-2-(2-(3-(trifluoromethyl)phenyl)hydrazineyl)ethyl)



Following the general procedure outlined, $N$-Boc-glycine ( $0.40 \mathrm{~g}, 2.28 \mathrm{mmol}$ ) and 3-trifluoromethylphenylhydrazine hydrochloride ( $0.53 \mathrm{~g}, 2.51 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a dark orange solid ( $0.49 \mathrm{~g}, 65 \%$ ) as a mixture of rotamers [4:1]; $\mathrm{Rf}_{\mathrm{f}}$ 0.44 (n-hex/EtOAc 4:1); m.p. $93-95^{\circ} \mathrm{C}$; $V_{\max } 3272$ (N-H), 1674, 1658 (C=O), 1525, 1339, 1252, 1116, $699 \mathrm{~cm}^{-1}$; all data provided for the major rotamer; $\delta_{H}$ (300 MHz, CDCl 3 ) 8.68 (1H, bs, CONHNH), 7.25 (1H, d, J 7, Ar-H), 7.10 (1H, d, J7, Ar-H), 7.00 (1H, s, Ar-H), 6.94 (1H, dd, J8, 2, Ar-H), 5.48 (1H, bs, BocNHCH2), 3.88 (2H, d, J 5, BocNHCH2), 1.44 (9H, s, ( $\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}$ ); סc ( $\left.100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 170.4$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.6 ( $\mathrm{C}=\mathrm{ONH}$ ), 148.1 (ipso-Ar-C), 131.7 (ipso-Ar-C), 129.6 (Ar-C), 125.4 (CF3), 117.5 (Ar-C), 116.4 ( $\mathrm{Ar}-\mathrm{C}$ ), 109.7 ( $\mathrm{Ar}-\mathrm{C}$ ), 80.9 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 43.3$ ( $\mathrm{BocNHCH}_{2}$ ), $\left.28.2\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; \delta_{F}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$-62.8 ( $\left.\mathrm{CF}_{3}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 356$ $\left(\mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}, 356.1203\left(\mathrm{C}_{14} \mathrm{H}_{18} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~F}_{3} \mathrm{Na}\right.$ requires 356.1193).
tert-Butyl (2-(2-(3-nitrophenyl)hydrazineyl)-2-oxoethyl)carbamate 67Dc


Following the general procedure outlined, $N$-Boc-glycine ( $0.40 \mathrm{~g}, 2.28 \mathrm{mmol}$ ) and 3-nitrophenylhydrazine hydrochloride ( $0.48 \mathrm{~g}, 2.51 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow solid ( $0.59 \mathrm{~g}, 83 \%$ ) as a mixture of rotamers [7:1]; Rf 0.39 ( $n$-hex/EtOAc 4:1); m.p. $116-119{ }^{\circ} \mathrm{C}$; $v_{\max } 3334,3237$ (N-H), 1709, 1684 (C=O), 1516, 1336, 1289, 1163,

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$733 \mathrm{~cm}^{-1}$; all data provided for the major rotamer; $\delta н\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.54(1 \mathrm{H}$, bs, CONHNH), 7.71 (1H, m, Ar-H), 7.61 (1H, t, J 2, Ar-H), 7.35 (1H, t, J 8, Ar-H), 7.11 (1H, m, Ar-H), 6.58 (1H, bd, J 2, Ar-NH), 5.40 (1H, bs, BocNHCH2), 3.94 (2H, d, J 6, BocNHCH2), 1.47 (9H, s, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta c\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 170.4$ (C=ONHNH), 156.5 ( $\mathrm{C}=\mathrm{ONH}$ ), 149.1 (ipso-Ar-C), 149.0 (ipso-Ar-C), 129.9 (Ar-C), 119.2 ( $\mathrm{Ar}-\mathrm{C}$ ), 115.7 ( $\mathrm{Ar}-\mathrm{C}$ ), 107.6 ( $\mathrm{Ar}-\mathrm{C}$ ), $81.2\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 43.5\left(\mathrm{BocNHCH}_{2}\right)$, $\left.28.3\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 333\left(\mathrm{MNa}^{+}\right)$; HRMS (ES$\left.{ }^{+}\right)$Found $\mathrm{MNa}^{+}, 333.1182$ ( $\mathrm{C}_{13} \mathrm{H}_{18} \mathrm{~N}_{4} \mathrm{O}_{5} \mathrm{Na}$ requires 333.1169).
tert-Butyl
(S)-(1-(2-(3-bromophenyl)hydrazineyl)-1-oxopropan-2-yl)
carbamate 67 Cl


Following the general procedure outlined, N -Boc-L-alanine ( $0.40 \mathrm{~g}, 2.11 \mathrm{mmol}$ ) and 3-bromophenylhydrazine hydrochloride ( $0.52 \mathrm{~g}, 2.33 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as an orange gummy solid ( $0.53 \mathrm{~g}, 70 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.31$ (DCM/EtOH/ $\mathrm{NH}_{3}$ 200:6:1); $v_{\max } 3425,3278$ (N-H), 2980 (C-H), 1689, 1676 (C=O), 1504, 1480, 1245, 1158, $778 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.73 (1H, bs, CONHNH), 7.01 (1H, s, Ar-H), $6.97(1 \mathrm{H}, \mathrm{t}, \mathrm{J} 1, \mathrm{Ar}-H), 6.91(1 \mathrm{H}, \mathrm{t}, \mathrm{J} 1, \operatorname{Ar}-H), 6.69(1 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}), 6.35$ (1H, d, J 3, Ar-NH), 5.33 (1H, m, BocNHCH), 4.29 (1H, t, J 7, BocNHCH), 1.45 (9H, s, ( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) 1.37$ (3H, d, J 7, NHCHCH3); ठc (100 MHz, CDCl3) 173.4 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.9 ( $\mathrm{C}=\mathrm{ONH}$ ), 149.2 (ipso-Ar-C), 130.5 (Ar-C), 123.7 (Ar-C), 116.1 ( $\mathrm{Ar}-\mathrm{C}$ ), $112.0(\mathrm{Ar}-\mathrm{C}), 80.7\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 43.6$ (BocNHCH$), 28.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $\left.17.9\left(\mathrm{NHCHCH}_{3}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 380\left({ }^{79} \mathrm{Br}\right] \mathrm{MNa}^{+}\right), 382$ ( $\left.{ }^{81} \mathrm{Br}\right] \mathrm{MNa}^{+}$), $737\left({ }^{79,79} \mathrm{Br}\right]$ $2 \mathrm{MNa}^{+}$); 739 ( $\left[^{79,81} \mathrm{Br}\right] 2 \mathrm{MNa}^{+}$), 741 ( $\left.{ }^{81,81} \mathrm{Br}\right] 2 \mathrm{MNa}^{+}$); HRMS (ES ${ }^{+}$) Found $\left.{ }^{79} \mathrm{Br}\right] \mathrm{MNa}^{+}, 380.0591\left(\mathrm{C}_{14} \mathrm{H}_{20}{ }^{79} \mathrm{BrN}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 380.0580).

## Chapter 6: Experimental section

## tert-Butyl (S)-(1-oxo-1-(2-(o-tolyl)hydrazineyl)propan-2-yl)carbamate 67Dp



Following the general procedure outlined, N -Boc-L-alanine ( $0.10 \mathrm{~g}, 0.54 \mathrm{mmol}$ ) and 2-methylphenylhydrazine hydrochloride $(0.07 \mathrm{~g}, \quad 0.57 \mathrm{mmol})$ were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as an orange solid ( $0.09 \mathrm{~g}, 61 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.39$ ( $\mathrm{DCM} / \mathrm{EtOH} / \mathrm{NH}_{3}$ 200:6:1); m.p. 134-136 ${ }^{\circ} \mathrm{C}$; $v_{\max } 3351,3279(\mathrm{~N}-\mathrm{H}), 2980(\mathrm{C}-\mathrm{H}), 1673,1606$ (C=O), 1534, 1479, 1286, 1152, 752, $627 \mathrm{~cm}^{-1}$; ठн ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.75 (1H, bs, CONHNH), 7.67 (1H, ddd, J 8, 16, 31, Ar-H), 7.05 (2H, m, Ar-H), 6.79 (1H, ddd, J 5, 7, 20, Ar-H), 5.39 (1H, bd, J8, BocNHCH), 4.33 (1H, m, BocNHCH), 2.18 (3H, s, Ar-CH3), 1.43 (9H, s, ( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.36\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 8, \mathrm{BocNHCHCH}_{3}\right) ; \delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 173.0$ (C=ONHNH), 155.8 (C=ONH), 145.5 (ipso-Ar-C), 130.5 (Ar-C), 126.9 (ipso-Ar-C), 123.2 (ipso-Ar-C), 120.9 (Ar-C), 112.0 (Ar-C), 80.4 ((CH3)3 CO$), 48.7$ (BocNHCH), $\left.28.4\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 18.2\left(\mathrm{Ar}^{-\mathrm{CH}_{3}}\right), 17.0\left(\mathrm{BocNHCHCH}_{3}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 332\left(\mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}, 332.1290\left(\mathrm{C}_{15} \mathrm{H}_{23} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 332.1275).

## tert-Butyl (S)-(1-oxo-1-(2-(m-tolyl)hydrazineyl)propan-2-yl)carbamate 67Dv



Following the general procedure outlined, $N$-Boc-L-alanine ( $0.30 \mathrm{~g}, 1.59 \mathrm{mmol}$ ) and 3-methylphenylhydrazine hydrochloride ( $0.28 \mathrm{~g}, 1.74 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a dark orange oil ( $0.30 \mathrm{~g}, 64 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.35$ (DCM/EtOH/NH3 200:6:1); $V_{\max } 3313$ (N-H), 2982 (C-H), 1683, 1660 (C=O), 1505, 1366, 1249, 1160, 1060, $779 \mathrm{~cm}^{-1}$; ठн (400 MHz, CDCl3) 8.61 (1H, bs, CONHNH), 7.06 (2H, m, Ar-H), 6.68 (1H, d, J7, Ar-H), 6.58 (1H, d, J 9, Ar-H), 5.33 (1H, bs, BocNHCH), 4.30 (1H, m, BocNHCH), 2.25 (3H, s, Ar- $\mathrm{CH}_{3}$ ), 1.45 ( $\left.9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.35(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 7$, $\mathrm{BocNHCHCH}_{3}$ ); $\mathrm{\delta c}_{\mathrm{c}}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 173.1$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.8 ( $\left.\mathrm{C}=\mathrm{ONH}\right), 147.9$ (ipso-Ar-C), 129.1 (ipso-Ar-C), 122.1 (Ar-C), 114.3 (Ar-C), 110.7 (Ar-C), 80.5 $\left.\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 48.7(\mathrm{BocNHCH}), 28.5\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 21.6 \quad\left(\mathrm{Ar}-\mathrm{CH}_{3}\right), 18.1$ ( $\mathrm{BocNHCHCH}_{3}$ ); m/z (ES') 316 ( $\mathrm{MNa}^{+}$); HRMS ( $\mathrm{ES}^{+}$) Found $\mathrm{MNa}^{+}$, 316.1643 ( $\mathrm{C}_{15} \mathrm{H}_{23} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{Na}$ requires 316.1632).

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tert-Butyl yl)carbamate 67 Cr
(S)-(1-(2-(4-bromophenyl)hydrazineyl)-1-oxopropan-2-


Following the general procedure outlined, N -Boc-L-alanine ( $0.40 \mathrm{~g}, 2.11 \mathrm{mmol}$ ) and 4-bromophenylhydrazine hydrochloride ( $0.52 \mathrm{~g}, 2.33 \mathrm{mmol}$ ) were transformed following flash column chromatography (DCM/EtOH/NH3 200:6:1) into the title compound which was isolated as a brown oil ( $0.33 \mathrm{~g}, 51 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.37$ (DCM/EtOH/NH3 200:6:1); $V_{\max }$ 3316, 3302 (N-H), 2987, 2937 (C-H), 1683, 1663 (C=O), 1520, 1486, 1320, 1249, 1161, $812 \mathrm{~cm}^{-1}$; $\delta$ н ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $8.43(1 \mathrm{H}$, bs, CONHNH), $7.30-7.27$ (2H, d, J 9, Ar-H), $6.69-6.66$ (2H, d, J 8, Ar-H), 6.15 (1H, bs, Ar-NH), 5.12 (1H, bd, J7, BocNHCH), 4.28 (1H, m, BocNHCH), 1.46 ( 9 H , s, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.39\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 7\right.$, BocNHCHCH 3 ); $\delta c\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 172.9$ ( $C=O N H N H$ ), 155.8 ( $C=O N H$ ), 146.9 (ipso-Ar-C), 131.9 (Ar-C), 129.2 (ipso-Ar-C), 115.1 (Ar-C), 113.1 (ipso-Ar-C), 80.8 (( $\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}$ ), 48.6 (BocNHCH), 28.3 $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 17.6$ ( $\mathrm{BocNHCHCH}_{3}$ ); m/z (ES $\left.\left.{ }^{+}\right) 380\left({ }^{79} \mathrm{Br}\right] \mathrm{MNa}^{+}\right), 382$ ([ $\left.{ }^{81} \mathrm{Br}^{2}\right] \mathrm{MH}^{+}$); HRMS (ES ${ }^{+}$) Found $\left[{ }^{79} \mathrm{Br}\right] \mathrm{MNa}^{+}, 380.0589 \quad\left(\mathrm{C}_{14} \mathrm{H}_{20}{ }^{79} \mathrm{BrN}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 380.0580).

## tert-Butyl (S)-(1-oxo-1-(2-(2-(trifluoromethyl)phenyl)hydrazineyl)propan-2yl )carbamate 67 Az



Following the general procedure outlined, $N$-Boc- $L$-alanine ( $0.40 \mathrm{~g}, 2.11 \mathrm{mmol}$ ) and 2-trifluoromethylphenylhydrazine hydrochloride ( $0.49 \mathrm{~g}, 2.33 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow gummy solid ( $0.52 \mathrm{~g}, 71 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.32$ (DCM/EtOH/NH3 200:6:1); $V_{\max } 3311$ (N-H), 2982 (C-H), 1696, 1657 (C=O), 1512, 1487, 1322, 1252, 1158, 1097, $757 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.47 ( $1 \mathrm{H}, \mathrm{bs}, \mathrm{CONHNH}$ ), 7.50 ( $1 \mathrm{H}, \mathrm{d}, ~ J 7$, Ar-H), 7.38 (1H, t, J 7, Ar-H), 7.00 (1H, d, J 8, Ar-H), 6.97 (1H, t, J 7, Ar-H), 6.53 (1H, bs, Ar-NH), 5.16 (1H, bd, J7, BocNHCH), 4.34 (1H, m, BocNHCH), 1.46 ( 9 H ,
 ( $C=O N H N H$ ), 155.9 ( $C=O N H$ ), 145.4 (ipso-Ar-C), 133.0 (Ar-C), 126.6 (Ar-C),

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120.2 (Ar-C), 113.4 (Ar-C), 80.8 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 48.7$ (BocNHCH), $\left.28.3\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $17.5\left(\mathrm{BocNHCHCH}_{3}\right) ; \delta \mathrm{F}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)-61.8\left(\mathrm{CF}_{3}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 370\left(\mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}, 370.1361\left(\mathrm{C}_{15} \mathrm{H}_{20} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~F}_{3} \mathrm{Na}\right.$ requires 370.1349).

## tert-Butyl (S)-(1-oxo-1-(2-(3-(trifluoromethyl)phenyl)hydrazineyl)propan-2yl)carbamate 67 Bf



Following the general procedure outlined, N -Boc-L-alanine ( $0.40 \mathrm{~g}, 2.11 \mathrm{mmol}$ ) and 3-trifluoromethylphenylhydrazine hydrochloride ( $0.49 \mathrm{~g}, 2.33 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow solid ( $0.51 \mathrm{~g}, 69 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.28$ (DCM/EtOH/NH3 200:6:1); m.p. $105-108{ }^{\circ} \mathrm{C}$; $v_{\max } 3328,3265(\mathrm{~N}-\mathrm{H}), 2987(\mathrm{C}-\mathrm{H}), 1683,1668(\mathrm{C}=\mathrm{O}), 1520,1335$, 1249, 1159, 1072, $785 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.61 (1H, bs, CONHNH), 7.30 (1H, t, J 7, Ar-H), 7.12 (1H, d, J 7, Ar-H), 7.01 (1H, s, Ar-H), 6.96 (1H, dd, J 8, 2, Ar-H), 6.37 (1H, bs, Ar-NH), 5.16 (1H, bd, J 7, BocNHCH), 4.32 ( $1 \mathrm{H}, \mathrm{m}$, BocNHCH), 1.45 (9H, s, ( $\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}$ ), 1.40 (3H, d, J 7, BocNHCHCH3); סc (100 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 173.2 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.9 ( $\mathrm{C}=\mathrm{ONH}$ ), 148.3 (ipso-Ar-C), 131.7 ( $\mathrm{Ar}-$ $\mathrm{CF}_{3}$ ), 129.6 ( $\mathrm{Ar}-\mathrm{C}$ ), 125.9 ( $\mathrm{Ar}_{-\mathrm{CF}_{3} \text { ), } 117.6 \text { ( } \mathrm{Ar}-\mathrm{C} \text { ), } 116.5 \text { ( } \mathrm{Ar}-\mathrm{C} \text { ), } 109.8 \text { ( } \mathrm{Ar}-\mathrm{C} \text { ), } 80.8 ~}^{8}$ $\left.\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 48.7(\mathrm{BocNHCH}), 28.3\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 17.6\left(\mathrm{BocNHCHCH}_{3}\right) ; \delta \mathrm{F}(300$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right)-62.8\left(\mathrm{CF}_{3}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 370\left(\mathrm{MNa}^{+}\right) ; \mathrm{HRMS}\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}$, $370.1365\left(\mathrm{C}_{15} \mathrm{H}_{20} \mathrm{~N}_{3} \mathrm{O}_{5} \mathrm{~F}_{3} \mathrm{Na}\right.$ requires 370.1349$)$.

## tert-Butyl (S)-(1-(2-(3-nitrophenyl)hydrazineyl)-1-oxopropan-2-yl)carbamate 67Dd



Following the general procedure outlined, $N$-Boc- $L$-alanine ( $0.40 \mathrm{~g}, 2.11 \mathrm{mmol}$ ) and 3 -nitrophenylhydrazine hydrochloride ( $0.44 \mathrm{~g}, 2.33 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow gummy solid ( $0.53 \mathrm{~g}, 78 \%$ ); Rf 0.23 (DCM/EtOH/NH3 200:6:1); $v_{\max } 3345,3328$, 3242 (N-H), 3003 (C-H), 1706, 1672 (C=O), 1509, 1343, 1302, 1253, 1160, 734

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$\mathrm{cm}^{-1}$; $\delta \mathrm{H}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.54$ (1H, bs, CONHNH), 7.72 (1H, dd, J 7, 1, Ar-H), 7.62 (1H, t, J2, Ar-H), 7.36 (1H, t, J 8, Ar-H), 7.12 (1H, dd, J8, 2, Ar-H), 6.43 (1H, bd, J3, Ar-NH), 5.08 (1H, bd, J7, BocNHCH), 4.35 (1H, m, BocNHCH), 1.48 ( 9 H , s, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, 1.44 (3H, d, J 7, BocNHCHCH ${ }_{3}$ ); $\delta c\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 173.1$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.9 ( $\mathrm{C}=\mathrm{ONH}$ ), 149.1 (ipso-Ar-C), 129.9 (Ar-C), 119.3 (Ar-C), 117.9 (ipso-Ar-C), 115.8 (Ar-C), 107.7 (Ar-C), 80.9 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 48.6$ (BocNHCH), $\left.28.3\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $17.2\left(\mathrm{BocNHCHCH}_{3}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 347\left(\mathrm{MNa}^{+}\right) ;$HRMS (ES$\left.{ }^{+}\right)$ Found $\mathrm{MNa}^{+}, 347.1344\left(\mathrm{C}_{14} \mathrm{H}_{20} \mathrm{~N}_{4} \mathrm{O}_{5} \mathrm{Na}\right.$ requires 347.1334).

## tert-Butyl (S)-(1-(2-(2-chlorophenyl)hydrazineyl)-3-methyl-1-oxobutan-2$\mathrm{yl})$ carbamate 67 Ai



Following the general procedure outlined, $N$-Boc- $L$-valine ( $0.40 \mathrm{~g}, 1.84 \mathrm{mmol}$ ) and 2-chlorophenylhydrazine hydrochloride ( $0.36 \mathrm{~g}, 2.03 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as an orange solid ( $0.45 \mathrm{~g}, 71 \%$ ); Rf 0.36 (DCM/EtOH/ $\mathrm{NH}_{3} 200: 6: 1$ ); m.p. $108-111^{\circ} \mathrm{C}$; $V_{\max } 3318$ (N-H), 2970 (C-H), 1661, 1651 (C=O), 1526, 1485, 1303, 1248, 1171, $745,646 \mathrm{~cm}^{-1}$; $\delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.59$ (1H, bs, CONHNH), 7.26 (1H, dd, J 8, 1, Ar-H), 7.09 (1H, td, J 8, 1, Ar-H), 6.88 (1H, dd, J 8, 1, Ar-H), 6.80 (1H, td, J 7, 1, Ar-H), 5.28 (1H, bd, J 8, BocNHCH), 4.02 (1H, dd, J7, 1, BocNHCH), 2.13 (1H, m, $\left.\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.44\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.00-0.96\left(6 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ;$ бc (100 MHz, CDCl3) 172.0 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.2 ( $\mathrm{C}=\mathrm{ONH}$ ), 143.7 (ipso-Ar-C), 129.4 (Ar-C), 127.6 (Ar-C), 121.3 (Ar-C), 113.7 ( $\mathrm{Ar}-\mathrm{C}$ ), 80.5 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 58.8$ ( BocNHCH ), $30.5\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 19.4\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 18.3$ $\left.\left.\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \quad m / z \quad\left(\mathrm{ES}^{+}\right) \quad 364 \quad\left({ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right), 366 \quad\left({ }^{37} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right), \quad 705$ $\left.\left(\left[{ }^{35,35} \mathrm{Cl}\right] 2 \mathrm{M}+\mathrm{Na}^{+}\right), 707\left(\left[{ }^{35,37} \mathrm{Cl}\right] 2 \mathrm{M}+\mathrm{Na}^{+}\right), 709\left({ }^{[37,37} \mathrm{Cl}\right] 2 \mathrm{M}+\mathrm{Na}^{+}\right) ;$HRMS (ES $\left.{ }^{+}\right)$ Found $\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}, 364.1415\left(\mathrm{C}_{16} \mathrm{H}_{24}{ }^{35} \mathrm{CIN}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 364.1404).

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tert-Butyl (S)-(3-methyl-1-oxo-1-(2-(p-tolyl)hydrazinyl)butan-2-yl)carbamate 67Ec


Following the general procedure outlined, $N$-Boc- $L$-valine ( $0.40 \mathrm{~g}, 1.84 \mathrm{mmol}$ ) and 4-methylphenylhydrazine hydrochloride ( $0.32 \mathrm{~g}, 2.03 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow solid ( $0.52 \mathrm{~g}, 88 \%$ ) as a mixture of rotamers [3:1]; Rf 0.39 (DCM/EtOH/NH3 200:6:1); m.p. $126-129{ }^{\circ} \mathrm{C}$; $V_{\max } 3334,3297$ (N-H), 2970 (C-H), 1686, 1650 (C=O), 1523, 1365, 1245, 1162, 1020, 818, $646 \mathrm{~cm}^{-1}$; all data provided for the major rotamer; $\delta_{H}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.65(1 \mathrm{H}, \mathrm{bs}, \mathrm{CONHNH}), 7.00-6.97(2 \mathrm{H}$, d, J 8, Ar-H), 6.73 - 6.70 (2H, d, J 8, Ar-H), 5.36 (1H, bd, J 8, BocNHCH), 4.02 (1H, t, J 7, BocNHCH), 2.24 (3H, s, Ar-CH3), 2.11 (1H, m, CHCH(CH3 $\left.)_{2}\right), 1.44$ (9H, s, ( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.97-0.93\left(6 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; ~ \delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 172.2$ (C=ONHNH), 156.2 (C=ONH), 145.3 (ipso-Ar-C), 129.6 (Ar-C), 113.9 (Ar-C), 80.4 $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 58.8(\mathrm{BocNHCH}), 30.5\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 20.6$ (Ar$\left.\mathrm{CH}_{3}\right), 19.3\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 17.5\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; m / z\left(\mathrm{ES}^{+}\right) 344\left(\mathrm{MNa}^{+}\right), 665$ $\left(2 \mathrm{M}+\mathrm{Na}^{+}\right)$; HRMS $\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}, 344.1967\left(\mathrm{C}_{17} \mathrm{H}_{27} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 344.1945).

## tert-Butyl (S)-(1-(2-(4-isopropylphenyl)hydrazinyl)-3-methyl-1-oxobutan-2yl)carbamate 67 Ei



Following the general procedure outlined, $N$-Boc- $L$-valine ( $0.40 \mathrm{~g}, 1.84 \mathrm{mmol}$ ) and 4-isopropylphenylhydrazine hydrochloride ( $0.38 \mathrm{~g}, 2.03 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as an orange solid ( $0.37 \mathrm{~g}, 58 \%$ ) as a mixture of rotamers [4:1]; Rf 0.42 (DCM/EtOH/NH3 200:6:1); m.p. 88-91 ${ }^{\circ} \mathrm{C}$; $V_{\max } 3328$ (N-H), 2959 (C-H), 1687, 1657 (C=O), 1513, 1304, 1248, 1168, $830 \mathrm{~cm}^{-1}$; all data provided for the major rotamer; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.88 (1H, bs, CONHNH), 7.04 - 7.01 (2H, d, J 8, Ar-H), 6.74 - 6.72 (2H, d, J 8, Ar-H), 5.44 (1H, bd, J 9, BocNHCH), 4.03 (1H, t, J

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7, BocNHCH), $2.84\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $2.09\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $1.43(9 \mathrm{H}, \mathrm{s}$, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.19-1.17\left(6 \mathrm{H}, \mathrm{d}, \mathrm{J} 7, \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) 0.94-0.89\left(6 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right)$; ठс ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 172.2 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.2 ( $\mathrm{C}=\mathrm{ONH}$ ), 145.6 (ipso-Ar-C), 141.6 (ipso-Ar-C), 129.9 (Ar-C), 113.8 (Ar-C), 80.2 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 58.7$ (BocNHCH), $33.3\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $30.8\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 24.2\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 19.3$ $\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $18.2\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 372\left(\mathrm{MNa}^{+}\right), 721\left(2 \mathrm{M}+\mathrm{Na}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}, 372.2285\left(\mathrm{C}_{19} \mathrm{H}_{31} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 372.2258).

## tert-Butyl (S)-(3-methyl-1-oxo-1-(2-(4-(trifluoromethyl)phenyl)hydrazinyl) butan-2-yl)carbamate 67Bm



Following the general procedure outlined, $N$-Boc- $L$-valine ( $0.40 \mathrm{~g}, 1.84 \mathrm{mmol}$ ) and 4-trifluoromethylphenylhydrazine hydrochloride ( $0.43 \mathrm{~g}, 2.03 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow solid ( $0.43 \mathrm{~g}, 62 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.41$ ( $\mathrm{DCM} / E t O H / \mathrm{NH}_{3}$ 200:6:1); m.p. 114-116 ${ }^{\circ} \mathrm{C}$; $v_{\max } 3322,3255$ (N-H), 1683, 1661 (C=O), 1521, 1507, 1267, 1251, $1159 \mathrm{~cm}^{-1}$; $\delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.75$ (1H, bs, CONHNH), $7.03-7.00(2 \mathrm{H}, \mathrm{d}, \mathrm{J}$ 8, Ar-H), $6.80-6.77$ (2H, d, J 8, Ar-H), 5.31 (1H, bd, J 8, BocNHCH), 4.02 (1H, m, BocNHCH), 2.11 (1H, m, $\left.\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, 1.44 ( $\left.9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.97-0.95$ (6H, d, J 6, CHCH $\left.\left(\mathrm{CH}_{3}\right)_{2}\right)$; бс ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 172.5 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.2 (C=ONH), 146.5 (ipso-Ar-C), 143.1 (ipso-Ar-C), 122.1 (Ar-C), 114.2 (Ar-C), 80.6 $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 58.8(\mathrm{BocNHCH}), 30.5\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 19.3$ $\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 18.2\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta$ ( $\left.300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)-58.4\left(\mathrm{CF}_{3}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right)$ $414\left(\mathrm{MK}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\mathrm{MK}^{+}, 414.1632\left(\mathrm{C}_{17} \mathrm{H}_{24} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~F}_{3}\right.$ requires 414.1630).

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## tert-Butyl (S)-(1-(2-(4-cyanophenyl)hydrazineyl)-3-methyl-1-oxobutan-2yl)carbamate 67Cy



Following the general procedure outlined, $N$-Boc- $L$-valine ( $0.40 \mathrm{~g}, 1.84 \mathrm{mmol}$ ) and 4-cyanophenylhydrazine hydrochloride ( $0.34 \mathrm{~g}, 2.03 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow solid (0.32 g, 52\%); Rf 0.34 (DCM/EtOH/NH3 200:6:1); m.p. 124-126 ${ }^{\circ} \mathrm{C}$; $V_{\max }$ 3288 (N-H), 2970(C-H), 2223 (C三N), 1667, 1607 (C=O), 1507, 1366, 1244, 1168, $829,648 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.71 (1H, bs, CONHNH), $7.44-7.41$ ( $2 \mathrm{H}, \mathrm{d}$, J 8, Ar-H), 6.79 - 6.77 (2H, d, J 8, Ar-H), 6.66 (1H, bs, Ar-NH), 5.24 (1H, bd, J 9, BocNHCH), 4.02 (1H, t, J 8, BocNHCH), 2.13 (1H, m, CHCH(CH3)2), 1.45 (9H, s, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.99-0.97\left(6 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ;$ бc (100 MHz, CDCl 3 ) 172.5 (C=ONHNH), 156.3 ( $\mathrm{C}=\mathrm{ONH}$ ), 151.5 (ipso-Ar-C), 133.6 (Ar-C), 119.5 (C三N), 112.9 (Ar-C), 103.0 (ipso-Ar-C), 80.7 ((CH3)3CO), 58.8 (BocNHCH), 30.4 $\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $19.4\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 18.3\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \mathrm{m} / \mathrm{z}$ (ES $\left.{ }^{+}\right) 355\left(\mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}, 355.1748\left(\mathrm{C}_{17} \mathrm{H}_{24} \mathrm{~N}_{4} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 355.1741).
tert-Butyl (S)-(3-methyl-1-(2-(4-(methylsulfonyl)phenyl)hydrazineyl)-1-oxobutan-2-yl)carbamate 67Dk


Following the general procedure outlined, $N$-Boc- $L$-valine ( $0.40 \mathrm{~g}, 1.84 \mathrm{mmol}$ ) and 4-(methylsulphonyl)phenylhydrazine hydrochloride ( $0.45 \mathrm{~g}, 2.03 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow solid ( $0.42 \mathrm{~g}, 59 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.32$ (DCM/EtOH/NH3 200:6:1); m.p. 155-158 ${ }^{\circ} \mathrm{C}$; $v_{\max } 3299$ (N-H), 2977 (C-H), 1668 (C=O), 1598, 1505, 1288, 1136, 1088, 769, $509 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.60 (1H, bs, CONHNH), $7.70-7.67$ (2H, d, J 8, Ar-H), $6.85-6.82(2 H, d, J 8, A r-H), 6.64(1 H, b s, A r-N H), 5.20(1 H$,

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bd, J 8, BocNHCH), 4.03 (1H, dd, J 7, 1, BocNHCH), 3.00 (3H, s, $\mathrm{SO}_{2} \mathrm{CH}_{3}$ ), 2.19 $\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.46\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.01-0.98(6 \mathrm{H}$, dd, J 6, 4, $\left.\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 172.4$ ( $\left.\mathrm{C}=\mathrm{ONHNH}\right), 156.2$ ( $\left.\mathrm{C}=\mathrm{ONH}\right), 152.4$ (ipso-Ar-C), 131.4 (ipso-Ar-C), 129.1 (Ar-C), 112.7 (Ar-C), 80.7 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 58.9$ ( BocNHCH ), $44.8\left(\mathrm{SO}_{2} \mathrm{CH}_{3}\right), 30.4\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 19.4$ $\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 18.2\left(\mathrm{CHCH}_{\left.\left(\mathrm{CH}_{3}\right)_{2}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 408\left(\mathrm{MNa}^{+}\right) ; ~ H R M S ~\left(\mathrm{ES}^{+}\right) \text {Found }}\right.$ $\mathrm{MNa}^{+}, 408.1575\left(\mathrm{C}_{17} \mathrm{H}_{27} \mathrm{~N}_{3} \mathrm{O}_{5} \mathrm{SNa}\right.$ requires 408.1563).

## tert-Butyl (S)-(1-(2-(3-bromophenyl)hydrazineyl)-3-methyl-1-oxobutan-2yl)carbamate 67 Cm



Following the general procedure outlined, $N$-Boc- $L$-valine ( $0.40 \mathrm{~g}, 1.84 \mathrm{mmol}$ ) and 3-bromophenylhydrazine hydrochloride ( $0.45 \mathrm{~g}, 2.03 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a white solid ( $0.32 \mathrm{~g}, 45 \%$ ); Rf 0.29 (DCM/EtOH/NH3 200:6:1); m.p. $107-110{ }^{\circ} \mathrm{C}$; $v_{\max }$ 3359, 3259 (N-H), 2972 (C-H), 1684, 1664 (C=O), 1523, 1471, 1306, 1246, 1169, 765, $649 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.55 (1H, bs, CONHNH), 7.03 (1H, d, J 7, Ar-H), 6.99 (1H, t, J 1, Ar-H), 6.96 (1H, m, Ar-H), 6.73 (1H, dt, J 7, 1, Ar-H), 6.28 (1H, bs, Ar-NH), 5.22 (1H, bd, J 8, BocNHCH), 4.02 (1H, dd, J 7, 1, BocNHCH), $2.17\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.45\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.99-0.97(6 \mathrm{H}$, d, J 6, $\left.\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right)$; ठc ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 172.3 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.2 ( $\mathrm{C}=\mathrm{ONH}$ ), 149.3 (ipso-Ar-C), 130.5 (Ar-C), 123.9 (ipso-Ar-C), 129.1 (Ar-C), 116.4 (Ar-C), 112.2 (Ar-C), $80.5\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 58.7$ (BocNHCH), $30.6\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.4$ $\left.\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 19.4\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 18.2\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 408\left({ }^{79} \mathrm{Br}\right] \mathrm{MNa}^{+}\right)$, $\left.410\left[{ }^{81} \mathrm{Br}\right] \mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\left[{ }^{79} \mathrm{Br}\right] \mathrm{MNa}^{+}, 408.0903\left(\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{~N}_{3} \mathrm{O}_{3}{ }^{79} \mathrm{BrNa}\right.$ requires 408.0893).

## Chapter 6: Experimental section

tert-Butyl (S)-(1-(2-(3,5-bis(trifluoromethyl)phenyl)hydrazineyl)-3-methyl-1-oxobutan-2-yl)carbamate 67 Bu


Following the general procedure outlined, $N$-Boc- $L$-valine ( $0.40 \mathrm{~g}, 1.84 \mathrm{mmol}$ ) and (3,5-bis(trifluoromethyl)phenyl)hydrazine hydrochloride ( $0.57 \mathrm{~g}, 2.03 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow solid ( $0.37 \mathrm{~g}, 46 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.35$ (DCM/EtOH/NH3 200:6:1); m.p. 142-144 ${ }^{\circ} \mathrm{C}$; $v_{\max } 3337$ (N-H), 1671, 1660 (C=O), 1524, 1373, 1277, 1167, 1121, 877, $682 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.55 ( $1 \mathrm{H}, \mathrm{bs}, \mathrm{CONHNH}$ ), 7.34 ( $1 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{H}$ ), 6.21 (2H, s, Ar-H), 6.58 (1H, bs, Ar-NH), 5.12 (1H, bd, J 8, BocNHCH), 4.02 (1H, d, J 7, BocNHCH), $2.18\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.45\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.02-$ $0.99\left(6 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ;$ бc ( $\left.100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 172.6$ ( $\left.\mathrm{C}=\mathrm{ONHNH}\right), 156.2$ ( $\mathrm{C}=\mathrm{ONH}$ ), 149.1 (ipso-Ar-C), 132.7 ( $\mathrm{Ar}^{2} \mathrm{CFF}_{3}, \mathrm{t}, \mathrm{J} 33$ ), 125.0 (ipso-Ar-C), 121.3 (ipso-Ar-C), 114.1 (Ar-C), 112.9 ( $\mathrm{Ar}-\mathrm{C}$ ), $80.7\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 58.7$ ( BocNHCH$), 30.4$ $\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.2\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $19.2\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 18.1\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta_{F}$ ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) -63.3 ( $\mathrm{CF}_{3}$ ); m/z ( $\mathrm{ES}^{+}$) 466 ( $\mathrm{MNa}^{+}$); HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}$, $466.1547\left(\mathrm{C}_{18} \mathrm{H}_{23} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~F}_{6} \mathrm{Na}\right.$ requires 466.1536$)$.

## tert-Butyl carbamate 67Dq

(S)-(3-methyl-1-oxo-1-(2-(o-tolyl)hydrazineyl)butan-2-yl)


Following the general procedure outlined, $N$-Boc- $L$-valine ( $0.40 \mathrm{~g}, 1.84 \mathrm{mmol}$ ) and 2-methylphenylhydrazine hydrochloride ( $0.32 \mathrm{~g}, 2.03 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a dark red gummy solid ( $0.47 \mathrm{~g}, 80 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.35$ (DCM/EtOH/NH3 200:6:1); $v_{\max } 3319$ (NH), 2967 (C-H), 1679, 1643 (C=O), 1525, 1494, 1304, 1247, 1171, 747, $647 \mathrm{~cm}^{-}$ ${ }^{1}$; $\delta н\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.47$ (1H, bs, CONHNH), $7.06-7.04$ (2H, d, J 7, Ar-H), $6.83-6.81$ (2H, d, J 7, Ar-H), 5.24 (1H, bd, J 9, BocNHCH), 4.04 (1H, dd, J 7, 1,

BocNHCH), 2.21 (3H, s, Ar- $\mathrm{CH}_{3}$ ), 2.16 ( $\left.1 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.44(9 \mathrm{H}, \mathrm{s}$, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.99-0.97\left(6 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right)$; $\delta c\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 171.7$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.1 ( $\mathrm{C}=\mathrm{ONH}$ ), 145.5 (ipso-Ar-C), 130.5 (Ar-C), 126.8 (Ar-C), 123.2 (ipso-Ar-C), 120.9 (Ar-C), 112.2 (Ar-C), 80.3 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 58.8$ (BocNHCH), $30.6\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 19.4\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 18.2\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $17.0\left(\mathrm{Ar}^{-} \mathrm{CH}_{3}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 344\left(\mathrm{MNa}^{+}\right) ; \mathrm{HRMS}\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}, 344.1964$ $\left(\mathrm{C}_{17} \mathrm{H}_{27} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 344.1945$)$.
tert-Butyl
(S)-(3-methyl-1-oxo-1-(2-(m-tolyl)hydrazineyl)butan-2-yl) carbamate 67Dw


Following the general procedure outlined, $N$-Boc- $L$-valine ( $0.40 \mathrm{~g}, 1.84 \mathrm{mmol}$ ) and 3-methylphenylhydrazine hydrochloride ( $0.32 \mathrm{~g}, 2.03 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a light brown solid ( $0.29 \mathrm{~g}, 50 \%$ ); $\mathrm{R}_{\mathrm{f}} 0.32$ (DCM/EtOH/NH3 200:6:1); m.p. $146-148{ }^{\circ} \mathrm{C}$; $V_{\max }$ 3325, 3269 (N-H), 2970 (C-H), 1685, 1664 (C=O), 1524, 1478, 1306, 1246, 1168, $693 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.87 ( $1 \mathrm{H}, \mathrm{bs}, \mathrm{CONHNH}$ ), $7.07(1 \mathrm{H}, \mathrm{t}, \mathrm{J} 7$, Ar-H), 6.68 (1H, d, J 7, Ar-H), 6.59 - 6.57 (2H, m, Ar-H), 6.25 (1H, bd, J 3, ArNH), 5.41 (1H, bd, J 9, BocNHCH), 4.06 (1H, t, J 8, BocNHCH), 2.22 (3H, s, Ar$\left.\mathrm{CH}_{3}\right)$, $2.08\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.43\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.94-0.92(6 \mathrm{H}, \mathrm{d}, \mathrm{J}$ 6, $\left.\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right)$; $\delta с\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 171.2$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.1 ( $\mathrm{C}=\mathrm{ONH}$ ), 147.9 (ipso-Ar-C), 138.9 (ipso-Ar-C), 128.9 (Ar-C), 121.9 (Ar-C), 114.4 (Ar-C), 110.7 ( $\mathrm{Ar}-\mathrm{C}$ ), $80.1\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 58.6$ ( BocNHCH$), 30.9\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.4$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 21.5\left(\mathrm{Ar}^{2} \mathrm{CH}_{3}\right)$, $19.3\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $18.2\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right)$ $344\left(\mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}$, $344.1970\left(\mathrm{C}_{17} \mathrm{H}_{27} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 344.1945).

## Chapter 6: Experimental section

## tert-Butyl (S)-(1-(2-(4-bromophenyl)hydrazineyl)-3-methyl-1-oxobutan-2-yl) carbamate 67Cs



Following the general procedure outlined, $N$-Boc- $L$-valine ( $0.40 \mathrm{~g}, 1.84 \mathrm{mmol}$ ) and 4-bromophenylhydrazine hydrochloride ( $0.45 \mathrm{~g}, 2.03 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a white solid ( $0.41 \mathrm{~g}, 58 \%$ ); as a mixture of rotamers [9:1]; $\mathrm{R}_{\mathrm{f}} 0.41$ (DCM/EtOH/NH3 200:6:1); m.p. $127-129{ }^{\circ} \mathrm{C}$; $V_{\max } 3323,3284$ (N-H), 2972 (C-H), 1686, 1651 (C=O), 1521, 1489, 1245, 1164, 826, $634 \mathrm{~cm}^{-1}$; all data provided for the major rotamer; $\delta_{H}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.33(1 \mathrm{H}, \mathrm{bs}, \mathrm{CONHNH}), 7.31-7.28(2 \mathrm{H}, \mathrm{d}, \mathrm{J} 9$, Ar-H), 6.73 - 6.70 (2H, d, J 8, Ar-H), 6.28 (1H, bs, Ar-NH), 5.15 (1H, bd, J 8, BocNHCH), 4.00 (1H, dd, J 7, 1, BocNHCH), 2.18 (1H, m, $\left.\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.48$ (9H, s, ( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.01-0.98\left(6 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ;$ бc $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 172.1$ (C=ONHNH), 155.9 (C=ONH), 146.9 (ipso-Ar-C), 131.9 (Ar-C), 115.3 (Ar-C), 80.4 $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 58.7$ (BocNHCH), $30.3\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 19.4$ $\left.\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), \quad 18.2 \quad\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \quad \mathrm{m} / \mathrm{z} \quad\left(\mathrm{ES}^{+}\right) 408 \quad\left({ }^{79} \mathrm{Br}\right] \mathrm{MNa}^{+}\right), 410$ $\left(\left[{ }^{81} \mathrm{Br}\right] \mathrm{MNa}^{+}\right) ;$HRMS (ES $\left.{ }^{+}\right)$Found $\left[{ }^{79} \mathrm{Br}\right] \mathrm{MNa}^{+}, 408.0906\left(\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{~N}_{3} \mathrm{O}_{3}{ }^{79} \mathrm{BrNa}\right.$ requires 408.0893).

## tert-Butyl (S)-(3-methyl-1-oxo-1-(2-(2(trifluoromethyl)phenyl)hydrazineyl)

 butan-2-yl)carbamate 67Ba

Following the general procedure outlined, $N$-Boc- $L$-valine ( $0.40 \mathrm{~g}, 1.84 \mathrm{mmol}$ ) and 2-trifluoromethylphenylhydrazine hydrochloride ( $0.43 \mathrm{~g}, 2.03 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow solid ( $0.48 \mathrm{~g}, 70 \%$ ); Rf 0.35 (DCM/EtOH/NH3 200:6:1); m.p. 114-116 ${ }^{\circ} \mathrm{C}$; $v_{\max } 3327$ (N-H), 2972 (C-H), 1679, 1652 (C=O), 1526, 1496, 1323, 1288, 1107, 1036, $753 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.44 (1H, bs, CONHNH), 7.49

## Chapter 6: Experimental section

(1H, d, J 8, Ar-H), 7.35 (1H, t, J 7, Ar-H), 7.01 (1H, d, J 8, Ar-H), 6.93 (1H, t, J 7, Ar-H), 6.58 (1H, bs, Ar-NH), 5.22 (1H, bd, J 8, BocNHCH), 4.05 (1H, m, BocNHCH), $2.16\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.44\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.03-0.98(6 \mathrm{H}$, $\left.\mathrm{m}, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right)$; $\delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 172.1$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.2 ( $\mathrm{C}=\mathrm{ONH}$ ), 145.4 (ipso-Ar-C), 132.9 (Ar-C), 126.5 (Ar-C), 120.1 (Ar-C), 117.7 (ipso-Ar-C), 113.5 (Ar-C), 80.6 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 58.8$ ( BocNHCH$), 30.4\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.3$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $19.4\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 18.3\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta_{F}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)-61.9$ $\left(\mathrm{CF}_{3}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 398\left(\mathrm{MNa}^{+}\right) ; \operatorname{HRMS}\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}, 398.1685$ ( $\mathrm{C}_{17} \mathrm{H}_{24} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~F}_{3} \mathrm{Na}$ requires 398.1662 ).

## tert-Butyl (S)-(3-methyl-1-oxo-1-(2-(3(trifluoromethyl)phenyl)hydrazineyl) butan-2-yl)carbamate 67Bg



Following the general procedure outlined, $N$-Boc- $L$-valine ( $0.40 \mathrm{~g}, 1.84 \mathrm{mmol}$ ) and 3-trifluoromethylphenylhydrazine hydrochloride ( $0.43 \mathrm{~g}, 2.03 \mathrm{mmol}$ ) were transformed following flash column chromatography ( $\mathrm{DCM} / \mathrm{EtOH} / \mathrm{NH}_{3}$ 200:6:1) into the title compound which was isolated as a light brown solid ( $0.29 \mathrm{~g}, 42 \%$ ); $R_{f} 0.37$ (DCM/EtOH/NH3 200:6:1); m.p. $117-119{ }^{\circ} \mathrm{C}$; $\quad V_{\max } 3322,3252(\mathrm{~N}-\mathrm{H})$, 1683, 1663 (C=O), 1524, 1333, 1120, 1162, 693 cm$^{-1} ;$ бн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.52 (1H, bs, CONHNH), 7.33 (1H, t, J 7, Ar-H), 7.15 (1H, d, J 7, Ar-H), 7.07 (1H, t, J 1, Ar-H), 7.00 (1H, dd, J 8, 2, Ar-H), 5.19 (1H, bd, J 9, BocNHCH), 4.06 (1H, dd, J 7, 1, BocNHCH), 2.19 (1H, m, CHCH(CH3 $)_{2}$ ), 1.47 ( $\left.9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.03-$ $1.00\left(6 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right)$; $\delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 172.4$ ( $\left.\mathrm{C}=\mathrm{ONHNH}\right), 156.2$ (C=ONH), 148.3 (ipso-Ar-C), 129.6 (Ar-C), 122.2 (ipso-Ar-C), 117.6 (Ar-C), 116.6 (Ar-C), $110.0(\mathrm{Ar}-\mathrm{C}), 80.6\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 58.8(\mathrm{BocNHCH}), 30.5\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 19.3\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $18.1\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta_{F}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ -62.9 (CF3); m/z (ES ${ }^{+}$) 398 ( $\mathrm{MNa}^{+}$); HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}$, 398.1677 ( $\mathrm{C}_{17} \mathrm{H}_{24} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~F}_{3} \mathrm{Na}$ requires 398.1662).

## Chapter 6: Experimental section

## tert-Butyl (S)-(3-methyl-1-(2-(3-nitrophenyl)hydrazineyl)-1-oxobutan-2yl)carbamate 67De



Following the general procedure outlined, $N$-Boc-L-valine ( $0.40 \mathrm{~g}, 1.84 \mathrm{mmol}$ ) and 3-nitrophenylhydrazine hydrochloride ( $0.38 \mathrm{~g}, 2.03 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow solid ( $0.51 \mathrm{~g}, 79 \%$ ); Rf 0.49 (DCM/EtOH/NH3 200:6:1); m.p. $135-137{ }^{\circ} \mathrm{C}$; $V_{\text {max }}$ 3333, 3249 (N-H), 2975 (C-H), 1679, 1654 (C=O), 1521, 1481, 1302, 1244, 1159, 1017, $734 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.47 (1H, bs, CONHNH), 7.64 (1H, dd, J 8, 2, Ar-H), 7.57 (1H, t, J 2, Ar-H), 7.27 (1H, t, J 8, Ar-H), 7.05 (1H, dd, J 8, 2, Ar-H), 5.19 (1H, bd, J 8, BocNHCH), 4.06 (1H, dd, J 8, 1, BocNHCH), 2.12 (1H, m, $\left.\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.39\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.96-0.94\left(6 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta c$ ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 172.5 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.2 ( $\mathrm{C}=\mathrm{ONH}$ ), 149.1 (ipso-Ar-C), 129.6 (Ar-C), 119.3 ( $\mathrm{Ar}-\mathrm{C}$ ), 115.7 ( $\mathrm{Ar}-\mathrm{C}$ ), 107.8 ( $\mathrm{Ar}-\mathrm{C}$ ), 80.7 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 58.8$ ( BocNHCH ), $30.4\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 19.4\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 18.1$ $\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 375\left(\mathrm{MNa}^{+}\right) ; \mathrm{HRMS}\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}, 375.1649$ $\left(\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{~N}_{4} \mathrm{O}_{5}\right.$ requires 375.1647$)$.
tert-Butyl (S)-(1-(2-(2-chlorophenyl)hydrazineyl)-4-methyl-1-oxopentan-2yl)carbamate 67Aj


Following the general procedure outlined, $N$-Boc-L-leucine ( $0.30 \mathrm{~g}, 1.30 \mathrm{mmol}$ ) and 2-chlorophenylhydrazine hydrochloride ( $0.26 \mathrm{~g}, 1.43 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a light brown solid ( $0.25 \mathrm{~g}, 54 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.32$ (DCM/EtOH/NH $300: 6: 1$ ); m.p. 119-121 ${ }^{\circ} \mathrm{C}$; $v_{\max } 3334$ (N-H), 2956 (C-H), 1722, 1658 (C=O), 1595, 1516, 1369, 1159, 837, $752 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.73 (1H, bs, CONHNH), 7.17 (1H, dd, J 7, 1, Ar-H), 7.02 (1H, td, J 8, 1, Ar-H), 6.78 (1H, dd, J 8, 1, Ar-H), 6.73 (1H, td, J 7, 1, Ar-H), 6.38 (1H, bs, CONHNH), 5.17 (1H, bd, J 8, BocNHCH),

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$4.25(1 \mathrm{H}, \mathrm{m}, \mathrm{BocNHCH}), 1.64\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $1.48(1 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.36\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.87-0.83\left(6 \mathrm{H}, \mathrm{t}, \mathrm{J} 6, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$; $\delta \mathrm{c}$ ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 172.8 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.1 ( $\mathrm{C}=\mathrm{ONH}$ ), 143.7 (ipso-Ar-C), 129.4 (Ar-C), 127.6 (Ar-C), 121.2 (Ar-C), 119.4 (ipso-Ar-C), 113.5 (Ar-C), 80.5 $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 51.5(\mathrm{BocNHCH}), 40.6\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 24.7$ $\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.7\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.2\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 378$ ( $\left.{ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}$), $380\left(\left[{ }^{37} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right), 733\left(\left[{ }^{35,35} \mathrm{Cl}\right] 2 \mathrm{M}+\mathrm{Na}^{+}\right), 735\left(\left[{ }^{35,37} \mathrm{Cl}\right] 2 \mathrm{M}+\mathrm{Na}^{+}\right), 737$ ( $\left.\left[{ }^{37,37} \mathrm{Cl}\right] 2 \mathrm{M}+\mathrm{Na}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}$, $378.1567\left(\mathrm{C}_{17} \mathrm{H}_{26}{ }^{35} \mathrm{CIN} \mathrm{N}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 378.1555 ).

## tert-Butyl (S)-(1-(2-(3-chlorophenyl)hydrazinyl)-4-methyl-1-oxopentan-2yl)carbamate 67t



Following the general procedure outlined, $N$-Boc-L-leucine ( $0.30 \mathrm{~g}, 1.30 \mathrm{mmol}$ ) and 3-chlorophenylhydrazine hydrochloride ( $0.26 \mathrm{~g}, 1.43 \mathrm{mmol}$ ) were transformed following flash column chromatography ( $n$-hex/EtOAc 4:1) into the title compound which was isolated as a brown solid ( $0.16 \mathrm{~g}, 34 \%$ ); Rf 0.39 ( $n-$ hex/EtOAc 4:1); m.p. $123-125{ }^{\circ} \mathrm{C}$; $v_{\max } 3287$ (N-H), 2959 (C-H), 1697, 1682 (C=O), 1594, 1475, 1387, 1367, 1163, 769, $685 \mathrm{~cm}^{-1}$; бн $^{\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.77}$ (1H, bs, CONHNH), 7.07 (1H, t, J 8, Ar-H), 6.82 (1H, d, J 8, Ar-H), 6.77 (1H, t, J 2, Ar-H), 6.66 (1H, dd, J 8, 2, Ar-H), 5.20 (1H, bd, J 8, BocNHCH), 4.27 (1H, m, BocNHCH), $1.69-1.61\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH} H \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $1.54(1 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CHHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.45\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.94-0.89\left(6 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$; $\delta c$ (100 MHz, CDCl3) 173.1 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.1 (C=ONH), 149.1 (ipso-Ar-C), 134.9 (ipso-Ar-C), 130.1 (Ar-C), 128.6 (ipso-Ar-C), 120.8 (Ar-C), 113.3 (Ar-C), 111.6 (Ar-C), $80.6\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 51.5(\mathrm{BocNHCH}), 40.6\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $24.7\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.87\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.1\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 378$ $\left(\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right)$, $380\left(\left[{ }^{37} \mathrm{CI}\right] \mathrm{MNa}^{+}\right) ;$HRMS (ES ${ }^{+}$) Found $\left[{ }^{35} \mathrm{CI}\right] \mathrm{MNa}^{+}, 378.1572$ ( $\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{~N}_{3} \mathrm{O}_{3}{ }^{35} \mathrm{CINa}$ requires 378.1572 ).

## Chapter 6: Experimental section

## tert-Butyl (S)-(1-(2-(2,6-dichlorophenyl)hydrazineyl)-4-methyl-1-oxopentan-2-yl)carbamate 67Av



Following the general procedure outlined, $N$-Boc-L-leucine ( $0.30 \mathrm{~g}, 1.30 \mathrm{mmol}$ ) and 2,6-dichlorophenylhydrazine hydrochloride ( $0.30 \mathrm{~g}, 1.43 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a brown oil ( $0.19 \mathrm{~g}, 37 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.40$ ( $n$-hex/EtOAc 4:1); $v_{\max } 3307,3264$ (N-H), 2979, 2961 (C-H), 1679, 1656 (C=O), 1517, 1446, 1281, 1162, 768, 618 $\mathrm{cm}^{-1}$; $\delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.59(1 \mathrm{H}, \mathrm{bs}, \mathrm{CONHNH}), 7.24(1 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{H}), 7.22(1 \mathrm{H}$, s, Ar-H), 6.92 (1H, t, J 8, Ar-H), 6.76 (1H, bd, J 4, CONHNH), 4.95 (1H, bd, J 8, BocNHCH), 4.21 (1H, m, BocNHCH), $1.67-1.62\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CHHCH}\left(\mathrm{CH}_{3}\right)_{2}\right.$, $\left.\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.50\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.44\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.92-0.88$ (6H, t, J 6, CH2CH(CH3)2); бс (100 MHz, CDCl3) 171.9 (C=ONHNH), 155.7 (C=ONH), 140.9 (ipso-Ar-C), 128.8 (Ar-C), 126.9 (ipso-Ar-C), 125.9 (ipso-Ar-C), 123.9 ( $\mathrm{Ar}-\mathrm{C}$ ), $80.2\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 51.2(\mathrm{BocNHCH}), 40.9\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.3$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $24.6\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.8\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.0\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; m / z$ (ES ${ }^{+}$) 412 ( $\left.{ }^{35,35} \mathrm{CI}\right] \mathrm{MNa}^{+}$), 414 ( $\left[{ }^{37,35} \mathrm{CI}\right] \mathrm{MNa}^{+}$), 416 ( $\left.{ }^{37,37} \mathrm{CI}\right] \mathrm{MNa}^{+}$); HRMS (ES ${ }^{+}$) Found $\left[{ }^{35,35} \mathrm{Cl}\right] \mathrm{MNa}^{+}, 412.1182\left(\mathrm{C}_{17} \mathrm{H}_{25}{ }^{35,35} \mathrm{CIN}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 412.1165$)$.

## tert-Butyl (S)-(1-(2-(2,4-difluorophenyl)hydrazineyl)-4-methyl-1-oxopentan-2-yl)carbamate 67 Cb



Following the general procedure outlined, $N$-Boc-L-leucine ( $0.30 \mathrm{~g}, 1.30 \mathrm{mmol}$ ) and 2,4-difluorophenylhydrazine hydrochloride ( $0.26 \mathrm{~g}, 1.43 \mathrm{mmol}$ ) were transformed following flash column chromatography ( $n$-hex/EtOAc 4:1) into the title compound which was isolated as a brown oil ( $0.11 \mathrm{~g}, 25 \%$ ); Rf 0.43 ( $n-$ hex/EtOAc 4:1); $V_{\max } 3283$ (N-H), 2960 (C-H), 1669 (C=O), 1506, 1469, 1367, 1251, 1163, $961 \mathrm{~cm}^{-1}$; $\delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.89(1 \mathrm{H}, \mathrm{bs}, \mathrm{CONHNH}), 6.83(1 \mathrm{H}$,

## Chapter 6: Experimental section

m, Ar-H), 6.76 (1H, m, Ar-H), 6.68 (1H, t, J 8, Ar-H), 6.22 (1H, bs, CONHNH), 5.31 (1H, bd, J 8, BocNHCH), 4.30 (1H, m, BocNHCH), $1.66-1.52$ (3H, m, $\left.\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.43\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.94-0.90(6 \mathrm{H}, \mathrm{t}, \mathrm{J} 6$, $\left.\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ;$ бc ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 173.1 ( $\left.\mathrm{C}=\mathrm{ONHNH}\right)$, 156.1 ( $\left.\mathrm{C}=\mathrm{ONH}\right), 132.4$ (1C, dd, J 10, 3, Ar-CF), 115.3 (Ar-C), 110.9 (1C, dd, J 22, 3, Ar-C), 104.1 (1C, q, J 22, Ar-C), $80.5\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 51.5$ (BocNHCH), $40.6\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.3$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $24.7\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.7\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.1\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta_{F}$ ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) -121.0 (Ar-F), -129.2 (Ar-F); m/z (ES ${ }^{+}$) 380 ( $\mathrm{MNa}^{+}$); HRMS $\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}, 380.1779\left(\mathrm{C}_{17} \mathrm{H}_{25} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~F}_{2} \mathrm{Na}\right.$ requires 380.1756).
tert-Butyl
(S)-(4-methyl-1-oxo-1-(2-(p-tolyl)hydrazineyl)pentan-2-yl) carbamate 67Ed


Following the general procedure outlined, $N$-Boc-L-leucine ( $0.40 \mathrm{~g}, 1.73 \mathrm{mmol}$ ) and 4-methylphenylhydrazine hydrochloride ( $0.30 \mathrm{~g}, 1.90 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow gummy solid ( $0.39 \mathrm{~g}, 69 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.46$ (DCM/EtOH/NH 3 200:6:1); $V_{\max } 3379$ (N-H), 2957, 2931 (C-H), 1675, 1614 (C=O), 1512, 1453, 1366, 1249, 1164, 1045, $814 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.55 (1H, bs, CONHNH), $7.00-6.97$ (2H, d, J 8, Ar-H), 6.71 - 6.68 (2H, d, J 8, Ar-H), 5.13 (1H, bs, BocNHCH), 4.23 (1H, m, BocNHCH), $2.24\left(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{CH}_{3}\right), 1.69-1.59\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CHHCH}\left(\mathrm{CH}_{3}\right)_{2}\right.$, $\left.\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.53\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.45\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.94-0.90$ (6H, t, J 6, CH2CH(CH3)2); ठc (100 MHz, CDCl3) 172.7 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.9 (C=ONH), 145.5 (ipso-Ar-C), 130.4 (ipso-Ar-C), 129.8 (Ar-C), 129.6 (Ar-C), 113.7 (Ar-C), 113.1 (Ar-C), $80.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 51.5(\mathrm{BocNHCH}), 40.7\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $28.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 24.7\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.8\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.1\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $20.6\left(\mathrm{Ar}-\mathrm{CH}_{3}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 358\left(\mathrm{MNa}^{+}\right), 693\left(2 \mathrm{M}+\mathrm{Na}^{+}\right) ;$HRMS (ES $\left.{ }^{+}\right)$Found $\mathrm{MNa}^{+}$, $358.2119\left(\mathrm{C}_{18} \mathrm{H}_{29} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 358.2101) .

## Chapter 6: Experimental section

## tert-Butyl (S)-(1-(2-(4-isopropylphenyl)hydrazineyl)-4-methyl-1-oxopentan-

## 2-yl)carbamate 67Ej



Following the general procedure outlined, $N$-Boc-L-leucine ( $0.40 \mathrm{~g}, 1.73 \mathrm{mmol}$ ) and 4-isopropylphenylhydrazine hydrochloride ( $0.36 \mathrm{~g}, 1.90 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a brown gummy solid ( $0.49 \mathrm{~g}, 78 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.41$ (DCM/EtOH/NH3 200:6:1); $V_{\max } 3357,3244,(\mathrm{~N}-\mathrm{H}), 2957,2933,2870(\mathrm{C}-\mathrm{H}), 1689,1661$ (C=O), 1511, 1467, 1366, 1254, 1168, 1048, $828 \mathrm{~cm}^{-1}$; $\delta \mathrm{H}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.39$ ( $1 \mathrm{H}, \mathrm{bs}, \mathrm{CONHNH}$ ), 7.07 - 7.04 (2H, d, J 8, Ar-H), 6.75 - 6.72 (2H, d, J 8, Ar-H), 5.07 (1H, bd, J 8, BocNHCH ), $4.22(1 \mathrm{H}, \mathrm{m}, \mathrm{BocNHCH}), 2.86\left(1 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.72-1.62(2 \mathrm{H}$, m, $\left.\mathrm{CH} H \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.54\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.46(9 \mathrm{H}, \mathrm{s}$, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 2.86\left(6 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{Ar}-\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 0.95-0.91\left(6 \mathrm{H}, \mathrm{t}, \mathrm{J} 6, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$; ठс ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 172.6 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.9 ( $\mathrm{C}=\mathrm{ONH}$ ), 145.6 (ipso-Ar-C), 141.7 (ipso-Ar-C), 127.0 ( $\mathrm{Ar}-\mathrm{C}$ ), 113.7 ( $\mathrm{Ar}-\mathrm{C}), 80.5\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 51.5$ ( BocNHCH ), $40.6\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $33.3\left(\mathrm{Ar}-\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 24.7\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $24.2\left(\mathrm{Ar}-\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.8\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.1\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 386$ $\left(\mathrm{MNa}^{+}\right), 749\left(2 \mathrm{M}+\mathrm{Na}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}, 386.2433\left(\mathrm{C}_{20} \mathrm{H}_{33} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 386.2414).

## tert-Butyl (S)-(4-methyl-1-oxo-1-(2-(4-(trifluoromethyl)phenyl)hydrazineyl) pentan-2-yl)carbamate 67Bn



Following the general procedure outlined, $N$-Boc-L-leucine ( $0.40 \mathrm{~g}, 1.73 \mathrm{mmol}$ ) and 4-trifluoromethylphenylhydrazine hydrochloride ( $0.40 \mathrm{~g}, 1.90 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow gummy solid ( $0.43 \mathrm{~g}, 64 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.38$ ( $\mathrm{DCM} / \mathrm{EtOH} / \mathrm{NH}_{3}$ 200:6:1);

## Chapter 6: Experimental section

$V_{\max } 2961$ (C-H), 1684, 1657 (C=O), 1513, 1468, 1321, 1160, 1114, 1067, 835 $\mathrm{cm}^{-1}$; $\delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.84$ (1H, bs, CONHNH), $7.39-7.37$ (2H, d, J 8, ArH), 6.79 - 6.77 (2H, d, J 8, Ar-H), 5.20 (1H, bd, J 8, BocNHCH), 4.27 (1H, m, BocNHCH), $1.68-1.53\left(3 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.45(9 \mathrm{H}, \mathrm{s}$, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.93-0.88\left(6 \mathrm{H}, \mathrm{t}, \mathrm{J} 6, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 173.2$ ( $C=O N H N H$ ), 156.1 ( $C=O N H$ ), 150.6 (ipso-Ar-C), 126.5 (Ar-C), 122.8 (ipso-ArC), 112.6 ( $\mathrm{Ar}-\mathrm{C}), 80.8\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 51.5(\mathrm{BocNHCH}), 40.5\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.3$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $24.7\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.7\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $21.9\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta_{F}$ ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) -61.6 ( $\mathrm{CF}_{3}$ ); m/z ( $\mathrm{ES}^{+}$) $412\left(\mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}$, $412.1835\left(\mathrm{C}_{18} \mathrm{H}_{26} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~F}_{3} \mathrm{Na}\right.$ requires 412.1818$)$.

## tert-Butyl (S)-(1-(2-(4-cyanophenyl)hydrazineyl)-4-methyl-1-oxopentan-2yl)carbamate 67 Cz



Following the general procedure outlined, $N$-Boc-L-leucine ( $0.40 \mathrm{~g}, 1.73 \mathrm{mmol}$ ) and 4-cyanophenylhydrazine hydrochloride $(0.32 \mathrm{~g}, 1.90 \mathrm{mmol})$ were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as an orange solid ( $0.38 \mathrm{~g}, 63 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.33$ (DCM/EtOH/NH3 200:6:1); m.p. 135-137 ${ }^{\circ} \mathrm{C}$; $v_{\max } 3315,3253$ (N-H), 2969 (C-H), 2214 (CミN), 1679, 1609 (C=O), 1501, 1368, 1250, 1161, 1049, $826 \mathrm{~cm}^{-1}$; $\delta н\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.72$ (1H, bs, CONHNH), 7.44 - 7.41 (2H, d, J 8, Ar-H), 6.78 - 6.75 (2H, d, J 8, Ar-H), 6.66 (1H, bd, J 3, CONHNH), 5.13 (1H, bd, J 8, BocNHCH), 4.27 (1H, m, BocNHCH), 1.67 $-1.53\left(3 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.46\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.95-0.93$ $\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 0.92-0.90\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$; ठc ( 100 MHz , $\mathrm{CDCl}_{3}$ ) 173.2 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.2 ( $\mathrm{C}=\mathrm{ONH}$ ), 151.4 (ipso-Ar-C), 133.6 (Ar-C), 119.6 ( $\mathrm{C} \equiv \mathrm{N}$ ), 112.9 (Ar-C), 102.9 (ipso-Ar-C), 80.9 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 51.6$ (BocNHCH), $40.4\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 24.7\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 22.8\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $21.9\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$; $m / z\left(\mathrm{ES}^{+}\right) 369\left(\mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}$, 369.1909 ( $\mathrm{C}_{18} \mathrm{H}_{26} \mathrm{~N}_{4} \mathrm{O}_{3} \mathrm{Na}$ requires 369.1897).

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## tert-Butyl (S)-(4-methyl-1-(2-(4-(methylsulfonyl)phenyl)hydrazineyl)-1-oxopentan-2-yl)carbamate 67DI



Following the general procedure outlined, $N$-Boc-L-leucine ( $0.40 \mathrm{~g}, 1.73 \mathrm{mmol}$ ) and 4-(methylsulphonyl)phenylhydrazine hydrochloride ( $0.42 \mathrm{~g}, 1.90 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow gummy solid ( $0.37 \mathrm{~g}, 54 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.45$ (DCM/EtOH/NH3 200:6:1); $V_{\max } 3399$ (N-H), 1688, 1647 (C=O), 1597, 1524, 1303, 1291, 1138, $764 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.71 (1H, bs, CONHNH), 7.59 - 7.57 (2H, d, J 8, Ar-H), 6.73 6.71 (2H, d, J 8, Ar-H), 5.12 (1H, bd, J 8, BocNHCH), 4.19 (1H, m, BocNHCH), $2.92\left(3 \mathrm{H}, \mathrm{s}, \mathrm{SO}_{2} \mathrm{CH}_{3}\right), 1.64-1.48\left(3 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.39$ (9H, s, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.89-0.87\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 0.86-0.84(3 \mathrm{H}, \mathrm{d}, \mathrm{J}$ $\left.6, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$; $\delta с\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 173.2$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.1 ( $\mathrm{C}=\mathrm{ONH}$ ), 152.4 (ipso-Ar-C), 131.1 (ipso-Ar-C), 129.1 (Ar-C), 112.5 ( $\mathrm{Ar}-\mathrm{C}), 80.8$ (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, 51.7 (BocNHCH), $44.8\left(\mathrm{SO}_{2} \mathrm{CH}_{3}\right), 40.5\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 24.7$ $\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.8\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $21.9\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 422$ $\left(\mathrm{MNa}^{+}\right)$; HRMS ( $\mathrm{ES}^{+}$) Found $\mathrm{MNa}^{+}$, $422.1724\left(\mathrm{C}_{18} \mathrm{H}_{29} \mathrm{~N}_{3} \mathrm{O}_{5} \mathrm{SNa}\right.$ requires 422.1720).

## tert-Butyl (S)-(1-(2-(2-bromophenyl)hydrazineyl)-4-methyl-1-oxopentan-2yl)carbamate 67Ch



Following the general procedure outlined, $N$-Boc-L-leucine ( $0.40 \mathrm{~g}, 1.73 \mathrm{mmol}$ ) and 2-bromophenylhydrazine hydrochloride ( $0.43 \mathrm{~g}, 1.90 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as an orange gummy solid ( $0.32 \mathrm{~g}, 46 \%$ ); Rf 0.39 (DCM/EtOH/NH3 200:6:1); $v_{\max } 3342,3253$ (N-H), 2959 (C-H), 1721, 1655 (C=O), 1514, 1364, 1253, 1157, 1020, $751 \mathrm{~cm}^{-1}$; $\delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.30$ (1H, bs, CONHNH), 7.44 (1H, dd, J7,1, Ar-H), 6.18 (1H, dt, J 8, 1, Ar-H), 6.86 (1H, dd, J 8,1, Ar-H), 6.78

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(1H, td, J7,1, Ar-H), 6.40 (1H, bd, J3, CONHNH), 4.97 (1H, bd, J 8, BocNHCH), $4.24(1 \mathrm{H}, \mathrm{m}, \mathrm{BocNHCH}), 1.74-1.67\left(3 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.47$ (9H, s, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.98-0.96\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 0.95-0.93(3 \mathrm{H}, \mathrm{d}, \mathrm{J}$ $\left.6, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$; $\delta с$ ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 172.4 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.9 ( $\mathrm{C}=\mathrm{ONH}$ ), 144.7 (ipso-Ar-C), 132.6 ( $\mathrm{Ar}-\mathrm{C}$ ), 128.3 ( $\mathrm{Ar}-\mathrm{C}$ ), 121.9 ( $\mathrm{Ar}-\mathrm{C}$ ), 113.7 ( $\mathrm{Ar}-\mathrm{C}$ ), 109.1 (Ar-C), $80.8\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 51.8(\mathrm{BocNHCH}), 40.3\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $24.7\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.7\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.1\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; m / z\left(\mathrm{ES}^{+}\right) 422$ $\left(\mathrm{MNa}^{+}\right)$; HRMS $\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}, 422.1063\left(\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{~N}_{3} \mathrm{O}_{3}{ }^{79} \mathrm{BrNa}\right.$ requires 422.1050).

## tert-Butyl (S)-(1-(2-(3-bromophenyl)hydrazineyl)-4-methyl-1-oxopentan-2yl)carbamate 67Cn



Following the general procedure outlined, $N$-Boc-L-leucine ( $0.40 \mathrm{~g}, 1.73 \mathrm{mmol}$ ) and 3-bromophenylhydrazine hydrochloride ( $0.43 \mathrm{~g}, 1.90 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a light brown solid ( $0.36 \mathrm{~g}, 53 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.36$ ( $\mathrm{DCM} / \mathrm{EtOH} / \mathrm{NH}_{3}$ 200:6:1); m.p. 167-169 ${ }^{\circ} \mathrm{C} ; v_{\max } 3283$ (N-H), 2959, 2928 (C-H), 1697, 1681 (C=O), 1589, $1473,1367,1249,1162,768 \mathrm{~cm}^{-1} ; \delta_{H}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.43(1 \mathrm{H}, \mathrm{bs}, \mathrm{CONHNH})$, 7.04 (1H, d, J 7, Ar-H), 7.01 (1H, t, J 1, Ar-H), 6.95 (1H, t, J 1, Ar-H), 6.73 (1H, dt, J 7,1, Ar-H), 5.02 (1H, bd, J 8, BocNHCH), 4.25 (1H, m, BocNHCH), 1.71 $1.66\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH} H \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.55\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.47$ (9H, s, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.97-0.95\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 0.94-0.92(3 \mathrm{H}, \mathrm{d}, \mathrm{J}$ 6, $\left.\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$; $\delta c\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 172.9$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.1 ( $\mathrm{C}=\mathrm{ONH}$ ), 149.3 (ipso-Ar-C), 130.5 (Ar-C), 123.9 (Ar-C), 123.1 (ipso-Ar-C), 116.3 (Ar-C), 112.2 ( $\mathrm{Ar}-\mathrm{C}$ ), $80.7\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 51.5(\mathrm{BocNHCH}), 40.4\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.4$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $24.7\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.7\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.1\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; m / z$ $\left(\mathrm{ES}^{+}\right) 422\left(\mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}, 422.1066\left(\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{~N}_{3} \mathrm{O}_{3}{ }^{79} \mathrm{BrNa}\right.$ requires 422.1050 ).

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tert-Butyl (S)-(1-(2-(3,5-bis(trifluoromethyl)phenyl)hydrazineyl)-4-methyl-1-oxopentan-2-yl)carbamate 67Bv


Following the general procedure outlined, $N$-Boc-L-leucine ( $0.40 \mathrm{~g}, 1.73 \mathrm{mmol}$ ) and (3,5-bis(trifluoromethyl)phenyl)hydrazine hydrochloride ( $0.53 \mathrm{~g}, 1.90 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a light brown solid ( $0.46 \mathrm{~g}, 58 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.46$ (DCM/EtOH/NH3 200:6:1); m.p. $112-114{ }^{\circ} \mathrm{C} ; V_{\max } 3329,3267(\mathrm{~N}-\mathrm{H}), 2972(\mathrm{C}-\mathrm{H}), 1669,1621$ (C=O), 1522, 1498, 1382, 1277, 1163, $867 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.72 ( $1 \mathrm{H}, \mathrm{bs}, \mathrm{CONHNH}$ ), 7.33 (1H, s, Ar-H), $6.65(2 \mathrm{H}, \mathrm{s}, \operatorname{Ar}-H), 6.65$ (1H, bs, CONHNH), 5.02 (1H, bd, J 7, BocNHCH), 4.25 (1H, m, BocNHCH), $1.69-1.62\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CHHCH}\left(\mathrm{CH}_{3}\right) 2\right.$, $\left.\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.54\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.45\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.95-0.93$ (3H, d, J 6, CH2CH(CH3 $)_{2}$ ), $0.91-0.89\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$; ठc ( 100 MHz , $\mathrm{CDCl}_{3}$ ) 173.3 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.2 ( $\mathrm{C}=\mathrm{ONH}$ ), 149.1 (ipso-Ar-C), 132.3 (ipso-ArC), 125.1 (ipso-Ar-C), 121.4 (ipso-Ar-C), 114.1 (Ar-C), 112.9 (Ar-C), 81.1 $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 51.6(\mathrm{BocNHCH}), 40.3\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.2\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 24.7$ $\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.4\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.0\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta \mathrm{F}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ -63.2 ( $\mathrm{CF}_{3}$ ); m/z (ES ${ }^{+}$) $480\left(\mathrm{MNa}^{+}\right)$; $\mathrm{HRMS}\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}, 480.1700$ ( $\mathrm{C}_{19} \mathrm{H}_{25} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~F}_{6} \mathrm{Na}$ requires 480.1692 ).

## tert-Butyl

(S)-(4-methyl-1-oxo-1-(2-(o-tolyl)hydrazineyl)pentan-2-yl)
carbamate 67Dr


Following the general procedure outlined, $N$-Boc-L-leucine ( $0.40 \mathrm{~g}, 1.73 \mathrm{mmol}$ ) and 2-methylphenylhydrazine hydrochloride ( $0.30 \mathrm{~g}, 1.90 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a brown gummy solid ( $0.47 \mathrm{~g}, 81 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.42$ (DCM/EtOH/NH3 200:6:1);

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$V_{\max } 3281$ (N-H), 2959, 2933 (C-H), 1667, 1608 (C=O), 1496, 1366, 1249, 1163, 1046, $748 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.57 (1H, bs, CONHNH), $6.98-6.96(2 \mathrm{H}$, d, J 7, Ar-H), $6.73-6.71$ (2H, d, J 7, Ar-H), 5.13 (1H, bd, J 8, BocNHCH), 4.22 (1H, m, BocNHCH), $2.12\left(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{CH}_{3}\right), 1.64-1.54\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CHHCH}\left(\mathrm{CH}_{3}\right)_{2}\right.$, $\left.\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.48\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.37\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.88-0.84$ (6H, t, J 6, CH2CH(CH3 $)_{2}$ ); бс (100 MHz, CDCl ${ }_{3}$ ) 172.5 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.0 (C=ONH), 145.6 (ipso-Ar-C), 130.4 (Ar-C), 126.8 (Ar-C), 123.2 (Ar-C), 120.8 (ArC), 112.1 ( $\mathrm{Ar}-\mathrm{C}), 80.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 51.5(\mathrm{BocNHCH}), 40.7\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.4$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $24.7\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.8\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.2\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, 16.9 ( $\mathrm{Ar}-\mathrm{CH}_{3}$ ); m/z (ES ${ }^{+}$) $358\left(\mathrm{MNa}^{+}\right) ; \operatorname{HRMS}\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}, 358.2124$ $\left(\mathrm{C}_{18} \mathrm{H}_{29} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 358.2101).

## tert-Butyl

(S)-(4-methyl-1-oxo-1-(2-(m-tolyl)hydrazineyl)pentan-2yl)carbamate 67Dx


Following the general procedure outlined, $N$-Boc-L-leucine ( $0.40 \mathrm{~g}, 1.73 \mathrm{mmol}$ ) and 3-methylphenylhydrazine hydrochloride ( $0.30 \mathrm{~g}, 1.90 \mathrm{mmol}$ ) were transformed following flash column chromatography (DCM/EtOH/NH3 200:6:1) into the title compound which was isolated as a brown oil ( $0.22 \mathrm{~g}, 37 \%$ ); Rf 0.56 (DCM/EtOH/NH3 200:6:1); $V_{\max } 3263$ (N-H), 2959, 2929, 2872 (C-H), 1698, 1681, 1664 (C=O), 1526, 1366, 1280, 1163, $772 \mathrm{~cm}^{-1}$; бн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.42 (1H, bs, CONHNH), 7.02 (1H, t, J7, Ar-H), 6.62 (1H, d, J 7, Ar-H), 6.53 (2H, s, Ar-H), 6.06 (1H, bs, CONHNH), 5.05 (1H, bd, J 8, BocNHCH), 4.17 (1H, m, BocNHCH), 2.18 (3H, s, Ar-CH3), $1.65-1.56\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CHHCH}\left(\mathrm{CH}_{3}\right)_{2}, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.42$ (1H, m, CHHCH (CH3 $)_{2}$ ), $1.38\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.88-0.83(6 \mathrm{H}, \mathrm{t}, \mathrm{J} 6$, $\left.\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ;$ бc ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 172.8 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.9 ( $\mathrm{C}=\mathrm{ONH}$ ), 147.8 (ipso-Ar-C), 138.9 (ipso-Ar-C), 129.0 (Ar-C), 122.0 (Ar-C), 114.3 (Ar-C), 110.6 (Ar-C), $80.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 51.5(\mathrm{BocNHCH}), 40.7\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $24.7\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.8\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.1\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $21.5\left(\mathrm{Ar}-\mathrm{CH}_{3}\right)$;

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$m / z\left(\mathrm{ES}^{+}\right) 336\left(\mathrm{MH}^{+}\right)$; HRMS $\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MH}^{+}, 336.1901\left(\mathrm{C}_{18} \mathrm{H}_{3} \mathrm{~N}_{3} \mathrm{O}_{3}\right.$ requires 336.1904).
tert-Butyl (S)-(1-(2-(4-bromophenyl)hydrazineyl)-4-methyl-1-oxopentan-2yl)carbamate 67Ct


Following the general procedure outlined, $N$-Boc-L-leucine ( $0.40 \mathrm{~g}, 1.73 \mathrm{mmol}$ ) and 4-bromophenylhydrazine hydrochloride $(0.43 \mathrm{~g}, 1.90 \mathrm{mmol})$ were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a light orange solid ( $0.41 \mathrm{~g}, 59 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.53$ (DCM/EtOH/NH3 200:6:1); m.p. 141-143 ${ }^{\circ} \mathrm{C} ; V_{\max } 3307$ (N-H), 2957 (C-H), 1688, 1653 (C=O), 1520, 1488, 1366, 1270, 1171, $821 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.66 (1H, bs, CONHNH), 7.26 - 7.24 (2H, d, J 8, Ar-H), 6.66 - 6.63 (2H, d, J 8, Ar-H), 6.24 (1H, bs, CONHNH), 5.16 (1H, bd, J 8, BocNHCH), 4.24 (1H, m, BocNHCH), 1.68 - 1.57 (2H, m, $\left.\mathrm{CHHCH}\left(\mathrm{CH}_{3}\right)_{2}, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.53\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.45(9 \mathrm{H}, \mathrm{s}$, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.94-0.89\left(6 \mathrm{H}, \mathrm{t}, \mathrm{J} 6, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 172.9$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.0 ( $\mathrm{C}=\mathrm{ONH}$ ), 147.0 (ipso-Ar-C), 131.9 (Ar-C), 115.1 (Ar-C), 112.9 (ipso-Ar-C), 80.6 ((CH3)3CO), 51.5 (BocNHCH), $40.6\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.3$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $24.7\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.8\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.1\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; m / z$ (ES ${ }^{+}$) 422 ( $\left.{ }^{79} \mathrm{Br}\right] \mathrm{MNa}^{+}$), 424 ( $\left.{ }^{81} \mathrm{Br}\right] \mathrm{MNa}^{+}$); HRMS (ES ${ }^{+}$) Found ${ }^{79} \mathrm{Br}^{7} \mathrm{MNa}^{+}$, $422.1058\left(\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{~N}_{3} \mathrm{O}_{3}{ }^{79} \mathrm{BrNa}\right.$ requires 422.1050$)$.

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## tert-Butyl (S)-(4-methyl-1-oxo-1-(2-(2-(trifluoromethyl)phenyl)hydrazineyl) pentan-2-yl)carbamate 67Bb



Following the general procedure outlined, $N$-Boc-L-leucine ( $0.40 \mathrm{~g}, 1.73 \mathrm{mmol}$ ) and 2-trifluoromethylphenylhydrazine hydrochloride ( $0.40 \mathrm{~g}, 1.90 \mathrm{mmol}$ ) were transformed following flash column chromatography (DCM/EtOH/NH3 200:8:1) into the title compound which was isolated as a yellow oil ( $0.29 \mathrm{~g}, 43 \%$ ) $\mathrm{Rf}_{\mathrm{f}} 0.32$ (DCM/EtOH/NH3 200:8:1); $v_{\max } 3288$ (N-H), 2961 (C-H), 1671, 1614 (C=O), 1524, 1322, 1276, 1164, 1109, $1035 \mathrm{~cm}^{-1}$; $\delta н\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 9.18$ (1H, bs, CONHNH), 7.36 - 7.31 (2H, t, J 8, Ar-H), 7.16 (1H, t, J 8, Ar-H), 6.87 (1H, d, J 8, Ar-H), 6.49 (1H, bs, CONHNH), 5.50 ( $1 \mathrm{H}, \mathrm{bd}, \mathrm{J} 8$, BocNHCH), $4.34(1 \mathrm{H}, \mathrm{m}$, BocNHCH $)$, $1.59-1.47\left(3 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)^{2} 1.33(9 \mathrm{H}, \mathrm{s}$, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.84-0.80\left(6 \mathrm{H}, \mathrm{t}, \mathrm{J} 5, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta c\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 173.0$ ( $C=O N H N H$ ), 156.1 ( $C=O N H$ ), 145.4 (ipso-Ar-C), 132.9 (Ar-C), 126.3 (1C, ipso-Ar-CCF ${ }_{3}$ ), 122.7 (Ar-C), 119.9 (Ar-C), 114.9 (1C, q, J 31, Ar-CF3), 113.5 (Ar-C), $80.5\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 51.4(\mathrm{BocNHCH}), 40.8\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 24.7$ $\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.6\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.1\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta$ ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) -61.9 ( $\mathrm{CF}_{3}$ ); m/z (ES ${ }^{+}$) $412\left(\mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}, 412.1798$ ( $\mathrm{C}_{18} \mathrm{H}_{26} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~F}_{3} \mathrm{Na}$ requires 412.1819 ).

## tert-Butyl (S)-(4-methyl-1-oxo-1-(2-(3-(trifluoromethyl)phenyl)hydrazineyl) pentan-2-yl)carbamate 67Bh



Following the general procedure outlined, $N$-Boc-L-leucine ( $0.40 \mathrm{~g}, 1.73 \mathrm{mmol}$ ) and 3-trifluoromethylphenylhydrazine hydrochloride ( $0.40 \mathrm{~g}, 1.90 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was

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isolated as a yellow oil ( $0.38 \mathrm{~g}, 56 \%$ ); Rf 0.38 (DCM/EtOH/NH3 200:6:1); $v_{\text {max }}$ 3327, 3278 (N-H), 2959 (C-H), 1680,1671, 1663 (C=O), 1519, 1333, 1249, 1112, $1071 \mathrm{~cm}^{-1}$; $\delta_{H}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.93$ (1H, bs, CONHNH), 7.15 (1H, d, J7, Ar-H), 7.01 (1H, d, J 7, Ar-H), 6.92 (1H, s, Ar-H), 6.84 (1H, dd, J 7, 1, Ar-H), 6.46 (1H, bs, CONHNH), 5.18 (1H, bd, J 8, BocNHCH), 4.26 (1H, m, BocNHCH), 1.57 $1.51\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CHHCH}\left(\mathrm{CH}_{3}\right)_{2}, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.45\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.35$ ( $\left.9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.84\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 0.81(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6$, $\left.\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ;$ ठc ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 173.4 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.1 ( $\left.\mathrm{C}=\mathrm{ONH}\right)$, 148.4 (ipso-Ar-C), 132.1 (1C, q, J 31, Ar-CF3), 129. 6 (Ar-C), 117.3 (Ar-C), 116.4 (ArC), $109.8(\mathrm{Ar}-\mathrm{C}), 80.6\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 51.5(\mathrm{BocNHCH}), 40.8\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.2$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $24.7\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.5\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.1\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta_{F}$ ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) -62.8 ( $\mathrm{CF}_{3}$ ); m/z ( $\left.\mathrm{ES}^{+}\right) 412$ ( $\mathrm{MNa}^{+}$); HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}$, $412.1838\left(\mathrm{C}_{18} \mathrm{H}_{26} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~F}_{3} \mathrm{Na}\right.$ requires 412.1818).

## tert-Butyl (S)-(4-methyl-1-(2-(3-nitrophenyl)hydrazineyl)-1-oxopentan-2-yl) carbamate 67Df



Following the general procedure outlined, $N$-Boc-L-leucine ( $0.40 \mathrm{~g}, 1.73 \mathrm{mmol}$ ) and 3-nitrophenylhydrazine hydrochloride ( $0.36 \mathrm{~g}, 1.90 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow solid ( $0.51 \mathrm{~g}, 81 \%$ ); $\mathrm{Rf}_{f} 0.51$ (DCM/EtOH/NH3 200:6:1); m.p. $127-129{ }^{\circ} \mathrm{C} ; \mathrm{V}_{\max }$ 3329 (N-H), 2956 (C-H), 1665, 1619 (C=O), 1519, 1348, 1322, 1279, 1161, 733, $647 \mathrm{~cm}^{-1}$; бн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.62 (1H, bs, CONHNH), 7.63 (1H, d, J 2, Ar-H), 7.53 (1H, s, Ar-H), 7.25 (1H, d, J 7, Ar-H), 7.04 (1H, m, Ar-H), 6.46 (1H, bs, CONHNH), 5.01 (1H, bs, BocNHCH), 4.20 (1H, m, BocNHCH), 1.65 - 1.57 (2H, m, $\left.\mathrm{CHHCH}\left(\mathrm{CH}_{3}\right)_{2}, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.51\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.39(9 \mathrm{H}, \mathrm{s}$, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 0.90\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 5, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 0.87\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 5, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta c$ (100 MHz, $\mathrm{CDCl}_{3}$ ) 173.3 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.2 ( $\mathrm{C}=\mathrm{ONH}$ ), 151.9 (ipso-Ar-C), 149.2 (ipso-Ar-C), 129.8 (Ar-C), 119.2 (Ar-C), 115.5 (Ar-C), 107.6 (Ar-C), 80.9 $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 51.5(\mathrm{BocNHCH}), 40.5\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 24.8$

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$\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.7\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $22.1\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 389$ $\left(\mathrm{MNa}^{+}\right)$; $\mathrm{HRMS}\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}, 389.1807\left(\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{~N}_{4} \mathrm{O}_{5} \mathrm{Na}\right.$ requires 389.1795).
tert-Butyl (S)-(1-(2-(2-chlorophenyl)hydrazineyl)-4-(methylthio)-1-oxobutan-2-yl)carbamate 67Ak


Following the general procedure outlined, N-Boc-L-methionine ( $0.30 \mathrm{~g}, 1.20$ mmol ) and 2-chlorophenylhydrazine hydrochloride ( $0.26 \mathrm{~g}, 1.33 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a light brown solid ( $0.25 \mathrm{~g}, 54 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.38$ ( $\mathrm{DCM} / \mathrm{EtOH} / \mathrm{NH}_{3}$ 200:6:1); m.p. 119-121 ${ }^{\circ} \mathrm{C}$; $v_{\max } 3276$ (N-H), 2986, 2918 (C-H), 1674 (C=O), 1489, 1367, 1249, 1161, 1034, $745 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.77 (1H, bs, CONHNH), 7.26 (1H, dd, J 7, 1, Ar-H), 7.12 (1H, td, J 7, 1, Ar-H), 6.86 (1H, dd, J 8, 1, Ar-H), 6.80 (1H, dd, J 7, 1, Ar-H), 6.48 (1H, bd, J 3, Ar-NH), 5.45 (1H, bd, J 8, BocNHCH), 4.47 (1H, m, BocNHCH), $2.58\left(2 \mathrm{H}, \mathrm{t}, \mathrm{J}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right), 2.09\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right)$, $2.01\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right), 1.44\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 172.3$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.0 ( $\mathrm{C}=\mathrm{ONH}$ ), 146.6 (ipso-Ar-C), 129.2 (Ar-C), 127.8 (Ar-C), 121.2 (ipso-Ar-C), 119.3 (ipso-Ar-C), 113.5 ( $\mathrm{Ar}-\mathrm{C}$ ), 80.5 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 52.0$ ( BocNHCH ), $31.5\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 30.1\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 15.3$ $\left.\left.\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right) ; \quad m / z \quad\left(\mathrm{ES}^{+}\right) \quad 396 \quad\left({ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right), 338 \quad\left({ }^{37} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right), 769$ $\left.\left({ }^{35,35} \mathrm{Cl}\right] 2 \mathrm{M}+\mathrm{Na}^{+}\right) 771$ ( $\left[^{35,37} \mathrm{Cl}\right] 2 \mathrm{M}+\mathrm{Na}^{+}$), 773 ( $\left.{ }^{37,37} \mathrm{Cl}\right] 2 \mathrm{M}+\mathrm{Na}^{+}$); HRMS (ES ${ }^{+}$) Found $\left[{ }^{35} \mathrm{CI}\right] \mathrm{MNa}^{+}, 396.1142\left(\mathrm{C}_{16} \mathrm{H}_{24}{ }^{35} \mathrm{CIN}_{3} \mathrm{O}_{3} \mathrm{SNa}\right.$ requires 396.1119).
(9H-fluoren-9-yl)methyl (methylthio)-1-oxobutan-2-yl)carbamate 68u


Following the general procedure outlined, $N$-Fmoc-L-methionine ( $0.40 \mathrm{~g}, 1.08$ mmol ) and 3-chlorophenylhydrazine hydrochloride ( $0.21 \mathrm{~g}, 1.19 \mathrm{mmol}$ ) were transformed following flash column chromatography (DCM/EtOH/NH3 200:6:1) into the title compound which was isolated as a light brown solid ( $0.28 \mathrm{~g}, 52 \%$ ); Rf 0.42 (DCM/EtOH/NH3 200:6:1); m.p. $186-189{ }^{\circ} \mathrm{C}$; $v \max 3278$ (N-H), 1686, 1655 (C=O), 1541, 1476, 1304, 1265, 756, 740, 679 cm-1; סн ( 300 MHz , DMSOd6) 9.93 (1H, bs, CONHNH), $8.07-8.06$ (1H, bd, J 2, Ar-NH), $7.91-7.89$ (2H, d, J 7, Ar-H), $7.75-7.72$ (2H, m, Ar-H), 7.45-7.40 (2H, t, J 7, Ar-H), 7.35-7.31 (2H, t, J 7, Ar-H), 7.12 (1H, t, J 7, Ar-H), $6.73-6.63(3 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}), 4.33(2 \mathrm{H}, \mathrm{m}$, $\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}$-Fluorene), $4.27\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene), $4.14(1 \mathrm{H}, \mathrm{m}$, FmocNHCH), $2.56\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right), 2.08\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 1.94$ ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{~S}$ ); $\delta \mathrm{c}(100 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d} 6) 172.1$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.6 (C=ONH), 151.3 (ipso-Ar-C), 144.2 (ipso-Ar-C), 141.2 (ipso-Ar-C), 133.9 (ipso-Ar-C), 130.8 (Ar-C), 128.1 (Ar-C), 127.5 (Ar-C), 125.8 (Ar-C), 120.6 (Ar-C), 118.3 (Ar-C), 111.9 (Ar-C), 111.3 (Ar-C), $66.3 \quad\left(\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene), 53.0 (FmocNHCH), 47.2 (C-Fluorene), $31.9\left(\mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right), 30.2\left(\mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right), 15.1$ $\left.\left(\mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 496\left(\left[{ }^{35} \mathrm{CI}\right] \mathrm{MH}^{+}\right), 498\left(\left[{ }^{37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 518\left({ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right)$, $520\left(\left[{ }^{37} \mathrm{CI}\right] \mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\left[{ }^{35} \mathrm{CI}\right] \mathrm{MH}^{+}, 496.1456\left(\mathrm{C}_{26} \mathrm{H}_{27}{ }^{35} \mathrm{ClN}_{3} \mathrm{O}_{3} \mathrm{~S}\right.$ requires 496.1456).

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tert-Butyl (S)-(1-(2-(4-isopropylphenyl)hydrazineyl)-4-(methylthio)-1-oxobutan-2-yl)carbamate 67Ek


Following the general procedure outlined, $N$-Boc-L-methionine $(0.30 \mathrm{~g}, 1.20$ mmol ) and 2-chlorophenylhydrazine hydrochloride ( $0.25 \mathrm{~g}, 1.32 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as an orange solid ( $0.22 \mathrm{~g}, 48 \%$ ); $\mathrm{R}_{\mathrm{f}} 0.36$ (DCM/EtOH/NH3 200:6:1); m.p. 147-149 ${ }^{\circ} \mathrm{C}$; $v_{\max } 3316$ (N-H), 2959, 2919 (C-H), 1686, 1656 (C=O), 1514, 1456, 1365, 1251, 1167, 824, $536 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.61 (1H, bs, CONHNH), 7.05 (2H, d, J 8, Ar-H), 6.73 (2H, d, J 8, Ar-H), 6.15 (1H, bs, Ar-NH), 5.39 (1H, bd, J 8, BocNHCH), 4.40 (1H, q, J 8, BocNHCH), 2.81 (1H, Hept, J 7, CH(CH3)2), $2.53\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right), 2.08\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 1.94\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right)$, $1.45\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.19\left(6 \mathrm{H}, \mathrm{d}, \mathrm{J} 1, \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$; $\delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 170.5$ (C=ONHNH), 155.9 ( $\mathrm{C=ONH}$ ), 146.3 (ipso-Ar-C), 138.9 (ipso-Ar-C), 127.0 (Ar-C), 112.9 ( $\mathrm{Ar}-\mathrm{C}$ ), $79.8\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 60.3(\mathrm{BocNHCH}), 31.7\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 30.2$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right)$, $28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $23.3\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 15.3\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right) ; \mathrm{m} / \mathrm{z}$ $\left(\mathrm{ES}^{+}\right) 404\left(\mathrm{MNa}^{+}\right) ;$HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}, 404.1996\left(\mathrm{C}_{19} \mathrm{H}_{3} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{SNa}\right.$ requires 404.1978).

## tert-Butyl (S)-(1-(2-(4-(methylsulfonyl)phenyl)hydrazineyl)-4-(methylthio)-1-oxobutan-2-yl)carbamate 67Dm



Following the general procedure outlined, $N$-Boc-L-methionine ( $0.40 \mathrm{~g}, 1.60$ mmol ) and 4-(methylsulphonyl)phenylhydrazine hydrochloride ( $0.39 \mathrm{~g}, 1.77 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow solid ( $0.41 \mathrm{~g}, 61 \%$ ); $\mathrm{R}_{\mathrm{f}} 0.43$ (DCM/EtOH/NH3 200:6:1); m.p. 141-143 ${ }^{\circ} \mathrm{C}$; $v_{\max } 3336,3256$ (N-H), 2984, 2915 (C-H), 1683, 1662 (C=O), 1598,

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1322, 1285, 1129, 1091, $772 \mathrm{~cm}^{-1}$; $\delta н\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.49$ (1H, bs, CONHNH), 7.76 - 7.73 (2H, d, J 8, Ar-H), $6.90-6.87$ (2H, d, J 8, Ar-H), 5.26 (1H, bd, J 7, BocNHCH), $4.44(1 \mathrm{H}, \mathrm{m}, \mathrm{BocNHCH}), 3.02\left(3 \mathrm{H}, \mathrm{s}, \mathrm{SO}_{2} \mathrm{CH}_{3}\right), 2.63(2 \mathrm{H}, \mathrm{t}, \mathrm{J} 6$, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~S}$ ), $2.14\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right)$, $2.06\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right), 1.50(9 \mathrm{H}, \mathrm{s}$, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta с\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 172.1$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 152.3 ( $\mathrm{C}=\mathrm{ONH}$ ), 131.7 (ipso-Ar-C), 129.2 (Ar-C), 112.7 (Ar-C), $44.9\left(\mathrm{SO}_{2} \mathrm{CH}_{3}\right), 30.2\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right)$, $28.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 15.5\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 440\left(\mathrm{MNa}^{+}\right) ;$HRMS (ES $\left.{ }^{+}\right)$ Found $\mathrm{MNa}^{+}, 440.1296\left(\mathrm{C}_{17} \mathrm{H}_{27} \mathrm{~N}_{3} \mathrm{O}_{5} \mathrm{SNa}\right.$ requires 417.1392).

## tert-Butyl <br> (S)-(1-(2-(2-bromophenyl)hydrazineyl)-4-(methylthio)-1-oxobutan-2-yl)carbamate 67Ci



Following the general procedure outlined, $N$-Boc-L-methionine ( $0.40 \mathrm{~g}, 1.60$ mmol ) and 2-bromophenylhydrazine hydrochloride ( $0.33 \mathrm{~g}, 1.77 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as an orange oil ( $0.36 \mathrm{~g}, 53 \%$ ); $\mathrm{R}_{\mathrm{f}} 0.40$ ( $\mathrm{DCM} / \mathrm{EtOH} / \mathrm{NH}_{3}$ 200:6:1); $v_{\text {max }}$ 3279 (N-H), 2977, 2918 (C-H), 1670 (C=O), 1594, 1486, 1366, 1249, 1160, 1022, $744 \mathrm{~cm}^{-1}$; $\delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 9.07$ (1H, bs, CONHNH), 7.38 (1H, dt, J 8, 2, ArH), 7.09 (1H, m, Ar-H), 6.80 (1H, dd, J8, 2, Ar-H), 6.70 (1H, dt, J 8, 2, Ar-H), 6.44 (1H, bd, J 3, Ar-H), 5.64 (1H, bd, J 8, BocNHCH), 4.47 (1H, q, J 8, BocNHCH), $2.53\left(2 \mathrm{H}, \mathrm{t}, \mathrm{J} 7, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right), 2.06\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 1.91\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right)$, $1.42\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; \delta c\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 171.6$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.9 (C=ONH), 144.4 (ipso-Ar-C), 132.7 (Ar-C), 128.1 (Ar-C), 121.5 (Ar-C), 113.8 (ArC), 108.6 (ipso-Ar-C), $81.1\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 51.6(\mathrm{BocNHCH}), 31.0\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right)$, $30.2\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 27.8\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $15.5\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 440$ ( $\left.{ }^{79} \mathrm{Br}\right] \mathrm{MNa}^{+}$), 442 ( $\left.{ }^{81} \mathrm{Br}\right] \mathrm{MNa}^{+}$); HRMS (ES ${ }^{+}$) Found $\left.{ }^{79}{ }^{3} \mathrm{Br}\right] \mathrm{MNa}^{+}, 440.0622$ $\left(\mathrm{C}_{16} \mathrm{H}_{24}{ }^{79} \mathrm{BrN}_{3} \mathrm{O}_{3} \mathrm{SNa}\right.$ requires 440.0614$)$.

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## tert-Butyl (S)-(1-(2-(3-bromophenyl)hydrazineyl)-4-(methylthio)-1-

 oxobutan-2-yl)carbamate 67Co

Following the general procedure outlined, $N$-Boc-L-methionine ( $0.30 \mathrm{~g}, 1.20$ mmol ) and 3-bromophenylhydrazine hydrochloride ( $0.25 \mathrm{~g}, 1.32 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a pale yellow solid ( $0.23 \mathrm{~g}, 45 \%$ ); $\mathrm{R}_{\mathrm{f}} 0.41$ (DCM/EtOH/NH3 200:6:1); m.p. $112-114{ }^{\circ} \mathrm{C}$; $v_{\max } 3332,3271$ (N-H), 2991 (C-H), 1683, 1659 (C=O), 1522, $1473,1245,1164,766 \mathrm{~cm}^{-1}$; $\delta$ н ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.70 (1H, bs, CONHNH), 7.03 (3H, m, Ar-H), $6.70(1 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-H), 6.30(1 \mathrm{H}, \mathrm{bd}, \mathrm{J} 4, \mathrm{Ar}-H), 5.38$ (1H, bd, J 8, BocNHCH), 4.40 (1H, q, J 8, BocNHCH), $2.55\left(2 \mathrm{H}, \mathrm{t}, \mathrm{J} 7, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right), 2.09(3 \mathrm{H}, \mathrm{s}$, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}$ ), $1.93\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right), 1.46\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta \mathrm{c}(100 \mathrm{MHz}$, $\mathrm{CDCl}_{3}$ ) 172.1 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.9 ( $\mathrm{C}=\mathrm{ONH}$ ), 149.2 (ipso-Ar-C), 130.0 (Ar-C), 123.9 (Ar-C), 123.1 (Ar-C), 116.1 (Ar-C), 112.6 (ipso-Ar-C), $81.5\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $53.2(\mathrm{BocNHCH}), 31.1\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 30.2\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 28.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $\left.\left.15.4\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right) ; m / z\left(\mathrm{ES}^{+}\right) 440\left({ }^{79} \mathrm{Br}\right] \mathrm{MNa}^{+}\right), 442\left({ }^{81} \mathrm{Br}\right] \mathrm{MNa}^{+}\right) ;$HRMS (ES $\left.{ }^{+}\right)$ Found $\left[{ }^{79} \mathrm{Br}\right] \mathrm{MNa}^{+}, 440.0626\left(\mathrm{C}_{16} \mathrm{H}_{24}{ }^{79} \mathrm{BrN}_{3} \mathrm{O}_{3} \mathrm{SNa}\right.$ requires 440.0614$)$.

## tert-Butyl

(S)-(1-(2-(3,5-bis(trifluoromethyl)phenyl)hydrazineyl)-4-(methylthio)-1-oxobutan-2-yl)carbamate 67Bw


Following the general procedure outlined, $N$-Boc-L-methionine ( $0.40 \mathrm{~g}, 1.60$ mmol ) and (3,5-bis(trifluoromethyl)phenyl)hydrazine hydrochloride ( $0.49 \mathrm{~g}, 1.77$ $\mathrm{mmol})$ were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a pale yellow solid ( $0.43 \mathrm{~g}, 69 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.39$ (DCM/EtOH/NH3 200:6:1); m.p. $128-130{ }^{\circ} \mathrm{C}$; $V_{\max }$ 3331, 3271 (N-H), 2994 (C-H), 1679, 1666 (C=O), 1521, 1378, 1275, 1118, $877 \mathrm{~cm}^{-1}$; $\delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.50(1 \mathrm{H}, \mathrm{bs}$, CONHNH), 7.36 (1H, s, Ar-H), 7.21 (2H, s, Ar-H), 6.45 (1H, bd, J 3, Ar-NH), 5.19

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(1H, bd, J7, BocNHCH), 4.44 (1H, q, J 7, BocNHCH), 2.62 (2H, t, J7, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~S}$ ), $2.22\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHHCH}_{2} \mathrm{~S}\right), 2.11\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 2.05\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH} \mathrm{CHH}_{2} \mathrm{~S}\right)$, 1.47 (9H, s, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 172.3$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 149.1 (ipso-ArC), 132.8 (ipso-Ar-C), 132.4 (ipso-Ar-C), 114.3 (Ar-C), 112.7 (Ar-C), 30.2 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 28.2\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 15.4\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right) ; \delta \mathrm{F}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)-$ $63.1\left(\mathrm{CF}_{3}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 498\left(\mathrm{MNa}^{+}\right) ; \mathrm{HRMS}\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}, 498.1265$ $\left(\mathrm{C}_{18} \mathrm{H}_{23} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~F}_{6} \mathrm{SNa}\right.$ requires 498.1257$)$.
tert-Butyl (S)-(4-(methylthio)-1-oxo-1-(2-(o-tolyl)hydrazineyl)butan-2yl)carbamate 67Ds


Following the general procedure outlined, $N$-Boc-L-methionine ( $0.40 \mathrm{~g}, 1.60$ mmol ) and 2-methylphenylhydrazine hydrochloride ( $0.28 \mathrm{~g}, 1.77 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a brown oil ( $0.39 \mathrm{~g}, 68 \%$ ); $\mathrm{R}_{\mathrm{f}} 0.42$ (DCM/EtOH/NH3 200:6:1); $v_{\text {max }}$ 3385 (N-H), 2977, 2918 (C-H), 1669, 1607 (C=O), 1495, 1366, 1248, 1161, 1048, $749 \mathrm{~cm}^{-1}$; $\delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.78$ (1H, bs, CONHNH), $7.00-6.95(2 \mathrm{H}, \mathrm{t}, \mathrm{J} 7$, Ar-H), 6.75 - 6.69 (2H, t, J 8, Ar-H), 5.49 (1H, bd, J 8, BocNHCH), 4.40 (1H, m, BocNHCH), $2.47\left(2 \mathrm{H}, \mathrm{t}, \mathrm{J} 7, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right), 2.10\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 2.05(1 \mathrm{H}, \mathrm{m}$, $\mathrm{CHHCH}_{2} \mathrm{~S}$ ), $1.99\left(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{CH}_{3}\right), 1.89\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHHCH}_{2} \mathrm{~S}\right), 1.36(9 \mathrm{H}, \mathrm{s}$, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta с$ ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 171.8 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.9 ( $\mathrm{C}=\mathrm{ONH}$ ), 145.5 (ipso-Ar-C), 130.5 (Ar-C), 126.9 (Ar-C), 123.2 (ipso-Ar-C), 120.9 (Ar-C), 112.1 (Ar-C), $80.5\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 52.0 \quad(\mathrm{BocNHCH}), 31.4 \quad\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), \quad 30.1$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 28.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 16.9\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 15.4\left(\mathrm{Ar}-\mathrm{CH}_{3}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right)$ $376\left(\mathrm{MNa}^{+}\right)$; HRMS $\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}, 376.1685\left(\mathrm{C}_{17} \mathrm{H}_{27} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{SNa}\right.$ requires 376.1665).

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## tert-Butyl (S)-(4-(methylthio)-1-oxo-1-(2-(m-tolyl)hydrazineyl)butan-2yl)carbamate 67Dy



Following the general procedure outlined, $N$-Boc-L-methionine ( $0.40 \mathrm{~g}, 1.60$ mmol ) and 2-methylphenylhydrazine hydrochloride ( $0.28 \mathrm{~g}, 1.77 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a brown oil ( $0.39 \mathrm{~g}, 69 \%$ ); $\mathrm{R}_{\mathrm{f}} 0.49$ ( $\mathrm{DCM} / E t O H / \mathrm{NH}_{3}$ 200:6:1); m.p. 132 - $134{ }^{\circ} \mathrm{C}$; $V_{\max }$ 3324, 3281, 3264 (N-H), 2971, 2918 (C-H), 1684, 1656 (C=O), 1523, 1498, 1304, 1243, 1164, $776 \mathrm{~cm}^{-1}$; $\delta \mathrm{H}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.58(1 \mathrm{H}$, bs, CONHNH), 7.07 (1H, m, Ar-H), 6.70 (1H, d, J 7, Ar-H), 6.59 (1H, bd, J 1, Ar-NH), 5.40 (1H, bd, J 8, BocNHCH), 4.40 (1H, d, J 8, BocNHCH), 2.54 (2H, t, J 7, $\left.\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right), 2.26\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 2.09\left(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{CH}_{3}\right), 1.89(2 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right), 1.45\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; ~ \delta c\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 171.9$ (C=ONHNH), 155.1 (C=ONH), 147.3 (ipso-Ar-C), 139.2 (ipso-Ar-C), 129.0 (Ar-C), 122.1 (Ar-C), 114.3 (Ar-C), 110.6 (Ar-C), $80.6\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 52.0$ (BocNHCH), 31.3 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 30.1\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right)$, $28.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, $21.5\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right)$, $15.4\left(\mathrm{Ar}-\mathrm{CH}_{3}\right) ; m / z\left(\mathrm{ES}^{+}\right) 376\left(\mathrm{MNa}^{+}\right) ; \mathrm{HRMS}\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}, 376.1685$ $\left(\mathrm{C}_{17} \mathrm{H}_{27} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{SNa}\right.$ requires 376.1665).

## tert-Butyl (S)-(1-(2-(4-bromophenyl)hydrazineyl)-4-(methylthio)-1-oxobutan-2-yl)carbamate 67 Cu



Following the general procedure outlined, $N$-Boc-L-methionine ( $0.40 \mathrm{~g}, 1.60$ mmol ) and 4-bromophenylhydrazine hydrochloride ( $0.39 \mathrm{~g}, 1.77 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a brown gummy solid ( $0.34 \mathrm{~g}, 51 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.41$ ( $\mathrm{DCM} / E t O H / \mathrm{NH}_{3} 200: 6: 1$ ); $V_{\max } 3302$ (N-H), 2979, 2918 (C-H), 1683, 1652 (C=O), 1518, 1488, 1377, 1251, $1163,1003,819 \mathrm{~cm}^{-1}$; бн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.63 (1H, bs, CONHNH), 7.21 - 7.18

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(2H, d, J 8, Ar-H), $6.60-6.57$ (2H, d, J 8, Ar-H), 5.35 (1H, bd, J 8, BocNHCH), $4.34(1 \mathrm{H}, \mathrm{m}, \mathrm{BocNHCH}), 2.48\left(2 \mathrm{H}, \mathrm{t}, \mathrm{J} 7, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right), 2.07(1 \mathrm{H}, \mathrm{m}, \mathrm{CHHCH} 2 \mathrm{~S})$, $2.01\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 1.92(1 \mathrm{H}, \mathrm{m}, \mathrm{CHHCH} 2 \mathrm{~S}), 1.38\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta c$ ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 172.2 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.9 ( $\mathrm{C}=\mathrm{ONH}$ ), 146.9 (ipso-Ar-C), 131.9 ( $\mathrm{Ar}-\mathrm{C}$ ), 115.1 ( $\mathrm{Ar}-\mathrm{C}$ ), 113.1 (ipso-Ar-C), 80.8 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 51.9$ ( BocNHCH ), 31.1 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 30.2\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 28.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 15.4\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right) ; \mathrm{m} / \mathrm{z}$ (ES ${ }^{+}$) $\left.440\left({ }^{79} \mathrm{Br}\right] \mathrm{MNa}^{+}\right), 442$ ( $\left.{ }^{81} \mathrm{Br}\right] \mathrm{MNa}^{+}$); HRMS (ES ${ }^{+}$) Found $\left.{ }^{79}{ }^{79} \mathrm{Br}\right] \mathrm{MNa}^{+}$, $440.0628\left(\mathrm{C}_{16} \mathrm{H}_{24}{ }^{79} \mathrm{BrN}_{3} \mathrm{O}_{3} \mathrm{SNa}\right.$ requires 440.0614).
tert-Butyl (S)-(4-(methylthio)-1-oxo-1-(2-(2-(trifluoromethyl)phenyl) hydrazineyl)butan-2-yl)carbamate 67Bc


Following the general procedure outlined, $N$-Boc-L-methionine ( $0.40 \mathrm{~g}, 1.60$ mmol ) and 2-trifluoromethylphenylhydrazine hydrochloride ( $0.38 \mathrm{~g}, 1.77 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a brown oil ( $0.33 \mathrm{~g}, 50 \%$ ); Rf 0.53 (DCM/EtOH/NH3 200:6:1); $V_{\text {max }}$ 3378, 3312 (N-H), 2976 (C-H), 1676, 1614 (C=O), 1523, 1474, 1322, 1285, 1166, 1033, 767, $640 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.95 (1H, bs, CONHNH), 7.38 (1H, d, J 7, Ar-H), 7.23 (1H, m, Ar-H), 6.89 (1H, d, J 8, Ar-H), 6.81 (1H, t, J7, Ar-H), 6.50 (1H, bs, Ar-NH), 5.54 (1H, bd, J8, BocNHCH), 4.47 (1H, m, BocNHCH), 2.51 ( 2 H , $\left.\mathrm{t}, \mathrm{J} 7, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right), 2.00\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 1.94\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right), 1.35(9 \mathrm{H}$, $\left.\mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 172.3$ ( $\left.\mathrm{C}=\mathrm{ONHNH}\right)$, 156.1 ( $\left.\mathrm{C}=\mathrm{ONH}\right), 145.3$ (ipso-Ar-C), 133.0 (Ar-CF3), 126.5 (1C, t, J 5, ipso-Ar-CCF3), 122.7 (Ar-C), 120.0 (Ar-C), 113.4 (Ar-C), 111.9 (Ar-C), $80.7\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 51.9$ (BocNHCH), 31.2 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 30.1\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 15.3\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right) ; \delta_{F}$ (300 MHz, CDCl 3 ) -61.8 (CF3); m/z (ES ${ }^{+}$) $408\left(\mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\mathrm{MH}^{+}$, $430.1383\left(\mathrm{C}_{18} \mathrm{H}_{24} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~F}_{3} \mathrm{SNa}\right.$ requires 430.1386$)$.

## Chapter 6: Experimental section

## tert-Butyl (S)-(4-(methylthio)-1-oxo-1-(2-(3-(trifluoromethyl)phenyl) hydrazineyl)butan-2-yl)carbamate 67Bi



Following the general procedure outlined, $N$-Boc-L-methionine ( $0.40 \mathrm{~g}, 1.60$ mmol ) and 3-trifluoromethylphenylhydrazine hydrochloride ( $0.38 \mathrm{~g}, 1.77 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow solid ( $0.43 \mathrm{~g}, 66 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.51$ (DCM/EtOH/NH3 200:6:1); m.p. 125-127 ${ }^{\circ} \mathrm{C}$; $v_{\max } 3329,3279$ (N-H), 1681, 1667 (C=O), 1517, 1334, 1246, 1112, 1071, $784 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.82 (1H, bs, CONHNH), 7.29 (1H, t, J 7, Ar-H), 7.11 (1H, d, J 7, Ar-H), 7.02 (1H, s, Ar-H), 6.96 (1H, d, J 8, Ar-H), 6.46 (1H, bs, Ar-NH), 5.42 (1H, bd, J 8, BocNHCH), 4.47 (1H, q, J 7, BocNHCH), 2.57 (2H, $\left.\mathrm{t}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right), 2.17\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHHCH}_{2} \mathrm{~S}\right), 2.08\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 2.00(1 \mathrm{H}$, $\mathrm{m}, \mathrm{CHHCH} 2 \mathrm{~S}), 1.45\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; ~ \delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 172.5$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.0 (C=ONH), 148.2 (ipso-Ar-C), 131.7 (1C, q, J 32, Ar-CF3), 129.7 (Ar-C), 125.8 (1C, q, J 271, ipso-Ar-CCF3), 117.6 (Ar-C), 116.5 (Ar-C), 109.9 (Ar-C), 80.9 $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 52.0(\mathrm{BocNHCH}), 31.0\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 30.1\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 28.3$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 15.3\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right) ; ~ \delta \mathrm{~F}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)-61.8\left(\mathrm{CF}_{3}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right)$ $408\left(\mathrm{MNa}^{+}\right)$; HRMS $\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MH}^{+}, 430.1383\left(\mathrm{C}_{18} \mathrm{H}_{24} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~F}_{3} \mathrm{SNa}\right.$ requires 430.1386).
tert-Butyl (S)-(4-(methylthio)-1-(2-(3-nitrophenyl)hydrazineyl)-1-oxobutan-

## 2-yl)carbamate 67Dg



Following the general procedure outlined, $N$-Boc-L-methionine ( $0.40 \mathrm{~g}, 1.60$ mmol ) and 3-nitrophenylhydrazine hydrochloride ( $0.33 \mathrm{~g}, 1.77 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow solid ( $0.41 \mathrm{~g}, 66 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.29$ (DCM/EtOH/NH3 200:6:1); m.p.

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124-126 ${ }^{\circ} \mathrm{C}$; $v_{\max } 3323$ (N-H), 1678, 1665 (C=O), 1517, 1349, 1295, 1051, 734, $653 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.64 (1H, bs, CONHNH), 7.74 (1H, dd, J 8, 1, ArH), 7.64 (1H, t, J 2, Ar-H), 7.37 (1H, t, J 8, Ar-H), 7.14 (1H, dd, J 8, 1, Ar-H), 6.50 (1H, bs, Ar-NH), 5.31 (1H, bd, J 8, BocNHCH), 4.49 (1H, q, J7, BocNHCH), 2.64 ( $2 \mathrm{H}, \mathrm{t}, \mathrm{J} 7, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~S}$ ), $2.24\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHHCH}_{2} \mathrm{~S}\right), 2.14\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 2.05$ (1H, m, CHHCH ${ }_{2}$ S), 1.49 ( $\left.9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; бc ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 172.4 ( $C=O N H N H$ ), 156.1 ( $C=O N H$ ), 149.2 (ipso-Ar-C), 129.9 (Ar-C), 119.2 (Ar-C), 115.8 (Ar-C), 107.7 (Ar-C), $81.2\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 52.0$ (BocNHCH), 30.6 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 30.2\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 15.4\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right) ; m / z$ ( $\mathrm{ES}^{+}$) 407 ( $\mathrm{MNa}^{+}$); HRMS ( $\mathrm{ES}^{+}$) Found $\mathrm{MNa}^{+}, 407.0738\left(\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{~N}_{4} \mathrm{O}_{5} \mathrm{SNa}\right.$ requires 407.1360 ).

## tert-Butyl (S)-(1-(2-(2-chlorophenyl)hydrazineyl)-1-oxo-3-phenylpropan-2-

## yl)carbamate 67AI



Following the general procedure outlined, $N$-Boc-L-phenylalanine ( $0.30 \mathrm{~g}, 1.13$ mmol ) and 2-chlorophenylhydrazine hydrochloride ( $0.22 \mathrm{~g}, 1.24 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a light brown gummy solid ( $0.41 \mathrm{~g}, 93 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.58$ (DCM/EtOH/NH3 200:6:1); m.p. 135-137 ${ }^{\circ} \mathrm{C}$; $v_{\max } 3272$ (N-H), 2978 (C-H), 1668 (C=O), 1596, 1492, 1366, 1249, 1161, 743, $698 \mathrm{~cm}^{-1}$; бн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 7.82 (1H, bs, CONHNH), $7.33-7.29$ (2H, m, Ar-H), $7.25-7.21$ (3H, m, Ar-H), 7.04 (1H, t, J 7, Ar-H), 6.82 (1H, td, J 7, 1, Ar-H), 6.43 (1H, d, J 7, Ar-H), 6.33 (1H, bd, J 3, ArNH), 5.08 (1H, bd, J 8, BocNHCH), 4.48 (1H, q, J 7, BocNHCH), 3.12 (2H, d, J7, $\left.\mathrm{Ar}-\mathrm{CH}_{2}\right), 1.44\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ;$ бc (100 MHz, CDCl 3 ) 171.3 ( $\left.\mathrm{C=ONHNH}\right), 143.3$ (ipso-Ar-C), 136.1 (ipso-Ar-C), 131.6 (ipso-Ar-C), 129.4 (Ar-C), 129.3 (Ar-C), 128.9 (Ar-C), 127.6 (Ar-C), 127.2 (Ar-C), 121.4 (Ar-C), 113.6 (Ar-C), 28.3 $\left.\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; m / z\left(\mathrm{ES}^{+}\right) 412\left(\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right), 414\left({ }^{[37} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right), 801\left(\left[{ }^{35,35} \mathrm{Cl}\right] 2 \mathrm{M}+\mathrm{Na}^{+}\right)$,

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$\left.803\left({ }^{35,37} \mathrm{Cl}\right] 2 \mathrm{M}+\mathrm{Na}^{+}\right)$, 805 ( $\left.\left[{ }^{37,37} \mathrm{Cl}\right] 2 \mathrm{M}+\mathrm{Na}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}$, $412.1416\left(\mathrm{C}_{20} \mathrm{H}_{24}{ }^{35} \mathrm{CIN}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 412.1398$)$.

## tert-Butyl

(S)-(1-oxo-3-phenyl-1-(2-(p-tolyl)hydrazineyl)propan-2-yl) carbamate 67Ef


Following the general procedure outlined, $N$-Boc-L-phenylalanine ( $0.30 \mathrm{~g}, 1.13$ mmol ) and 4-methylphenylhydrazine hydrochloride ( $0.19 \mathrm{~g}, 1.24 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a light brown solid ( $0.26 \mathrm{~g}, 61 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.43$ (DCM/EtOH/NH3 200:6:1); m.p. 116-118 ${ }^{\circ} \mathrm{C}$; $v_{\max } 3318$ (N-H), 2962 (C-H), 1679, 1662, 1651 (C=O), 1526, 1485, 1303, 1249, 1172, 746, $647 \mathrm{~cm}^{-1}$; $\delta$ н ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.09 (1H, bs, CONHNH), $7.28-7.26$ (3H, t, J 6, Ar-H), $7.18-7.16$ (2H, d, J 6, Ar-H), $6.94-$ 6.92 (2H, d, J 7, Ar-H), $6.49-6.47$ (2H, d, J 8, Ar-H), 5.94 (1H, bs, Ar-NH), 5.21 (1H, bd, J 8, BocNHCH), 4.47 (1H, m, BocNHCH), 3.06 (2H, m, Ar-CH2), 2.23 (3H, s, Ar-CH3), $1.41\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; ~ \delta c\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 171.6(\mathrm{C}=\mathrm{ONHNH})$, 155.6 (C=ONH), 145.1 (ipso-Ar-C), 136.3 (ipso-Ar-C), 129.6 (Ar-C), 129.4 (Ar-C), 128.8 (Ar-C), 127.07 (Ar-C), 113.7 ( $\mathrm{Ar}-\mathrm{C}$ ), 80.6 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 54.5$ ( BocNHCH ), $38.6\left(\mathrm{Ar}^{\left.-\mathrm{CH}_{2}\right)}\right.$, $28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 20.6\left(\mathrm{Ar}^{2} \mathrm{CH}_{3}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 392\left(\mathrm{MNa}^{+}\right), 761$ $\left(2 \mathrm{M}+\mathrm{Na}^{+}\right)$; HRMS $\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}, 392.1966\left(\mathrm{C}_{21} \mathrm{H}_{27} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 392.1945).

## tert-Butyl (S)-(1-(2-(4-cyanophenyl)hydrazineyl)-1-oxo-3-phenylpropan-2yl)carbamate 67Db



Following the general procedure outlined, $N$-Boc-L-phenylalanine ( $0.40 \mathrm{~g}, 1.51$ mmol ) and 4-cyanophenylhydrazine hydrochloride ( $0.28 \mathrm{~g}, 1.66 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was

## Chapter 6: Experimental section

isolated as a light brown solid ( $0.39 \mathrm{~g}, 68 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.48$ (DCM/EtOH/NH3 200:6:1); m.p. $138-140{ }^{\circ} \mathrm{C}$; $v_{\max } 3345,3313,3271$ (N-H), 2986 (C-H), 2117 (C三N), 1691, 1655 (C=O), 1529, 1485, 1270, 1165, 1047, 834, $704 \mathrm{~cm}^{-1}$; $\delta \mathrm{H}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ 7.83 (1H, bs, CONHNH), 7.42 - 7.39 (2H, d, J 8, Ar-H), 7.35 - 7.31 (3H, m, ArH), $7.23-7.20(2 \mathrm{H}, \mathrm{dd}, \mathrm{J} 6,1, \mathrm{Ar}-H), 6.56-6.53(2 \mathrm{H}, \mathrm{d}, \mathrm{J}$ 8, Ar-H), $6.18(1 \mathrm{H}$, bd, J 2, Ar-NH), 5.04 (1H, bd, J 7, BocNHCH), 4.45 (1H, q, J 7, BocNHCH), 3.11 $\left(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{CH}_{2}\right), 1.45\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 171.5$ (C=ONHNH), 151.1 (ipso-Ar-C), 133.6 (Ar-C), 129.4 (Ar-C), 128.9 (Ar-C), 127.4 (Ar-C), 113.0 (Ar-C), $28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 403\left(\mathrm{MNa}^{+}\right) ;$HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}, 403.1746\left(\mathrm{C}_{21} \mathrm{H}_{24} \mathrm{~N}_{4} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 403.1741).

## tert-Butyl

(S)-(1-(2-(4-(methylsulfonyl)phenyl)hydrazineyl)-1-oxo-3-phenylpropan-2-yl)carbamate 67Dn


Following the general procedure outlined, $N$-Boc-L-phenylalanine ( $0.40 \mathrm{~g}, 1.51$ mmol ) and 4-(methylsulphonyl)phenylhydrazine hydrochloride ( $0.37 \mathrm{~g}, 1.66 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a white solid ( $0.24 \mathrm{~g}, 36 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.41$ (DCM/EtOH/NH3 200:6:1); m.p. 124-126 ${ }^{\circ} \mathrm{C}$; $v_{\text {max }} 3349,3273$ (N-H), 1713, 1668 (C=O), 1597, 1518, 1291, 1144, $766 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 7.88 (1zH, bs, CONHNH), $7.72-7.69(2 \mathrm{H}, \mathrm{d}, \mathrm{J}$ 8, Ar-H), 7.37 (3H, m, Ar-H), 7.27 (2H, m, Ar-H), 6.67 - 6.65 (2H, d, J 8, Ar-H), 5.05 (1H, bd, J 8, BocNHCH), 4.48 (1H, q, J 7, BocNHCH), 3.14 (2H, d, J 7, Ar$\mathrm{CH}_{2}$ ), $3.02\left(3 \mathrm{H}, \mathrm{s}, \mathrm{SO}_{2} \mathrm{CH}_{3}\right), 1.48\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d}_{6}\right)$ 172.1 (C=ONHNH), 155.8 ( $\mathrm{C}=\mathrm{ONH}$ ), 153.7 (ipso-Ar-C), 138.2 (ipso-Ar-C), 129.8 ( $\mathrm{Ar}-\mathrm{C}$ ), 128.8 ( $\mathrm{Ar}-\mathrm{C}$ ), 128.7 ( $\mathrm{Ar}-\mathrm{C}$ ), $126.8(\mathrm{Ar}-\mathrm{C}), 111.6$ ( $\mathrm{Ar}-\mathrm{C}), 78.7\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, 54.9 ( BocNHCH ), $44.8\left(\mathrm{SO}_{2} \mathrm{CH}_{3}\right), 37.6\left(\mathrm{Ar}^{-} \mathrm{CH}_{2}\right), 28.6\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 456$ $\left(\mathrm{MNa}^{+}\right)$; HRMS $\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}, 456.0877\left(\mathrm{C}_{21} \mathrm{H}_{27} \mathrm{~N}_{3} \mathrm{O}_{5} \mathrm{SNa}\right.$ requires 456.1564).

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## tert-Butyl (S)-(1-(2-(2-bromophenyl)hydrazineyl)-1-oxo-3-phenylpropan-2yl)carbamate 67Cj



Following the general procedure outlined, $N$-Boc-L-phenylalanine ( $0.40 \mathrm{~g}, 1.51$ mmol ) and 2-bromophenylhydrazine hydrochloride ( $0.37 \mathrm{~g}, 1.66 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a brown solid ( $0.47 \mathrm{~g}, 72 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.38$ (DCM/EtOH/NH3 200:6:1); m.p. 131-133 ${ }^{\circ} \mathrm{C}$; $v_{\max } 3325,3291$ (N-H), 1679, 1661 (C=O), 1523, 1484, 1291, 1166, 1022, $745 \mathrm{~cm}^{-1}$; $\delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.62$ (1H, bs, CONHNH), 7.36 (1H, d, J 7, Ar-H), 7.27 - 7.24 (2H, m, Ar-H), 7.20 - 7.17 (2H, m, Ar-H), 7.01 (1H, t, J 7, ArH), 6.69 (1H, t, J 7, Ar-H), 6.31 (2H, m, Ar-H), 5.47 (1H, bd, J 8, BocNHCH), 4.62 (1H, m, BocNHCH), 3.07 (2H, d, J 6, Ar-CH2), 1.37 (9H, s, ( $\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}$ ); סc (100 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 171.7 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.8 ( $\mathrm{C}=\mathrm{ONH}$ ), 144.3 (ipso-Ar-C), 136.3 (ipso-Ar-C), 129.5 (Ar-C), 128.8 ( $\mathrm{Ar}-\mathrm{C}$ ), 128.3 ( $\mathrm{Ar}-\mathrm{C}$ ), 127.0 ( $\mathrm{Ar}-\mathrm{C}$ ), 121.7 ( $\mathrm{Ar}-\mathrm{C}$ ), 113.8 (Ar-C), $80.6\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 54.3(\mathrm{BocNHCH}), 38.5\left(\mathrm{Ar}-\mathrm{CH}_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; \mathrm{m} / \mathrm{z}$ $\left(\mathrm{ES}^{+}\right) 456\left(\mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}, 456.0911\left(\mathrm{C}_{20} \mathrm{H}_{24}{ }^{79} \mathrm{BrN}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 456.0893).

## tert-Butyl (S)-(1-(2-(3-bromophenyl)hydrazineyl)-1-oxo-3-phenylpropan-2yl)carbamate 67 Cp



Following the general procedure outlined, $N$-Boc-L-phenylalanine ( $0.33 \mathrm{~g}, 1.23$ mmol ) and 3-bromophenylhydrazine hydrochloride ( $0.25 \mathrm{~g}, 1.35 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a pale orange solid ( $0.22 \mathrm{~g}, 41$ \%); $\mathrm{Rf}_{\mathrm{f}} 0.94$ (EtOAc); m.p. 129-131 ${ }^{\circ}$ C; $V_{\max }$ 3334, 3247 (N-H), 2987 (C-H), 1693, 1673 (C=O), 1525, 1487, 1320, 1248, 1165, 859, $678 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.07 ( $1 \mathrm{H}, \mathrm{bs}, \mathrm{CONHNH}$ ), 7.29 (5H, m, Ar-H), 7.19 (2H, dd, J 7, 2, Ar-H), 6.97 (2H, d, J 4, Ar-H), 6.81 (1H, s, Ar-

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H), 6.49 (1H, bs, Ar-NH), 5.15 (1H, bd, J 8, BocNHCH), 4.45 (1H, q, J 7, BocNHCH), $3.07\left(2 \mathrm{H}, \mathrm{t}, \mathrm{J} 7, \mathrm{Ar}-\mathrm{CH}_{2}\right), 1.43\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ 172.2 (C=ONHNH), 149.4 (ipso-Ar-C), 136.5 (ipso-Ar-C), 130.9 (Ar-C), 129.7 (ArC), 129.4 ( $\mathrm{Ar}-\mathrm{C}$ ), 127.7 ( $\mathrm{Ar}-\mathrm{C}$ ), 124.5 ( $\mathrm{Ar}-\mathrm{C}$ ), 123.5 ( $\mathrm{Ar}-\mathrm{C}$ ), 116.8 ( $\mathrm{Ar}-\mathrm{C}$ ), 112.6 (Ar-C), $81.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 54.9(\mathrm{BocNHCH}), 38.3\left(\mathrm{Ar}-\mathrm{CH}_{2}\right), 28.8\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; ~ m / z$ (ES ${ }^{+}$) 456 ( $\left.{ }^{79}{ }^{\mathrm{Br}}\right] \mathrm{MNa}^{+}$), 458 ( $\left.{ }^{81} \mathrm{Br}\right] \mathrm{MNa}^{+}$); HRMS (ES ${ }^{+}$) Found $\left.{ }^{79} \mathrm{Br}\right] \mathrm{MNa}^{+}$, $456.0904\left(\mathrm{C}_{20} \mathrm{H}_{24}{ }^{79} \mathrm{BrN}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 456.0893$)$.

## tert-Butyl (S)-(1-(2-(3,5-bis(trifluoromethyl)phenyl)hydrazineyl)-1-oxo-3-phenylpropan-2-yl)carbamate 67Bx



Following the general procedure outlined, $N$-Boc-L-phenylalanine ( $0.40 \mathrm{~g}, 1.51$ mmol ) and (3,5-bis(trifluoromethyl)phenyl)hydrazine hydrochloride ( $0.47 \mathrm{~g}, 1.66$ mmol ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a brown gummy solid ( $0.39 \mathrm{~g}, 52 \%$ ); Rf 0.44 (DCM/EtOH/NH3 200:6:1); $V_{\max }$ 3331, 3275 (N-H), 2964, 2936 (C-H), 1685, 1666 (C=O), 1524, 1468, 1379, 1276, 1121, $681 \mathrm{~cm}^{-1}$; $\delta \mathrm{H}\left(300 \mathrm{MHz}, \mathrm{iCDCl}_{3}\right) 8.09(1 \mathrm{H}$, bs, CONHNH), 7.34 - 7.29 (4H, m, Ar-H), 7.21 - 7.18 (2H, J7, 2, Ar-H), 7.10 (2H, s, Ar-H), 6.38 (1H, bd, J 3, Ar-NH), 5.06 (1H, bd, J 7, BocNHCH), 4.48 (1H, q, J 7, BocNHCH), 3.12 (2H, dd, J6, 2, Ar-CH2), 1.43 (9H, s, ( $\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}$ ); סc ( 100 MHz , $\mathrm{CDCl}_{3}$ ) 172.2 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 148.9 (ipso-Ar-C), 135.8 (ipso-Ar-C), 132.7 (ipso-ArC), 132.3 (ipso-Ar-C), 129.1 (Ar-C), 128.9 (Ar-C), 127.4 (Ar-C), 112.9 (Ar-C), 81.4 $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 28.2\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; \delta \mathrm{F}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)-63.1\left(\mathrm{CF}_{3}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 514$ $\left(\mathrm{MNa}^{+}\right)$; HRMS $\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}, 514.1542\left(\mathrm{C}_{22} \mathrm{H}_{23} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~F}_{6} \mathrm{Na}\right.$ requires 514.1536).

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## tert-Butyl yl)carbamate 67Dt

(S)-(1-oxo-3-phenyl-1-(2-(o-tolyl)hydrazineyl)propan-2-


Following the general procedure outlined, $N$-Boc-L-phenylalanine ( $0.20 \mathrm{~g}, 0.75$ mmol ) and 2-methylphenylhydrazine hydrochloride ( $0.10 \mathrm{~g}, 0.83 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as an orange gummy solid ( $0.22 \mathrm{~g}, 78 \%$ ); $R_{f} 0.95$ (EtOAc); $v_{\max } 3330$, 3273 (N-H), 2977 (C-H), 1683, 1652 (C=O), 1522, 1496, 1294, 1165, $745 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.20 (1H, bs, CONHNH), 7.27 (3H, t, J 6, Ar-H), 7.20 (2H, d, J6, Ar-H), 7.01 (1H, d, J 7, Ar-H), $6.96(1 \mathrm{H}, \mathrm{t}, J 7, \mathrm{Ar}-H), 6.78$ (1H, t, J 7, ArH), 6.36 (1H, d, J 7, Ar-H), 5.27 (1H, bd, J 8, BocNHCH), 4.51 (1H, q, J 7, BocNHCH), 3.08 (2H, t, J7, Ar-CH2), 2.16 (3H, s, Ar-CH3), 1.41 (9H, s, ( $\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}$ ); ठс (100 MHz, CDCl3) 171.5 ( $\mathrm{C=ONHNH}$ ), 155.8 (C=ONH), 145.2 (ipso-Ar-C), 136.3 (ipso-Ar-C), 130.5 (Ar-C), 129.5 (Ar-C), 128.9 (Ar-C), 127.2 (Ar-C), 126.9 (Ar-C), 123.2 (Ar-C), 121.0 ( $\mathrm{Ar}-\mathrm{C}$ ), 112.2 ( $\mathrm{Ar}-\mathrm{C}$ ), $80.7\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 54.6$ (BocNHCH), $38.3\left(\mathrm{Ar}_{-} \mathrm{CH}_{2}\right), 28.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 17.1$ (Ar-CH3); m/z (ES $\left.{ }^{+}\right) 392$ $\left(\mathrm{MNa}^{+}\right)$; $\mathrm{HRMS}\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}, 392.1963\left(\mathrm{C}_{21} \mathrm{H}_{27} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 392.1945).

## tert-Butyl

(S)-(1-oxo-3-phenyl-1-(2-(m-tolyl)hydrazineyl)propan-2yl)carbamate 67Dz


Following the general procedure outlined, $N$-Boc-L-phenylalanine ( $0.33 \mathrm{~g}, 1.23$ mmol ) and 3-methylphenylhydrazine hydrochloride ( $0.16 \mathrm{~g}, 1.35 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as an orange oil ( $0.41 \mathrm{~g}, 91 \%$ ); $\mathrm{R}_{\mathrm{f}} 0.86$ (EtOAc); $v_{\max } 3327,3285(\mathrm{~N}-\mathrm{H})$, 1691, 1653 (C=O), 1518, 1495, 1250, 1164, 771, $699 \mathrm{~cm}^{-1}$; ठн ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.14 (1H, bs, CONHNH), 7.27 (3H, t, J7, Ar-H), 7.18 (2H, d, J7, Ar-H), 7.01 (1H, t, J 7, Ar-H), 6.67 (1H, d, J 7, Ar-H), 6.46 (1H, s, Ar-H), 6.35 (1H, d, J 7, Ar-H),

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5.23 (1H, bd, J 8, BocNHCH), 4.48 (1H, q, J 7, BocNHCH), 3.05 (2H, t, J 8, Ar$\mathrm{CH}_{2}$ ), $2.23\left(3 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{CH}_{3}\right), 1.41\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 171.6$ (C=ONHNH), 155.7 ( $\mathrm{C}=\mathrm{ONH}$ ), 147.5 (ipso-Ar-C), 139.0 (ipso-Ar-C), 136.4 (Ar-C), 129.5 (Ar-C), 129.1 (Ar-C), 128.9 (Ar-C), 127.1 (Ar-C), 122.3 (Ar-C), 114.4 (ArC), 110.7 ( $\mathrm{Ar}-\mathrm{C}$ ), $80.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 54.7$ ( BocNHCH ), $38.2\left(\mathrm{Ar}_{2} \mathrm{CH}_{2}\right), 28.4$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 21.6\left(\mathrm{Ar}-\mathrm{CH}_{3}\right) ; m / z\left(\mathrm{ES}^{+}\right) 392\left(\mathrm{MNa}^{+}\right) ;$HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}$, $392.1975\left(\mathrm{C}_{21} \mathrm{H}_{27} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 392.1945).
tert-Butyl (S)-(1-(2-(4-bromophenyl)hydrazineyl)-1-oxo-3-phenylpropan-2yl)carbamate 67 Cv


Following the general procedure outlined, $N$-Boc-L-phenylalanine ( $0.40 \mathrm{~g}, 1.51$ mmol ) and 4-bromophenylhydrazine hydrochloride ( $0.37 \mathrm{~g}, 1.66 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as an orange solid ( $0.37 \mathrm{~g}, 56 \%$ ); $\mathrm{R}_{\mathrm{f}} 0.37$ (DCM/EtOH/NH ${ }_{3}$ 200:6:1); m.p. 127-129 ${ }^{\circ} \mathrm{C}$; $v_{\max } 3308(\mathrm{~N}-\mathrm{H}), 2984(\mathrm{C}-\mathrm{H}), 1687,1654$ (C=O), 1521, 1488, 1366, 1252, 1167, 1003, $820 \mathrm{~cm}^{-1}$; $\delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.22$ (1H, bs, CONHNH), 7.32 - 7.28 (3H, m, Ar-H), 7.23 - 7.18 (4H, m, Ar-H), 6.41 - 6.38 (2H, d, J 8, Ar-H), 5.27 (1H, bd, J 8, BocNHCH), 4.62 (1H, q, J 7, BocNHCH), 3.08 (2H, d, J 7, Ar$\mathrm{CH}_{2}$ ), $1.44\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 171.7$ ( $\left.\mathrm{C}=\mathrm{ONHNH}\right), 155.6$ (C=ONH), 146.6 (ipso-Ar-C), 136.1 (ipso-Ar-C), 131.8 (Ar-C), 129.4 (Ar-C), 127.1 (Ar-C), 115.1 (Ar-C), $113.0(\mathrm{Ar}-\mathrm{C}), 80.8\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 54.4$ (BocNHCH), 38.2 (Ar$\mathrm{CH}_{2}$ ), $\left.\left.28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 456\left({ }^{79} \mathrm{Br}\right] \mathrm{MNa}^{+}\right), 458\left({ }^{81} \mathrm{Br}\right] \mathrm{MNa}^{+}\right) ;$HRMS $\left(\mathrm{ES}^{+}\right)$Found $\left[^{79} \mathrm{Br}\right] \mathrm{MNa}^{+}, 456.0897\left(\mathrm{C}_{20} \mathrm{H}_{24}{ }^{79} \mathrm{BrN}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 456.0893).

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## tert-Butyl (S)-(1-oxo-3-phenyl-1-(2-(2-(trifluoromethyl)phenyl)hydrazineyl) propan-2-yl)carbamate 67Bd



Following the general procedure outlined, $N$-Boc-L-phenylalanine ( $0.40 \mathrm{~g}, 1.51$ mmol ) and 2-trifluoromethylphenylhydrazine hydrochloride ( $0.35 \mathrm{~g}, 1.66 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow gummy solid ( $0.42 \mathrm{~g}, 66 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.47$ (DCM/EtOH/NH 3 200:6:1); $V_{\max } 3331,3283(\mathrm{~N}-\mathrm{H}), 1665,1618(\mathrm{C}=\mathrm{O}), 1524,1497,1325,1282,1162,1107$, $751 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.66 ( $1 \mathrm{H}, \mathrm{bs}, \mathrm{CONHNH}$ ), 7.37 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}$ ), 7.20 - 7.16 (3H, m, Ar-H), 7.13 - 7.02 (3H, m, Ar-H), 6.76 (1H, m, Ar-H), 6.36 (1H, m, Ar-H), 5.50 (1H, bd, J 8, BocNHCH), 4.60 (1H, m, BocNHCH), 3.00 (2H, m, Ar$\mathrm{CH}_{2}$ ), 1.27 ( $\left.9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 171.9$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.8 ( $\mathrm{C}=\mathrm{ONH}$ ), 145.1 (ipso-Ar-C), 136.2 (ipso-Ar-C), 132.9 (Ar-CF3), 129.4 (Ar-C), 128.7 (Ar-C), 127.0 (Ar-C), 126.4 (1C, q, J 5, ipso-Ar-CCF3), 122.7 (Ar-C), 119.9 (Ar-C), 117.7 (Ar-C), 113.5 (Ar-C), 111.9 (Ar-C), $80.7\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 54.3$ ( BocNHCH ), $38.5\left(\mathrm{Ar}-\mathrm{CH}_{2}\right), 28.2\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta_{\mathrm{F}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)-61.9\left(\mathrm{CF}_{3}\right)$; $m / z\left(\mathrm{ES}^{+}\right) 424\left(\mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\mathrm{MNa}^{+}, 446.1009\left(\mathrm{C}_{21} \mathrm{H}_{24} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~F}_{3} \mathrm{Na}\right.$ requires 446.1662).

## tert-Butyl (S)-(1-oxo-3-phenyl-1-(2-(3-(trifluoromethyl)phenyl)hydrazineyl) propan-2-yl)carbamate 67Bj



Following the general procedure outlined, $N$-Boc-L-phenylalanine ( $0.40 \mathrm{~g}, 1.51$ mmol ) and 3-trifluoromethylphenylhydrazine hydrochloride ( $0.35 \mathrm{~g}, 1.66 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow solid ( $0.48 \mathrm{~g}, 76 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.46$ (DCM/EtOH/NH3 200:6:1); m.p. $141-143{ }^{\circ} \mathrm{C}$; $v_{\max } 3338$ (N-H), 1676 (C=O), 1519, 1334, 1249, 1161, $696 \mathrm{~cm}^{-1}$; бн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.13 (1H, bs, CONHNH), 7.23 - 7.17 (3H, m, Ar-H), $7.15-$ 7.10 (3H, m, Ar-H), 7.03 (1H, d, J 7, Ar-H), 6.85 (1H, s, Ar-H), 6.63 (1H, d, J 7,

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Ar-H), 5.12 (1H, bs, BocNHCH), 4.45 (1H, q, J 7, BocNHCH), 3.02 (2H, dd, J 7, 3, $\left.\mathrm{Ar}-\mathrm{CH}_{2}\right), 1.35\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; \delta c\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 171.9$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.7 (C=ONH), 147.9 (ipso-Ar-C), 135.9 (ipso-Ar-C), 131.7 (1C, q, J 317, ipso-
 127.0 ( $\mathrm{Ar}-\mathrm{C}$ ), 125.8 ( $\mathrm{Ar}-\mathrm{C}$ ), 122.2 (1C, ipso- $\mathrm{Ar}-\mathrm{CCF}_{3}$ ), 117.6 ( $\mathrm{Ar}-\mathrm{C}$ ), 116.4 ( $\mathrm{Ar}-\mathrm{C}$ ), 110.0 ( $\mathrm{Ar}-\mathrm{C}), 80.9\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 54.5(\mathrm{BocNHCH}), 37.9\left(\mathrm{Ar}-\mathrm{CH}_{2}\right), 28.2\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta_{\mathrm{F}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)-62.7\left(\mathrm{CF}_{3}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 424\left(\mathrm{MNa}^{+}\right) ;$HRMS (ES$\left.{ }^{+}\right)$Found $\mathrm{MNa}^{+}, 446.1009\left(\mathrm{C}_{21} \mathrm{H}_{24} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~F}_{3} \mathrm{Na}\right.$ requires 446.1662).

## tert-Butyl (S)-(1-(2-(3-nitrophenyl)hydrazineyl)-1-oxo-3-phenylpropan-2-yl) carbamate 67Dh



Following the general procedure outlined, $N$-Boc-L-phenylalanine ( $0.40 \mathrm{~g}, 1.51$ mmol ) and 3-nitrophenylhydrazine hydrochloride ( $0.31 \mathrm{~g}, 1.66 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a pale yellow solid ( $0.43 \mathrm{~g}, 70 \%$ ); $\mathrm{R}_{\mathrm{f}} 0.32$ (DCM/EtOH/ $\mathrm{NH}_{3}$ 200:6:1); m.p. 134-136 ${ }^{\circ} \mathrm{C}$; $V_{\max } 3359,3336,3239$ (N-H), 1694, 1664 (C=O), 1525, 1351, 1253, 1165, 1047, $733 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.13 (1H, bs, CONHNH), 7.73 (1H, dd, J 8, 1, Ar-H), 7.54 (1H, t, J 2, Ar-H), $7.35-7.30$ (3H, m, Ar-H), 7.26 7.21 (2H, m, Ar-H), 6.89 (1H, d, J 8, Ar-H), 5.14 (1H, bd, J 8, BocNHCH), 4.53 (1H, q, J 7, BocNHCH), 3.02 (2H, d, J 7, 3, Ar-CH2), 1.46 (9H, s, ( $\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}$ ); סc (100 MHz, CDCl3) 171.9 (C=ONHNH), 149.1 (ipso-Ar-C), 148.8 (ipso-Ar-C), 135.9 (ipso-Ar-C), 129.8 (Ar-C), 129.3 (Ar-C), 128.9 (Ar-C), 127.3 (Ar-C), 119.1 (Ar-C), 115.8 (Ar-C), 107.9 (Ar-C), 81.2 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 51.1$ (BocNHCH), 37.7 (Ar$\left.\mathrm{CH}_{2}\right), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 423\left(\mathrm{MNa}^{+}\right) ; \mathrm{HRMS}\left(\mathrm{ES}^{+}\right)$Found $\mathrm{MNa}^{+}$, $423.1056\left(\mathrm{C}_{20} \mathrm{H}_{24} \mathrm{~N}_{4} \mathrm{O}_{5} \mathrm{Na}\right.$ requires 423.1639$)$.

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## tert-Butyl <br> (S)-4-((((9H-fluoren-9-yl)methoxy)carbonyl)amino)-5-(2-(3-chlorophenyl)hydrazinyl)-5-oxopentanoate 68n



Following the general procedure outlined, N-Fmoc-O-(tert-butyl)-L-glutamic acid $(0.30 \mathrm{~g}, 0.71 \mathrm{mmol})$ and 3-chlorophenylhydrazine hydrochloride ( $0.14 \mathrm{~g}, 0.78$ mmol ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow solid ( $0.27 \mathrm{~g}, 70 \%$ ); Rf 0.75 ( n -hex/EtOAc 4:1); m.p. 139-141 ${ }^{\circ} \mathrm{C}$; $v \max 3275$ (N-H), 1686, 1659 (C=O), 1541, 1477, 1257, 1154, 1048, 735 cm-1; бн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.31 (1H, bs, CONHNH), 7.76 - 7.74 (2H, d, J 7, Ar-H), 7.58 - 7.55 (2H, d, J 7, Ar-H), 7.41 - 7.36 (2H, t, J 7, Ar-H), 7.32 7.28 (2H, dt, J7, 1, Ar-H), 7.09 (1H, t, J7, Ar-H), 6.85 (1H, d, J7, Ar-H), 6.79 (1H, s, Ar-H), 6.68 (1H, d, J 8, Ar-H), 5.82 (1H, bd, J 7, FmocNHCH), $4.46(2 \mathrm{H}, \mathrm{m}$, $\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}$-Fluorene), $4.37\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene), $4.22(1 \mathrm{H}, \mathrm{m}$, FmocNHCH), $2.36\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2}\right)$, $2.17\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2}\right), 1.46$ $\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; ~ \delta c \quad\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 172.9\left(\mathrm{C}=\mathrm{OO}\left(\mathrm{CH}_{3}\right)_{3}\right), 171.7$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.5 ( $\mathrm{C}=\mathrm{ONH}$ ), 148.9 (ipso-Ar-C), 143.6 (ipso-Ar-C), 141.3 (ipso-Ar-C), 135.0 (ipso-Ar-C), 130.3 (Ar-C), 127.3 (Ar-C), 127.1 (Ar-C), 124.9 (Ar-C), 121.2 (Ar-C), 120.0 (Ar-C), 113.6 (Ar-C), 111.7 (Ar-C), 81.4 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 67.2$ $\left(\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene), $53.1 \quad(\mathrm{FmocNHCH}), \quad 47.1$ (C-Fluorene), 31.6 $\left(\mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2}\right), 28.1\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 27.3\left(\mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 572$ ( $\left[{ }^{35} \mathrm{CI}\right] \mathrm{MNa}^{+}$), 574 ( $\left[{ }^{37} \mathrm{C}\right] \mathrm{MNa}^{+}$); HRMS (ES ${ }^{+}$) Found $\left[{ }^{35} \mathrm{CI}\right] \mathrm{MNa}^{+}$, 572.1934 ( $\mathrm{C}_{30} \mathrm{H}_{32}{ }^{35} \mathrm{ClN}_{3} \mathrm{O}_{5} \mathrm{Na}$ requires 572.1923 ).

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## tert-Butyl <br> (S)-(3-(1-benzyl-1 H-imidazol-5-yl)-1-(2-(3-chlorophenyl)

hydrazineyl)-1-oxopropan-2-yl)carbamate 67y


Following the general procedure outlined, N -Fmoc- N -(im)-benzyl-L-histidine $(0.40 \mathrm{~g}, 1.16 \mathrm{mmol})$ and 3-chlorophenylhydrazine hydrochloride ( $0.23 \mathrm{~g}, 1.27$ mmol ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a reddish brown oil ( $0.40 \mathrm{~g}, 74 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.35$ (DCM/EtOH/ $\mathrm{NH}_{3}$ 200:6:1); vmax 3269 (N-H), 2979 (C-H), 1678 (C=O), 15497, 1479, 1366, 1249, 1160, $729 \mathrm{~cm}-1$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 9.37 (1H, bs, CONHNH), 7.97 (1H, bs, ArNH), 7.38 - 7.32 (5H, m, Ar-H), 7.18 - 7.15 (2H, m, Ar-H), 7.06 (1H, t, J 7, Ar-H), 6.79 (1H, m, Ar-H), 6.68 (1H, t, J 2, Ar-H), 6.65 (1H, dd, J 8, 1, Ar-H), 6.32 (1H, bd, J 7, BocNHCH), 5.09 (2H, s, Ar-CH2lm), 4.65 (1H, m, BocNHCH), 3.11 (2H, $\left.\mathrm{t}, \mathrm{J} 6, \mathrm{CHCH}_{2} \mathrm{Im}\right), 1.42\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ;$ бc ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 171.6 (C=ONHNH), 149.4 (ipso-Ar-C), 134.8 (ipso-Ar-C), 130.1 (Ar-C), 129.3 (Ar-C), 128.9 (Ar-C), 127.9 (Ar-C), 120.9 (Ar-C), 113.2 (Ar-C), 111.8 (Ar-C), 80.1 $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 52.1\left(\mathrm{Ar}-\mathrm{CH}_{2} \mathrm{Im}\right)$, $29.5\left(\mathrm{CHCH}_{2} \mathrm{Im}\right), 28.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 470$ $\left(\left[{ }^{35} \mathrm{C}\right] \mathrm{MH}^{+}\right), 472\left(\left[{ }^{37} \mathrm{CI}\right] \mathrm{MH}^{+}\right) ;$HRMS (ES $\left.{ }^{+}\right)$Found $\left[{ }^{35} \mathrm{CI}\right] \mathrm{MH}^{+}, 470.1952$ $\left(\mathrm{C}_{24} \mathrm{H}_{29}{ }^{35} \mathrm{CIN}_{5} \mathrm{O}_{3}\right.$ requires 470.1953 ).

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(9H-fluoren-9-yl)methyl (S)-(1-(2-(3-chlorophenyl)hydrazinyl)-1-oxo-3-(tritylthio)propan-2-yl)carbamate 68p


Following the general procedure outlined, $N$-Fmoc-S-(trityl)-L-cysteine ( 0.40 g , 0.68 mmol ) and 3-chlorophenylhydrazine hydrochloride ( $0.15 \mathrm{~g}, 0.82 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a light brown solid ( $0.35 \mathrm{~g}, 72 \%$ ); $\mathrm{R}_{\mathrm{f}} 0.24$ ( n -hex/EtOAc 4:1); m.p. 90 - $93^{\circ} \mathrm{C}$; vmax 3271 (N-H), 1677 (C=O), 1597, 1486, 1445, 1219, 1033, 738, 699 cm-1; ठн (300 MHz, CDCl 3 ) $7.76-7.73(2 \mathrm{H}, \mathrm{d}, J 7, \mathrm{Ar}-H), 7.56-7.53(2 \mathrm{H}, \mathrm{d}, ~ J$ 7, Ar-H), 7.43 - 7.39 (4H, m, Ar-H), 7.31 - 7.18 (15H, m, Ar-H), 7.06 (1H, t, J 7, Ar-H), $6.84(1 \mathrm{H}, \mathrm{dd}, J 7,2, \operatorname{Ar}-H), 6.74(1 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{H}), 6.62(1 \mathrm{H}, \mathrm{d}, \mathrm{J} 7, \operatorname{Ar}-\mathrm{H}), 5.92$ (1H, bs, Ar-NH), $4.92(1 \mathrm{H}, \mathrm{bd}, \mathrm{J} 7, \mathrm{FmocNHCH}), 4.46\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}-\right.$ Fluorene), 4.17 (1H, m, $\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}$-Fluorene), 3.76 (1H, m, FmocNHCH), 2.68 (2H, m, CHCH2S-(trityl)); $\delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 148.7$ (ipso-Ar-C), 144.2 (ipso-ArC), 143.6 (ipso-Ar-C), 143.5 (ipso-Ar-C), 141.4 (ipso-Ar-C), 135.0 (ipso-Ar-C), 130.2 ( $\mathrm{Ar}-\mathrm{C}$ ), 129.5 ( $\mathrm{Ar}-\mathrm{C}$ ), 128.2 ( $\mathrm{Ar}-\mathrm{C}$ ), 127.9 ( $\mathrm{Ar}-\mathrm{C}$ ), 127.8 ( $\mathrm{Ar}-\mathrm{C}$ ), 127.1 ( $\mathrm{Ar}-$ C), 124.8 ( $\mathrm{Ar}-\mathrm{C}$ ), 121.3 ( $\mathrm{Ar}-\mathrm{C}$ ), 120.1 ( $\mathrm{Ar}-\mathrm{C}$ ), 113.6 ( $\mathrm{Ar}-\mathrm{C}$ ), 111.9 ( $\mathrm{Ar}-\mathrm{C}$ ), 67.6 ( $\mathrm{C}-$ (trityl), $67.1 \quad\left(\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene), 47.2 (C-Fluorene); m/z (ES ${ }^{+}$) 732 $\left(\left[{ }^{35} \mathrm{C}\right] \mathrm{MNa}^{+}\right), 734$ ( $\left.{ }^{37} \mathrm{C} \mid\right] \mathrm{MNa}^{+}$); HRMS (ES ${ }^{+}$) Found $\left[{ }^{35} \mathrm{CI}\right] \mathrm{MNa}^{+}, 732.2071$ $\left(\mathrm{C}_{43} \mathrm{H}_{36} \mathrm{ClN}_{3} \mathrm{O}_{3} \mathrm{SNa}\right.$ requires 732.2058$)$.

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(9H-fluoren-9-yl)methyl (S)-(1-(2-(3-chlorophenyl)hydrazineyl)-1-oxo-5-(3-((2,2,4,6,7-pentamethyl-2,3-dihydrobenzofuran-5-yl)sulfonyl)guanidino) pentan-2-yl)carbamate 68t


Following the general procedure outlined, $N$-Fmoc- $N-((2,2,4,6,7-$ pentamethyldihydro-benzofuran-5-sulfonyl)-L-arginine ( $0.40 \mathrm{~g}, 0.62 \mathrm{mmol}$ ) and 3chlorophenylhydrazine hydrochloride ( $0.12 \mathrm{~g}, 0.68 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a light brown solid ( $0.14 \mathrm{~g}, 30 \%$ ); Rf 0.32 (DCM/EtOH/NH3 200:6:1); m.p. $120-122{ }^{\circ} \mathrm{C}$; vmax 3282 (N-H), 2972, 2935 (C-H), 1684, 1658 (C=O), 1537, 1451, 1256, 1090, 758, 641 cm-1; бн ( 300 MHz , DMSO-d6) 9.89 (1H, bd, J 2, CONHNH), 8.06 (1H, bd, J 2, SO2NH), 7.91-7.89 (2H, d, J7, Ar-H), 7.75 - 7.71 (2H, t, J 6, Ar-H), 7.44 - 7.39 (2H, t, J 7, Ar-H), 7.35 - 7.30 (2H, d, J 7, Ar-H), 7.11 (1H, t, J 7, Ar-H), 6.74 - 6.70 (2H, m, Ar-H), 6.68 (1H, m, Ar-H), 6.45 (1H, bs, Ar-NH), 4.29 (2H, m, $\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}$-Fluorene), $4.26\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene), $4.06(1 \mathrm{H}, \mathrm{m}$, FmocNHCH), 3.09 (2H, m, CH2NHCNHNH), 2.95 ( $2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2}$-benzofuran), 2.50 $\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right.$-benzofuran), $2.44\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right.$-benzofuran), $2.00\left(3 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3}-\right.$ benzofuran), $1.69(2 \mathrm{H}, \mathrm{m}, \mathrm{FmocNHCHCH}), 1.49\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}_{2} \mathrm{CH}_{2}\right), 1.40(6 \mathrm{H}$, s, ( $\mathrm{CH}_{3}$ )2-benzofuran); $\delta c$ ( 100 MHz , DMSO-d6) 172.3 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 157.9 (C=NHNH) 156.5 (C=ONH), 151.3 (ipso-Ar-C), 144.4 (ipso-Ar-C), 144.2 (ipso-ArC), 141.2 (ipso-Ar-C), 131.9 (Ar-C), 130.7 (Ar-C), 128.1 (Ar-C), 127.5 (Ar-C), 127.2 (Ar-C), 125.7 (Ar-C), 124.8 (Ar-C), 120.6 (Ar-C), 86.8 (C-benzofuran), 66.2 $\left(\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene), 53.7 ( FmocNHCH ), 47.1 (C-Fluorene), $42.9\left(\mathrm{CH}_{2}\right.$ benzofuran), $40.9\left(\mathrm{CH}_{2} \mathrm{NHCNHNH}\right), 29.3\left(\mathrm{FmocNHCHCH}_{2}\right), 28.8\left(\mathrm{CH}_{3}\right)_{2}-$

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benzofuran $), \quad 25,2 \quad\left(\mathrm{CHCH}_{2} \mathrm{CH}_{2}\right), \quad 19.5 \quad\left(\mathrm{CH}_{3}\right.$-benzofuran $), \quad 18.1 \quad\left(\mathrm{CH}_{3}-\right.$ benzofuran), 12.8 ( $\mathrm{CH}_{3}$-benzofuran); $m / z$ (ES ${ }^{+}$) 773 ( $\left.{ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}$), $\left.775\left({ }^{[37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right)$, 795 ( $\left.{ }^{35} \mathrm{CI}\right] \mathrm{MNa}^{+}$), 797 ( $\left.{ }^{37} \mathrm{CI}\right] \mathrm{MNa}^{+}$); HRMS (ES ${ }^{+}$) Found $\left[{ }^{35} \mathrm{CI}\right] \mathrm{MH}^{+}, 773.2899$ $\left(\mathrm{C}_{40} \mathrm{H}_{46}{ }^{35} \mathrm{ClN}_{6} \mathrm{O}_{6} \mathrm{~S}\right.$ requires 773.2883).
(9H-fluoren-9-yl)methyl (S)-(1-(2-(3-chlorophenyl)hydrazineyl)-1,4-dioxo-4 (tritylamino)butan-2-yl)carbamate 680


Following the general procedure outlined, $N$-Fmoc- $N$-(trityl)-L-asparagine ( 0.40 g , 0.67 mmol ) and 3-chlorophenylhydrazine hydrochloride ( $0.13 \mathrm{~g}, 0.74 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a white solid ( $0.19 \mathrm{~g}, 37 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.39$ (DCM/EtOH/NH3 200:6:1); m.p. 168-171 ${ }^{\circ} \mathrm{C}$; $v \max 3304$ (N-H), 1692, 1654 (C=O), 1527, 1489, 1259, 1033, 736, 701 cm-1; бн (300 MHz, DMSO-d6) 9.92 (1H, bd, J 2, CONHNH), 8.60 (1H, bs, CONH-trityl), 8.05 (1H, bd, J 2, CONHCH), 7.92 - 7.90 (2H, d, J 7, Ar-H), 7.76 7.74 (3H, d, J 7, Ar-H), 7.45 - 7.38 (3H, m, Ar-H), 7.23 - 7.19 (15H, m, Ar-H), $7.13(1 \mathrm{H}, \mathrm{t}, \mathrm{J}$ 8, Ar-H), $6.73(1 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}), 6.69-6.63(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}), 4.43(2 \mathrm{H}$, m, $\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}$-Fluorene), $4.32-4.25\left(2 \mathrm{H}, \mathrm{m}, \mathrm{FmocNHCH}, \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}-\right.$ Fluorene), 2.83 (1H, m, FmocNHCHCHH), 2.59 (1H, d, J 4, FmocNHCHCHH); ठc (100 MHz, DMSO-d6) 171.8 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 169.0 ( $\mathrm{C}=\mathrm{NH}-$ trityl), 156.3 (C=ONH), 151.3 (ipso-Ar-C), 145.2 (ipso-Ar-C), 144.4 (ipso-Ar-C), 144.2 (ipso-Ar-C), 141.2 (ipso-Ar-C), 133.9 (ipso-Ar-C), 130.7 (Ar-C), 129.1 (Ar-C), 128.1 (ArC), 127.9 (Ar-C), 127.6 (Ar-C), 126.8 (Ar-C), 125.8 (Ar-C), 125.7 (Ar-C), 120.6 (Ar-C), 118.3 (Ar-C), 111.9 (Ar-C), 111.2 (Ar-C), 69.9 (C-(trityl), 66.4 $\left(\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene), $51.4 \quad$ (FmocNHCH), 47.2 (C-Fluorene) 39.7 (FmocNHCHCH2); m/z (ES $\left.{ }^{+}\right) 743\left(\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right), 745$ ( $\left.{ }^{37} \mathrm{Cl}\right] \mathrm{MNa}^{+}$), 1443 $\left.\left.\left(\left[{ }^{35,35} \mathrm{Cl}\right] 2 \mathrm{M}+\mathrm{H}^{+}\right), \quad 1445 \quad\left({ }^{[35,37} \mathrm{Cl}\right] 2 \mathrm{M}+\mathrm{H}^{+}\right), \quad 1447 \quad\left({ }^{37,37} \mathrm{Cl}\right] 2 \mathrm{M}+\mathrm{H}^{+}\right), \quad 1463$

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$\left.\left(\left[^{35,35} \mathrm{Cl}\right] 2 \mathrm{M}+\mathrm{Na}^{+}\right), 1465\left({ }^{35,37} \mathrm{Cl}\right] 2 \mathrm{M}+\mathrm{Na}^{+}\right), 1467\left({ }^{\left.\left[{ }^{37,37} \mathrm{Cl}\right] 2 \mathrm{M}+\mathrm{Na}^{+}\right) ; \text {HRMS }\left(\mathrm{ES}^{+}\right)}\right.$ Found $\left[{ }^{35} \mathrm{C}\right] \mathrm{MNa}^{+}, 743.2403\left(\mathrm{C}_{44} \mathrm{H}_{37}{ }^{35} \mathrm{ClN}_{4} \mathrm{O}_{4} \mathrm{Na}\right.$ requires 743.2396$)$.

## tert-Butyl <br> (S)-3-(2-((((9H-fluoren-9-yl)methoxy)carbonyl)amino)-3-(2-(3-

 chlorophenyl)hydrazineyl)-3-oxopropyl)-1H-indole-1-carboxylate 68s

Following the general procedure outlined, $N$-Fmoc- $N$-(tert-butyl)-L-tryptophan $(0.40 \mathrm{~g}, 0.76 \mathrm{mmol})$ and 3-chlorophenylhydrazine hydrochloride $(0.15 \mathrm{~g}, 0.84$ mmol ) were transformed following Flash column chromatography (DCM/EtOH/NH3 200:6:1) into the title compound which was isolated as a brown gum ( $0.14 \mathrm{~g}, 27 \%$ ); Rf 0.53 (DCM/EtOH/NH3 200:6:1); vmax 3274 (N-H), 2975 (C-H), 1693, 1662 (C=O), 1597, 1452, 1369, 1269, 1158, 757, $735 \mathrm{~cm}-1$; ठн (300 MHz, DMSO-d6) 10.10 (1H, bd, J2, CONHNH), 8.10 - 8.05 (2H, m, Ar-H), 7.89 - 7.86 (2H, d, J 7, Ar-H), 7.82 (1H, d, J 7, Ar-H), $7.66-7.63$ (3H, t, J 5, Ar-H), 7.43 - 7.35 (3H, m, Ar-H), 7.28 - 7.20 (3H, m, Ar-H), 7.10 (1H, t, J 8, Ar-H), 6.70 - 6.67 (2H, m, Ar-H), 6.56 (1H, m, Ar-H), 4.50 (1H, m, FmocNHCH), 4.22 - 4.19 $\left(3 \mathrm{H}, \mathrm{m}, \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene, $\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}$-Fluorene), 3.14 (1H, dd, J 14, 5, FmocNHCHCHH), 3.07 ( $1 \mathrm{H}, \mathrm{dd}, \mathrm{J} 14,9$, FmocNHCHCHH), 1.59 ( $9 \mathrm{H}, \mathrm{s}$, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta c(100 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d} 6) 171.8$ ( $\left.\mathrm{C}=\mathrm{ONHNH}\right)$, 156.4 ( $\left.\mathrm{C}=\mathrm{ONH}\right), 151.1$ (ipso-Ar-C), 149.5 (ipso-Ar-C), 144.3 (ipso-Ar-C), 144.1 (ipso-Ar-C), 141.1 (ipso-Ar-C), 135.2 (ipso-Ar-C), 133.9 (ipso-Ar-C), 130.9 (Ar-C), 130.5 (Ar-C), 128.1 (ArC), 127.5 (Ar-C), 125.7 (Ar-C), 124.8 (Ar-C), 124.7 (Ar-C), 122.9 (Ar-C), 120.6 (Ar-C), 119.9 (Ar-C), 118.3 (Ar-C), 116.8 (ipso-Ar-C), 115.2 (Ar-C), 111.9 (Ar-C), 111.1 ( $\mathrm{Ar}-\mathrm{C}), 84.0\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 66.3\left(\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}\right.$-Fluorene), 53.5 ( FmocNHCH ), 47.0 (C-Fluorene), $28.1\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$, 27.6 ( $\mathrm{FmocNHCHCH}_{2}$ ); m/z (ES $\left.{ }^{+}\right) 550$ (MBoc $+\mathrm{H}^{+}$).

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(9H-fluoren-9-yl)methyl hydrazinyl)-1-oxobutan-2-yl)carbamate 681


Following the general procedure outlined, N -Fmoc-O-(tert-butyl)-L-threonine ( $0.40 \mathrm{~g}, 1.01 \mathrm{mmol}$ ) and 3-chlorophenylhydrazine hydrochloride ( $0.19 \mathrm{~g}, 1.11$ mmol ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a light brown solid ( $0.35 \mathrm{~g}, 66 \%$ ); $\mathrm{R}_{\mathrm{f}} 0.35$ (n-hex/EtOAc 4:1); m.p. 88-90²C; vmax 2973 (C-H), 1675 (C=O), 1598, 1478, 1190, 1077, $758,739 \mathrm{~cm}-1$; ठн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.47 (1H, bs, CONHNH), 7.77 - 7.74 (2H, d, J 7, Ar-H), $7.60-7.58$ (2H, d, J 7, Ar-H), 7.42 - 7.37 (2H, t, J 7, Ar-H), 7.33 7.28 (2H, m, Ar-H), 7.14 (1H, t, J 7, Ar-H), 6.91 (1H, m, Ar-H), 6.86 (1H, s, Ar-H), 6.76 (1H, dd, J 8, 1, Ar-H), 5.91 (1H, bd, J 5, FmocNHCH), 4.44 (2H, m, $\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}$-Fluorene), $4.32(1 \mathrm{H}, \mathrm{m}, \mathrm{FmocNHCH}), 4.24-4.17(2 \mathrm{H}, \mathrm{m}$, FmocNHCHCH, $\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}$-Fluorene), $1.31\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.10(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6$, $\left.\mathrm{NHCHCHCH}_{3}\right) ; ~ \delta c\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 170.2$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.0 ( $\left.\mathrm{C}=\mathrm{ONH}\right), 149.1$ (ipso-Ar-C), 143.8 (ipso-Ar-C), 143.6 (ipso-Ar-C), 141.4 (ipso-Ar-C), 141.3 (ipso-Ar-C), 135.1 (ipso-Ar-C), 130.3 (Ar-C), 127.8 (Ar-C), 127.1 (Ar-C), 125.1 (Ar-C), 121.3 (Ar-C), 120.0 (Ar-C), 113.8 (Ar-C), 111.9 (Ar-C), 75.9 (( $\left.\left.\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 67.1$ $\left(\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}-\right.$ Fluorene), 66.5 (FmocNHCHCH), 58.2 (FmocNHCH), 47.2 (CFluorene), $28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) 17.2\left(\mathrm{NHCHCHCH}_{3}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 544\left(\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right)$, $\left.\left.546 \quad\left({ }^{37} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right), \quad 1065\left(\left[{ }^{35,35} \mathrm{Cl}\right] 2 \mathrm{M}+\mathrm{Na}^{+}\right), \quad 1067 \quad\left({ }^{35,37} \mathrm{CI}\right] 2 \mathrm{M}+\mathrm{Na}^{+}\right), \quad 1069$ $\left(\left[{ }^{37,37} \mathrm{Cl}\right] 2 \mathrm{M}+\mathrm{Na}^{+}\right)$; HRMS (ES $\left.{ }^{+}\right)$Found $\left[{ }^{35} \mathrm{CI}\right] \mathrm{MNa}^{+}$, $544.2015\left(\mathrm{C}_{29} \mathrm{H}_{32} \mathrm{~N}_{3} \mathrm{O}_{4}{ }^{35} \mathrm{CINa}\right.$ requires 544.1974 ).

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tert-Butyl (S)-(2-(2-(3-chlorophenyl)hydrazinyl)-2-oxo-1-phenylethyl) carbamate 67Ab


Following the general procedure outlined, $N$-Boc-L-phenylglycine ( $0.40 \mathrm{~g}, 1.59$ mmol ) and 3-chlorophenylhydrazine hydrochloride ( $0.31 \mathrm{~g}, 1.75 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a pale yellow solid ( $0.28 \mathrm{~g}, 47 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.45$ (DCM/EtOH/NH3 200:6:1); m.p. 146-149 ${ }^{\circ} \mathrm{C}$; $v \max 3302$ (N-H), 1657 (C=O), 1597, 1486, 1362, 1251, 1170, 1051, 763, 696 cm-1; бн ( 300 MHz , DMSO-d6) 10.08 (1H, bd, J 2, CONHNH), 8.12 (1H, bd, J 2, CONHCH), 7. $51-7.49$ (2H, m, Ar-H), $7.40-7.33$ (3H, m, Ar-H), 7.05 (1H, t, J 8, Ar-H), 6.68 (1H, m, Ar-H), 6.53 (1H, m, Ar-H), 6.51 (1H, m, Ar-H), 5.25 (1H, d, J 8, CONHCH), 1.42 ( $\left.9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta \mathrm{c}(100 \mathrm{MHz}$, DMSOd6) 151.1 (ipso-Ar-C), 133.9 (ipso-Ar-C), 130.6 (Ar-C), 128.8 (Ar-C), 128.3 (ArC), $127.8(\mathrm{Ar}-\mathrm{C}), 118.3(\mathrm{Ar}-\mathrm{C}), 111.2(\mathrm{Ar}-\mathrm{C}), 78.7\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 57.2(\mathrm{BocNHCH})$, $\left.\left.28.6 \quad\left(\left(\mathrm{CH}_{3}\right) 3 \mathrm{CO}\right) ; \quad m / z \quad\left(\mathrm{ES}^{+}\right) 398 \quad\left({ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right), 400 \quad\left({ }^{37} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right), \quad 773$ $\left.\left.\left(\left[{ }^{35,35} \mathrm{Cl}\right] 2 \mathrm{M}+\mathrm{Na}^{+}\right), 775\left({ }^{35,37} \mathrm{Cl}\right] 2 \mathrm{M}+\mathrm{Na}^{+}\right), 777\left({ }^{[37,37} \mathrm{Cl}\right] 2 \mathrm{M}+\mathrm{Na}^{+}\right)$; HRMS (ES $\left.{ }^{+}\right)$ Found $\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}$, $398.1257\left(\mathrm{C}_{19} \mathrm{H}_{22}{ }^{35} \mathrm{ClN}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 398.1242).
tert-Butyl (S)-(1-(2-(3-chlorophenyl)hydrazinyl)-1-oxo-4-phenylbutan-2yl)carbamate 67 Ac


Following the general procedure outlined, N -Boc-L-homophenylalanine ( 0.40 g , 1.43 mmol ) and 3-chlorophenylhydrazine hydrochloride ( $0.28 \mathrm{~g}, 1.58 \mathrm{mmol}$ ) were transformed following work up with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a light brown solid ( $0.28 \mathrm{~g}, 48 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.45$ (DCM/EtOH/NH3 200:6:1); m.p. 112-115 ${ }^{\circ} \mathrm{C}$; $v \max 3287$ (N-H), 2979 (C-H), 1676, 1661 (C=O), 1597, 1495, 1367, 1251, 1161, 768, $699 \mathrm{~cm}-1$; $\boldsymbol{\delta}_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.58$ (1H, bs, CONHNH),

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7. $27-7.25$ (2H, m, Ar-H), 7. 21 (1H, m, Ar-H), $7.15-7.12$ (2H, m, Ar-H), 7.06 (1H, t, J 8, Ar-H), 6.83 (1H, m, Ar-H), 6.77 (1H, t, J 2, Ar-H), 6.66 (1H, m, Ar-H), 5.27 (1H, bd, J 8, CONHCH), $4.20(1 \mathrm{H}, \mathrm{m}, \mathrm{CONHCH}), 2.69\left(2 \mathrm{H}, \mathrm{t}, \mathrm{J} 8, \mathrm{CH}_{2} \mathrm{CH}_{2}-\right.$ Ar), 2.19 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{CONHCHCHH}$ ), $1.98(1 \mathrm{H}, \mathrm{m}, \mathrm{CONHCHCHH}), 1.46(9 \mathrm{H}, \mathrm{s}$, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; ~ \delta c\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 172.5$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 156.0 ( $\mathrm{C}=\mathrm{ONH}$ ), 149.1 (ipso-Ar-C), 140.5 (ipso-Ar-C), 135.0 (ipso-Ar-C), 130.2 (Ar-C), 128.6 (Ar-C), 128.4 (Ar-C), 128.3 (Ar-C), 126.3 (Ar-C), 121.0 (Ar-C), 113.4 (Ar-C), 111.7 (ArC), $80.8\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 52.5(\mathrm{BocNHCH}), 33.3\left(\mathrm{CONHCHCH}_{2}\right), 31.8\left(\mathrm{CH}_{2} \mathrm{CH}_{2}-\mathrm{Ar}\right)$, $\left.\left.28.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; m / z\left(\mathrm{ES}^{+}\right) 426\left({ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right), 428\left({ }^{37} \mathrm{CI}\right] \mathrm{MNa}^{+}\right) ;$HRMS (ES ${ }^{+}$) Found $\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}, 426.1573\left(\mathrm{C}_{21} \mathrm{H}_{26}{ }^{35} \mathrm{ClN}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 426.1555).

## tert-Butyl (R)-(1-(2-(3-chlorophenyl)hydrazineyl)-1-oxopropan-2-yl) carbamate 67Ad



Following the general procedure outlined, $N$-Boc- $D$-alanine ( $0.40 \mathrm{~g}, 2.11 \mathrm{mmol}$ ) and 3-chlorophenylhydrazine hydrochloride ( $0.42 \mathrm{~g}, 2.33 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow gummy solid ( $0.32 \mathrm{~g}, 48 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.42$ ( $\mathrm{DCM} / E t O H / \mathrm{NH}_{3} 200: 6: 1$ ); vmax 3428, 3279 (N-H), 1687 (C=O), 1598, 1484, 1245, 11159, 1023, $855 \mathrm{~cm}-1$; бн ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.65 (1H, bs, CONHNH), 7.03 (1H, t, J 7, Ar-H), 6.76 (1H, m, Ar-H), 6.66 (1H, t, J 2, Ar-H), 6.58 (1H, m, Ar-H), 6.28 (1H, bd, J 3, NH-Ar), 5.24 (1H, m, CONHCH), 4.24 (1H, m, CONHCH), 1.38 ( $\left.9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.30$ (1H, d, J 7, CONHCHCH3); $\delta c\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 173.3$ ( $\mathrm{C}=\mathrm{ONHNH}$ ), 155.9 (C=ONH), 149.1 (ipso-Ar-C), 134.9 (ipso-Ar-C), 130.2 (Ar-C), 120.8 (Ar-C), 113.2 (Ar-C), 111.6 (Ar-C), $80.7\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 48.6(\mathrm{BocNHCH}), 28.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) 17.9$ $\left(\mathrm{CONHCHCH}_{3}\right) ; ~ m / z\left(E S^{+}\right) 336\left(\left[{ }^{35} \mathrm{C}\right] \mathrm{MNa}^{+}\right), 338$ ( $\left.{ }^{[37} \mathrm{CI}\right] \mathrm{MNa}^{+}$); HRMS (ES ${ }^{+}$) Found $\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}$, $336.1097\left(\mathrm{C}_{14} \mathrm{H}_{20}{ }^{35} \mathrm{ClN}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 336.1085).

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## tert-Butyl $\quad(R)$-(1-(2-(3-chlorophenyl)hydrazineyl)-4-(methylthio)-1-oxobutan-2-yl)carbamate 67Ae



Following the general procedure outlined, $N$-Boc-D-methionine ( $0.40 \mathrm{~g}, 1.60$ mmol ) and 3-chlorophenylhydrazine hydrochloride ( $0.32 \mathrm{~g}, 1.77 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a yellow gummy solid ( $0.27 \mathrm{~g}, 45 \%$ ); $\mathrm{R}_{\mathrm{f}} 0.47$ (DCM/EtOH/NH3 200:6:1); vmax 3332, 3269 (N-H), 1683, 1659 (C=O), 1597, 1477, 1246, 1219, 1164, 736 cm-1; $\mathrm{\delta H}_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.76$ (1H, bs, CONHNH), 7.03 (1H, t, J 8, Ar-H), 6.76 (1H, d, J 8, Ar-H), 6.69 (1H, t, J 2, Ar-H), 6.58 (1H, dd, J 8, 2, Ar-H), 6.32 (1H, bd, J 3, NH-Ar), 5.41 (1H, bd, J 8, CONHCH), 4.38 (1H, q, J7, CONHCH), 2.49 (2H, $\left.\mathrm{t}, \mathrm{J} 7, \mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right), 2.00\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 1.92\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right)$, 1.38 (9H, s, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right)$; $\delta с$ ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 172.3 (C=ONHNH), 155.9 (C=ONH), 149.1 (ipso-Ar-C), 134.9 (ipso-Ar-C), 130.2 (Ar-C), 120.9 (Ar-C), 113.3 (Ar-C), 111.6 (Ar-C), $80.8\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 51.9(\mathrm{BocNHCH}), 31.2\left(\mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right)$, $30.2\left(\mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right), 28.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 15.4\left(\mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 396$ $\left(\left[{ }^{35} \mathrm{C}\right] \mathrm{MNa}^{+}\right)$, 398 ( $\left.\left.{ }^{37} \mathrm{C}\right]\right] \mathrm{MNa}^{+}$); HRMS (ES ${ }^{+}$) Found $\left[{ }^{35} \mathrm{CI}\right] \mathrm{MNa}^{+}, 396.1136$ $\left(\mathrm{C}_{16} \mathrm{H}_{24}{ }^{35} \mathrm{ClN}_{3} \mathrm{O}_{3} \mathrm{SNa}\right.$ requires 396.1119).

## tert-Butyl (R)-(1-(2-(3-chlorophenyl)hydrazineyl)-1-oxo-3-phenylpropan-2$y \mathrm{y})$ carbamate 67Af



Following the general procedure outlined, $N$-Boc-D-phenylalanine ( $0.40 \mathrm{~g}, 1.51$ mmol ) and 3-chlorophenylhydrazine hydrochloride ( $0.29 \mathrm{~g}, 1.66 \mathrm{mmol}$ ) were transformed following trituration with $\mathrm{Et}_{2} \mathrm{O}$ into the title compound which was isolated as a white solid gum ( $0.26 \mathrm{~g}, 45 \%$ ); $\mathrm{Rf}_{\mathrm{f}} 0.55$ (DCM/EtOH/NH3 200:6:1); m.p. 148-150 ${ }^{\circ} \mathrm{C}$; $v \max 3355,3337,3244$ (N-H), 2987 (C-H), 1692, 1673 (C=O), 1596, 1525, 1166, 863, 772, $679 \mathrm{~cm}-1$; бн $\left.^{(300 ~ M H z, ~ C D C l} 3\right) 7.92$ (1H, bs, CONHNH), 7.38 - 7.30 (3H, m, Ar-H), 7.24 - 7.21 (2H, m, Ar-H), 7.11 (1H, t, J

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8, Ar-H), 6.87 (1H, dd, J 8, 1, Ar-H), 6.67 (1H, t, J 2, Ar-H), 6.51 (1H, d, J 8, ArH), 5.11 (1H, bd, J 8, CONHCH), 4.49 (1H, q, J 7, CONHCH), 3.12 (1H, d, J 7, $\left.\mathrm{CONHCHCH}_{2}\right), 1.46\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; ~ \delta \mathrm{c}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 171.6(\mathrm{C}=\mathrm{ONHNH})$, 148.8 (ipso-Ar-C), 136.0 (ipso-Ar-C), 134.9 (ipso-Ar-C), 130.1 (Ar-C), 129.3 (ArC), 128.9 (Ar-C), 127.3 (Ar-C), 121.2 (Ar-C), 113.4 (Ar-C), 111.7 (Ar-C), 28.4 $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 412\left(\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right), 414\left(\left[{ }^{37} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}, 412.1410\left(\mathrm{C}_{20} \mathrm{H}_{24}{ }^{35} \mathrm{ClN}_{3} \mathrm{O}_{3} \mathrm{Na}\right.$ requires 412.1398).

### 6.1.2.4 Synthesis of Benzoxa-[2,1,3]-diazole Peptidomimetics -

## General Procedure

Under a nitrogen atmosphere $N$-Boc amino acid hydrazides (1.10 equiv.) were dissolved in 4 M HCl solution in dioxane ( 3 mL ) and stirred at room temperature. After 90 minutes the mixture was evaporated, dried in vacuo and precipitated by using (EtOH / Et ${ }_{2} \mathrm{O}$ ). The quat was directly used in the next step. the resulting solid (1 equiv.) was suspended in THF ( 3 mL ) and triethylamine (3 equiv.) was added. After stirring for 10 minutes at room temperature, the solution was treated with 7-chlorobenzoxa-[2,1,3]-diazole-4-sulfonyl chloride (1.10 equiv.) and continued stirring for 5 hours. Reaction mixed with ethyl acetate ( 8 mL ) and distilled water ( 5 mL ). After separation of the two phases the organic layer washed again with distilled water ( $5 \mathrm{~mL} \times 3 \mathrm{~mL}$ ) followed by a wash with sat. aq. $\mathrm{NH}_{4} \mathrm{Cl}(6 \mathrm{~mL})$ then sat. aq. $\mathrm{NaHCO}_{3}(6 \mathrm{~mL})$ followed by brine ( 8 mL ). Organic layer dried over $\mathrm{MgSO}_{4}$, filtered, evaporated and dried in vacuo. The precipitation (EtOAc / n-hexane) and then the flash chromatography (DCM/EtOH/NH3 [600:8:1], [400:8:1], [200:6:1]) afforded the desired sulfonamides 65.

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## N -(2-(2-(2-bromophenyl)hydrazineyl)-2-oxoethyl)-7-chlorobenzo[c][1,2,5] oxadiazole-4-sulfonamide 65h



Following the general procedure outlined, tert-Butyl (2-(2-(2-bromophenyl)hydrazineyl)-2-oxoethyl)carbamate $67 \mathrm{Ce}(0.27 \mathrm{~g}, 0.78 \mathrm{mmol})$ was transformed following precipitation (EtOAc / n-hexane) and then the flash chromatography (DCM/EtOH/NH3 200:6:1) into the title compound which was isolated as a brown gummy solid ( $0.19 \mathrm{~g}, 48 \%$ ) as a mixture of rotamers [4:1]; Rf 0.35 (DCM/EtOH/NH3 200:6:1); $V_{\max } 3156$ (N-H), 1660 (C=O), 1591, 1488, 1340, 1159, 1040, $948 \mathrm{~cm}^{-1}$; all data provided for the major rotamer $\delta_{H}(300 \mathrm{MHz}$, DMSO-d6) 9.93 (1H, bd, J 2, CONHNH), 8.73 (1H, bs, Ar-NH), 8.04 (1H, d, J 7, Ar-H), 7.93 (1H, bd, J 5, SO2NHCH), 7.88 (1H, d, J 7, Ar-H), 7.42 (1H, dd, J 7, 1, Ar-H), 7.20 (1H, td, J 8, 1, Ar-H), 7.05 (1H, d, J 2, Ar-H), 6.62 (1H, dd, J 8, 1, Ar-H), 6.72 (1H, dt, J 7, 1, Ar-H), 3.88 (2H, s, SO2NHCH2); $\delta c(300 \mathrm{MHz}, ~ D M S O-~$ d6) 168.1 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 149.2 (ipso-Ar-C), 145.9 (ipso-Ar-C), 145.5 (ipso-Ar-C), 134.3 (Ar-C), 132.8 (Ar-C), 131.3 (Ar-C), 128.7 (Ar-C), 125.4 (ipso-Ar-C), 120.8 (Ar-C), 113.5 (Ar-C), 107.3 (ipso-Ar-C), $44.2\left(\mathrm{SNHCH}_{2}\right) ; m / z\left(\mathrm{ES}^{+}\right) 459\left({ }^{35} \mathrm{Cl}\right.$, $\left.\left.{ }^{79} \mathrm{Br}\right] \mathrm{MH}^{+}\right), 461\left(\left[{ }^{35} \mathrm{Cl},{ }^{81} \mathrm{Br}\right] \mathrm{MH}^{+}\right), 461\left(\left[{ }^{37} \mathrm{Cl},{ }^{79} \mathrm{Br}\right] \mathrm{MH}^{+}\right), 463\left(\left[{ }^{37} \mathrm{Cl},{ }^{81} \mathrm{Br}^{2}\right] \mathrm{MH}^{+}\right)$; HRMS (ES ${ }^{+}$) Found ( $\left[^{35} \mathrm{Cl},{ }^{79} \mathrm{Br}\right] \mathrm{MH}^{+}$), $459.8853\left(\mathrm{C}_{14} \mathrm{H}_{12}{ }^{35} \mathrm{Cl}^{79} \mathrm{BrN}_{5} \mathrm{O}_{4} \mathrm{~S}\right.$ requires 459.9477).

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## N -(2-(2-(3-bromophenyl)hydrazineyl)-2-oxoethyl)-7-chlorobenzo[c][1,2,5] oxadiazole-4-sulfonamide 65i



Following the general procedure outlined, tert-Butyl (2-(2-(3-bromophenyl)hydrazineyl)-2-oxoethyl)carbamate 67Ck ( $0.17 \mathrm{~g}, 0.49 \mathrm{mmol}$ ) was transformed following precipitation (EtOAc / n-hexane) and then the flash chromatography ( $\mathrm{DCM} / \mathrm{EtOH} / \mathrm{NH}_{3}$ 200:6:1) into the title compound which was isolated as a brown solid ( $0.17 \mathrm{~g}, 68 \%$ ) as a mixture of rotamers [5:1]; Rf 0.33 (DCM/EtOH/NH3 200:6:1); m.p. $247-249{ }^{\circ} \mathrm{C}$; $V_{\max } 3312,3263$ (N-H), 1688, 1658 (C=O), 1589, 1520, 1342, 1159, 1039, $950 \mathrm{~cm}^{-1}$; all data provided for the major rotamer бн ( $300 \mathrm{MHz}, ~ D M S O-d 6) 9.79$ (1H, bd, J 2, CONHNH), 8.72 (1H, bt, J 5, $\mathrm{SO}_{2} \mathrm{NHCH}$ ), 8.03 (1H, dd, J 7, 1, Ar-H), 7.91 (1H, m, Ar-NH), 7.86 (1H, d, J 7, Ar-H), 7.08 (1H, t, J 8, Ar-H), 6.86 (1H, m, Ar-H), 6.80 (1H, t, J 2, Ar-H), 6.61 (1H, dd, J 8, 1, Ar-H), 3.87 (2H, d, J 6, SO2NHCH2); $\delta c\left(300 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d}_{6}\right) 168.1$ (C=ONHNH), 151.0 (ipso-Ar-C), 149.2 (ipso-Ar-C), 145.9 (ipso-Ar-C), 134.1 (ArC), 131.2 (Ar-C), 128.9 (Ar-C), 125.4 (ipso-Ar-C), 122.5 (Ar-C), 121.4 (Ar-C), 114.7 (Ar-C), 111.5 (Ar-C), $44.2\left(\mathrm{SNHCH}_{2}\right) ; m / z\left(\mathrm{ES}^{+}\right) 459\left({ }^{35} \mathrm{Cl},{ }^{79} \mathrm{Br}^{2} \mathrm{MH}^{+}\right), 461$ ( $\left.{ }^{35} \mathrm{Cl},{ }^{81} \mathrm{Br}\right] \mathrm{MH}^{+}$), 461 ( $\left.{ }^{[37} \mathrm{Cl},{ }^{79} \mathrm{Br}\right] \mathrm{MH}^{+}$), $\left.463\left({ }^{37} \mathrm{Cl},{ }^{81} \mathrm{Br}\right] \mathrm{MH}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\left.\left({ }^{35} \mathrm{Cl},{ }^{79} \mathrm{Br}\right] \mathrm{MH}^{+}\right), 461.8837\left(\mathrm{C}_{14} \mathrm{H}_{12}{ }^{35} \mathrm{Cl}^{79} \mathrm{BrN}_{5} \mathrm{O}_{4} \mathrm{~S}\right.$ requires 459.9477).

## N -(2-(2-(4-bromophenyl)hydrazineyl)-2-oxoethyl)-7-chlorobenzo[c][1,2,5] oxadiazole-4-sulfonamide 65j



Following the general procedure outlined, tert-Butyl (2-(2-(4-bromophenyl)hydrazineyl)-2-oxoethyl)carbamate 67Cq (0.22 g, 0.64 mmol ) was

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transformed following precipitation (EtOAc / n-hexane) and then the flash chromatography (DCM/EtOH/NH3 200:6:1) into the title compound which was isolated as a brown gummy solid ( $0.25 \mathrm{~g}, 76 \%$ ) as a mixture of rotamers [4:1]; Rf 0.39 (DCM/EtOH/NH3 200:6:1); $v_{\max } 3302$ (N-H), 1683 (C=O), 1524, 1488, 1340, 1156, 1042, $949 \mathrm{~cm}^{-1}$; all data provided for the major rotamer $\delta_{H}(300 \mathrm{MHz}$, DMSO-d6) 9.76 (1H, bd, J 2, CONHNH), 8.68 (1H, bt, J 5, SO2NHCH), 8.02 (1H, d, J 7, Ar-H), 7.86 (1H, d, J7, Ar-H), 7.80 (1H, bd, J 2, Ar-NH), 7.27 - 7.24 (2H, d, J 8, Ar-H), 6.56 - 6.53 (2H, d, J 8, Ar-H), 3.86 (2H, d, J 5, SO2NHCH2); ठc (300 MHz, DMSO-d6) 168.0 (C=ONHNH), 149.2 (ipso-Ar-C), 148.7 (ipso-Ar-C), 145.9 (ipso-Ar-C), 134.2 (Ar-C), 131.7 (Ar-C), 128.9 (Ar-C), 125.4 (ipso-Ar-C), 114.5 (Ar-C), 109.8 (Ar-C), $44.1\left(\mathrm{SNHCH}_{2}\right) ; m / z\left(\mathrm{ES}^{+}\right) 459\left(\left[{ }^{35} \mathrm{Cl},{ }^{79} \mathrm{Br}\right] \mathrm{MH}^{+}\right), 461\left({ }^{[35} \mathrm{Cl}\right.$, $\left.\left.{ }^{81} \mathrm{Br}\right] \mathrm{MH}^{+}\right), 461\left(\left[{ }^{37} \mathrm{Cl},{ }^{79} \mathrm{Br}\right] \mathrm{MH}^{+}\right), 463\left(\left[{ }^{37} \mathrm{Cl},{ }^{81} \mathrm{Br}\right] \mathrm{MH}^{+}\right)$; HRMS (ES $)$Found $\left({ }^{[35} \mathrm{Cl}\right.$, $\left.\left.{ }^{79} \mathrm{Br}\right] \mathrm{MH}^{+}\right), 459.8880\left(\mathrm{C}_{14} \mathrm{H}_{12}{ }^{35} \mathrm{Cl}^{79} \mathrm{BrN}_{5} \mathrm{O}_{4} \mathrm{~S}\right.$ requires 459.9477).

## (S)-7-chloro-N-(1-oxo-1-(2-phenylhydrazineyl)propan-2-yl)benzo[c][1,2,5] oxadiazole-4-sulfonamide 65k



Following the general procedure outlined, tert-Butyl (S)-(1-oxo-1-(2-phenylhydrazineyl)propan-2-yl)carbamate 67b ( $0.22 \mathrm{~g}, 0.79 \mathrm{mmol}$ ) was transformed following precipitation (EtOAc / n-hexane) and then the flash chromatography (DCM/EtOH/NH3 200:6:1) into the title compound which was isolated as a brown oil (0.09 g, 26\%); Rf 0.34 (DCM/EtOH/NH3 200:6:1); $v_{\max }$ 3083 (N-H), 2917, 2849 (C-H), 1703, 1601 (C=O), 1524, 1345, 1162, 1040, 946 $\mathrm{cm}^{-1}$; $\delta \mathrm{H}(300 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d} 6) 9.75$ (1H, bs, CONHNH), 8.80 (1H, bd, J 8, $\mathrm{SO}_{2} \mathrm{NHCH}$ ), 8.02 (1H, d, J 7, Ar-H), 7.83 (1H, d, J7, Ar-H), 7.59 (1H, bs, Ar-NH), 7.11 - 7.06 (2H, td, J 7, 1, Ar-H), 6.72 (1H, t, J 7, Ar-H), $6.55-6.52$ (1H, d, J 8, Ar-H), $4.22\left(1 \mathrm{H}, \mathrm{m}, \mathrm{SO}_{2} \mathrm{NHCH}\right), 1.30\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 7, \mathrm{NHCHCH}_{3}\right)$; $\delta \mathrm{c}(300 \mathrm{MHz}$, DMSO-d6) 171.3 (C=ONHNH), 149.3 (ipso-Ar-C), 149.2 (ipso-Ar-C), 145.6 (ipso-Ar-C), 134.4 (Ar-C), 131.2 (Ar-C), 129.1 (Ar-C), 125.5 (ipso-Ar-C), 119.1 (Ar-C),

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112.5 ( $\mathrm{Ar}-\mathrm{C}$ ), 51.3 (SNHCH), $19.9\left(\mathrm{NHCHCH}_{3}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 396\left(\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 398$ $\left(\left[{ }^{37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right)$; HRMS $\left(E S^{+}\right)$Found $\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}$, $396.0027\left(\mathrm{C}_{15} \mathrm{H}_{15}{ }^{35} \mathrm{CIN}_{5} \mathrm{O}_{4} \mathrm{~S}\right.$ requires 396.0528).
(S)-7-chloro-N-(1-(2-(3-chlorophenyl)hydrazineyl)-1-oxopropan-2-yl)benzo [c][1,2,5]oxadiazole-4-sulfonamide 65I


Following the general procedure outlined, tert-Butyl (S)-(1-(2-(3-chlorophenyl)hydrazineyl)-1-oxopropan-2-yl)carbamate 67r ( $0.12 \mathrm{~g}, 0.38 \mathrm{mmol}$ ) was transformed following precipitation (EtOAc / n-hexane) and then the flash chromatography (DCM/EtOH/NH3 200:6:1) into the title compound which was isolated as a brown oil ( $0.08 \mathrm{~g}, 48 \%$ ); Rf 0.35 ( $\mathrm{DCM} / \mathrm{EtOH} / \mathrm{NH}_{3}$ 200:6:1); $V_{\max }$ 3273 (N-H), 1674 (C=O), 1597, 1476, 1346, 1151, 1041, $947 \mathrm{~cm}^{-1}$; $\delta_{\text {н }}(300 \mathrm{MHz}$, DMSO-d6) 9.86 (1H, bs, CONHNH), 8.83 (1H, bd, J 8, SO2NHCH), 8.03 (1H, d, J 7, Ar-H), 7.94 (1H, bs, Ar-NH), 7.85 (1H, d, J 7, Ar-H), 7.14 (1H, t, J 8, Ar-H), 6.73 (1H, dd, J 7, 1, Ar-H), 6.63 (1H, t, J 2, Ar-H), 6.53 (1H, dd, J 8, 1, Ar-H), 4.25 (1H, m, SO2NHCH), 1.29 (3H, d, J 7, NHCHCH3); ठc (300 MHz, DMSO-d6) 171.5 (C=ONHNH), 151.0 (ipso-Ar-C), 149.2 (ipso-Ar-C), 145.6 (ipso-Ar-C), 134.3 (Ar-C), 133.9 (ipso-Ar-C), 131.2 (Ar-C), 130.8 (ipso-Ar-C), 128.9 (Ar-C), 125.5 (ipso-Ar-C), 118.5 (Ar-C), 111.9 ( $\mathrm{Ar}-\mathrm{C}$ ), 111.0 ( $\mathrm{Ar}-\mathrm{C}$ ), 51.3 (SNHCH), 19.8 $\left(\mathrm{NHCHCH}_{3}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 430\left(\left[{ }^{35,35} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 433\left(\left[{ }^{35,37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 435\left(\left[{ }^{37,37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\left[{ }^{35,35} \mathrm{Cl}\right] \mathrm{MH}^{+}, 429.9606\left(\mathrm{C}_{15} \mathrm{H}_{14}{ }^{35,35} \mathrm{Cl}_{2} \mathrm{~N}_{5} \mathrm{O}_{4} \mathrm{~S}\right.$ requires 430.0139).

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## (R)-7-chloro-N-(1-(2-(3-chlorophenyl)hydrazineyl)-1-oxopropan-2-yl) benzo[c][1,2,5]oxadiazole-4-sulfonamide 65 m



Following the general procedure outlined, tert-Butyl (R)-(1-(2-(3-chlorophenyl)hydrazineyl)-1-oxopropan-2-yl)carbamate 67Ad ( $0.12 \mathrm{~g}, 0.38 \mathrm{mmol}$ ) was transformed following precipitation (EtOAc / n-hexane) and then the flash chromatography ( $\mathrm{DCM} / \mathrm{EtOH} / \mathrm{NH}_{3}$ 200:6:1) into the title compound which was isolated as a brown gummy solid ( $0.15 \mathrm{~g}, 86 \%$ ); Rf 0.35 (DCM/EtOH/NH3 200:6:1); $V_{\max } 3263$ (N-H), 1666, 1648 (C=O), 1597, 1475, 1348, 1162, 1051, $947 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d}_{6}$ ) 9.86 (1H, bs, CONHNH), 8.83 (1H, bd, J 8, SO2NHCH), 8.03 (1H, d, J 7, Ar-H), 7.93 (1H, bs, Ar-NH), 7.83 (1H, d, J 7, Ar-H), 7.14 (1H, t, J 8, Ar-H), 6.72 (1H, dd, J 7, 1, Ar-H), 6.63 (1H, t, J 2, Ar-H), 6.54 (1H, dd, J 8, 1, Ar-H), 4.26 (1H, m, SO2NHCH), 1.30 (3H, d, J 7, NHCHCH ${ }_{3}$ ); $\delta c(300 \mathrm{MHz}$, DMSO-d6) 171.5 (C=ONHNH), 151.0 (ipso-Ar-C), 149.2 (ipso-Ar-C), 145.6 (ipso-Ar-C), 134.3 (Ar-C), 133.9 (ipso-Ar-C), 131.2 (Ar-C), 130.8 (ipso-Ar-C), 128.9 (Ar-C), 125.6 (ipso-Ar-C), 118.5 (Ar-C), 111.9 (Ar-C), 111.0 (Ar-C), 51.3 (SNHCH), 19.8 ( $\mathrm{NHCHCH}_{3}$ ); m/z (ES $\left.{ }^{+}\right) 430\left(\left[{ }^{35,35} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 433\left(\left[{ }^{35,37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 435$ ( $\left.{ }^{37,37} \mathrm{Cl}\right] \mathrm{MH}^{+}$); HRMS (ES $)$Found $\left[{ }^{35,35} \mathrm{Cl}\right] \mathrm{MH}^{+}, 429.9618\left(\mathrm{C}_{15} \mathrm{H}_{14}{ }^{35,35} \mathrm{Cl}_{2} \mathrm{~N}_{5} \mathrm{O}_{4} \mathrm{~S}\right.$ requires 430.0139).

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(S)-7-chloro-N-(1-(2-(4-chlorophenyl)hydrazineyl)-1-oxopropan-2-yl) benzo[c][1,2,5]oxadiazole-4-sulfonamide 65n


Following the general procedure outlined, tert-butyl (S)-(1-(2-(4-chlorophenyl)hydrazineyl)-1-oxopropan-2-yl)carbamate 67 An ( $0.17 \mathrm{~g}, 0.54 \mathrm{mmol}$ ) was transformed following precipitation (EtOAc / n-hexane) and then the flash chromatography (DCM/EtOH/ $\mathrm{NH}_{3}$ 200:6:1) into the title compound which was isolated as a brown gummy solid ( $0.14 \mathrm{~g}, 54 \%$ ); Rf 0.35 (DCM/EtOH/NH3 200:6:1); $V_{\max } 3276$ (N-H), 1679 (C=O), 1598, 1491, 1338, 1151, $948 \mathrm{~cm}^{-1}$; ठн ( 300 MHz , DMSO-d6) 9.82 (1H, bd, J 2, CONHNH), 8.828 .83 (1H, bd, J8, SO2NHCH), 8.02 (1H, d, J 7, Ar-H), 7.85 (1H, d, J 7, Ar-H), 7.81 (1H, bd, J 2, Ar-NH), 7.14 - 7.11 (2H, d, J 8, Ar-H), 6.57 - 6.54 (2H, d, J 8, Ar-H), 4.20 (1H, m, SO2NHCH), 1.29 (3H, d, J7, NHCHCH3); סc (300 MHz, DMSO-d 6 ) 171.4 (C=ONHNH), 149.2 (ipso-Ar-C), 148.4 (ipso-Ar-C), 145.6 (ipso-Ar-C), 134.4 (Ar-C), 131.3 (Ar-C), 128.9 (ArC), 125.5 (ipso-Ar-C), 122.4 (Ar-C), 113.9 (Ar-C), 51.3 (SNHCH), 19.9 (NHCHCH3); m/z (ES $) 430\left(\left[{ }^{35,35} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 433\left(\left[{ }^{35,37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 435\left(\left[{ }^{37,37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right)$; HRMS (ES $)$ Found $\left[{ }^{35,35} \mathrm{Cl}\right] \mathrm{MH}^{+}$, $429.9630\left(\mathrm{C}_{15} \mathrm{H}_{14}{ }^{35,35} \mathrm{Cl}_{2} \mathrm{~N}_{5} \mathrm{O}_{4} \mathrm{~S}\right.$ requires 430.0139).
(S)-7-chloro-N-(1-(2-(2,6-dichlorophenyl)hydrazineyl)-1-oxopropan-2-yl) benzo[c][1,2,5]oxadiazole-4-sulfonamide 650


Following the general procedure outlined, tert-Butyl (S)-(1-(2-(2,6-dichlorophenyl)hydrazinyl)-1-oxopropan-2-yl)carbamate 67At (0.12 g, 0.35 mmol ) was transformed following precipitation (EtOAc / n-hexane) and then the

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flash chromatography (DCM/EtOH/NH3 200:6:1) into the title compound which was isolated as a brown gummy solid ( $0.16 \mathrm{~g}, 97 \%$ ); Rf 0.28 (DCM/EtOH/NH3 200:6:1); $v_{\max } 2920(\mathrm{C}-\mathrm{H}), 1688,1602$ (C=O), 1524, 1496, 1216, 1153, 1051, $\mathrm{cm}^{-}$ ${ }^{1}$; $\delta_{H}\left(300 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d}_{6}\right) 10.08$ (1H, bd, J 2, CONHNH), 8.65 (1H, bd, J 8, SO2NHCH), 7.90 (1H, d, J 7, Ar-H), 7.77 (1H, d, J 7, Ar-H), 7.27 - 7.24 (2H, d, J 8, Ar-H), 6.95 (1H, m, Ar-H), 6.70 (1H, bd, J 2, Ar-NH), 4.15 (1H, dd, J 7, 1, $\mathrm{SO}_{2} \mathrm{NHCH}$ ), 1.22 (3H, d, J 7, NHCHCH3); ठc ( $300 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d}_{6}$ ) 170.5 (C=ONHNH), 149.0 (ipso-Ar-C), 141.4 (ipso-Ar-C), 134.2 (Ar-C), 129.3 (Ar-C), 125.5 (ipso-Ar-C), 124.5 (Ar-C), 123.5 (Ar-C), 50.9 (SNHCH), 19.6 ( $\mathrm{NHCHCH}_{3}$ ); $\mathrm{m} / \mathrm{z}\left(\right.$ ES $\left.^{+}\right) 463\left(\left[{ }^{35,35,35} \mathrm{CI}\right] \mathrm{MH}^{+}\right), 465\left(\left[{ }^{35,35,37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 467\left(\left[{ }^{35,37,37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 469$ $\left(\left[{ }^{37,37,37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right) ; \quad$ HRMS $\left(\mathrm{ES}^{+}\right) \quad$ Found $\quad\left[{ }^{35,35,35} \mathrm{Cl}\right] \mathrm{MH}^{+}$, 463.9211 $\left(\mathrm{C}_{15} \mathrm{H}_{12}{ }^{35,35,35} \mathrm{Cl}_{3} \mathrm{~N}_{5} \mathrm{O}_{4} \mathrm{~S}\right.$ requires 463.9749).
(S)-N-(1-(2-(3-bromophenyl)hydrazineyl)-1-oxopropan-2-yl)-7-chlorobenzo [c][1,2,5]oxadiazole-4-sulfonamide 65p


Following the general procedure outlined, tert-Butyl (S)-(1-(2-(3-bromophenyl)hydrazineyl)-1-oxopropan-2-yl)carbamate $\mathbf{6 7 C l}(0.12 \mathrm{~g}, 0.34 \mathrm{mmol})$ was transformed following precipitation (EtOAc / n-hexane) and then the flash chromatography (DCM/EtOH/NH3 200:6:1) into the title compound which was isolated as a brown gummy solid ( $0.13 \mathrm{~g}, 79 \%$ ); Rf 0.33 (DCM/EtOH/NH3 200:6:1); $V_{\max } 3261$ (N-H), 1665 (C=O), 1595, 1474, 1347, 1149, 1040, $946 \mathrm{~cm}^{-1}$; ठн (300 MHz, DMSO-d6) 9.86 (1H, bd, J 2, CONHNH), 8.83 (1H, bd, J 8, SO2NHCH), 8.02 (1H, d, J 7, Ar-H), 7.93 (1H, bd, J 2, Ar-NH), 7.84 (1H, d, J 7, Ar-H), 7.08 (1H, t, J 8, Ar-H), 6.86 (1H, m, Ar-H), 6.79 (1H, t, J 2, Ar-H), 6.57 (1H, dd, J 8, 1, Ar-H), $4.25\left(1 \mathrm{H}, \mathrm{m}, \mathrm{SO}_{2} \mathrm{NHCH}\right), 1.29\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 7, \mathrm{NHCHCH}_{3}\right) ; \delta c(300 \mathrm{MHz}$, DMSO-d6) 171.5 (C=ONHNH), 151.2 (ipso-Ar-C), 149.2 (ipso-Ar-C), 145.6 (ipso-Ar-C), 134.3 (Ar-C), 131.2 (Ar-C), 131.1 (Ar-C), 129.0 (Ar-C), 125.5 (ipso-Ar-C), 122.5 (ipso-Ar-C), 121.4 (Ar-C), 114.8 (Ar-C), 111.4 (Ar-C), 51.3 (SNHCH), 19.8

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$\left.\left(\mathrm{NHCHCH}_{3}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 473\left({ }^{35} \mathrm{Cl},{ }^{79} \mathrm{Br}\right] \mathrm{MH}^{+}\right), 475\left(\left[{ }^{35} \mathrm{Cl},{ }^{81} \mathrm{Br}\right] \mathrm{MH}^{+}\right), 475\left({ }^{37} \mathrm{Cl}\right.$, $\left.\left.{ }^{79} \mathrm{Br}\right] \mathrm{MH}^{+}\right), 477\left(\left[{ }^{37} \mathrm{Cl},{ }^{81} \mathrm{Br}\right] \mathrm{MH}^{+}\right)$; HRMS (ES $\left.{ }^{+}\right)$Found $\left.\left({ }^{35} \mathrm{Cl},{ }^{79} \mathrm{Br}\right] \mathrm{MH}^{+}\right), 473.9089$ $\left(\mathrm{C}_{15} \mathrm{H}_{14}{ }^{35} \mathrm{Cl}^{79} \mathrm{BrN}_{5} \mathrm{O}_{4} \mathrm{~S}\right.$ requires 473.9633$)$.
(S)-N-(1-(2-(4-bromophenyl)hydrazineyl)-1-oxopropan-2-yl)-7-chlorobenzo [c][1,2,5]oxadiazole-4-sulfonamide 65q


Following the general procedure outlined, tert-Butyl (S)-(1-(2-(4-bromophenyl)hydrazineyl)-1-oxopropan-2-yl)carbamate $67 \mathrm{Cr}(0.17 \mathrm{~g}, 0.48 \mathrm{mmol})$ was transformed following precipitation (EtOAc / n-hexane) and then the flash chromatography (DCM/EtOH/ $\mathrm{NH}_{3}$ 200:6:1) into the title compound which was isolated as a brown oil ( $0.16 \mathrm{~g}, 67 \%$ ); Rf 0.39 ( $\mathrm{DCM} / \mathrm{EtOH} / \mathrm{NH}_{3} 200: 6: 1$ ); $v_{\text {max }}$ 3319 (N-H), 1678 (C=O), 1521, 1447, 1335, 1245, 1137, 1041, $945 \mathrm{~cm}^{-1}$; ठн (300 MHz, DMSO-d6) 9.82 (1H, bs, CONHNH), 8.82 (1H, bd, J 8, SO2NHCH), 8.03 (1H, d, J 7, Ar-H), 7.85 (1H, d, J 7, Ar-H), 7.26 - 7.23 (2H, d, J 8, Ar-H), 6.52 6.49 (2H, d, J 8, Ar-H), 4.23 (1H, m, SO2NHCH), 1.29 (3H, d, J 7, NHCHCH3); ठс ( $300 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d} 6$ ) 171.4 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 149.2 (ipso-Ar-C), 148.8 (ipso-ArC), 145.6 (ipso-Ar-C), 134.4 (Ar-C), 131.7 (Ar-C), 131.2 (Ar-C), 129.1 (Ar-C), 128.9 (ipso-Ar-C), 125.5 (ipso-Ar-C), 114.5 (Ar-C), 109.9 (Ar-C), 51.3 (SNHCH), $19.9\left(\mathrm{NHCHCH}_{3}\right) ; m / z\left(\mathrm{ES}^{+}\right) 473\left(\left[{ }^{35} \mathrm{Cl},{ }^{79} \mathrm{Br}\right] \mathrm{MH}^{+}\right), 475\left(\left[{ }^{35} \mathrm{Cl},{ }^{81} \mathrm{Br}\right] \mathrm{MH}^{+}\right), 475$ ( $\left.{ }^{37} \mathrm{Cl},{ }^{79} \mathrm{Br}\right] \mathrm{MH}^{+}$), 477 ( $\left.{ }^{37} \mathrm{Cl},{ }^{81} \mathrm{Br}\right] \mathrm{MH}^{+}$); HRMS (ES ${ }^{+}$) Found ( $\left[{ }^{35} \mathrm{Cl},{ }^{79} \mathrm{Br}\right] \mathrm{MH}^{+}$), $473.9099\left(\mathrm{C}_{15} \mathrm{H}_{14}{ }^{35} \mathrm{Cl}^{79} \mathrm{BrN}_{5} \mathrm{O}_{4} \mathrm{~S}\right.$ requires 473.9633).

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(S)-7-chloro-N-(1-oxo-1-(2-(p-tolyl)hydrazineyl)propan-2-yl)benzo[c][1,2,5] oxadiazole-4-sulfonamide 65r


Following the general procedure outlined, tert-Butyl (S)-(1-oxo-1-(2-(p-tolyl)hydrazinyl)propan-2-yl)carbamate 67Eb ( $0.17 \mathrm{~g}, 0.58 \mathrm{mmol}$ ) was transformed following precipitation (EtOAc / n-hexane) and then the flash chromatography ( $\mathrm{DCM} / \mathrm{EtOH} / \mathrm{NH}_{3}$ 200:6:1) into the title compound which was isolated as a brown gummy solid ( $0.23 \mathrm{~g}, 86 \%$ ); Rf 0.34 (DCM/EtOH/NH3 200:6:1); $V_{\max } 3312,3097$ (N-H), 1677, 1641 (C=O), 1512, 1450, 1335, 1137, $945 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d}_{6}$ ) 9.71 (1H, bs, CONHNH), 8.78 (1H, bd, J 8, SO2NHCH), 8.01 (1H, d, J 7, Ar-H), 7.82 (1H, d, J 7, Ar-H), 7.40 (1H, bs, Ar-NH), 6.90 - 6.88 (2H, d, J 8, Ar-H), 6.45-6.43 (2H, d, J 8, Ar-H), 4.20 (1H, dd, J 7, 1, SO2NHCH), 2.18 (3H, s, Ar-CH3), 1.29 (3H, d, J7, NHCHCH 3 ); סc ( 300 MHz , DMSO-d6) 171.1 (C=ONHNH), 147.1 (ipso-Ar-C), 145.6 (ipso-Ar-C), 134.3 (Ar-C), 131.2 (Ar-C), 129.5 (Ar-C), 127.6 (Ar-C), 125.4 (ipso-Ar-C), 112.7 (Ar-C), 51.3 (SNHCH), 20.6 (Ar-CH3), $20.0\left(\mathrm{NHCHCH}_{3}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 410\left(\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 412\left(\left[{ }^{37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right) ; ~ H R M S$ (ES ${ }^{+}$) Found $\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}, 410.0227\left(\mathrm{C}_{16} \mathrm{H}_{17}{ }^{35} \mathrm{CIN}_{5} \mathrm{O}_{4} \mathrm{~S}\right.$ requires 410.0685).

## (S)-7-chloro-N-(1-(2-(4-isopropylphenyl)hydrazineyl)-1-oxopropan-2-yl) benzo[c][1,2,5]oxadiazole-4-sulfonamide 65s



Following the general procedure outlined, tert-Butyl (S)-(1-(2-(4-isopropylphenyl)hydrazinyl)-1-oxopropan-2-yl)carbamate 67Eh (0.22 g, 0.68 mmol ) was transformed following precipitation (EtOAc / n-hexane) and then the

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flash chromatography (DCM/EtOH/NH3 200:6:1) into the title compound which was isolated as a brown gummy solid ( $0.23 \mathrm{~g}, 68 \%$ ) as a mixture of rotamers [6:1]; Rf 0.36 (DCM/EtOH/NH3 200:6:1); $V_{\max } 3270$ (N-H), 2959 (C-H), 1674, 1615 (C=O), 1513, 1339, 1151, 1041, $947 \mathrm{~cm}^{-1}$; all data provided for the major rotamer $\delta_{н}\left(300 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d}_{6}\right) 9.73$ (1H, bs, CONHNH), 8.79 (1H, bd, J 8, $\mathrm{SO}_{2} \mathrm{NHCH}$ ), 8.02 (1H, d, J7, Ar-H), 7.81 (1H, d, J 7, Ar-H), 6.97 - 6.94 (2H, d, J 8, Ar-H), 6.47 - 6.44 (2H, d, J 8, Ar-H), 4.24 (1H, m, SO2NHCH), 2.80 (1H, m, Ar-CH(CH3)2), $1.30\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 7, \mathrm{NHCHCH}_{3}\right), 1.17-1.14\left(6 \mathrm{H}, \mathrm{d}, \mathrm{J} 7, \mathrm{Ar}-\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$; ठc ( 300 MHz , DMSO-d6) 171.2 (C=ONHNH), 149.2 (ipso-Ar-C), 147.4 (ipso-Ar-C), 145.6 (ipso-Ar-C), 139.1 (ipso-Ar-C), 134.4 (Ar-C), 131.2 (Ar-C), 128.9 (Ar-C), 126.8 (Ar-C), 125.4 (ipso-Ar-C), 112.6 (Ar-C), 51.3 (SNHCH), 33.1 ( $\mathrm{Ar}-\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ ), 24.7 (Ar$\left.\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $20.0\left(\mathrm{NHCHCH}_{3}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 437\left(\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 439\left(\left[{ }^{37} \mathrm{C}\right] \mathrm{MH}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}, 436.0531\left(\mathrm{C}_{18} \mathrm{H}_{21}{ }^{35} \mathrm{CIN}_{5} \mathrm{O}_{4} \mathrm{~S}\right.$ requires 437.0998).
(S)-7-chloro-N-(1-(2-(4-cyanophenyl)hydrazineyl)-1-oxopropan-2-yl)benzo [c][1,2,5]oxadiazole-4-sulfonamide 65t


Following the general procedure outlined, tert-Butyl (S)-(1-(2-(4-cyanophenyl)hydrazinyl)-1-oxopropan-2-yl)carbamate 67Cx ( $0.22 \mathrm{~g}, 0.72 \mathrm{mmol}$ ) was transformed following precipitation (EtOAc / n-hexane) and then the flash chromatography (DCM/EtOH/ $\mathrm{NH}_{3}$ 200:6:1) into the title compound which was isolated as a brown oil ( $0.22 \mathrm{~g}, 62 \%$ ); Rf 0.41 (DCM/EtOH/NH3 200:6:1); $v_{\text {max }}$ 3278 (N-H), 2220 (C=N), 1686, 1607 (C=O), 1511, 1336, 1154, 1041, $948 \mathrm{~cm}^{-1}$; ठн ( 300 MHz , DMSO-d6) 9.99 (1H, bs, CONHNH), 8.86 (1H, bs, SO2NHCH), 8.53 (1H, bs, NH-Ar), 8.04 (1H, d, J 7, Ar-H), 7.89 (1H, d, J 7, Ar-H), 7.54 - 7.52 (2H, d, J 8, Ar-H), 6.65 - 6.62 (2H, d, J 8, Ar-H), 4.22 (1H, m, SO2NHCH), 1.30 (3H, d, J 7, NHCHCH3); ठc (300 MHz, DMSO-d6) 171.5 (C=ONHNH), 152.9 (ipso-ArC), 149.2 (ipso-Ar-C), 145.6 (ipso-Ar-C), 134.4 (Ar-C), 133.8 (Ar-C), 131.3 (Ar-C),

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128.9 (Ar-C), 125.5 (ipso-Ar-C), 120.5 ( $C \equiv N$ ), 112.0 (Ar-C), 99.5 (Ar-C), 51.3 (SNHCH), $\left.19.7\left(\mathrm{NHCHCH}_{3}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 421\left({ }^{[35} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 423\left(\left[{ }^{37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right) ;$HRMS (ES ${ }^{+}$) Found $\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}, 421.0025\left(\mathrm{C}_{16} \mathrm{H}_{14}{ }^{35} \mathrm{CIN}_{6} \mathrm{O}_{4} \mathrm{~S}\right.$ requires 421.0481).

## (S)-7-chloro-N-(1-(2-(3-nitrophenyl)hydrazineyl)-1-oxopropan-2-yl)benzo [c][1,2,5]oxadiazole-4-sulfonamide 65u



Following the general procedure outlined, tert-Butyl (S)-(1-(2-(3-nitrophenyl)hydrazineyl)-1-oxopropan-2-yl)carbamate 67Dd ( $0.22 \mathrm{~g}, 0.68 \mathrm{mmol}$ ) was transformed following precipitation (EtOAc / n-hexane) and then the flash chromatography (DCM/EtOH/ $\mathrm{NH}_{3}$ 200:6:1) into the title compound which was isolated as a brown gummy solid ( $0.19 \mathrm{~g}, 56 \%$ ); Rf 0.24 (DCM/EtOH/NH3 200:6:1); $V_{\max } 3348,3267$ (N-H), 1669, 1652 (C=O), 1532, 1340, 1149, $946 \mathrm{~cm}^{-1}$; $\delta_{H}(300$ MHz, DMSO-d6) 10.0 (1H, bs, CONHNH), 8.86 (1H, bd, J 8, SO2NHCH), 8.03 (1H, d, J7, Ar-H), 7.83 (1H, d, J 7, Ar-H), 7.56 (1H, dd, J7, 1, Ar-H), 7.43 (1H, t, J 2, Ar-H), 7.40 (1H, d, J 8, Ar-H), 7.02 (1H, dd, J 7, 1, Ar-H), 4.28 (1H, m, $\mathrm{SO}_{2} \mathrm{NHCH}$ ), 1.31 (3H, d, J 7, $\mathrm{NHCHCH}_{3}$ ); ठc ( $300 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d} 6$ ) 171.6 (C=ONHNH), 150.7 (ipso-Ar-C), 149.0 (ipso-Ar-C), 145.6 (ipso-Ar-C), 134.3 (ArC), 131.1 ( $\mathrm{Ar}-\mathrm{C}$ ), 130.6 (Ar-C), 128.9 (Ar-C), 125.6 (ipso-Ar-C), 118.6 (Ar-C), 113.3 (Ar-C), 106.2 ( $\mathrm{Ar}-\mathrm{C}$ ), 51.3 (SNHCH), $19.7\left(\mathrm{NHCHCH}_{3}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 441$ $\left(\left[{ }^{35} \mathrm{CI}\right] \mathrm{MH}^{+}\right), 443$ ( $\left[^{37} \mathrm{CI}\right] \mathrm{MH}^{+}$); HRMS (ES $)$Found $\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}, 440.9915$ $\left(\mathrm{C}_{15} \mathrm{H}_{14}{ }^{35} \mathrm{CIN}_{6} \mathrm{O}_{6} \mathrm{~S}\right.$ requires 441.0379 ).

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(S)-7-chloro-N-(4-(methylthio)-1-oxo-1-(2-phenylhydrazineyl)butan-2-yl) benzo[c][1,2,5]oxadiazole-4-sulfonamide 65v


Following the general procedure outlined, tert-Butyl (S)-(4-(methylthio)-1-oxo-1-(2-phenylhydrazinyl)butan-2-yl)carbamate $67 \mathrm{~g}(0.22 \mathrm{~g}, 0.65 \mathrm{mmol}$ ) was transformed following precipitation (EtOAc / n-hexane) and then the flash chromatography (DCM/EtOH/NH3 200:6:1) into the title compound which was isolated as a brown oil ( $0.17 \mathrm{~g}, 53 \%$ ); Rf 0.38 (DCM/EtOH/NH3 200:6:1); $v_{\text {max }}$ 3348, 3319 (N-H), 2855 (C-H), 1663, 1603 (C=O), 1524, 1494, 1332, 1154, 954 $\mathrm{cm}^{-1}$; $\delta_{H}\left(300 \mathrm{MHz}, \mathrm{DMSO}^{-d}\right) 9.84$ (1H, bd, J 2, CONHNH), 8.87 (1H, bs, NHAr), 8.05 (1H, d, J 7, Ar-H), 7.88 (1H, d, J 7, Ar-H), 7.63 (1H, bd, J 2, SO2NHCH), $7.12-7.07(2 \mathrm{H}, \mathrm{t}, J 7, \mathrm{Ar}-\mathrm{H}), 6.72(1 \mathrm{H}, \mathrm{t}, J 7$, Ar-H), $6.57-6.54$ (2H, d, J 7, ArH), 4.11 (1H, m, SO2NHCH), 2.43 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}_{2} \mathrm{CHHS}$ ), 2.32 ( $1 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CHCH}_{2} \mathrm{CHHS}\right), 1.92\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 1.87\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right)$; $\delta \mathrm{c}(300$ MHz, DMSO-d6) 170.5 (C=ONHNH), 149.4 (ipso-Ar-C), 145.6 (ipso-Ar-C), 134.9 (Ar-C), 131.9 (Ar-C), 129.1 (Ar-C), 128.5 (Ar-C), 125.6 (ipso-Ar-C), 119.0 (Ar-C), 112.5 (Ar-C), $54.4(\mathrm{SNHCH}), 32.3\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 29.8\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 14.8$ $\left.\left.\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right) ; m / z\left(\mathrm{ES}^{+}\right) 456\left({ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 458\left({ }^{37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right)$; HRMS (ES $\left.{ }^{+}\right)$Found $\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}, 456.0096\left(\mathrm{C}_{17} \mathrm{H}_{19}{ }^{35} \mathrm{CIN}_{5} \mathrm{O}_{4} \mathrm{~S}_{2}\right.$ requires 456.0562$)$.

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( $R$ )-7-chloro-N-(4-(methylthio)-1-oxo-1-(2-phenylhydrazineyl)butan-2-yl)benzo[c][1,2,5]oxadiazole-4-sulfonamide 65w


Following the general procedure outlined, tert-Butyl (R)-(4-(methylthio)-1-oxo-1-(2-phenylhydrazinyl)butan-2-yl)carbamate 67 o ( $0.12 \mathrm{~g}, 0.35 \mathrm{mmol}$ ) was transformed following precipitation (EtOAc / n-hexane) and then the flash chromatography (DCM/EtOH/ $\mathrm{NH}_{3}$ 200:6:1) into the title compound which was isolated as a brown gummy solid ( $0.12 \mathrm{~g}, 72 \%$ ); Rf 0.35 (DCM/EtOH/NH3 200:6:1); $V_{\max } 3227$ (N-H), 2853 (C-H), 1662 (C=O), 1524, 1331, 1153, 1099, $951 \mathrm{~cm}^{-1}$; $\delta_{\mathrm{H}}$ ( 300 MHz , DMSO-d6) 9.85 (1H, bs, CONHNH), 8.90 (1H, bd, J 8, NH-Ar), 8.06 (1H, d, J7, Ar-H), 7.87 (1H, d, J7, Ar-H), $7.12-7.07$ (2H, t, J7, Ar-H), 6.72 (1H, $\mathrm{t}, J 7, \mathrm{Ar}-H), 6.57-6.55(2 \mathrm{H}, \mathrm{d}, J 7, \mathrm{Ar}-\mathrm{H}), 4.13\left(1 \mathrm{H}, \mathrm{m}, \mathrm{SO}_{2} \mathrm{NHCH}\right), 2.45(1 \mathrm{H}$, $\mathrm{m}, \mathrm{CHCH}_{2} \mathrm{CHHS}$ ), $2.33\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}_{2} \mathrm{CHHS}\right), 1.92\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 1.88$ (2H, m, CHCH2CH2S); ठc (300 MHz, DMSO-d6) 170.6 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 149.4 (ipso-Ar-C), 145.6 (ipso-Ar-C), 134.9 (Ar-C), 131.3 (Ar-C), 129.1 (Ar-C), 128.5 (Ar-C), 125.6 (ipso-Ar-C), 119.1 (Ar-C), 112.5 (Ar-C), 54.4 (SNHCH), 32.3 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 29.8\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right)$, $14.8\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 456$ $\left.\left.\left({ }^{35} \mathrm{C} \mid\right] \mathrm{MH}^{+}\right), 458 \quad\left({ }^{37} \mathrm{C}\right] \mathrm{MH}^{+}\right)$; HRMS (ES $\left.{ }^{+}\right)$Found $\left[{ }^{35} \mathrm{CI}\right] \mathrm{MH}^{+}, 456.0105$ $\left(\mathrm{C}_{17} \mathrm{H}_{19}{ }^{35} \mathrm{CIN}_{5} \mathrm{O}_{4} \mathrm{~S}_{2}\right.$ requires 456.0562 ).

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(S)-7-chloro-N-(4-(methylthio)-1-oxo-1-(2-(4-(trifluoromethyl)phenyl) hydrazineyl)butan-2-yl)benzo[c][1,2,5]oxadiazole-4-sulfonamide 65x


Following the general procedure outlined, tert-Butyl (S)-(4-(methylthio)-1-oxo-1-(2-(4-(trifluoromethyl)phenyl)hydrazineyl)butan-2-yl)carbamate 67Bo (0.32 g, 0.79 mmol ) was transformed following precipitation (EtOAc / $n$-hexane) and then the flash chromatography ( $\mathrm{DCM} / \mathrm{EtOH} / \mathrm{NH}_{3} 200: 6: 1$ ) into the title compound which was isolated as a yellow gummy solid ( $0.20 \mathrm{~g}, 43 \%$ ); Rf 0.40 (DCM/EtOH/NH3 200:6:1); $V_{\max } 3259$ (N-H), 1682, 1617 (C=O), 1524, 1323, 1156, 1108, 1065, $953 \mathrm{~cm}^{-1}$; ठн ( 300 MHz , DMSO-d6) 10.02 (1H, bs, CONHNH), 8.95 (1H, bd, J 7, NH-Ar), 8.31 (1H, bd, J 4, SO2NHCH), 8.07 (1H, d, J 8, Ar-H), 7.90 (1H, d, J 8, Ar-H), $7.44-7.41$ (2H, d, J 8, Ar-H), $6.68-6.65$ (2H, d, J 8, Ar-H), $4.12\left(1 \mathrm{H}, \mathrm{m}, \mathrm{SO}_{2} \mathrm{NHCH}\right), 2.44\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}_{2} \mathrm{CHHS}\right), 2.32(1 \mathrm{H}, \mathrm{m}$, $\mathrm{CHCH}_{2} \mathrm{CHHS}$ ), $1.92\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 1.89\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right)$; $\delta \mathrm{c}(300$ MHz, DMSO-d6) 170.8 (C=ONHNH), 152.5 (ipso-Ar-C), 149.2 (ipso-Ar-C), 145.6 (ipso-Ar-C), 134.9 (Ar-C), 131.3 (Ar-C), 128.5 (ipso-Ar-C), 126.6 (Ar-C), 126.5 (Ar-C), 125.6 (Ar-C), 119.0 (1C, q, J 31, Ar-CF3), 111.8 (Ar-C), 54.3 (SNHCH), $32.1\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right)$, $29.8\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right)$, $14.8\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right)$; $\delta$ F $(300 \mathrm{MHz}$, DMSO-d 6 ) -59.4 (CF3); m/z (ES $\left.{ }^{+}\right) 524\left(\left[{ }^{35} \mathrm{CI}\right] \mathrm{MH}^{+}\right), 526\left(\left[{ }^{37} \mathrm{CI}\right] \mathrm{MH}^{+}\right)$; HRMS (ES $\left.{ }^{+}\right)$ Found $\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}, 523.9919\left(\mathrm{C}_{18} \mathrm{H}_{18}{ }^{35} \mathrm{CIF}_{3} \mathrm{~N}_{5} \mathrm{O}_{4} \mathrm{~S}_{2}\right.$ requires 524.0436).

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(S)-7-chloro-N-(1-(2-(3-chlorophenyl)hydrazineyl)-4-(methylthio)-1-oxobutan-2-yl)benzo[c][1,2,5]oxadiazole-4-sulfonamide 65y


Following the general procedure outlined, tert-Butyl (S)-(1-(2-(3-chlorophenyl)hydrazinyl)-4-(methylthio)-1-oxobutan-2-yl)carbamate 67w ( 0.22 g , 0.65 mmol ) was transformed following precipitation (EtOAc / $n$-hexane) and then the flash chromatography (DCM/EtOH/ $\mathrm{NH}_{3}$ 200:6:1) into the title compound which was isolated as a brown gummy solid ( $0.22 \mathrm{~g}, 69 \%$ ); Rf 0.41 (DCM/EtOH/NH3 200:6:1); $V_{\max }$ 3232, 3076 (N-H), 2919 (C-H), 1683 (C=O), 1597, 1525, 1340, 1156, 1023, $952 \mathrm{~cm}^{-1}$; бн ( 300 MHz , DMSO-d6) 9.95 (1H, bs, CONHNH), 8.94 (1H, bd, J 8, SO2NHCH), 8.06 (1H, d, J 7, Ar-H), 7.96 (1H, bs, NH-Ar), 7.88 (1H, d, J 7, Ar-H), 7.14 (1H, t, J 8, Ar-H), 6.74 (1H, dd, J 7, 1, ArH), 6.67 (1H, t, J 2, Ar-H), 6.56 (1H, dd, J7, 1, Ar-H), 4.11 (1H, q, J 7, SO2NHCH), 2.41 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}_{2} \mathrm{CHHS}$ ), 2.31 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}_{2} \mathrm{CHHS}$ ), $1.90(3 \mathrm{H}, \mathrm{s}$, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}$ ), 1.86 (2H, m, $\mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{~S}$ ); бс ( $300 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d}_{6}$ ) 170.8 (C=ONHNH), 151.0 (ipso-Ar-C), 149.3 (ipso-Ar-C), 145.6 (ipso-Ar-C), 134.9 (ArC), 133.9 (ipso-Ar-C), 131.2 (Ar-C), 130.8 (Ar-C), 128.4 (ipso-Ar-C), 125.7 (ipso-Ar-C), 118.5 (Ar-C), 111.9 (Ar-C), 111.1 (Ar-C), 54.3 (SNHCH), 32.0 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 29.7\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right)$, $14.8\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 490$ $\left(\left[{ }^{35,35} \mathrm{CI}\right] \mathrm{MH}^{+}\right), 492\left(\left[{ }^{35,37} \mathrm{CI}\right] \mathrm{MH}^{+}\right), 494$ ( $\left.{ }^{37,37} \mathrm{C}\right] \mathrm{MH}^{+}$); HRMS (ES ${ }^{+}$) Found $\left[{ }^{35,35} \mathrm{Cl}\right] \mathrm{MH}^{+}, 489.9696\left(\mathrm{C}_{17} \mathrm{H}_{18}{ }^{35,35} \mathrm{Cl}_{2} \mathrm{~N}_{5} \mathrm{O}_{4} \mathrm{~S}_{2}\right.$ requires 490.0172).

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(R)-7-chloro-N-(1-(2-(3-chlorophenyl)hydrazineyl)-4-(methylthio)-1-oxobutan-2-yl)benzo[c][1,2,5]oxadiazole-4-sulfonamide $\mathbf{6 5 z}$


Following the general procedure outlined, tert-Butyl (R)-(1-(2-(3-chlorophenyl)hydrazinyl)-4-(methylthio)-1-oxobutan-2-yl)carbamate 67Ae (0.12 $\mathrm{g}, 0.32 \mathrm{mmol}$ ) was transformed following precipitation (EtOAc / n-hexane) and then the flash chromatography ( $\mathrm{DCM} / \mathrm{EtOH} / \mathrm{NH}_{3} 200: 6: 1$ ) into the title compound which was isolated as a brown gummy solid ( $0.09 \mathrm{~g}, 56 \%$ ); Rf 0.41 (DCM/EtOH/NH3 200:6:1); $V_{\max }$ 3357, 3322 (N-H), 3077, 2855 (C-H), 1665, 1600 (C=O), 1524, 1328, 1154, 952 cm$^{-1}$; $\delta \mathrm{H}(300 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d} 6) 9.96$ (1H, bd, J 2, CONHNH), 8.06 (1H, d, J 7, Ar-H), 7.97 (1H, bd, J 2, SO2NHCH), 7.88 (1H, d, J 7, Ar-H), 7.14 (1H, t, J 8, Ar-H), 6.74 (1H, dd, J 7, 1, Ar-H), 6.66 (1H, t, J 2, ArH), 6.56 (1H, dd, J 7, 1, Ar-H), 4.10 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{SO}_{2} \mathrm{NHCH}$ ), 2.43 (1H, m, $\mathrm{CHCH}_{2} \mathrm{CHHS}$ ), $2.32\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}_{2} \mathrm{CHHS}\right), 1.90\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 1.86(2 \mathrm{H}$, $\mathrm{m}, \mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{~S}$ ); $\delta \mathrm{c}\left(300 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d}_{6}\right) 170.8$ (C=ONHNH), 151.0 (ipso-ArC), 149.3 (ipso-Ar-C), 145.6 (ipso-Ar-C), 134.9 (Ar-C), 133.9 (ipso-Ar-C), 131.2 (Ar-C), 130.8 (Ar-C), 128.4 (ipso-Ar-C), 125.7 (ipso-Ar-C), 118.5 (Ar-C), 111.9 (Ar-C), 111.1 (Ar-C), $54.3(\mathrm{SNHCH}), 32.0\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 29.7\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right)$, $14.8\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 490\left(\left[{ }^{35,35} \mathrm{CI}\right] \mathrm{MH}^{+}\right), 492\left(\left[{ }^{35,37} \mathrm{CI}\right] \mathrm{MH}^{+}\right), 494$ $\left(\left[{ }^{37,37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right)$; HRMS $\left(E S^{+}\right)$Found $\left[{ }^{35,35} \mathrm{Cl}\right] \mathrm{MH}^{+}$, $489.9699\left(\mathrm{C}_{17} \mathrm{H}_{18}{ }^{35,35} \mathrm{Cl}_{2} \mathrm{~N}_{5} \mathrm{O}_{4} \mathrm{~S}_{2}\right.$ requires 490.0172).

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(S)-7-chloro-N-(3-methyl-1-oxo-1-(2-phenylhydrazineyl)butan-2-yl)benzo
[c][1,2,5]oxadiazole-4-sulfonamide 65Aa


Following the general procedure outlined, tert-Butyl (S)-(3-methyl-1-oxo-1-(2-phenylhydrazineyl)butan-2-yl)carbamate 67c (0.12 g, 0.39 mmol ) was transformed following precipitation (EtOAc / n-hexane) and then the flash chromatography (DCM/EtOH/NH3 200:6:1) into the title compound which was isolated as a light brown solid ( $0.13 \mathrm{~g}, 72 \%$ ); Rf 0.39 (DCM/EtOH/NH3 200:6:1); m.p. 212-215 ${ }^{\circ} \mathrm{C}$; $v_{\max } 1672,1603$ (C=O), 1524, 1496, 1340, 1161, 1041, 952 $\mathrm{cm}^{-1}$; бн $^{(300 \mathrm{MHz}, ~ D M S O-d 6)} 9.79$ (1H, bs, CONHNH), 8.67 (1H, bd, J 9, SO2NHCH), 8.06 (1H, d, J 7, Ar-H), 7.89 (1H, d, J 7, Ar-H), $7.06-7.01$ (2H, t, J 7, Ar-H), 6.69 (1H, t, J 7, Ar-H), 6.47 - 6.45 (2H, d, J 7, Ar-H), 3.88 (1H, dd, J 7, 2, $\mathrm{SO}_{2} \mathrm{NHCH}$ ), $2.06\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 0.91\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 0.81$ (3H, d, J 6, CHCH ( $\left.\mathrm{CH}_{3}\right)_{2}$ ); $\delta \mathrm{c}(300 \mathrm{MHz}$, DMSO-d6) 170.2 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 149.4 (ipso-Ar-C), 145.5 (ipso-Ar-C), 134.6 (Ar-C), 131.3 (Ar-C), 128.9 (Ar-C), 125.6 (ipso-Ar-C), 118.9 (Ar-C), 112.4 (Ar-C), 61.2 (SNHCH), $31.3\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right) 2\right), 19.6$ $\left.\left.\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 18.5\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 424\left({ }^{[35} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 426\left({ }^{37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\left[{ }^{35} \mathrm{CI}\right] \mathrm{MH}^{+}, 424.0449\left(\mathrm{C}_{17} \mathrm{H}_{19}{ }^{35} \mathrm{CIN}_{5} \mathrm{O}_{4} \mathrm{~S}\right.$ requires 424.0841).

## (S)-7-chloro-N-(3-methyl-1-oxo-1-(2-(4-(trifluoromethyl)phenyl)hydrazineyl) butan-2-yl)benzo[c][1,2,5]oxadiazole-4-sulfonamide 65Ab



Following the general procedure outlined, tert-Butyl (S)-(3-methyl-1-oxo-1-(2-(4-(trifluoromethyl)phenyl)hydrazinyl)butan-2-yl)carbamate 67Bm (0.22 g, 0.59

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mmol) was transformed following precipitation (EtOAc / n-hexane) and then the flash chromatography (DCM/EtOH/NH3 200:6:1) into the title compound which was isolated as a yellow oil ( $0.19 \mathrm{~g}, 61 \%$ ); Rf 0.29 (DCM/EtOH/NH3 200:6:1); $v_{\text {max }}$ 3248 (N-H), 2969 (C-H), 1679, 1611 (C=O), 1507, 1342, 1253, 1152, 1025, 951 $\mathrm{cm}^{-1}$; $\delta н\left(300 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d}_{6}\right) 9.86$ (1H, bd, J 2, CONHNH), 8.69 (1H, bs, NHAr), 8.06 (1H, d, J 7, Ar-H), 7.91 (1H, d, J 7, Ar-H), 7.06 - 7.03 (2H, d, J 8, ArH), $6.54-6.51(2 H, d, J 8, A r-H), 3.84(1 H, d, J 7, S O 2 N H C H), 2.04(1 H, ~ m$, $\left.\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 0.89\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 0.78\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ;$ סc (300 MHz, DMSO-d6) 170.3 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 149.1 (ipso-Ar-C), 148.6 (ipso-Ar-C), 145.5 (ipso-Ar-C), 140.8 (ipso-Ar-C), 134.6 (Ar-C), 131.3 (Ar-C), 128.9 (ipso-ArC), 125.5 (ipso-Ar-C), 122.2 (Ar-C), 112.9 (Ar-C), 61.2 (SNHCH), 31.2 $\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 19.6\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 18.5\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta_{F}(300 \mathrm{MHz}$, DMSOd6) -57.3 (CF3); m/z (ES ${ }^{+}$) $509\left(\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MNH}_{4}{ }^{+}\right), 511$ ([ $\left.{ }^{37} \mathrm{Cl}\right] \mathrm{MNH}_{4}{ }^{+}$); HRMS (ES ${ }^{+}$) Found $\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MNH}_{4}{ }^{+}, 508.0201\left(\mathrm{C}_{18} \mathrm{H}_{21}{ }^{35} \mathrm{CIF}_{3} \mathrm{~N}_{6} \mathrm{O}_{4} \mathrm{~S}\right.$ requires 509.0981).

## (S)-7-chloro-N-(1-(2-(2-chlorophenyl)hydrazineyl)-3-methyl-1-oxobutan-2-yl) benzo[c][1,2,5]oxadiazole-4-sulfonamide 65Ac



Following the general procedure outlined, tert-Butyl (S)-(1-(2-(2-chlorophenyl)hydrazinyl)-3-methyl-1-oxobutan-2-yl)carbamate 67Ai (0.12 g, 0.35 mmol ) was transformed following precipitation (EtOAc / n-hexane) and then the flash chromatography (DCM/EtOH/NH3 200:6:1) into the title compound which was isolated as a brown gummy solid ( $0.12 \mathrm{~g}, 72 \%$ ); Rf 0.31 (DCM/EtOH/NH3 200:6:1); $V_{\max } 3276$ (N-H), 2962 (C-H), 1667 (C=O), 1522, 1442, 1344, 1159, 1037, $953 \mathrm{~cm}^{-1}$; ठн ( 300 MHz , DMSO-d6) 9.98 (1H, bs, CONHNH), 8.72 (1H, bd, J 9, SO2NHCH), 8.07 (1H, d, J 7, Ar-H), 7.92 (1H, d, J 7, Ar-H), 7.27 (1H, dd, J 7, 1, Ar-H), 7.03 (1H, td, J 8, 1, Ar-H), 6.77 (1H, td, J 8, 1, Ar-H), 6.40 (1H, dd, J 8, 1, Ar-H), $3.86\left(1 \mathrm{H}, \mathrm{dd}, \mathrm{J} 9,2, \mathrm{SO}_{2} \mathrm{NHCH}\right), 2.05\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 0.91(3 \mathrm{H}$,

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d, J 6, $\left.\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 0.79\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right)$; $\delta \mathrm{c}(300 \mathrm{MHz}$, DMSO-d6) 170.2 (C=ONHNH), 149.1 (ipso-Ar-C), 144.7 (ipso-Ar-C), 134.7 (Ar-C), 131.3 (ArC), 129.6 (ipso-Ar-C), 128.9 (Ar-C), 127.9 (Ar-C), 125.5 (ipso-Ar-C), 120.2 (Ar-C), 117.7 (ipso-Ar-C), 112.9 (Ar-C), 61.2 (SNHCH), $31.2\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 19.6$ $\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $18.5\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \quad \mathrm{m} / \mathrm{z} \quad\left(\mathrm{ES}^{+}\right) 458\left(\left[{ }^{35,35} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 460$ ( $\left.{ }^{35,37} \mathrm{Cl}\right] \mathrm{MH}^{+}$), 462 ( $\left.{ }^{37,37} \mathrm{Cl}\right] \mathrm{MH}^{+}$); HRMS (ES ${ }^{+}$) Found $\left[{ }^{35,35} \mathrm{Cl}^{3} \mathrm{MH}^{+}, 458.0037\right.$ $\left(\mathrm{C}_{17} \mathrm{H}_{18}{ }^{35,35} \mathrm{Cl}_{2} \mathrm{~N}_{5} \mathrm{O}_{4} \mathrm{~S}\right.$ requires 458.0452 ).

## (S)-7-chloro-N-(1-(2-(3-chlorophenyl)hydrazineyl)-3-methyl-1-oxobutan-2-yl) benzo[c][1,2,5]oxadiazole-4-sulfonamide 65Ad



Following the general procedure outlined, tert-Butyl (S)-(1-(2-(3-chlorophenyl)hydrazinyl)-3-methyl-1-oxobutan-2-yl)carbamate 67s (0.12 g, 0.35 mmol ) was transformed following precipitation (EtOAc / n-hexane) and then the flash chromatography (DCM/EtOH/NH3 200:6:1) into the title compound which was isolated as a brown gummy solid ( $0.11 \mathrm{~g}, 66 \%$ ); Rf 0.31 (DCM/EtOH/NH3 200:6:1); $v_{\max } 3368,3328$ (N-H), 1677, 1601 (C=O), 1524, 1485, 1332, 1164, 1039, $952 \mathrm{~cm}^{-1}$; ठн ( 300 MHz , DMSO-d6) 9.88 (1H, bs, CONHNH), 8.74 (1H, bd, J 8, SO2NHCH), 8.05 (1H, d, J 7, Ar-H), 7.90 (1H, d, J 7, Ar-H), 7.10 (1H, t, J 8, Ar-H), 6.72 (1H, dd, J7, 1, Ar-H), 6.63 (1H, t, J 2, Ar-H), 6.47 (1H, dd, J 7, 1, ArH), 3.86 (1H, dd, J7, 1, $\mathrm{SO}_{2} \mathrm{NHCH}$ ), $2.04\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 0.89(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6$, $\left.\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 0.76\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right)$; $\delta \mathrm{c}\left(300 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d}_{6}\right) 170.4$ (C=ONHNH), 150.1 (ipso-Ar-C), 149.1 (ipso-Ar-C), 145.5 (ipso-Ar-C), 134.5 (ArC), 133.9 (ipso-Ar-C), 131.2 (Ar-C), 130.7 (Ar-C), 128.9 (ipso-Ar-C), 125.5 (ipso-Ar-C), 118.4 (Ar-C), 111.9 (Ar-C), 110.9 (Ar-C), 61.2 (SNHCH), 31.2 $\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $19.6\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $18.5\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 458$ $\left.\left.\left({ }^{35,35} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 460\left({ }^{35,37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 462$ ( $\left.{ }^{37,37} \mathrm{Cl}\right] \mathrm{MH}^{+}$); HRMS (ES ${ }^{+}$) Found $\left[{ }^{35,35} \mathrm{Cl}\right] \mathrm{MH}^{+}, 458.0037\left(\mathrm{C}_{17} \mathrm{H}_{18}{ }^{35,35} \mathrm{Cl}_{2} \mathrm{~N}_{5} \mathrm{O}_{4} \mathrm{~S}\right.$ requires 458.0452) .

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( $R$ )-7-chloro-N-(1-oxo-3-phenyl-1-(2-phenylhydrazineyl)propan-2-yl)benzo [c][1,2,5]oxadiazole-4-sulfonamide 65Ae


Following the general procedure outlined, tert-Butyl (R)-(1-oxo-3-phenyl-1-(2-phenylhydrazineyl)propan-2-yl)carbamate 67p ( $0.22 \mathrm{~g}, 0.62 \mathrm{mmol}$ ) was transformed following precipitation (EtOAc / n-hexane) and then the flash chromatography (DCM/EtOH/NH3 200:6:1) into the title compound which was isolated as a yellow solid ( $0.19 \mathrm{~g}, 61$ \%); Rf 0.27 (DCM/EtOH/NH3 200:6:1); $v_{\text {max }}$ 3385 (N-H), 1675 (C=O), 1599, 1496, 1343, 1161, 1045, $955 \mathrm{~cm}^{-1}$; ठн ( 300 MHz , DMSO-d6) 9.96 (1H, bd, J 2, CONHNH), 7.79 (1H, d, J 7, Ar-H), 7.77 (1H, bs, $\left.\mathrm{SO}_{2} \mathrm{NHCH}\right), 7.74(1 \mathrm{H}, \mathrm{d}, \mathrm{J} 7, \mathrm{Ar}-\mathrm{H}), 7.15-7.09(2 \mathrm{H}, \mathrm{t}, J 7, \mathrm{Ar}-\mathrm{H}), 7.02-6.99(2 \mathrm{H}$, dd, J 7, 3, Ar-H), $6.84-6.82$ (2H, dd, J 5, 1, Ar-H), 6.68 (1H, t, J 7, Ar-H), 6.66 - 6.64 (2H, d, J 7, Ar-H), 4.18 (1H, dd, J 10, 4, SO2NHCH), 2.91 (1H, dd, J 13, 3, CHCHH-Ar), 2.74 (1H, dd, J 13, 2, CHCHH-Ar); סc (300 MHz, DMSO-d6) 170.8 (C=ONHNH), 149.5 (ipso-Ar-C), 142.4 (ipso-Ar-C), 137.1 (ipso-Ar-C), 133.9 (ArC), 130.8 ( $\mathrm{Ar}-\mathrm{C}$ ), 129.5 ( $\mathrm{Ar}-\mathrm{C}$ ), 129.2 ( $\mathrm{Ar}-\mathrm{C}$ ), 127.6 ( $\mathrm{Ar}-\mathrm{C}$ ), 126.2 ( $\mathrm{Ar}-\mathrm{C}$ ), 125.5 (Ar-C), 119.0 (Ar-C), 112.6 (Ar-C), 57.6 (SNHCH), 38.2 ( $\mathrm{CHCH}_{2}-\mathrm{Ar}$ ); m/z (ES ${ }^{+}$) $472\left(\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 474\left(\left[{ }^{37} \mathrm{CI}\right] \mathrm{MH}^{+}\right) ;$HRMS (ES $\left.{ }^{+}\right)$Found $\left.{ }^{[35} \mathrm{Cl}^{2}\right] \mathrm{MH}^{+}, 472.0429$ $\left(\mathrm{C}_{21} \mathrm{H}_{19}{ }^{35} \mathrm{CIN}_{5} \mathrm{O}_{4} \mathrm{~S}\right.$ requires 472.0841).

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(S)-7-chloro-N-(1-(2-(3-chlorophenyl)hydrazineyl)-1-oxo-3-phenylpropan-2-yl)benzo[c][1,2,5]oxadiazole-4-sulfonamide 65Af


Following the general procedure outlined, tert-Butyl (S)-(1-(2-(3-chlorophenyl)hydrazinyl)-1-oxo-3-phenylpropan-2-yl)carbamate 67x (0.18 g, 0.46 mmol ) was transformed following precipitation (EtOAc / $n$-hexane) and then the flash chromatography (DCM/EtOH/ $\mathrm{NH}_{3}$ 200:6:1) into the title compound which was isolated as a brown gummy solid ( $0.16 \mathrm{~g}, 68 \%$ ); Rf 0.30 (DCM/EtOH/NH3 200:6:1); $v_{\max } 3283$ (N-H), 1683 (C=O), 1596, 1523, 1157, 1079, $968 \mathrm{~cm}^{-1}$; ठн (300 MHz, DMSO-d6) 10.07 (1H, bd, J 2, CONHNH), 9.09 (1H, bd, J 8, SO2NHCH), 8.14 (1H, bd, J 2, Ar-NH), 7.80 (1H, d, J 7, Ar-H), 7.74 (1H, d, J 7, Ar-H), 7.16 (1H, t, J 8, Ar-H), 7.00 - 6.97 (2H, dd, J 7, 2, Ar-H), 6.82 - 6.80 (2H, dd, J 5, 1, Ar-H), 6.75 - 6.72 (2H, m, Ar-H), 6.62 (1H, m, Ar-H), 4.17 (1H, m, $\mathrm{SO}_{2} \mathrm{NHCH}$ ), 2.91 (1H, dd, J 13, 3, CHCHH-Ar), 2.73 (1H, dd, J 13, 2, CHCHHAr); ठc (300 MHz, DMSO-d6) 170.9 (C=ONHNH), 151.1 (ipso-Ar-C), 148.9 (ipso-Ar-C), 144.7 (ipso-Ar-C), 136.9 (ipso-Ar-C), 133.9 (Ar-C), 130.7 (Ar-C), 129.5 (ArC), 128.2 (Ar-C), 127.5 (Ar-C), 126.2 (Ar-C), 125.6 (Ar-C), 118.5 (Ar-C), 111.9 (Ar-C), 111.1 (Ar-C), 57.5 (SNHCH), $37.9\left(\mathrm{CHCH}_{2}-\mathrm{Ar}\right) ; \mathrm{m} / \mathrm{z}\left(\mathrm{ES}^{+}\right) 506$ $\left.\left(\left[{ }^{35,35} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 508\left(\left[{ }^{35,37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 510\left({ }^{37,37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right)$; HRMS (ES $)$Found $\left.{ }^{[35,35} \mathrm{Cl}\right] \mathrm{MH}^{+}, 506.0003\left(\mathrm{C}_{21} \mathrm{H}_{18}{ }^{35,35} \mathrm{Cl}_{2} \mathrm{~N}_{5} \mathrm{O}_{4} \mathrm{~S}\right.$ requires 506.0457).

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(R)-7-chloro-N-(1-(2-(3-chlorophenyl)hydrazineyl)-1-oxo-3-phenylpropan-2-yl)benzo[c][1,2,5]oxadiazole-4-sulfonamide 65Ag


Following the general procedure outlined, tert-Butyl (R)-(1-(2-(3-chlorophenyl)hydrazinyl)-1-oxo-3-phenylpropan-2-yl)carbamate 67Af (0.18 g, 0.46 mmol ) was transformed following precipitation (EtOAc / $n$-hexane) and then the flash chromatography (DCM/EtOH/ $\mathrm{NH}_{3}$ 200:6:1) into the title compound which was isolated as a brown gummy solid ( $0.15 \mathrm{~g}, 64 \%$ ); Rf 0.32 (DCM/EtOH/NH3 200:6:1); $v_{\max } 3265,3029(\mathrm{~N}-\mathrm{H}), 1672$ (C=O), 1597, 1477, 1155, $1042 \mathrm{~cm}^{-1}$; ठн (300 MHz, DMSO-d6) 10.07 (1H, bd, J2, CONHNH), 8.14 (1H, bd, J 2, Ar-NH), 7.80 (1H, d, J 7, Ar-H), 7.74 (1H, d, J 7, Ar-H), 7.33 (1H, dd, J 4, 2, Ar-H), 7.17 (1H, t, J 8, Ar-H), $7.00-6.97$ (2H, dd, J 7, 2, Ar-H), $6.82-6.80$ (2H, dd, J 5, 1, Ar-H), 6.75 - 6.72 (2H, m, Ar-H), 6.62 (1H, m, Ar-H), 4.17 (1H, dd, J 10, 4, SO2NHCH), 2.90 (1H, dd, J 13, 3, CHCHH-Ar), 2.73 (1H, dd, J 13, 2, CHCHH-Ar); סc (300 MHz, DMSO-d6) 170.9 (C=ONHNH), 151.1 (ipso-Ar-C), 148.9 (ipso-Ar-C), 144.7 (ipso-Ar-C), 136.9 (ipso-Ar-C), 133.9 (Ar-C), 130.7 (ArC), 129.5 ( $\mathrm{Ar}-\mathrm{C}$ ), 128.2 ( $\mathrm{Ar}-\mathrm{C}$ ), 127.5 ( $\mathrm{Ar}-\mathrm{C}$ ), 126.2 ( $\mathrm{Ar}-\mathrm{C}$ ), 125.6 ( $\mathrm{Ar}-\mathrm{C}$ ), 118.5 (Ar-C), 111.9 (Ar-C), 111.1 (Ar-C), 57.5 (SNHCH), 37.9 ( $\mathrm{CHCH}_{2}-\mathrm{Ar}$ ); m/z (ES ${ }^{+}$) $506\left(\left[{ }^{35,35} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 508\left(\left[{ }^{35,37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 510\left(\left[{ }^{37,37} \mathrm{C}\right] \mathrm{MH}^{+}\right)$; HRMS (ES $\left.{ }^{+}\right)$Found $\left.{ }^{[35,35} \mathrm{Cl}\right] \mathrm{MH}^{+}, 506.0009\left(\mathrm{C}_{21} \mathrm{H}_{18}{ }^{35,35} \mathrm{Cl}_{2} \mathrm{~N}_{5} \mathrm{O}_{4} \mathrm{~S}\right.$ requires 506.0457).

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(S)-7-chloro-N-(1-(2-(3-chlorophenyl)hydrazineyl)-1-oxo-4-phenylbutan-2-yl) benzo[c][1,2,5]oxadiazole-4-sulfonamide 65Ah


Following the general procedure outlined, tert-Butyl (S)-(1-(2-(3-chlorophenyl)hydrazinyl)-1-oxo-4-phenylbutan-2-yl)carbamate 67Ac (0.22 g, 0.54 mmol ) was transformed following precipitation (EtOAc / $n$-hexane) and then the flash chromatography (DCM/EtOH/ $\mathrm{NH}_{3}$ 200:6:1) into the title compound which was isolated as a brown gummy solid ( $0.14 \mathrm{~g}, 45 \%$ ); Rf 0.38 (DCM/EtOH/NH3 200:6:1); $v_{\max } 3328,3075(\mathrm{~N}-\mathrm{H}), 1669$ (C=O), 1598, 1454, 1333, 1156, 1040, $947 \mathrm{~cm}^{-1}$; ठн ( $300 \mathrm{MHz}, \mathrm{DMSO}_{6}$ ) 9.96 (1H, bd, J 2, CONHNH), 8.03 (1H, d, J7, Ar-H), 7.97 (1H, bd, J 2, SO2NHCH), 7.83(1H, d, J 7, Ar-H), 7.24 - 7.22 (2H, d, J 7, Ar-H), 7.18 - 7.13 (2H, m, Ar-H), 7.08 - 7.05 (2H, m, Ar-H), 6.73 - 6.70 (2H, m, Ar-H), 6.67 (1H, t, J 2, Ar-H), 6.55 (1H, dd, J 7, 1, Ar-H), 4.12 (1H, m, SO2NHCH), 2.69 (2H, m, CH2CH2-Ar), 1.98 (2H, m, CH2CH2-Ar); סc (300 MHz, DMSO-d6) 170.8 (C=ONHNH), 151.1 (ipso-Ar-C), 149.2 (ipso-Ar-C), 145.5 (ipso-Ar-C), 141.1 (ipso-Ar-C), 134.5 (Ar-C), 133.9 (Ar-C), 131.2 (Ar-C), 130.8 (Ar-C), 128.8 (Ar-C), 128.5 (Ar-C), 126.4 (Ar-C), 125.6 (Ar-C), 118.5 (Ar-C), 111.9 ( $\mathrm{Ar}-\mathrm{C}$ ), 111.1 ( $\mathrm{Ar}-\mathrm{C}$ ), 55.4 ( SNHCH ), $34.7\left(\mathrm{CH}_{2} \mathrm{CH}_{2}-\mathrm{Ar}\right), 31.6\left(\mathrm{CH}_{2} \mathrm{CH}_{2}-\mathrm{Ar}\right) ; ~ m / z$ (ES ${ }^{+}$) $520\left(\left[{ }^{35,35} \mathrm{CI}\right] \mathrm{MH}^{+}\right), 523\left(\left[{ }^{35,37} \mathrm{C}\right] \mathrm{MH}^{+}\right), 525\left(\left[{ }^{37,37} \mathrm{CI}\right] \mathrm{MH}^{+}\right)$; HRMS (ES $\left.{ }^{+}\right)$ Found $\left[{ }^{35,35} \mathrm{Cl}\right] \mathrm{MH}^{+}, 520.0156\left(\mathrm{C}_{22} \mathrm{H}_{20}{ }^{35,35} \mathrm{Cl}_{2} \mathrm{~N}_{5} \mathrm{O}_{4} \mathrm{~S}\right.$ requires 520.0613).

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(S)-7-chloro-N-(4-methyl-1-oxo-1-(2-(4-(trifluoromethyl)phenyl)hydrazineyl) pentan-2-yl)benzo[c][1,2,5]oxadiazole-4-sulfonamide 65Ai


Following the general procedure outlined, tert-Butyl (S)-(4-methyl-1-oxo-1-(2-(4-(trifluoromethyl)phenyl)hydrazineyl)pentan-2-yl)carbamate 67Bn ( $0.22 \mathrm{~g}, 0.57$ mmol ) was transformed following precipitation (EtOAc / n-hexane) and then the flash chromatography (DCM/EtOH/ $\mathrm{NH}_{3}$ 200:6:1) into the title compound which was isolated as a brown oil ( $0.18 \mathrm{~g}, 59 \%$ ); Rf 0.27 (DCM/EtOH/NH 3 200:6:1); $v_{\text {max }}$ 2961 (C-H), 1682, 1617 (C=O), 1524, 1323, 1156, 1110, 1065, $950 \mathrm{~cm}^{-1}$; סн (300 MHz, DMSO-d6) 9.88 (1H, bd, J2, CONHNH), 8.84 (1H, bd, J8, SO2NHCH), 8.26 (1H, bd, J 2, Ar-NH), 8.06 (1H, d, J 7, Ar-H), 7.89 (1H, d, J 7, Ar-H), 7.42 - 7.40 (2H, d, J 8, Ar-H), 6.62 - 6.60 (2H, d, J 8, Ar-H), 4.09 (1H, m, SO2NHCH), $1.57-$ $1.48\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CHHCH}\left(\mathrm{CH}_{3}\right)_{2}, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.45\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 0.87$ (3H, d, J 6, CH2CH(CH3 2) , $0.76\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$; $\delta \mathrm{c}(300 \mathrm{MHz}$, DMSOd6) 171.2 (C=ONHNH), 152.5 (ipso-Ar-C), 149.1 (ipso-Ar-C), 145.5 (ipso-Ar-C), 134.5 (Ar-C), 131.3 (Ar-C), 128.8 (ipso-Ar-C), 126.5 (Ar-C), 125.5 (ipso-Ar-C), 111.7 (Ar-C), 55.4 (SNHCH), $41.7\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 24.4\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 23.2$ $\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $21.4\left(\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; ~ \delta \mathrm{~F}\left(300 \mathrm{MHz}\right.$, DMSO-d6) -59.4 $\left(\mathrm{CF}_{3}\right) ; m / z$ (ES $\left.\left.\left.{ }^{+}\right) 506\left({ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 508\left({ }^{37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right)$; HRMS (ES $\left.{ }^{+}\right)$Found $\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}, 506.0444$ $\left(\mathrm{C}_{19} \mathrm{H}_{20}{ }^{35} \mathrm{CIF}_{3} \mathrm{~N}_{5} \mathrm{O}_{4} \mathrm{~S}\right.$ requires 506.0877).

## Chapter 6: Experimental section

## 7-chloro-N-((3S)-3-methyl-1-oxo-1-(2-phenylhydrazineyl)pentan-2-yl)benzo

 [c][1,2,5]oxadiazole-4-sulfonamide 65Aj

Following the general procedure outlined, tert-Butyl ((2S,3S)-3-methyl-1-oxo-1-(2-phenylhydrazineyl)pentan-2-yl)carbamate 67e ( $0.22 \mathrm{~g}, 0.68 \mathrm{mmol}$ ) was transformed following precipitation (EtOAc / n-hexane) and then the flash chromatography (DCM/EtOH/NH3 200:6:1) into the title compound which was isolated as a brown oil ( $0.23 \mathrm{~g}, 68 \%$ ); Rf 0.28 (DCM/EtOH/NH3 200:6:1); $v_{\text {max }}$ 3272 (N-H), 2968 (C-H), 1668, 1603 (C=O), 1524, 1496, 1338, 1159, 1041, 951 $\mathrm{cm}^{-1}$; ठн (300 MHz, DMSO-d6) 9.78 (1H, bs, CONHNH), 8.70 (1H, bd, J 9, SO2NHCH), 8.05 (1H, d, J7, Ar-H), 7.88 (1H, d, J7, Ar-H), 7.60 (1H, bs, Ar-NH), 7.06 - 7.00 (2H, dd, J 7, 1, Ar-H), 6.69 (1H, t, J 7, Ar-H), $6.47-6.45$ (2H, d, J 7, Ar-H), 3.88 (1H, dd, J 7, 1, SO2NHCH), 1.82 (1H, m, NHCHCH), 1.45 (1H, m, $\left.\mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CHHCH}_{3}\right), 1.09\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CHHCH}_{3}\right), 0.87(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6$, $\left.\mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 0.76\left(3 \mathrm{H}, \mathrm{t}, \mathrm{J} 7, \mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$; $\delta c\left(300 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d}_{6}\right)$ 170.1 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 149.4 (ipso-Ar-C), 145.5 (ipso-Ar-C), 134.6 (Ar-C), 131.3 (ArC), 128.9 (Ar-C), 128.9 (Ar-C), 125.5 (ipso-Ar-C), 118.9 (Ar-C), 112.4 (Ar-C), 59.9 (SNHCH), $37.3(\mathrm{NHCHCH}), 24.3\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 15.7\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, $\left.10.9\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 438\left({ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 440\left({ }^{37} \mathrm{Cl}^{2} \mathrm{MH}^{+}\right)$; HRMS (ES $\left.{ }^{+}\right)$ Found $\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}, 438.1011\left(\mathrm{C}_{18} \mathrm{H}_{21}{ }^{35} \mathrm{CIN}_{5} \mathrm{O}_{4} \mathrm{~S}\right.$ requires 438.1003).

## Chapter 6: Experimental section

## 7-chloro-N-((3S)-1-(2-(3-chlorophenyl)hydrazineyl)-3-methyl-1-oxopentan-2-yl)benzo[c][1,2,5]oxadiazole-4-sulfonamide 65Ak



Following the general procedure outlined, tert-Butyl ((2S,3S)-1-(2-(3-chlorophenyl)hydrazineyl)-3-methyl-1-oxopentan-2-yl)carbamate 67u ( 0.12 g , 0.34 mmol ) was transformed following precipitation (EtOAc / $n$-hexane) and then the flash chromatography (DCM/EtOH/NH3 200:6:1) into the title compound which was isolated as a brown gummy solid ( $0.09 \mathrm{~g}, 54 \%$ ); Rf 0.30 (DCM/EtOH/NH3 200:6:1); $V_{\max } 3265$ (N-H), 2927 (C-H), 1699 (C=O), 1597, 1478, 1302, 1075, 993 cm- ; бн ( $300 \mathrm{MHz}, ~ D M S O-d 6$ ) 9.90 (1H, bd, J 2, CONHNH), 8.05 (1H, d, J 7, Ar-H), 7.95 (1H, bd, J2, SO2NHCH), 7.89 (1H, d, J7, Ar-H), 7.73 (1H, bd, J2, Ar-NH), 7.09 (1H, t, J 8, Ar-H), 6.71 (1H, dd, J 8, 1, Ar-H), 6.63 (1H, t, J 2, Ar-H), 6.48 (1H, dd, J 8, 1, Ar-H), 3.87 (1H, d, J 7, SO2NHCH), 1.81 (1H, m, NHCHCH), $1.40\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CHHCH}_{3}\right), 1.26\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CHHCH}_{3}\right)$, $0.85\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 0.72\left(3 \mathrm{H}, \mathrm{t}, \mathrm{J} 7, \mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$; $\delta \mathrm{c}(300$ MHz, DMSO-d6) 170.4 (C=ONHNH), 151.0 (ipso-Ar-C), 149.1 (ipso-Ar-C), 145.5 (ipso-Ar-C), 134.6 (Ar-C), 133.9 (Ar-C), 131.2 (Ar-C), 130.6 (Ar-C), 128.9 (ipso-Ar-C), 125.6 (ipso-Ar-C), 118.4 (Ar-C), 111.9 (Ar-C), 111.1 (Ar-C), 59.9 (SNHCH), $37.1(\mathrm{NHCHCH}), 24.3\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 15.7\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 10.9$ $\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; \quad m / z\left(\mathrm{ES}^{+}\right) 472\left(\left[{ }^{35,35} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 474\left(\left[{ }^{35,37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 476$ ( $\left.{ }^{37,37} \mathrm{Cl}\right] \mathrm{MH}^{+}$); HRMS (ES ${ }^{+}$) Found $\left[{ }^{35,35} \mathrm{CI}\right] \mathrm{MH}^{+}, 472.0206\left(\mathrm{C}_{18} \mathrm{H}_{20}{ }^{35,35} \mathrm{Cl}_{2} \mathrm{~N}_{5} \mathrm{O}_{4} \mathrm{~S}\right.$ requires 472.0613).

## Chapter 6: Experimental section

### 6.1.2.5 Synthesis of Benzoxa-[2,1,3]-diazole Peptidomimetics -

## General Procedure

Under a nitrogen atmosphere $N$-Boc amino acid hydrazides (1.10 equiv.) were dissolved in 4 M HCl solution in dioxane ( 3 mL ) and stirred at room temperature. After 90 minutes the mixture was evaporated, dried in vacuo and precipitated by using (EtOH / $\mathrm{Et}_{2} \mathrm{O}$ ). The quat was directly used in the next step. the resulting solid (1 equiv.) was suspended in THF ( 3 mL ) and diisopropylethylamine (1.50 equiv.) was added. After stirring for 10 minutes at room temperature, the solution was treated with 7 -chlorobenzoxa-[2,1,3]-diazole-4-sulfonyl chloride (1.10 equiv.) and continued stirring for 4 hours. Reaction mixed with ethyl acetate ( 8 mL ) and distilled water ( 5 mL ). After separation of the two phases the organic layer washed again with distilled water ( $5 \mathrm{~mL} \times 3 \mathrm{~mL}$ ) followed by a wash with sat. aq. $\mathrm{NH}_{4} \mathrm{Cl}(6 \mathrm{~mL})$ then sat. aq. $\mathrm{NaHCO}_{3}(6 \mathrm{~mL})$ followed by brine ( 8 mL ). Organic layer dried over $\mathrm{MgSO}_{4}$, filtered, evaporated and dried in vacuo. The flash chromatography (DCM/EtOH/NH3 [600:8:1], [400:8:1], [200:8:1]) afforded the desired sulfonamides 65.

## (S)-7-chloro-N-(1-(2-(2-chlorophenyl)hydrazinyl)-1-oxopropan-2-

 yl)benzo[c][1,2,5]oxadiazole-4-sulfonamide 65Am

Following the general procedure outlined, tert-Butyl (S)-(1-(2-(2-chlorophenyl)hydrazinyl)-1-oxopropan-2-yl)carbamate 67Ah ( $0.05 \mathrm{~g}, 0.16 \mathrm{mmol}$ ) was transformed following flash chromatography (DCM/EtOH/NH3 400:8:1) into the title compound which was isolated as a golden brown gummy solid ( 0.01 g , 14\%) as a mixture of rotamers [3:1]; Rf 0.23 (DCM/EtOH/NH3 400:8:1); ठн (700 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.17 (1H, bs, CONHNH), 8.01 (1H, d , J 7, Ar-H), 7.51 (1H, d, J 7, Ar-H), 7.25 (1H, dd, J 8, 1, Ar-H), 7.13 (1H, dt, J 8, 1, Ar-H), 6.84 (1H, dt, J 8, 1, Ar-H), 4.38 (1H, m, SO2NHCH), 1.40 (3H, d, J7, $\mathrm{NHCHCH}_{3}$ ); $\delta \mathrm{c}\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$

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171.0 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 148.9 (ipso-Ar-C), 144.9 (ipso-Ar-C), 143.2 (ipso-Ar-C), 133.9 (Ar-C), 129.5 (Ar-C), 129.1 (Ar-C), 128.5 (Ar-C), 127.8 (Ar-C), 121.8 (ArC), 113.6 (Ar-C), $52.1(\mathrm{SNHCH}), 19.7\left(\mathrm{NHCHCH}_{3}\right) ; m / z\left(\mathrm{ES}^{+}\right) 430\left(\left[{ }^{35,35} \mathrm{Cl}^{2} \mathrm{MH}^{+}\right)\right.$, $432\left(\left[{ }^{35,37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 434\left(\left[{ }^{37,37} \mathrm{CI}\right] \mathrm{MH}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\left[{ }^{35,35} \mathrm{Cl}\right] \mathrm{MH}^{+}, 430.0146$ $\left(\mathrm{C}_{15} \mathrm{H}_{14}{ }^{35,35} \mathrm{Cl}_{2} \mathrm{~N}_{5} \mathrm{O}_{4} \mathrm{~S}\right.$ requires 430.0144 ).

## (S)-N-(1-(2-(2-bromophenyl)hydrazinyl)-1-oxopropan-2-yl)-7-chlorobenzo

 [c][1,2,5]oxadiazole-4-sulfonamide 65An

Following the general procedure outlined, tert-Butyl (S)-(1-(2-(2-bromophenyl)hydrazinyl)-1-oxopropan-2-yl)carbamate 67Cf ( $0.10 \mathrm{~g}, 0.28 \mathrm{mmol}$ ) was transformed following flash chromatography (DCM/EtOH/NH3 200:8:1) into the title compound which was isolated as a golden brown oil ( $0.03 \mathrm{~g}, 26 \%$ ) as a mixture of rotamers [3:1]; Rf 0.3 (DCM/EtOH/NH3 200:8:1); $v_{\max } 3292$ (N-H), 1668 $(\mathrm{C}=\mathrm{O}), 1522,1349,1162,947,744,635 \mathrm{~cm}^{-1}$; all data provided for the major rotamer; $\delta_{H}\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.39$ (1H, bs, NH), 8.02 (1H, d, J 7, Ar-H), 7.51 (1H, d, J7, Ar-H), 7.43 (1H, dd, J 7, 1, Ar-H), 7.17 (1H, dt, J 7, 1, Ar-H), 6.80 (2H, m, Ar-H), 6.22 (1H, bd, J 7, Ar-NH), 4.40 (1H, m, SO2NHCH), 1.41 (3H, d, J 7, $\mathrm{NHCHCH}_{3}$ ); $\delta с\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 171.2$ (C=ONHNH), 148.9 (Ar-C), 144. 9 (ArC), 144.1 (ipso-Ar-C), 134.0 (Ar-C), 132.7 (Ar-C), 129.2 (Ar-C), 128.4 (Ar-C), 127.4 (ipso-Ar-C), 122.3 (Ar-C), 113.7 (Ar-C), 109.2 (Ar-C), 52.2 ( $\mathrm{SO}_{2} \mathrm{NHCH}$ ), 19.6 ( $\mathrm{NHCHCH}_{3}$ ); m/z (ES $\left.\left.\left.{ }^{+}\right) 473\left({ }^{79} \mathrm{Br}^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 475\left({ }^{79} \mathrm{Br}^{37} \mathrm{Cl}^{3}\right] \mathrm{MH}^{+}\right), 475$ ( $\left[^{81} \mathrm{Br}^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}$), $477\left({ }^{81} \mathrm{Br}^{37} \mathrm{Cl}\right] \mathrm{MH}^{+}$); HRMS (ES ${ }^{+}$) Found [ $\left.{ }^{79} \mathrm{Br}^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}, 473.9633$ $\left(\mathrm{C}_{15} \mathrm{H}_{14} \mathrm{~N}_{5} \mathrm{O}_{4} \mathrm{~S}^{79} \mathrm{Br}^{35} \mathrm{CI}\right.$ requires 473.9638$)$.

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(S)-N-(1-(2-(tert-butyl)hydrazinyl)-1-oxopropan-2-yl)-7-chlorobenzo[c][1,2,5] oxadiazole-4-sulfonamide 65Ao


Following the general procedure outlined, tert-Butyl (S)-(1-(2-(tert-butyl)hydrazinyl)-1-oxopropan-2-yl)carbamate 89a ( $0.06 \mathrm{~g}, 0.75 \mathrm{mmol}$ ) was transformed following flash chromatography (DCM/EtOH/NH3 200:8:1) into the title compound which was isolated as a light yellow solid ( $0.04 \mathrm{~g}, 50 \%$ ); Rf 0.33 (DCM/EtOH/NH3 200:8:1); m.p. $190-192{ }^{\circ} \mathrm{C}$; $v \max 3364$ (N-H), 2972 (C-H), 1674 (C=O), 1466, 1350, 1171, 946, 745, $634 \mathrm{~cm}-1$; бн ( $^{(700 \mathrm{MHz}, ~ D M S O-d 6) ~}$ 9.23 (1H, bs, CONHNH), 7.70 (1H, bd, , J 8, SO2NHCH), 7.98 (1H, d, J7, Ar-H), 7.90 (1H, d, J 7, Ar-H), 4.10 (1H, m, SO2NHCH), 2.09 ( $\left.9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CNH}\right), 1.22$ (3H, d, J 8, NHCHCH3); ठc (700 MHz, DMSO-d6) 170.0 (C=ONHNH), 149.2 (ipso-Ar-C), 145.5 (ipso-Ar-C), 134.4 (ipso-Ar-C), 131.2 (ipso-Ar-C), 128.9 (Ar-C), 125.5 ( $\mathrm{Ar}-\mathrm{C}$ ), $79.6\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CNH}\right), 51.3(\mathrm{SNHCH}), 27.2\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CNH}\right) 20.2$ $\left.\left.\left(\mathrm{NHCHCH}_{3}\right) ; ~ m / z\left(E S^{+}\right) 376\left({ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 378\left({ }^{[37} \mathrm{CI}\right] \mathrm{MH}^{+}\right) ;$HRMS (ES $\left.{ }^{+}\right)$Found $\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MNa}^{+}, 376.0836\left(\mathrm{C}_{13} \mathrm{H}_{19}{ }^{35} \mathrm{ClN}_{5} \mathrm{O}_{4} \mathrm{~S}\right.$ requires 376.0846).

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### 6.1.2.6 Synthesis of Benzoxa-[2,1,3]-diazole Peptidomimetics -

## General Procedure

Under a nitrogen atmosphere $N$-Boc amino acid hydrazides (1.10 equiv.) were dissolved in 4 M HCl solution in dioxane ( 3 mL ) and stirred at room temperature for 90 minutes. The solvent was removed in vacuo, the resulting solid (1 equiv.) was directly suspended in MeCN ( 5 mL ) and pyridine ( 1.20 equiv.) was added. After stirring for 10 minutes at room temperature, the solution was treated with 7-chlorobenzoxa-[2,1,3]-diazole-4-sulfonyl chloride (1.20 equiv.) and continued stirring for 4 hours. Reaction mixed with ethyl acetate ( 8 mL ) and distilled water ( 5 mL ). After separation of the two phases the organic layer washed again with distilled water ( $5 \mathrm{~mL} \times 3 \mathrm{~mL}$ ) followed by a wash with sat. aq. $\mathrm{NH}_{4} \mathrm{Cl}(6 \mathrm{~mL})$ then sat. aq. $\mathrm{NaHCO}_{3}(6 \mathrm{~mL})$ followed by brine ( 8 mL ). Organic layer dried over $\mathrm{MgSO}_{4}$, filtered, evaporated and dried in vacuo. The recrystalysation (EtOH / $\mathrm{H}_{2} \mathrm{O}$ ) or the flash chromatography (DCM/MeOH [100:5]) afforded the desired sulfonamides 65.
(S)-7-chloro-N-(2-(2-(2,6-dichlorophenyl)hydrazinyl)-2-oxoethyl)benzo [c][1,2,5]oxadiazole-4-sulfonamide 65Ap


Following the general procedure outlined, tert-Butyl (2-(2-(2,6-dichlorophenyl)hydrazinyl)-2-oxoethyl)carbamate 67 As ( $0.05 \mathrm{~g}, 0.15 \mathrm{mmol}$ ) was transformed following recrystalysation $\left(\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}\right)$ into the title compound which was isolated as a light brown gummy solid ( $0.03 \mathrm{~g}, 39 \%$ ) as a mixture of rotamers [6:1]; Rf 0.29 (DCM/EtOH/NH3 400:8:1); $v_{\max } 3344$ (N-H), 2921, 2257 (C-H), 1676 (C=O), 1547, 1428, 1358, 1157, 1023, $996 \mathrm{~cm}^{-1}$; all data provided for the major rotamer $\delta_{\mathrm{H}}(400 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d} 6) 10.01$ (1H, bs, NH), 8.58 (1H, bt, J 6, CH2NH), 7.92 (1H, d, J 7, Ar-H), 7.78 (1H, d, J 7, Ar-H), 7.27 (1H, d, J 8, Ar-H), 6.97-6.93 (2H, m, Ar-H), 3.77 (2H, d, J 6, SO $\mathrm{NHCH}_{2}$ ); $\delta c$ ( $100 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d}_{6}$ ) 172.5 (C=ONHNH), 167.3 (ipso-Ar-C), 145.8 (ipso-Ar-C), 141.3 (ipso-Ar-C), 134.0 (Ar-

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C), 131.1 (Ar-C), 129.4 (Ar-C), 128.8 (Ar-C), 125.3 (ipso-Ar-C), 124.3 (ipso-Ar-C), 123.3 (Ar-C), $43.5\left(\mathrm{SNHCH}_{2}\right) ; m / z\left(\mathrm{ES}^{+}\right) 450\left(\left[{ }^{[35,35,35} \mathrm{Cl}^{2}\right] \mathrm{MH}^{+}\right), 452\left(\left[{ }^{35,35,37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right)$, $454\left(\left[{ }^{35,37,37} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 456$ ( $\left.{ }^{37,37,37} \mathrm{C}\right] \mathrm{MH}^{+}$); HRMS (ES ${ }^{+}$) Found $\left[{ }^{35,35,35} \mathrm{CI}\right] \mathrm{MH}^{+}$, $449.9609\left(\mathrm{C}_{14} \mathrm{H}_{11} \mathrm{~N}_{5} \mathrm{O}_{4} \mathrm{~S}^{35,35,35} \mathrm{Cl}\right.$ requires 449.9597).
(S)-7-chloro-N-(1-(2-(2,4-difluorophenyl)hydrazinyl)-3-methyl-1-oxobutan-2-yl)benzo[c][1,2,5]oxadiazole-4-sulfonamide 65Aq


Following the general procedure outlined, tert-Butyl(S)-(1-(2-(2,4-difluorophenyl)hydrazinyl)-3-methyl-1-oxobutan-2-yl)carbamate 67Ca ( 0.05 g , 0.15 mmol ) was transformed following flash column chromatography (DCM/MeOH 100:5) into the title compound which was isolated as an orange oil (0.04 g, 54\%); Rf 0.70 (DCM/EtOH/NH3 200:8:1); $v_{\max } 3257$ (N-H), 2962, 2928 (C-H), 1702, 1610 (C=O), 1506, 1354, 1268, 1159, 1026, 956 cm$^{-1}$; бн ( 400 MHz , DMSO-d6) 9.93 (1H, bs, NH), 8.69 (1H, bd, J 9, SO2NHCH), 8.05 (1H, d, J 7, ArH), 7.91 (1H, d, J 7, Ar-H), 7.60 (1H, bs, NH), 7.11 (1H, dt, J 8, 2, Ar-H), 6.77 (1H, m, Ar-H), 6.49 (1H, m, Ar-H), $3.82\left(1 \mathrm{H}, \mathrm{m}, \mathrm{SO}_{2} \mathrm{NHCH}\right), 2.01\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $0.87\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 0.74\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6, \mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right)$; $\delta \mathrm{c}(100 \mathrm{MHz}$, DMSO-d6) 172.1 (C=ONHNH), 149.1 (ipso-Ar-C), 145.5 (ipso-Ar-C), 134.6 (ArC), 132.2 (ipso-Ar-C), 132.1 (ipso-Ar-C), 131.3 (Ar-C), 128.9 (ipso-Ar-C), 125.4 (ipso-Ar-C), 114.4 (Ar-C), 111.0 (Ar-C), 104.1 (Ar-C), 61.2 ( $\mathrm{SO}_{2} \mathrm{NHCH}$ ), 31.2 $\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 19.6\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right), 18.5\left(\mathrm{CHCH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta$ ( 282 MHz , DMSOd6) -123.6 (Ar-F), -128.2 (Ar-F); m/z (ES $\left.{ }^{+}\right) 460\left(\left[{ }^{35} \mathrm{CI}\right] \mathrm{MH}^{+}\right) ;$HRMS (ES ${ }^{+}$) Found $\left.{ }^{[35} \mathrm{CI}\right] \mathrm{MH}^{+}, 460.0680\left(\mathrm{C}_{17} \mathrm{H}_{17} \mathrm{~N}_{5} \mathrm{O}_{4} \mathrm{~F}_{2} \mathrm{~S}^{35} \mathrm{CI}\right.$ requires 460.0658).

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(S)-7-chloro-N-(1-(2-(4-cyanophenyl)hydrazineyl)-4-(methylthio)-1-oxobutan-2-yl)benzo[c][1,2,5]oxadiazole-4-sulfonamide 65Ar


Following the general procedure outlined, tert-Butyl (S)-(1-(2-(4-cyanophenyl)hydrazinyl)-4-(methylthio)-1-oxobutan-2-yl)carbamate 67Da ( 0.04 g , 0.11 mmol ) was transformed following recrystalysation ( $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ ) into the title compound which was isolated as a brown gummy solid ( $0.05 \mathrm{~g}, 72 \%$ ); Rf 0.31 (DCM/EtOH/NH $200: 6: 1$ ); $v_{\max } 3305,3088(\mathrm{~N}-\mathrm{H}), 2227$ (C=N), 1679, 1606 (C=O), 1520, 1349, 1167, 1154, 1083, $947 \mathrm{~cm}^{-1}$; ठн ( 300 MHz , DMSO-d6) 10.07 (1H, bs, CONHNH), 8.96 (1H, bd, J 8, SO2NHCH), 8.55 (1H, bs, NH-Ar), 8.06 (1H, d, J 7, Ar-H), 7.91 (1H, d, J 7, Ar-H), $7.53-7.51$ (2H, d, J 8, Ar-H), $6.66-6.64$ (2H, d, $J 8, \operatorname{Ar}-H), 4.09\left(1 \mathrm{H}, \mathrm{m}, \mathrm{SO}_{2} \mathrm{NHCH}\right), 2.42\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}_{2} \mathrm{CHHS}\right), 2.31(1 \mathrm{H}, \mathrm{m}$, $\mathrm{CHCH}_{2} \mathrm{CHHS}$ ), $1.89\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 1.86\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right)$; $\delta \mathrm{c}(300$ MHz, DMSO-d6) 168.8 (C=ONHNH), 150.9 (ipso-Ar-C), 147.1 (ipso-Ar-C), 143.5 (ipso-Ar-C), 132.9 (Ar-C), 131.7 (Ar-C), 129.2 (Ar-C), 126.2 (ipso-Ar-C), 123.6 (ipso-Ar-C), 118.6 (C三N), 109.9 (ipso-Ar-C), 97.4 (ipso-Ar-C), 52.1 (SNHCH), $29.8\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 27.6\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right), 12.6\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{SCH}_{3}\right) ; ~ m / z\left(\mathrm{ES}^{+}\right) 481$ $\left.\left(\left[{ }^{35} \mathrm{Cl}\right] \mathrm{MH}^{+}\right), 483\left({ }^{37} \mathrm{C} \mid\right] \mathrm{MH}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\left[{ }^{35,35} \mathrm{CI}\right] \mathrm{MH}^{+}, 481.0105$ $\left(\mathrm{C}_{18} \mathrm{H}_{18}{ }^{35} \mathrm{CIN}_{6} \mathrm{O}_{4} \mathrm{~S}_{2}\right.$ requires 481.0514 ).

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### 6.1.2.7 Synthesis of Benzoxa-[2,1,3]-diazole Peptidomimetics -

## General Procedure

Under a nitrogen atmosphere $N$-Boc-azido-amino acid hydrazides (1.10 equiv.) were dissolved in $15 \% \mathrm{v} / \mathrm{v} \mathrm{HCl}$ in DCM ( 1.5 mL ) and stirred at room temperature for a hour. The solvent was removed in vacuo, the resulting solid (1 equiv.) was directly suspended in MeCN ( 5 mL ) and pyridine (1.20 equiv.) was added. After stirring for 10 minutes at room temperature, the solution was treated with 7-chlorobenzoxa-[2,1,3]-diazole-4-sulfonyl chloride (1.20 equiv.) and continued stirring for 19 hours. Reaction mixed with ethyl acetate ( 8 mL ) and distilled water ( 5 mL ). After separation of the two phases the organic layer washed again with distilled water ( $5 \mathrm{~mL} \times 3 \mathrm{~mL}$ ) followed by a wash with sat. aq. $\mathrm{NH}_{4} \mathrm{Cl}(6 \mathrm{~mL})$ then sat. aq. $\mathrm{NaHCO}_{3}(6 \mathrm{~mL})$ followed by brine ( 8 mL ). Organic layer dried over $\mathrm{MgSO}_{4}$, filtered, evaporated and dried in vacuo. The recrystalysation (EtOH / H2O) afforded the desired sulfonamides 65 .

## (S)-N-(3-(4-azidophenyl)-1-(2-(3-chlorophenyl)hydrazinyl)-1-oxopropan-2-yl) -7-chlorobenzo[c][1,2,5]oxadiazole-4-sulfonamide 65As



Following the general procedure outlined, tert-Butyl(S)-(3-(4-azidophenyl)-1-(2-(3-chlorophenyl)hydrazinyl)-1-oxopropan-2-yl)carbamate 67En (0.07 g, 0.16 mmol) was transformed following recrystalysation (EtOH / $\mathrm{H}_{2} \mathrm{O}$ ) into the title compound which was isolated as a brown solid ( $0.03 \mathrm{~g}, 35 \%$ ); Rf 0.41 (DCM/EtOH/NH3 200:8:1); $V_{\max }$ 3229, 3076 (N-H), 2925, 2856 (C-H), 2115 ( $\mathrm{N}_{3}$ ), 1677, 1596, 1346, 1288, 1156, 1024, $997 \mathrm{~cm}^{-1}$; бн ( 400 MHz , DMSO-d6) 10.08 (1H, bs, NH), 9.10 (1H, bd, J 8, SO2NHCH), 7.81 - 7.75 (2H, q, J 7, Ar-H), 7.17

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(1H, t, J 8, Ar-H), 6.98 - 6.96 (2H, d, J 8, Ar-H), $6.74-6.72$ (2H, m, Ar-H), 6.67 (1H, d, J 8, Ar-H), $6.46-6.44$ (2H, d, J 8, Ar-H), 4.10 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{SO}_{2} \mathrm{NHCH}$ ), 2.86 ( $1 \mathrm{H}, \mathrm{dd}, \mathrm{J} 13,3, \mathrm{SO}_{2} \mathrm{NHCH}_{2}$ ), $2.68\left(1 \mathrm{H}, \mathrm{t}, J 12, \mathrm{SO}_{2} \mathrm{NHCH}_{2}\right)$; $\delta \mathrm{c}(100 \mathrm{MHz}$, DMSO-d6) 170.9 ( $\mathrm{C}=\mathrm{ONHNH}$ ), 151.1 ( $\mathrm{C}=\mathrm{ONH}$ ), 148.8 (ipso-Ar-C), 144.6 (ipso-Ar-C), 137.5 (ipso-Ar-C), 134.1 (Ar-C), 134.0 (Ar-C), 133.9 (Ar-C), 131.0 (Ar-C), 130.9 (Ar-C), 128.1 (ipso-Ar-C), 125.4 (ipso-Ar-C), 118.6 (Ar-C), 117.9 (Ar-C), 111.9 (Ar-C), 111.2 (Ar-C), 57.6 ( $\mathrm{SO}_{2} \mathrm{NHCH}$ ), 37.1 ( $\mathrm{Ar}-\mathrm{CH}_{2}$ ); m/z (ES $\left.{ }^{+}\right) 547$
 $\left[{ }^{35,35} \mathrm{Cl}\right] \mathrm{MH}^{+}, 547.0463\left(\mathrm{C}_{21} \mathrm{H}_{17} \mathrm{~N}_{8} \mathrm{O}_{4} \mathrm{~S}^{35,35} \mathrm{Cl}_{2}\right.$ requires 547.0435).

### 6.1.2.8 Synthesis of Benzoxa-[2,1,3]-diazole Peptidomimetics -

## General Procedure

Under a nitrogen atmosphere $N$-Fmoc amino acid hydrazides (1.80 equiv.) were reacted with 1-octanethiol (7 equiv.) in THF ( 3 mL ) and stirred at room temperature. After 10 minutes 1,8-diazabicyclo[5.4.0]undec-7-ene (1.10 equiv.) was added and continued stirring for 5 hours. The mixture was evaporated, dried in vacuo and precipitated by using ( $\mathrm{EtOH} / \mathrm{Et}_{2} \mathrm{O}$ ). The resulting solid (1 equiv.) was suspended in THF ( 4 mL ) and $\mathrm{Et}_{3} \mathrm{~N}$ (3 equiv.) was added. After stirring for 10 minutes at room temperature, the solution was treated with 7-chlorobenzoxa-[2,1,3]-diazole-4-sulfonyl chloride (1.10 equiv.) and continued stirring for 5 hours. Reaction mixed with ethyl acetate ( 8 mL ) and distilled water ( 5 mL ). After separation of the two phases the organic layer washed again with distilled water ( $5 \mathrm{~mL} \times 3 \mathrm{~mL}$ ) followed by a wash with sat. aq. $\mathrm{NH}_{4} \mathrm{Cl}(6 \mathrm{~mL})$ then sat. aq. $\mathrm{NaHCO}_{3}(6 \mathrm{~mL})$ followed by brine ( 8 mL ). Organic layer dried over $\mathrm{MgSO}_{4}$, filtered, evaporated and dried in vacuo. The precipitation (EtOAc / n-hexane) and then the flash chromatography (DCM/EtOH/NH3 [600:8:1], [400:8:1], [200:6:1]) afforded the desired sulfonamides 65 .

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## $\mathrm{N}-((2 \mathrm{~S}, 3 \mathrm{~S})$-3-(tert-butoxy)-1-(2-(3-chlorophenyl)hydrazineyl)-1-oxobutan-2-yl)-7-chlorobenzo[c][1,2,5]oxadiazole-4-sulfonamide 65AI



Following the general procedure outlined, (9H-fluoren-9-yl)methyl ((2S,3S)-3-(tert-butoxy)-1-(2-(3-chlorophenyl)hydrazinyl)-1-oxobutan-2-yl)carbamate 68I ( $0.18 \mathrm{~g}, 0.60 \mathrm{mmol}$ ) was transformed following precipitation (EtOAc / $n$-hexane) and then the flash chromatography ( $\mathrm{DCM} / \mathrm{EtOH} / \mathrm{NH}_{3}$ 200:6:1) into the title compound which was isolated as a brown gummy solid ( $0.19 \mathrm{~g}, 61 \%$ ); Rf 0.58 (DCM/EtOH/NH3 200:6:1); $V_{\max } 3276$ (N-H), 2979, 2932 (C-H), 1729, 1688 (C=O), 1599, 1477, 1366, 11189, 1041, $950 \mathrm{~cm}^{-1}$; ठн ( 300 MHz , DMSO-d6) 9.64 (1H, bd, J 2, CONHNH), 8.49 (1H, bd, J 8, SO2NHCH), 8.07 (1H, d, J7, Ar-H), 7.97 (1H, bd, J 2, NH-Ar), 7.93 (1H, d, J 7, Ar-H), 7.09 (1H, t, J 7, Ar-H), 6.78 (1H, t, J 2, Ar-H), 6.70 (1H, dd, J 8, 1, Ar-H), 6.58 (1H, dd, J 8, 1, Ar-H), 4.07 (1H, m, $\mathrm{SO}_{2} \mathrm{NHCH}$ ), $3.85\left(1 \mathrm{H}, \mathrm{m}, \mathrm{SO}_{2} \mathrm{NHCHCH}\right), 1.10\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 1.04(3 \mathrm{H}, \mathrm{d}, \mathrm{J}$ 6, $\mathrm{NHCHCHCH}_{3}$ ); $\delta c(300 \mathrm{MHz}$, DMSO-d6) 168.5 (C=ONHNH), 151.0 (ipso-ArC), 149.2 (ipso-Ar-C), 145.5 (ipso-Ar-C), 134.6 (Ar-C), 133.9 (Ar-C), 131.3 (Ar-C), 130.5 (Ar-C), 128.9 (ipso-Ar-C), 125.6 (ipso-Ar-C), 118.4 (Ar-C), 112.1 (Ar-C), 111.3 (Ar-C), $74.5\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right), 68.6\left(\mathrm{SO}_{2} \mathrm{NHCHCH}\right), 60.2\left(\mathrm{SO}_{2} \mathrm{NHCH}\right), 28.3$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CO}\right) \quad 18.9 \quad\left(\mathrm{NHCHCHCH}_{3}\right) ; \quad \mathrm{m} / \mathrm{z} \quad\left(\mathrm{ES}^{+}\right) \quad 538 \quad\left(\left[{ }^{35,35} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right), 540$ ( $\left.\left[{ }^{35,37} \mathrm{Cl}\right] \mathrm{MNa}^{+}\right), 540\left(\left[{ }^{37,37} \mathrm{CI}\right] \mathrm{MNa}^{+}\right)$; HRMS (ES ${ }^{+}$) Found $\left[{ }^{35,35} \mathrm{CI}\right] \mathrm{MNa}^{+}, 538.0236$ $\left(\mathrm{C}_{20} \mathrm{H}_{23}{ }^{35,35} \mathrm{Cl}_{2} \mathrm{~N}_{5} \mathrm{O}_{5} \mathrm{SNa}\right.$ requires 538.0231).

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### 6.2 General Experimental Information of Biological Assessment

### 6.2.1 Cell Lines

The mycobacterial strains and cell lines used in this study can be found in Table 9.

| Strain | Genotype | Comments | Reference |
| :---: | :---: | :---: | :---: |
| $\begin{gathered} M t b \\ \mathrm{mc}^{27} 7000 \end{gathered}$ | $\Delta R D 1($ esx1):GFP <br> $\triangle p a n C D$ | Deletion of esx1 or RD1 responsible for virulence factor secretion, ESAT-6 and CFP-10 replaced with GFP. Pantothenate auxotroph. | W. Jacobs, <br> Albert Einstein <br> College of <br> Medicine. |
| $\begin{gathered} M t b \\ \mathrm{mc}^{2} 7902 \end{gathered}$ | $\triangle l e u C D \triangle p a n C D \triangle a r g B$ | $\mathrm{mc}^{2} 6206$ derived, arginine auxotroph | $\begin{gathered} \hline \text { Vilcheze et al. } \\ 10.1128 / \mathrm{mbio} .00 \\ 938-18 \end{gathered}$ |
| $\begin{gathered} \text { Mtb } \\ \mathrm{mc}^{2} 8245 \end{gathered}$ | $\Delta p a n C D \Delta l e u C D \Delta a r g B$ <br> $\Delta 2116169-2162530$ | $\mathrm{mc}^{2} 7902$ derived, $\Delta 2116169-$ <br> 2162530; INH-Resistant | $\begin{aligned} & \text { Vilcheze et al. } \\ & \text { 10.1128/mbio. } 00 \\ & 938-18 \end{aligned}$ |
| $\begin{gathered} M t b \\ \mathrm{mc}^{2} 8250 \end{gathered}$ | $\begin{aligned} & \Delta \text { panCD } \Delta \text { leuCD } \Delta \operatorname{argB} r \\ & \text { poB }(\mathrm{H} 445 \mathrm{Y}) \Delta 2122397- \\ & 2170320 \end{aligned}$ | $\mathrm{mc}^{2} 8247 \text { derived, } \Delta 2122397-$ <br> 2170320; INH-Resistant, rpoB His445 --> <br> Lys; RIF resistant | $\begin{gathered} \text { Vilcheze et al. } \\ 10.1128 / \mathrm{mbio} .00 \\ 938-18 \end{gathered}$ |
| $\begin{gathered} \text { Mtb } \\ \mathrm{mc}^{2} 8257 \end{gathered}$ | $\Delta p a n C D \Delta l e u C D \Delta a r g B$ $r p o B(\mathrm{H} 445 \mathrm{Y}) \operatorname{katG}(\mathrm{V} 1$ <br> A) | $\mathrm{mc}^{2} 8247$ derived, $r p o B$ His 445 --> Lys; RIF resistant, kat $G$ Val1 --> Ala; INH- <br> Resistant | $\begin{gathered} \text { Vilcheze et al. } \\ 10.1128 / \mathrm{mbio} .00 \\ 938-18 \end{gathered}$ |
| Mtb <br> AKB7002 | $\begin{gathered} \Delta \text { RD1 }^{(\mathrm{esx} 1): G F P} \\ \Delta p a n C D ; \text { Kan } \end{gathered}$ | As Mtb mc²7000 pMV261, generated spontaneous mutant, rifampicin-resistant, rpoB His $445 \rightarrow$ Lys. | Brown, A.K. et <br> al. <br> Molecules, 25 $\text { (10), p. } 2387$ |

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|  | $\Delta R D 1(e s x 1): G F P$ | As Mtb mc²7000 pMV261, generated | Brown, A.K. et |
| :---: | :---: | :---: | :---: |
| Mtb | AKanCD; Kan ${ }^{R}$ | spontaneous mutant, isoniazid- | al. |
| AKB721 | katG $(\mathrm{S} 315 \mathrm{~N})$ | resistant, katG Ser315 $\rightarrow$ Asp. | Molecules, 25 |
|  |  |  | $(10)$, p.2387 |

### 6.2.2 Bacterial Growth Conditions

Mycobacterial species were cultured in either Middlebrook 7H9 broth or Middlebrook 7H10 agar media supplemented with albumin-dextrose-catalase (ADC) or oleic acid-albumin-dextrose-catalase (OADC) enrichments and 0.2\% glycerol, $0.2 \%$ casamino acids, $24 \mu \mathrm{~g} / \mathrm{mL}$ pantothenate, $1 \mu \mathrm{~g} / \mathrm{mL}$ penicillin $\mathrm{G}, 10$ $\mu \mathrm{g} / \mathrm{mL}$ cyclohexamide, purchased from BD Biosciences. All reagents were purchased from Sigma-Aldrich, Gillingham, UK unless stated otherwise.

### 6.2.3 In vitro bacterial inhibitory assay

Minimum inhibitory concentrations (MICs) of the intermediates 67, 68, 89 and 91 and final compounds 65 on different mycobacterial strains were determined using broth micro-dilution assay in 96-well plates. Bacterial colonies from LJ slants were scraped and re-suspended in Middlebrook 7H9GC medium. Homogeneous mycobacterial suspension was prepared in sterile water, vortexed with glass beads (1-5 mm) for five minutes. After aerosol settlement, the colony-forming unit (CFU) of the suspension was determined by the measurement of optical density (OD) using the SmartSpecTM Plus spectrophotometer (BIO-RAD, CA, USA) at an excitation of 530 nm with an emission of 590 nm . The concentration of stock for each test compound is $10 \mathrm{mg} / \mathrm{mL}$ and then a 2-fold serial dilution was prepared (250 to $0.24 \mu \mathrm{~g} / \mathrm{mL}$ ) in Middlebrook 7H9 medium supplemented with OADC and glycerol as aforementioned. The correct mycobacterial strain suspension was added to an equal volume ( $100 \mu \mathrm{~L}$ ) of test compounds solutions in each well to

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obtain the final concentration of $1 \times 107 \mathrm{CFU} / \mathrm{mL}$ for Mtb strains. The untreated bacterial suspension and sterile medium were used as positive and negative growth controls. The plates were then incubated at $37^{\circ} \mathrm{C}$ for 7 days in an oven incubator. A volume of $20 \mu \mathrm{~L}$ of $0.02 \%$ (w/v) resazurin dye was added to each well at the end of the incubation period. The color change was evaluated after $\sim 48 \mathrm{~h}$ for Mtb strains. Resazurin assay is a colorimetric assay in which the viable cells reduce blue resazurin into fluorescent pink resorufin. MIC of each compound on the susceptible, mono-resistant and double-resistant Mtb strains was determined visually and defined as the lowest concentration remained in blue.

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### 8.1 The biological data for in vitro evaluating antitubercular activity

As observed for in vitro antitubercular screening, the MIC results of the Boc- $N$ hydrazides 67 and Fmoc- $N$-hydrazides 68 Ag-Ar against wild-type Mtb, isoniazid-resistant Mtb strain and rifampicin-resistant Mtb were obtained.

Table A1: The MIC results of the antimycobacterial susceptibility testing of the unsubstituted Boc- $N$-hydrazides 67a-j and Fmoc- $N$-hydrazides 68a-j; where a (-) indicates to a lack of inhibition activity.

|  |  | MIC ( $\mu \mathrm{M}$ ) |  |  |
| :---: | :---: | :---: | :---: | :---: |
| Entry | Structure | $\begin{gathered} \text { WT. } \\ M t b \\ \mathrm{mc}^{2} 7000 \end{gathered}$ | $\begin{gathered} \mathrm{INH} . \\ \text { resistant } \\ M t b \\ \mathrm{mc}^{2} 7021 \end{gathered}$ | $\begin{gathered} \text { RIF. } \\ \text { resistant } \\ M t b \\ \text { mc}^{2} 7002 \end{gathered}$ |
| 1 |  | 241.22 | - | - |
| 2 |  <br> 67b | 114.56 | - | - |
| 3 |  | 13.01 | 208.20 | - |
| 4 |  | 12.44 | 199.12 | - |
| 5 |  | - | 199.12 | - |
| 6 |  <br> $67 f$ | 104.79 | 52.39 | 104.79 |
| 7 |  | 5.89 | 188.54 | - |

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| 8 |  | 2.81 | 90.03 | - |
| :---: | :---: | :---: | :---: | :---: |
| 9 |  | - | - | - |
| 10 |  | 138.66 | 69.33 | 138.66 |
| 11 |  | - | - | - |
| 12 |  | 32.81 | 32.81 | 65.63 |
| 13 |  | 127.60 | 127.60 | 127.60 |
| 14 |  | 31.03 | 31.03 | 31.03 |
| 15 |  | - | - | - |

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| 16 |  | - | - | - |
| :---: | :---: | :---: | :---: | :---: |
| 17 |  | - | 91.32 | - |
| 18 |  | - | 69.78 | 69.78 |
| 19 |  | - | - | - |
| 20 |  | - | - | - |

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Table A2: The MIC results of the antimycobacterial susceptibility testing of the unsubstituted Boc- $N$-hydrazides $\mathbf{6 7 k} \mathbf{k}$; where a (-) indicates to a lack of inhibition activity.

|  |  | MIC ( $\mu \mathrm{M}$ ) |  |  |
| :---: | :---: | :---: | :---: | :---: |
| Entry | Structure | $\begin{gathered} \text { WT. } \\ M t b \\ \mathrm{mc}^{2} 7000 \end{gathered}$ | $\begin{gathered} \text { INH. } \\ \text { resistant } \\ M t b \\ \text { mc}^{2} 7021 \end{gathered}$ | $\begin{gathered} \hline \text { RIF. } \\ \text { resistant } \\ M t b \\ \text { mc}^{2} 7002 \end{gathered}$ |
| 21 |  | - | 289.25 | - |
| 22 |  | 57.28 | - | - |
| 23 |  | 86.61 | 173.22 | 173.22 |
| 24 |  | - | - | - |
| 25 |  | - | - | - |
| 26 |  | 90.03 | - | - |

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Table A3: The MIC results of the antimycobacterial susceptibility testing of the substituted Boc-$N$-hydrazides $\mathbf{6 7} \mathbf{q}$-Af and Fmoc- $N$-hydrazides $\mathbf{6 8 k}$-u; where a (-) indicates to a lack of inhibition activity.

|  |  | MIC ( $\mu \mathrm{M}$ ) |  |  |
| :---: | :---: | :---: | :---: | :---: |
| Entry | Structure | $\begin{gathered} \text { WT. } \\ M t b \\ \mathrm{mc}^{2} 7000 \end{gathered}$ | $\begin{gathered} \hline \text { INH. } \\ \text { resistant } \\ M t b \\ \mathrm{mc}^{2} 7021 \end{gathered}$ | $\begin{gathered} \text { RIF. } \\ \text { resistant } \\ \text { Mtb } \\ \text { mc}^{2} 7002 \end{gathered}$ |
| 27 |  | 13.34 | - | - |
| 28 |  | 3.19 | 12.75 | 101.98 |
| 29 |  | 93.61 | 93.61 | 93.61 |
| 30 |  | 89.92 | 179.85 | 89.92 |
| 31 |  | 44.96 | 44.96 | 89.92 |
| 32 |  | 94.17 | 47.08 | 94.17 |
| 33 |  | 2.67 | 42.79 | - |
| 34 |  | 5.13 | 82.08 | 41.04 |

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| 35 |  | 34.04 | 17.02 | 34.04 |
| :---: | :---: | :---: | :---: | :---: |
| 36 |  | 129.03 | 64.52 | 129.03 |
| 37 |  | 95.13 | 95.13 | 95.13 |
| 38 |  | 85.14 | 42.57 | 85.14 |
| 39 |  | 23.40 | 23.40 | 46.81 |
| 40 |  <br> 67Ad | 101.98 | 101.98 | 101.98 |
| 41 |  | 85.58 | 85.58 | 85.58 |
| 42 |  | - | 82.08 | - |
| 43 |  | 62.99 | 15.75 | 62.99 |

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| 44 |  | 15.32 | 15.32 | 15.32 |
| :---: | :---: | :---: | :---: | :---: |
| 45 |  | 59.70 | 59.70 | 59.70 |
| 46 |  | 58.18 | 29.09 | 29.09 |
| 47 |  | 44.37 | - | - |
| 48 |  | - | 45.05 | - |
| 49 |  | 10.88 | 5.44 | 10.88 |
| 50 |  | 26.98 | - | 13.49 |
| 51 |  | - | - | - |
| 52 |  | 82.76 | 82.76 | 82.76 |

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|  | 68t |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
| 53 |  | 32.26 | 129.03 | 32.26 |

Table A4: The MIC results of the antimycobacterial susceptibility testing of the chlorinated BocN -hydrazides 67Ag-Ar at different positions on aryl hydrazines; where a (-) indicates to a lack of inhibition activity.

|  |  | MIC ( $\mu \mathrm{M}$ ) |  |  |
| :---: | :---: | :---: | :---: | :---: |
| Entry | Structure | $\begin{gathered} \text { WT. } \\ M t b \\ \mathrm{mc}^{2} 7000 \end{gathered}$ | $\begin{gathered} \text { INH. } \\ \text { resistant } \\ M t b \\ \mathrm{mc}^{2} 7021 \end{gathered}$ | $\begin{gathered} \text { RIF. } \\ \text { resistant } \\ M t b \\ \text { mc}^{2} 7002 \end{gathered}$ |
| 54 |  <br> 67 Ag | 106.75 | 53.38 | - |
| 55 |  | 50.99 | 101.98 | - |
| 56 |  <br> 67Ai | 187.22 | 187.22 | 187.22 |
| 57 |  | 179.85 | 179.85 | 179.85 |
| 58 |  | 21.40 | 171.17 | 171.17 |
| 59 |  | 164.15 | 164.15 | 164.15 |
| 60 |  | 213.51 | 13.34 | 53.38 |

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|  | 67Am |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
| 61 |  | 1.59 | 50.99 | 25.50 |
| 62 |  | 93.61 | 46.81 | 93.61 |
| 63 |  | 89.92 | 179.85 | 89.92 |
| 64 |  | 21.40 | 21.40 | 85.58 |
| 65 |  | 41.04 | 41.04 | 82.08 |

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Table A5: The MIC results of the antimycobacterial susceptibility testing of the di-ortho chlorinated Boc- $N$-hydrazides 67As-Ax; where a (-) indicates to a lack of inhibition activity.

|  |  | MIC ( $\mu \mathrm{M}$ ) |  |  |
| :---: | :---: | :---: | :---: | :---: |
| Entry | Structure | WT. $M t b$ $\mathrm{mc}^{2} 7000$ | $\begin{gathered} \text { INH. } \\ \text { resistant } \\ M t b \\ \mathrm{mc}^{2} 7021 \end{gathered}$ | $\begin{gathered} \text { RIF. } \\ \text { resistant } \\ M t b \\ \mathrm{mc}^{2} 7002 \end{gathered}$ |
| 66 |  <br> 67As | 23.94 | 191.50 | 191.50 |
| 67 |  | 45.95 | - | - |
| 68 |  <br> 67 Au | 5.32 | 85.04 | - |
| 69 |  | 81.99 | 163.97 | 81.99 |
| 70 |  <br> 67Aw | 19.59 | 156.73 | - |
| 71 |  <br> 67 Ax | 18.85 | 150.83 | - |

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Table A6: The MIC results of the antimycobacterial susceptibility testing of the trifluoromethylated Boc- $N$-hydrazides 67Ay-Bp; where a (-) indicates to a lack of inhibition activity.

|  |  | MIC ( $\mu \mathrm{M}$ ) |  |  |
| :---: | :---: | :---: | :---: | :---: |
| Entry | Structure | WT. $M t b$ mc$^{27} 7000$ | $\begin{gathered} \text { INH. } \\ \text { resistant } \\ M t b \\ \text { mc'}^{2} 7021 \end{gathered}$ | $\begin{gathered} \hline \text { RIF. } \\ \text { resistant } \\ M t b \\ \text { mc'}^{2} 7002 \end{gathered}$ |
| 72 |  <br> 67Ay | 96.01 | 192.01 | 192.01 |
| 73 |  | - | 184.26 | - |
| 74 |  | - | 170.49 | - |
| 75 |  | - | 164.35 | - |
| 76 |  | 78.54 | 78.54 | 78.54 |
| 77 |  | - | 151.14 | - |
| 78 |  | - | 48.00 | 96.01 |
| 79 |  | 92.13 | 92.13 | 92.13 |
| 80 |  | - | 85.24 | - |

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|  | 67Bg |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
| 81 |  | 20.54 | 41.09 | 41.09 |
| 82 |  <br> 67 Bi | - | - | - |
| 82 |  <br> 67Bj | 37.79 | 37.79 | 75.57 |
| 83 |  | 12.01 | NA | NA |
| 84 |  | 46.06 | 23.03 | 92.13 |
| 85 |  | - | - | - |
| 86 |  | 82.17 | - | 82.17 |
| 87 |  | 39.27 | 39.27 | 39.27 |
| 88 |  | 151.14 | NA | NA |

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Table A7: The MIC results of the antimycobacterial susceptibility testing of the di-meta trifluoromethylated Boc- $N$-hydrazides 67Bs-Bx; where a (-) indicates to a lack of inhibition activity.

|  |  | MIC ( $\mu \mathrm{M}$ ) |  |  |
| :---: | :---: | :---: | :---: | :---: |
| Entry | Structure | $\begin{gathered} \text { WT. } \\ M t b \\ \mathrm{mc}^{2} 7000 \end{gathered}$ | $\begin{gathered} \text { INH. } \\ \text { resistant } \\ M t b \\ \text { mc}^{2} 7021 \end{gathered}$ | $\begin{gathered} \hline \text { RIF. } \\ \text { resistant } \\ \text { Mtb } \\ \text { mc}^{2} 7002 \end{gathered}$ |
| 89 |  <br> 67Bs | 79.74 | 79.74 | 79.74 |
| 90 |  | 19.26 | - | - |
| 91 |  | NA | NA | NA |
| 92 |  <br> 67Bv | - | 139.92 | - |
| 93 |  | - | - | - |
| 94 |  | - | - | - |

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Table A8: The MIC results of the antimycobacterial susceptibility testing of the di-substituted hydrazides with fluorine 67By-Cd; where a (-) indicates to a lack of inhibition activity.

|  |  | MIC ( $\mu \mathrm{M}$ ) |  |  |
| :---: | :---: | :---: | :---: | :---: |
| Entry | Structure | $\begin{gathered} \text { WT. } \\ M t b \\ \mathrm{mc}^{2} 7000 \end{gathered}$ | $\begin{gathered} \mathrm{INH} . \\ \text { resistant } \\ \text { Mtb } \\ \mathrm{mc}^{2} 7021 \end{gathered}$ | $\begin{gathered} \text { RIF. } \\ \text { resistant } \\ \text { Mtb } \\ \text { mc}^{2} 7002 \\ \hline \end{gathered}$ |
| 95 |  | 106.21 | 26.55 | 106.21 |
| 96 |  | 25.37 | - | - |
| 97 |  | - | - | - |
| 98 |  | - | 179.07 | - |
| 99 |  | - | - | - |
| 100 |  | 10.22 | - | - |

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Table A9: The MIC results of the antimycobacterial susceptibility testing of the brominated
hydrazides $67 \mathrm{Ce}-\mathrm{Cv}$; where a (-) indicates to a lack of inhibition activity.

|  |  | MIC ( $\mu \mathrm{M}$ ) |  |  |
| :---: | :---: | :---: | :---: | :---: |
| Entry | Structure | $\begin{gathered} \text { WT. } \\ \text { Mtb } \\ \mathrm{mc}^{2} 7000 \end{gathered}$ | $\begin{gathered} \mathrm{INH} . \\ \text { resistant } \\ M t b \\ \mathrm{mc}^{2} 7021 \end{gathered}$ | $\begin{gathered} \text { RIF. } \\ \text { resistant } \\ M t b \\ \text { mc }^{2} 7002 \end{gathered}$ |
| 101 |  <br> 67Ce | - | 185.93 | - |
| 102 |  | 44.66 | 178.65 | - |
| 103 |  | 10.86 | 86.89 | - |
| 104 |  | 159.87 | 159.87 | 159.87 |
| 105 |  | 159.87 | 159.87 | - |
| 106 |  | 147.35 | 147.35 | 147.35 |
| 107 |  | 5.81 | 23.24 | 46.48 |
| 108 |  | 44.66 | 22.33 | 44.66 |
| 109 |  | 42.98 | 42.98 | 42.98 |

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|  | 67 Cm |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
| 110 |  | 159.87 | 19.98 | 39.97 |
| 111 |  | 19.98 | 19.98 | 39.97 |
| 112 |  | 73.68 | 18.42 | 36.84 |
| 113 |  | 23.24 | 5.81 | 23.24 |
| 114 |  | 44.66 | 11.17 | 44.66 |
| 115 |  <br> 67Cs | 85.96 | 42.98 | 85.96 |
| 116 |  | 39.97 | 39.97 | 39.97 |
| 117 |  | NA | NA | NA |
| 118 |  | 36.84 | 36.84 | 18.42 |

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Table A10: The MIC results of the antimycobacterial susceptibility testing of the cyanated hydrazides 67 Cw -Db; where a (-) indicates to a lack of inhibition activity.

|  |  | MIC ( $\mu \mathrm{M}$ ) |  |  |
| :---: | :---: | :---: | :---: | :---: |
| Entry | Structure | $\begin{gathered} \text { WT. } \\ M t b \\ \mathrm{mc}^{2} 7000 \end{gathered}$ | $\begin{gathered} \text { INH. } \\ \text { resistant } \\ M t b \\ \text { mc }^{2} 7021 \end{gathered}$ | $\begin{gathered} \text { RIF. } \\ \text { resistant } \\ M t b \\ \text { mc}^{2} 7002 \end{gathered}$ |
| 119 |  | 110.22 | 110.22 | 220.45 |
| 120 |  | 26.29 | - | - |
| 121 |  | - | - | - |
| 122 |  | - | 184.74 | 184.74 |
| 123 |  | 10.98 | 87.80 | - |
| 124 |  | - | - | - |

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Table A11: The MIC results of the antimycobacterial susceptibility testing of the 3-nitro hydrazides 67Dc-Dh; where a (-) indicates to a lack of inhibition activity.

|  |  | MIC ( $\mu \mathrm{M}$ ) |  |  |
| :---: | :---: | :---: | :---: | :---: |
| Entry | Structure | $\begin{gathered} \text { WT. } \\ M t b \\ \mathrm{mc}^{2} 7000 \end{gathered}$ | $\begin{gathered} \text { INH. } \\ \text { resistant } \\ M t b \\ \text { mc}^{2} 7021 \end{gathered}$ | $\begin{gathered} \text { RIF. } \\ \text { resistant } \\ M t b \\ \text { mc}^{2} 7002 \end{gathered}$ |
| 125 |  | - | - | - |
| 126 |  | 98.66 | 98.66 | 98.66 |
| 127 |  <br> 67De | - | - | - |
| 128 |  | 174.66 | 87.33 | 87.33 |
| 129 |  | - | 90.81 | 90.81 |
| 130 |  | 79.91 | 79.91 | 79.91 |

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Table A12: The MIC results of the antimycobacterial susceptibility testing of the 4-methylsulfonyl hydrazides 67Di-Dn; where a (-) indicates to a lack of inhibition activity.

|  |  | MIC ( $\mu \mathrm{M}$ ) |  |  |
| :---: | :---: | :---: | :---: | :---: |
| Entry | Structure | $\begin{gathered} \text { WT. } \\ M t b \\ \mathrm{mc}^{2} 7000 \end{gathered}$ | INH. resistant Mtb $\mathrm{mc}^{27} 021$ | $\begin{gathered} \text { RIF. } \\ \text { resistant } \\ M t b \\ \text { mc}^{2} 7002 \end{gathered}$ |
| 131 |  | - | - | - |
| 132 |  | 89.53 | - | - |
| 133 |  <br> 67Dk | - | - | - |
| 134 |  | - | - | - |
| 135 |  | - | - | - |
| 136 |  | - | - | - |

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Table A13: The MIC results of the antimycobacterial susceptibility testing of the methylated hydrazides 67Do-Ef; where a (-) indicates to a lack of inhibition activity.

|  |  | MIC ( $\mu \mathrm{M}$ ) |  |  |
| :---: | :---: | :---: | :---: | :---: |
| Entry | Structure | $\begin{gathered} \text { WT. } \\ M t b \\ \mathrm{mc}^{2} 7000 \end{gathered}$ | $\begin{gathered} \text { INH. } \\ \text { resistant } \\ M t b \\ \text { mc}^{2} 7021 \end{gathered}$ | $\begin{gathered} \text { RIF. } \\ \text { resistant } \\ M t b \\ \mathrm{mc}^{2} 7002 \end{gathered}$ |
| 137 |  <br> 67Do | 229.11 | 229.11 | - |
| 138 |  | 109.08 | 109.08 | - |
| 139 |  | 99.56 | 199.12 | 199.12 |
| 140 |  | 190.79 | 190.79 | 190.79 |
| 141 |  | 181.06 | - | - |
| 142 |  | 173.22 | 86.61 | 173.22 |
| 143 |  | 114.56 | 114.56 | 114.56 |
| 144 |  | 109.08 | 218.15 | 109.08 |
| 145 |  | 99.56 | 24.89 | 99.56 |

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|  | 67Dw |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
| 146 |  | 95.39 | 95.39 | 190.79 |
| 147 |  <br> 67Dy | - | 181.06 | 181.06 |
| 148 |  <br> 67Dz | 86.61 | 86.61 | 86.61 |
| 149 |  | - | 229.11 | 229.11 |
| 150 |  | 3.41 | 109.08 | 109.08 |
| 151 |  | 199.12 | 199.12 | 199.12 |
| 152 |  | 190.79 | 95.39 | 95.39 |
| 153 |  | 1.41 | 45.26 | 181.06 |
| 154 |  | 173.22 | 173.22 | 173.22 |

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Table A14: The MIC results of the antimycobacterial susceptibility testing of the 4-isopropyl hydrazides 67 Eg -EI; where a (-) indicates to a lack of inhibition activity.

|  |  | MIC ( $\mu \mathrm{M}$ ) |  |  |
| :---: | :---: | :---: | :---: | :---: |
| Entry | Structure | WT. $M t b$ $\mathrm{mc}^{27} 7000$ | $\begin{gathered} \mathrm{INH} . \\ \text { resistant } \\ M t b \\ \mathrm{mc}^{2} 7021 \end{gathered}$ | RIF. <br> resistant <br> Mtb <br> $\mathrm{mc}^{2} 7002$ |
| 155 |  | 104.10 | 104.10 | 104.10 |
| 156 |  | 3.11 | 99.56 | 199.12 |
| 157 |  | 183.13 | 91.56 | 91.56 |
| 158 |  | - | 176.07 | 176.07 |
| 159 |  | - | 167.74 | 167.74 |
| 160 |  | 80.50 | - | 161.00 |

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Table A15: The MIC results of the antimycobacterial susceptibility testing of the protected amino acids 70a-70Aa; where a (-) indicates to a lack of inhibition activity.

|  |  | MIC ( $\mu \mathrm{M}$ ) |  |  |
| :---: | :---: | :---: | :---: | :---: |
| Entry | Structure | $\begin{gathered} \text { WT. } \\ M t b \\ \mathrm{mc}^{2} 7000 \end{gathered}$ | $\begin{gathered} \text { INH. } \\ \text { resistant } \\ M t b \\ \mathrm{mc}^{2} 7021 \end{gathered}$ | RIF. resistant $M t b$ $\mathrm{mc}^{2} 7002$ |
| 165 |  | - | - | - |
| 166 |  | - | - | - |
| 167 |  | - | - | - |
| 168 |  | - | - | - |
| 169 |  | - | - | - |
| 170 |  | - | - | - |
| 171 |  | - | - | - |
| 172 |  | - | - | - |
| 173 |  | - | - | - |

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|  | 70i |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
| 174 |  | - | - | - |
| 175 |  | - | - | 166.91 |
| 176 |  | - | - | - |
| 177 |  | - | - | - |
| 178 |  | - | - | - |
| 179 |  | - | - | - |
| 180 |  | - | - | - |
| 181 |  | - | - | - |

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| 182 |  | - | - | - |
| :---: | :---: | :---: | :---: | :---: |
| 183 |  | 121.54 | 121.54 | 121.54 |
| 184 |  <br> $70 t$ | - | - | - |
| 185 |  | - | - | - |
| 186 |  | - | - | - |
| 187 |  | - | - | - |
| 188 |  | - | - | - |
| 189 |  | 338.25 | 338.25 | 338.25 |
| 200 |  | - | - | - |

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### 8.2 The Crystal data and structure refinement

Crystal data and structure refinement for the resulting amino acid hydrazides.
tert-Butyl-(S)-(1-(2-(2,4-difluorophenyl)hydrazinyl)-3-methyl-1-oxobutan-2$\mathrm{yl})$ carbamate 67 Ca


tert-Butyl-(S)-(1-(2-(2,4-difluorophenyl)hydrazinyl)-4-(methylthio)-1-oxobutan-2-yl)carbamate 67Cc


tert-Butyl-(2-(2-(4-chlorophenyl)hydrazinyl)-2-oxoethyl)carbamate 67Am



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| Compound | 67Ca | 67Cc | 67Am |
| :---: | :---: | :---: | :---: |
| Empirical formula | $\mathrm{C}_{16} \mathrm{H}_{23} \mathrm{~F}_{2} \mathrm{~N}_{3} \mathrm{O}_{3}$ | $\mathrm{C}_{16} \mathrm{H}_{23} \mathrm{~F}_{2} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}$ | $\mathrm{C}_{13} \mathrm{H}_{18} \mathrm{ClN}_{3} \mathrm{O}_{3}$ |
| Formula weight | 343.37 | 375.43 | 299.75 |
| Temperature/K | 150.0(2) | 150.0(2) | 200.0(2) |
| Crystal system | orthorhombic | monoclinic | monoclinic |
| Space group | $\mathrm{P} 2{ }_{12}{ }_{21}{ }_{1}$ | P21 | $\mathrm{P} 21 / \mathrm{c}$ |
| a/ $\AA$ | 9.0949(2) | 12.7631(5) | 6.30898(15) |
| b/Å | 10.4986(2) | 5.04890(19) | 18.1728(4) |
| c/ $\AA$ | 18.9737(5) | 15.3008(7) | 28.7539(10) |
| $\alpha /{ }^{\circ}$ | 90 | 90 | 90 |
| $\beta /{ }^{\circ}$ | 90 | 109.201(5) | 92.489(3) |
| $\gamma^{1}{ }^{\circ}$ | 90 | 90 | 90 |
| Volume/ ${ }^{\text {a }}$ | 1811.68(7) | 931.13(7) | 3293.57(16) |
| Z | 4 | 2 | 8 |
| $\rho_{\text {calcg }} / \mathrm{cm}^{3}$ | 1.259 | 1.339 | 1.209 |
| $\mu / \mathrm{mm}^{-1}$ | 0.857 | 1.901 | 2.150 |
| $\mathrm{F}(000)$ | 728.0 | 396.0 | 1264.0 |
| Crystal size/mm ${ }^{3}$ | $0.34 \times 0.1 \times 0.06$ | $0.45 \times 0.07 \times 0.03$ | $0.34 \times 0.16 \times 0.04$ |
| Radiation | $\mathrm{CuK} \alpha(\lambda=1.54184)$ | $\mathrm{CuK} \alpha(\lambda=1.54184)$ | $\operatorname{CuK} \alpha(\lambda=1.54184)$ |
| $2 \Theta$ range for data collection $/{ }^{\circ}$ | 9.322 to 133.936 | 7.334 to 133.64 | 7.846 to 133.648 |
| Index ranges | $\begin{aligned} & -8 \leq \mathrm{h} \leq 10,-12 \leq \mathrm{k} \leq \\ & 12,-22 \leq 1 \leq 22 \end{aligned}$ | $\begin{aligned} -15 & \leq h \leq 15,-5 \leq k \leq 5, \\ -18 & \leq 1 \leq 17 \end{aligned}$ | $\begin{aligned} & -4 \leq \mathrm{h} \leq 7,-20 \leq \mathrm{k} \leq 21, \\ & -33 \leq 1 \leq 34 \end{aligned}$ |
| Reflections collected | 25502 | 7414 | 23718 |
| Independent reflections | $\begin{aligned} & 3206\left[R_{\text {int }}=0.0463,\right. \\ & \left.R_{\text {sigma }}=0.0224\right] \end{aligned}$ | $\begin{aligned} & 2897 \quad\left[\mathrm{R}_{\text {int }}=0.0441,\right. \\ & \left.\mathrm{R}_{\text {sigma }}=0.0439\right] \end{aligned}$ | $\begin{aligned} & 5839 \quad\left[\mathrm{R}_{\text {int }}=0.0410,\right. \\ & \left.\mathrm{R}_{\text {sigma }}=0.0323\right] \end{aligned}$ |
| Data/restraints/parameters | 3206/0/232 | 2897/4/239 | 5839/442/422 |
| Goodness-of-fit on $\mathrm{F}^{2}$ | 1.046 | 1.065 | 1.034 |
| Final R indexes $[\mathrm{I}>=2 \sigma(\mathrm{I})$ ] | $\begin{aligned} & \mathrm{R}_{1}=0.0281, \quad \mathrm{wR}_{2}= \\ & 0.0687 \end{aligned}$ | $\begin{aligned} & \mathrm{R}_{1}=0.0382, \quad \mathrm{wR}_{2}= \\ & 0.0899 \end{aligned}$ | $\begin{aligned} & \mathrm{R}_{1}=0.0551, \quad \mathrm{wR}_{2}= \\ & 0.1512 \end{aligned}$ |
| Final R indexes [all data] | $\begin{aligned} & \mathrm{R}_{1}=0.0323, \mathrm{wR}_{2}= \\ & 0.0719 \end{aligned}$ | $\begin{aligned} & \mathrm{R}_{1}=0.0440, \quad \mathrm{wR}_{2}= \\ & 0.0954 \end{aligned}$ | $\begin{aligned} & \mathrm{R}_{1}=0.0782, \quad \mathrm{wR}_{2}= \\ & 0.1716 \end{aligned}$ |


| Largest diff. peak/hole/e $\AA^{-}$ <br> 3 | $0.16 /-0.13$ | $0.23 /-0.22$ | $0.35 /-0.31$ |
| :--- | :--- | :--- | :--- |
| Flack parameter | $0.11(7)$ | $0.00(3)$ | - |

tert-Butyl (2-(2-(3-chlorophenyl)hydrazinyl)-2-oxoethyl)carbamate 67q


tert-Butyl (S)-(1-oxo-1-(2-phenylhydrazineyl)propan-2-yl)carbamate 67b



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## tert-Butyl

(2-oxo-2-(2-(2-(trifluoromethyl)phenyl)hydrazineyl)ethyl)
carbamate 67Ay



| Compound | 67q | 67b | 67Ay |
| :---: | :---: | :---: | :---: |
| Empirical formula | $\mathrm{C}_{13} \mathrm{H}_{18} \mathrm{ClN}_{3} \mathrm{O}_{3}$ | $\mathrm{C}_{14} \mathrm{H}_{21} \mathrm{~N}_{3} \mathrm{O}_{3}$ | $\mathrm{C}_{14} \mathrm{H}_{18} \mathrm{~F}_{3} \mathrm{~N}_{3} \mathrm{O}_{3}$ |
| Formula weight | 299.75 | 279.34 | 333.31 |
| Temperature/K | 150.0(2) | 150.0(2) | 150.0(2) |
| Crystal system | triclinic | orthorhombic | monoclinic |
| Space group | P-1 | $\mathrm{P} 2{ }_{1} 2_{1} 2_{1}$ | $\mathrm{P} 21 / \mathrm{n}$ |
| a/Å | 5.6650(3) | 9.1696(2) | 7.3634(2) |
| b/Å | 9.1347(4) | 9.6329(3) | 14.2093(4) |
| c/Å | 14.1624(5) | 17.2044(5) | 15.6570(5) |
| $\alpha /{ }^{\circ}$ | 91.485(3) | 90 | 90 |
| $\beta /{ }^{\circ}$ | 97.703(3) | 90 | 102.288(3) |
| $\gamma^{\prime}{ }^{\circ}$ | 102.737(4) | 90 | 90 |
| Volume/ $\AA^{3}$ | 707.23(5) | 1519.66(7) | 1600.64(8) |
| Z | 2 | 4 | 4 |
| $\rho \mathrm{calcg} / \mathrm{cm}^{3}$ | 1.408 | 1.221 | 1.383 |
| $\mu / \mathrm{mm}^{-1}$ | 2.503 | 0.712 | 1.048 |
| F(000) | 316.0 | 600.0 | 696.0 |
| Crystal size/mm ${ }^{3}$ | $0.19 \times 0.14 \times 0.05$ | $0.18 \times 0.12 \times 0.09$ | $0.18 \times 0.05 \times 0.03$ |

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| Radiation | $\mathrm{CuK} \alpha(\lambda=1.54184)$ | $\mathrm{CuK} \alpha(\lambda=1.54184)$ | $\mathrm{CuK} \alpha(\lambda=1.54184)$ |
| :---: | :---: | :---: | :---: |
| $2 \Theta$ range for data collection $/{ }^{\circ}$ | 9.944 to 133.852 | 10.284 to 133.806 | 8.492 to 133.806 |
| Index ranges | $\begin{aligned} & -6 \leq \mathrm{h} \leq 6,-10 \leq \mathrm{k} \leq \\ & 10,-16 \leq 1 \leq 16 \end{aligned}$ | $\begin{aligned} & -8 \leq \mathrm{h} \leq 10,-11 \leq \mathrm{k} \leq \\ & 11,-20 \leq 1 \leq 18 \end{aligned}$ | $\begin{aligned} & -6 \leq h \leq 8,-16 \leq k \leq 16, \\ & -18 \leq 1 \leq 18 \end{aligned}$ |
| Reflections collected | 19294 | 10840 | 22137 |
| Independent reflections | $\begin{aligned} & 2502\left[R_{\text {int }}=0.0382,\right. \\ & \left.R_{\text {sigma }}=0.0191\right] \end{aligned}$ | $\begin{aligned} & 2693 \quad\left[\mathrm{R}_{\text {int }}=0.0523,\right. \\ & \left.\mathrm{R}_{\text {sigma }}=0.0380\right] \end{aligned}$ | $\begin{aligned} & 2835 \quad\left[\mathrm{R}_{\text {int }}=0.0537,\right. \\ & \left.\mathrm{R}_{\text {sigma }}=0.0264\right] \end{aligned}$ |
| Data/restraints/parameters | 2502/0/194 | 2693/0/195 | 2835/0/221 |
| Goodness-of-fit on $\mathrm{F}^{2}$ | 1.026 | 1.068 | 1.043 |
| Final R indexes [ $\mathrm{I}>=2 \sigma(\mathrm{I})$ ] | $\begin{aligned} & \mathrm{R}_{1}=0.0282, \quad \mathrm{wR}_{2}= \\ & 0.0685 \end{aligned}$ | $\begin{aligned} & \mathrm{R}_{1}=0.0362, \quad \mathrm{wR}_{2}= \\ & 0.0907 \end{aligned}$ | $\begin{aligned} & \mathrm{R}_{1}=0.0398, \quad \mathrm{wR}_{2}= \\ & 0.1051 \end{aligned}$ |
| Final R indexes [all data] | $\begin{aligned} & \mathrm{R}_{1}=0.0338, \mathrm{wR}_{2}= \\ & 0.0716 \end{aligned}$ | $\begin{aligned} & \mathrm{R}_{1}=0.0409, \quad \mathrm{wR}_{2}= \\ & 0.0963 \end{aligned}$ | $\begin{aligned} & \mathrm{R}_{1}=0.0502, \quad \mathrm{wR}_{2}= \\ & 0.1143 \end{aligned}$ |
| Largest diff. peak/hole/eÅ 3 | 0.18/-0.19 | 0.13/-0.17 | 0.47/-0.34 |
| Flack parameter | - | 0.20(15) | - |

## tert-Butyl (S)-(1-oxo-1-(2-(2-(trifluoromethyl)phenyl)hydrazineyl)propan-2yl)carbamate 67 Az




## tert-Butyl (S)-(3-methyl-1-oxo-1-(2-(2(trifluoromethyl)phenyl)hydrazineyl)

 butan-2-yl)carbamate 67Ba

tert-Butyl
(S)-(1-(2-(2-bromophenyl)hydrazineyl)-1-oxopropan-2-
yl)carbamate 67Cf



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| Compound | 67Az | 67Ba | 67Cf |
| :---: | :---: | :---: | :---: |
| Empirical formula | $\mathrm{C}_{15} \mathrm{H}_{20} \mathrm{~F}_{3} \mathrm{~N}_{3} \mathrm{O}_{3}$ | $\mathrm{C}_{17} \mathrm{H}_{24} \mathrm{~F}_{3} \mathrm{~N}_{3} \mathrm{O}_{3}$ | $\mathrm{C}_{29} \mathrm{H}_{42} \mathrm{Br}_{2} \mathrm{Cl}_{2} \mathrm{~N}_{6} \mathrm{O}_{6}$ |
| Formula weight | 347.34 | 375.39 | 801.40 |
| Temperature/K | 150.0(2) | 150.0(2) | 150.0(2) |
| Crystal system | Tetragonal | orthorhombic | orthorhombic |
| Space group | P41 | $\mathrm{P} 2{ }_{1} 2_{1} 2_{1}$ | $\mathrm{P} 2{ }_{1} 2_{1} 2_{1}$ |
| a/Å | 10.2512(2) | 4.99930(10) | 13.4621(3) |
| b/Å | 10.2512(2) | 19.5052(5) | 16.4034(3) |
| c/ $\AA$ | 16.4278(7) | 19.9425(5) | 16.6639(3) |
| $\alpha /{ }^{\circ}$ | 90 | 90 | 90 |
| $\beta /{ }^{\circ}$ | 90 | 90 | 90 |
| $\gamma{ }^{\circ}$ | 90 | 90 | 90 |
| Volume/ $\AA^{3}$ | 1726.35(10) | 1944.64(8) | 3679.79(12) |
| Z | 4 | 4 | 4 |
| $\rho_{\text {calcg }} / \mathrm{cm}^{3}$ | 1.336 | 1.282 | 1.447 |
| $\mu / \mathrm{mm}^{-1}$ | 0.993 | 0.919 | 4.519 |
| F(000) | 728.0 | 792.0 | 1640.0 |
| Crystal size/mm ${ }^{3}$ | $0.26 \times 0.2 \times 0.06$ | $0.28 \times 0.15 \times 0.06$ | $0.38 \times 0.08 \times 0.06$ |
| Radiation | $\mathrm{Cu} \mathrm{K} \mathrm{\alpha}(\lambda=1.54184)$ | $\mathrm{CuK} \alpha(\lambda=1.54184)$ | $\operatorname{CuK} \alpha(\lambda=1.54184)$ |
| $2 \Theta$ range for data collection $/{ }^{\circ}$ | 8.626 to 133.76 | 8.868 to 133.842 | 7.562 to 133.626 |
| Index ranges | $\begin{aligned} & -12 \leq h \leq 11,-12 \leq k \\ & \leq 11,-19 \leq 1 \leq 18 \end{aligned}$ | $\begin{aligned} & -5 \leq \mathrm{h} \leq 5,-23 \leq \mathrm{k} \leq 23, \\ & -23 \leq 1 \leq 23 \end{aligned}$ | $\begin{aligned} & -13 \leq \mathrm{h} \leq 16,-19 \leq \mathrm{k} \leq \\ & 16,-19 \leq 1 \leq 14 \end{aligned}$ |
| Reflections collected | 13849 | 20746 | 26356 |
| Independent reflections | $\begin{aligned} & 3026\left[\mathrm{R}_{\text {int }}=0.0575,\right. \\ & \left.\mathrm{R}_{\text {sigma }}=0.0394\right] \end{aligned}$ | $\begin{aligned} & 3427 \quad\left[\mathrm{R}_{\text {int }}=0.0625,\right. \\ & \left.\mathrm{R}_{\text {sigma }}=0.0373\right] \end{aligned}$ | $\begin{aligned} & 6503 \quad\left[\mathrm{R}_{\text {int }}=0.0532,\right. \\ & \left.\mathrm{R}_{\text {sigma }}=0.0412\right] \end{aligned}$ |
| Data/restraints/parameters | 3026/1/231 | 3427/360/297 | 6503/373/447 |
| Goodness-of-fit on $\mathrm{F}^{2}$ | 1.064 | 1.079 | 1.052 |
| Final R indexes [ $\mathrm{I}>=2 \sigma$ (I)] | $\begin{aligned} & \mathrm{R}_{1}=0.0363, \mathrm{wR}_{2}= \\ & 0.0901 \end{aligned}$ | $\begin{aligned} & \mathrm{R}_{1}=0.0374, \quad \mathrm{wR}_{2}= \\ & 0.0924 \end{aligned}$ | $\begin{aligned} & \mathrm{R}_{1}=0.0556, \quad \mathrm{wR}_{2}= \\ & 0.1384 \end{aligned}$ |
| Final R indexes [all data] | $\begin{aligned} & \mathrm{R}_{1}=0.0416, \mathrm{wR}_{2}= \\ & 0.0953 \end{aligned}$ | $\begin{aligned} & \mathrm{R}_{1}=0.0419, \quad \mathrm{wR}_{2}= \\ & 0.0966 \end{aligned}$ | $\begin{aligned} & \mathrm{R}_{1}=0.0627, \quad \mathrm{wR}_{2}= \\ & 0.1439 \end{aligned}$ |


| Largest diff. peak/hole/e $\AA^{-}$ <br> 3 | $0.12 /-0.13$ | $0.15 /-0.16$ | $1.17 /-0.82$ |
| :--- | :--- | :--- | :--- |
| Flack parameter | $-0.14(11)$ | $-0.07(8)$ | $-0.023(15)$ |

tert-Butyl (S)-(1-(2-(3-bromophenyl)hydrazineyl)-3-methyl-1-oxobutan-2yl)carbamate 67 Cm


tert-Butyl (S)-(1-(2-(2,6-dichlorophenyl)hydrazineyl)-4-methyl-1-oxopentan-2-yl)carbamate 67Av


tert-Butyl (S)-(1-(2-(2-chlorophenyl)hydrazineyl)-1-oxo-3-phenylpropan-2yl)carbamate 67AI


tert-Butyl (S)-(1-(2-benzylhydrazinyl)-1-oxopropan-2-yl)carbamate 91



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| Compound | 67Cm | 67Av | 67AI | 91 |
| :---: | :---: | :---: | :---: | :---: |
| Empirical formula | $\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{BrN}_{3} \mathrm{O}_{3}$ | $\mathrm{C}_{17} \mathrm{H}_{25} \mathrm{Cl}_{2} \mathrm{~N}_{3} \mathrm{O}_{3}$ | $\mathrm{C}_{20} \mathrm{H}_{24} \mathrm{ClN}_{3} \mathrm{O}_{3}$ | $\mathrm{C}_{30} \mathrm{H}_{46} \mathrm{~N}_{6} \mathrm{O}_{6}$ |
| Formula weight | 386.29 | 390.30 | 389.87 | 586.73 |
| Temperature/K | 150.0(2) | 150.0(2) | 150.0(2) | 150.0(2) |
| Crystal system | monoclinic | triclinic | orthorhombic | monoclinic |
| Space group | C2 | P1 | $\mathrm{P} 2_{1} 2_{1} 2_{1}$ | $\mathrm{P} 2_{1}$ |
| a/ $\AA$ | 25.7185(4) | 9.2189(4) | 5.0159(2) | 10.1442(5) |
| b/Å | 5.04190(10) | 11.0840(4) | 16.3832(7) | 4.9901(3) |
| c/Å | 15.0967(3) | 11.7550(5) | 24.6159(9) | 15.7321(7) |
| $\alpha /{ }^{\circ}$ | 90 | 62.645(4) | 90 | 90 |
| $\beta /{ }^{\circ}$ | 104.452(2) | 71.907(4) | 90 | 98.307(4) |
| $\gamma^{\prime}{ }^{\circ}$ | 90 | 88.350(3) | 90 | 90 |
| Volume/ $\AA^{3}$ | 1895.65(6) | 1004.81(8) | 2022.85(14) | 788.01(7) |
| Z | 4 | 2 | 4 | 1 |
| $\rho_{\text {calcg }} / \mathrm{cm}^{3}$ | 1.354 | 1.290 | 1.280 | 1.236 |
| $\mu / \mathrm{mm}^{-1}$ | 3.099 | 3.076 | 1.876 | 0.709 |
| F(000) | 800.0 | 412.0 | 824.0 | 316.0 |
| Crystal size/mm ${ }^{3}$ | $0.2 \times 0.09 \times 0.04$ | $0.21 \times 0.18 \times 0.14$ | $0.32 \times 0.11 \times 0.05$ | $\begin{aligned} & 0.43 \times 0.1 \times \\ & 0.03 \end{aligned}$ |
| Radiation | $\begin{array}{ll} \mathrm{Cu} \quad \mathrm{~K} \alpha \quad(\lambda= \\ 1.54184) & \end{array}$ | $\begin{array}{lrll} \hline \mathrm{Cu} & \mathrm{~K} \alpha & (\lambda & = \\ 1.54184) & & \end{array}$ | $\begin{aligned} & \mathrm{CuK} \alpha \quad(\lambda= \\ & 1.54184) \end{aligned}$ | $\begin{aligned} & \mathrm{Cu} \mathrm{~K} \alpha(\lambda= \\ & 1.54184) \end{aligned}$ |
| $2 \Theta$ range for data collection $/{ }^{\circ}$ | 7.098 to 133.61 | 8.99 to 133.526 | 7.182 to 133.716 | $\begin{aligned} & 8.81 \quad \text { to } \\ & 133.146 \end{aligned}$ |
| Index ranges | $\begin{aligned} & -30 \leq \mathrm{h} \leq 30,-5 \\ & \leq \mathrm{k} \leq 5,-17 \leq 1 \leq \\ & 17 \end{aligned}$ | $\begin{aligned} & -10 \leq \mathrm{h} \leq 10,-13 \leq \\ & \mathrm{k} \leq 13,-13 \leq 1 \leq \\ & 13 \end{aligned}$ | $\begin{aligned} & -5 \leq h \leq 4,-19 \leq k \\ & \leq 19,-28 \leq 1 \leq 29 \end{aligned}$ | $\begin{aligned} & -12 \leq \mathrm{h} \leq \\ & 12,-5 \leq \mathrm{k} \leq \\ & 5,-18 \leq 1 \leq \\ & 18 \end{aligned}$ |
| Reflections collected | 26168 | 26701 | 14161 | 10833 |
| Independent reflections | $\begin{aligned} & 3230 \quad\left[\mathrm{R}_{\mathrm{int}}=\right. \\ & 0.0504, \\ & 0.0228] \end{aligned}$ | $\begin{array}{lr} 6773 & {\left[\mathrm{R}_{\text {int }}=\right.} \\ 0.0604, & \mathrm{R}_{\text {sigma }}= \\ 0.0452] & \end{array}$ | $\begin{array}{lr} 3557 & {\left[\mathrm{R}_{\text {int }}=\right.} \\ 0.0553, & \mathrm{R}_{\text {sigma }}= \\ 0.0450] & \end{array}$ | $\begin{aligned} & 2639\left[\mathrm{R}_{\text {int }}=\right. \\ & 0.0606, \\ & \mathrm{R}_{\text {sigma }}= \\ & 0.0437] \end{aligned}$ |
| Data/restraints/parameters | 3230/4/222 | 6773/3/479 | 3557/425/311 | 2639/1/203 |
| Goodness-of-fit on $\mathrm{F}^{2}$ | 1.043 | 1.053 | 1.068 | 1.069 |

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| Final R indexes [ $\mathrm{I}>=2 \sigma$ ( I$)$ ] | $\begin{aligned} & \mathrm{R}_{1}=0.0261, \\ & \mathrm{wR}_{2}=0.0665 \end{aligned}$ | $\begin{aligned} & \mathrm{R}_{1}=0.0616 \\ & \mathrm{wR}_{2}=0.1565 \end{aligned}$ | $\begin{aligned} & \mathrm{R}_{1}=0.0431 \\ & \mathrm{wR}_{2}=0.0953 \end{aligned}$ | $\begin{aligned} & \hline \mathrm{R}_{1}= \\ & 0.0460, \\ & \mathrm{wR}_{2}= \\ & 0.1137 \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: |
| Final R indexes [all data] | $\begin{aligned} & \mathrm{R}_{1}=0.0279 \\ & \mathrm{wR}_{2}=0.0683 \end{aligned}$ | $\begin{aligned} & \mathrm{R}_{1}=0.0671, \\ & \mathrm{wR}_{2}=0.1647 \end{aligned}$ | $\begin{aligned} & \mathrm{R}_{1}=0.0589 \\ & \mathrm{wR}_{2}=0.1057 \end{aligned}$ | $\begin{aligned} & \mathrm{R}_{1}= \\ & 0.0558, \\ & \mathrm{wR}_{2}= \\ & 0.1256 \end{aligned}$ |
| Largest diff. peak/hole/e $\AA$ 3 | 0.31/-0.36 | 0.55/-0.39 | 0.26/-0.19 | 0.20/-0.19 |
| Flack parameter | -0.020(11) | -0.01(2) | 0.008(15) | -0.1(2) |

Crystal data and structure refinement for the resulting benzoxa-[2,1,3]-diazoles.
t-tert-Butyl ((7-chlorobenzo[c][1,2,5]oxadiazol-4-yl)sulfonyl)glycinate 63a


(R)-7-chloro-N-(1-oxo-3-phenyl-1-(2-phenylhydrazineyl)propan-2-yl)benzo[c][1,2,5]oxadiazole-4-sulfonamide 65Ae


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| Compound | 63a | 65Ae |
| :---: | :---: | :---: |
| Empirical formula | $\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{ClN}_{3} \mathrm{O}_{5} \mathrm{~S}$ | $\mathrm{C}_{21} \mathrm{H}_{18} \mathrm{ClN}_{5} \mathrm{O}_{4} \mathrm{~S}$ |
| Formula weight | 347.77 | 471.91 |
| Temperature/K | 150.0(2) | 150.0(2) |
| Crystal system | orthorhombic | triclinic |
| Space group | Pca2 ${ }_{1}$ | P1 |
| a/Å | 11.2682(2) | 7.4611(3) |
| b/Å | 12.9378(3) | 9.9630(4) |
| c/Å | 10.1060(2) | 14.5199(4) |
| $\alpha /{ }^{\circ}$ | 90 | 87.046(3) |
| $\beta /{ }^{\circ}$ | 90 | 79.873(3) |
| $\gamma{ }^{\circ}$ | 90 | 79.780(3) |
| Volume/ $\AA^{3}$ | 1473.32(6) | 1045.43(7) |
| Z | 4 | 2 |
| $\rho$ calcg/ $\mathrm{cm}^{3}$ | 1.568 | 1.499 |
| $\mu / \mathrm{mm}^{-1}$ | 3.892 | 2.906 |
| $F(000)$ | 720.0 | 488.0 |
| Crystal size/mm ${ }^{3}$ | $0.21 \times 0.18 \times 0.04$ | $0.28 \times 0.12 \times 0.09$ |
| Radiation | $\operatorname{CuK} \alpha(\lambda=1.54184)$ | $\mathrm{Cu} \mathrm{K} \alpha(\lambda=1.54184)$ |
| $2 \Theta$ range for data collection/ ${ }^{\circ}$ | 6.832 to 133.55 | 9.022 to 133.668 |
| Index ranges | $\begin{aligned} & -13 \leq \mathrm{h} \leq 11,-14 \leq \mathrm{k} \leq 15,-11 \leq 1 \\ & \leq 12 \end{aligned}$ | $-8 \leq \mathrm{h} \leq 8,-11 \leq \mathrm{k} \leq 11,-17 \leq 1 \leq 17$ |
| Reflections collected | 9999 | 27923 |
| Independent reflections | $\begin{aligned} & 2475 \quad\left[\mathrm{R}_{\text {int }}=0.0372, \quad \mathrm{R}_{\text {sigma }}=\right. \\ & 0.0305] \end{aligned}$ | $7049\left[\mathrm{R}_{\text {int }}=0.0666, \mathrm{R}_{\text {sigma }}=0.0488\right]$ |

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| Data/restraints/parameters | $2475 / 1 / 206$ | $7049 / 18 / 595$ |
| :--- | :--- | :--- |
| Goodness-of-fit on $\mathrm{F}^{2}$ | 1.047 | 1.032 |
| Final R indexes $[\mathrm{I}>=2 \sigma(\mathrm{I})]$ | $\mathrm{R}_{1}=0.0263, \mathrm{wR}_{2}=0.0608$ | $\mathrm{R}_{1}=0.0425, \mathrm{wR}_{2}=0.1046$ |
| Final R indexes [all data] | $\mathrm{R}_{1}=0.0298, \mathrm{wR}_{2}=0.0634$ | $\mathrm{R}_{1}=0.0484, \mathrm{wR}_{2}=0.1112$ |
| Largest diff. peak/hole/e $\AA^{-3}$ | $0.21 /-0.23$ | $0.21 /-0.35$ |
| Flack parameter | $-0.004(10)$ | $-0.008(14)$ |


[^0]:    * Incidence is a number of new cases arising during a defined period.

